

Most abundant isotopes in the periodic table																									
SPIN=1/2									He																
INTEGER SPINS						HALF-INTEGER QUADRUPOLAR SPINS																			
QUADRUPOLAR SPINS												B	C	N	O	F	Ne								
Li	Be											Al	Si	P	S	Cl	Ar								
Na	Mg																								
K	Ca	Se	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr								
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	St	Te	I	Xe								
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn								
Fr	Ra	Ac										Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
												Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr



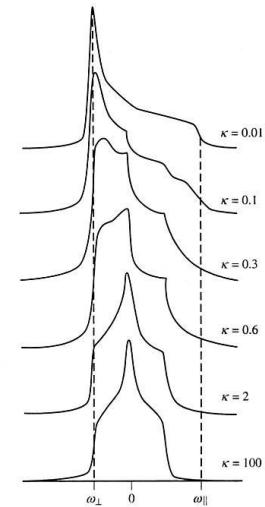
Nuclear Magnetic Resonance (NMR): solution, solid state, imaging



christian.bonhomme@upmc.fr

Christian BONHOMME, Professor

Laboratoire de Chimie de la Matière Condensée
UMR CNRS 7574 - Sorbonne Université, Paris



Spin: from 1925 to 2004



W. Pauli, Physics 1945

"for the discovery of the Exclusion Principle, also called the Pauli Principle"

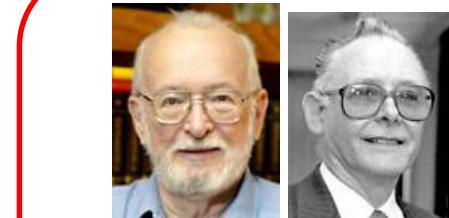


G. Uhlenbeck, S. Goudsmit
« fathers of the spin »



I. I. Rabi, Physique 1944

"for his resonance method for recording the magnetic properties of atomic nuclei"



P. C. Lauterbur, P. Mansfield, Medicine 2003

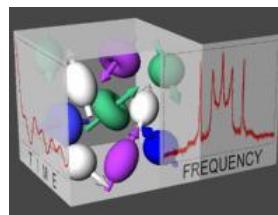


"for their discoveries concerning magnetic resonance imaging"



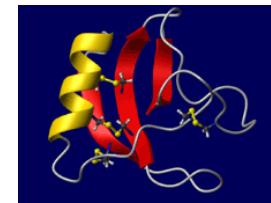
F. Bloch, E. M. Purcell, Physics 1952

"for the development of new methods for nuclear magnetic precision measurements and discoveries in connection therewith"



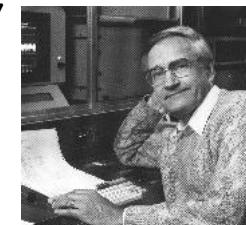
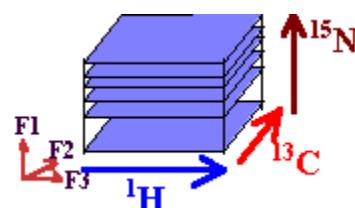
K. Wüthrich, Chemistry 2002

"for his development of NMR spectroscopy for determining the three dimensional structure of biological macromolecules in solution"



R. R. Ernst, Chemistry 1991

"for his contribution to the development of the methodology of high resolution NMR spectroscopy"



Spin quantum number I

$I \neq 0 \longrightarrow \text{NMR}...$

Most abundant isotopes in the periodic table																	
H	SPIN-I/2 INTEGER SPINS												He				
Li	Be	HALF-INTEGER QUADRUPOLAR SPINS												B	C	N	O
Na	Mg												Ne	Al	Si	P	S
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

receptivity :

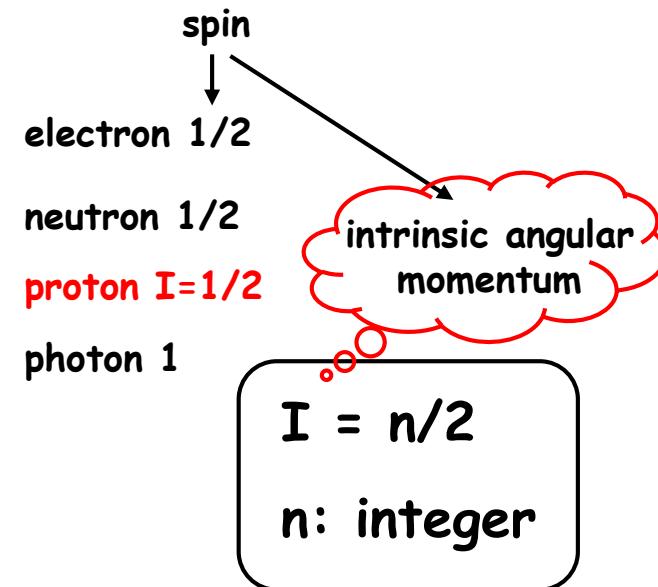
$$D^P = \frac{|\gamma_X|^3}{\gamma_{^{1H}}^3} \frac{(\%X)}{(\%^{1H})} \frac{(I_x+1)I_x}{(I_{^{1H}}+1)I_{^{1H}}}$$

$$^{13}C: I = \frac{1}{2} (1.1\%)$$

$$^{12}C: I = 0 (98.9\%)$$

$$D^P(^1H) = 1$$

$$D^P(^{13}C) = 0.00017\dots !$$



identity card:

isotope

^{13}C

spin I (m_I)

natural abundance (%)

gyromagnetic ratio

(rad s⁻¹ T⁻¹)

Nuclear spins and ... magnetic fields

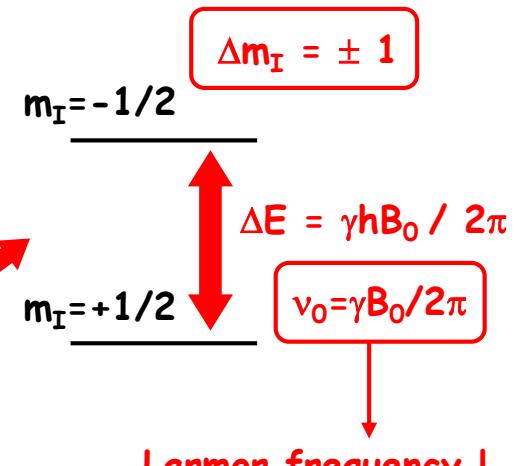
$$\hat{\mu} = \gamma \hbar \hat{I}$$

magnetic moment

spin angular momentum

gyromagnetic ratio

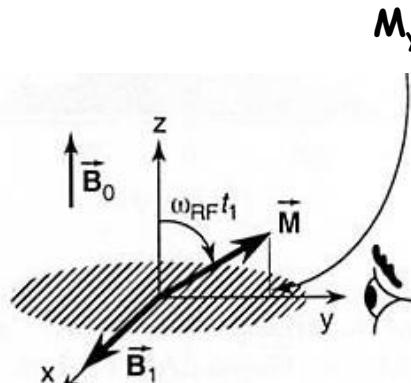
$E = -\mu \cdot B$



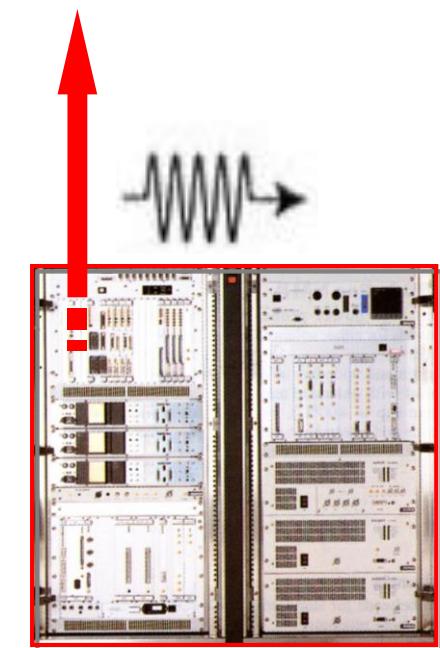
Boltzmann equation

Curie's law

$$M = \frac{N \gamma^2 h^2 B_0 I(I+1)}{12 \pi^2 kT}$$



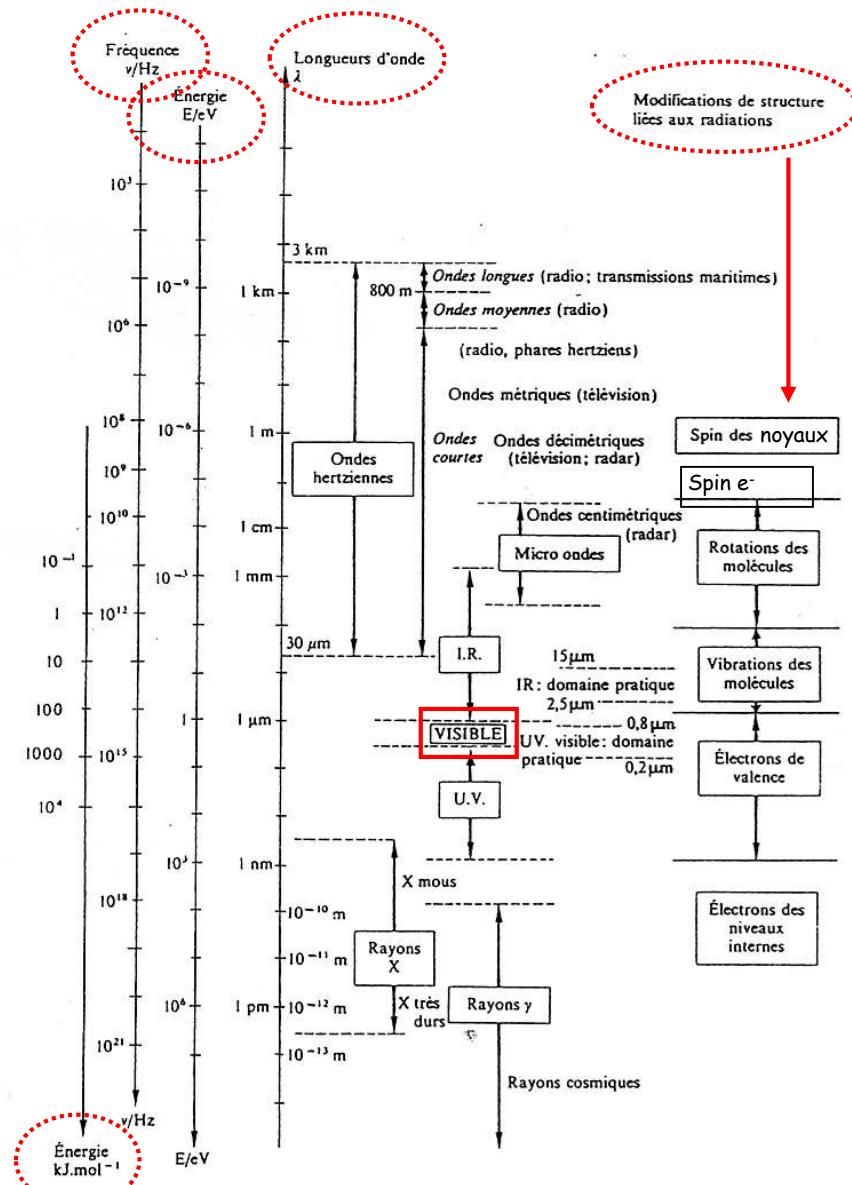
order of magnitude : very small !...
sensitivity ?



Man, Encyclopedia of analytical chemistry, 2000, 12228.

$B_1(RF)$ "at resonance" !

Electromagnetic spectrum



frequency ν (Hz)

wave length $\lambda = c/\nu$ (m)

wave number $\nu \equiv 1/\lambda$ (cm⁻¹)

energy $h\nu$ (J)

units : eV, kJ.mol⁻¹, cm⁻¹

NMR

B_0 (T)

7

14

21

ν_0 (¹H) (MHz)

300

600

900



Les radiations électromagnétiques.

Electron Paramagnetic Resonance

The case of the proton

$$\hat{\mu} = \gamma \hbar \hat{I}$$

$$\gamma \hbar = g_N \beta_N$$

$$g_N = 5,5855$$

$$\beta_N = e\hbar/(2m_p)$$

$$= 5,051 \cdot 10^{-27} \text{ J.T}^{-1}$$

The case of the electron

$$g_e = 2,0023$$

$$\beta_e = e\hbar/(2m_e)$$

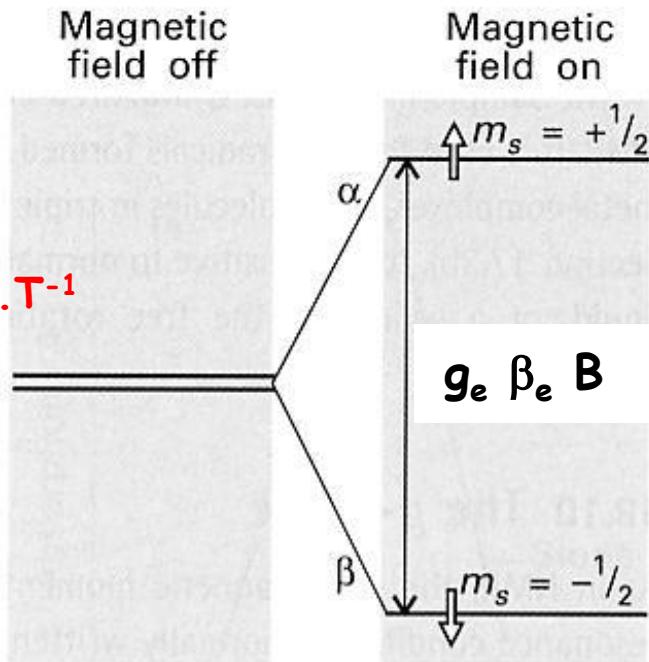
$$= 9,274 \cdot 10^{-24} \text{ J.T}^{-1}$$

Order of magnitude:

$$B \approx 0,3 \text{ T}$$

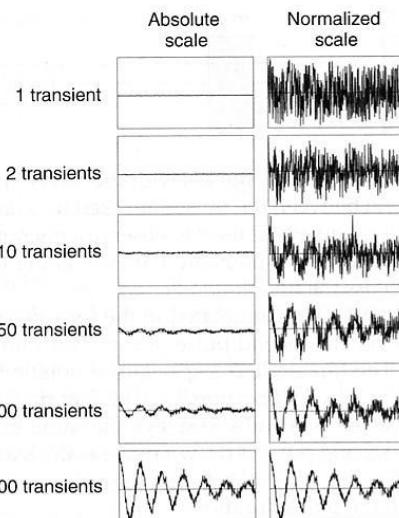
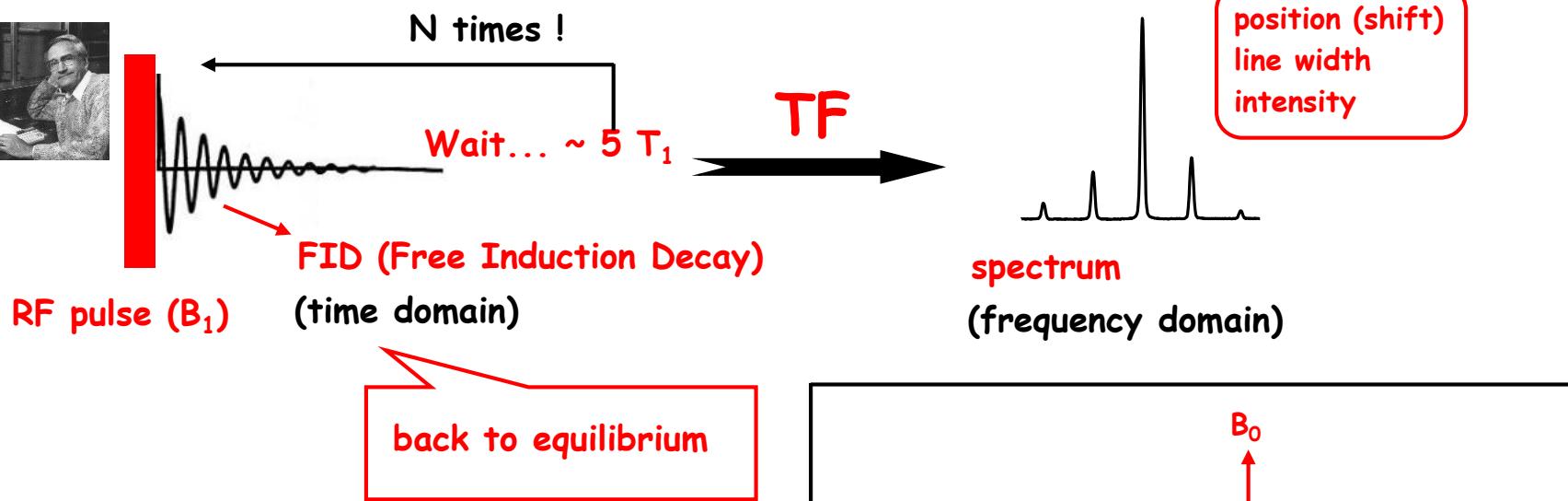
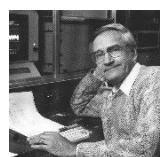
$$\nu = 9 \text{ GHz} = 9 \cdot 10^9 \text{ Hz} ; \lambda \approx 3 \text{ cm}$$

microwaves

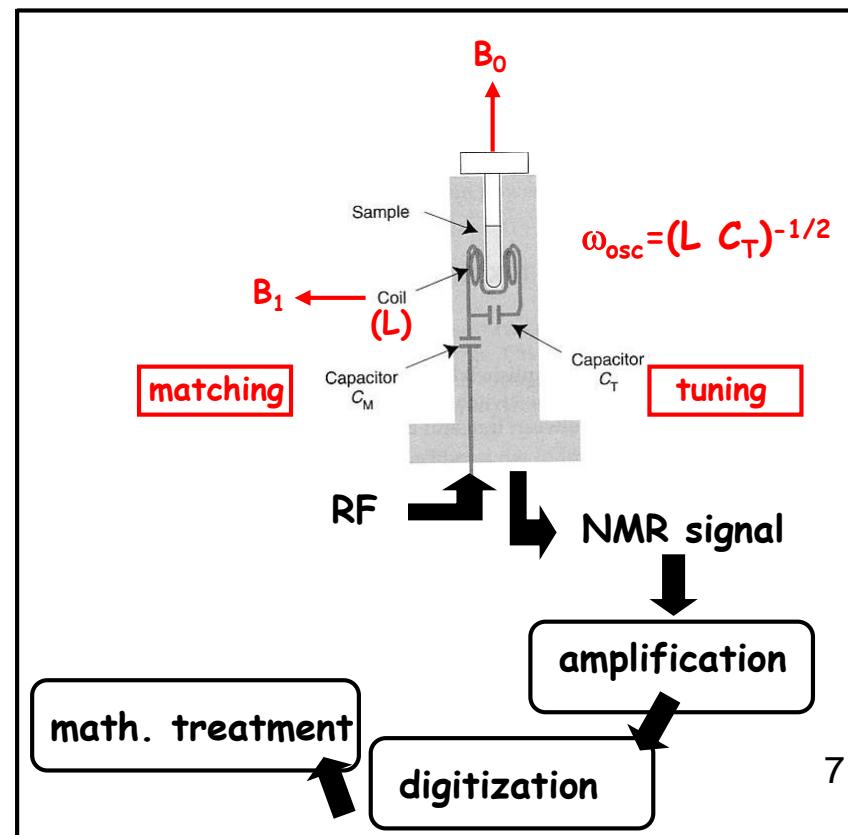


Fourier transform NMR

« The idea » :



Levitt, Spin dynamics, 2002.

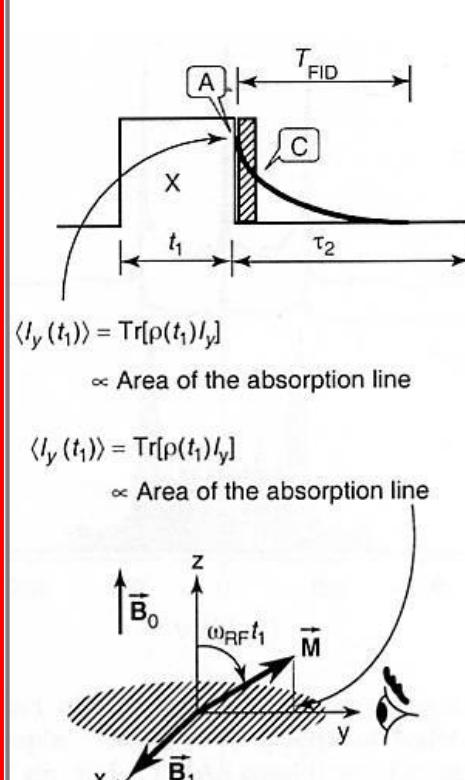
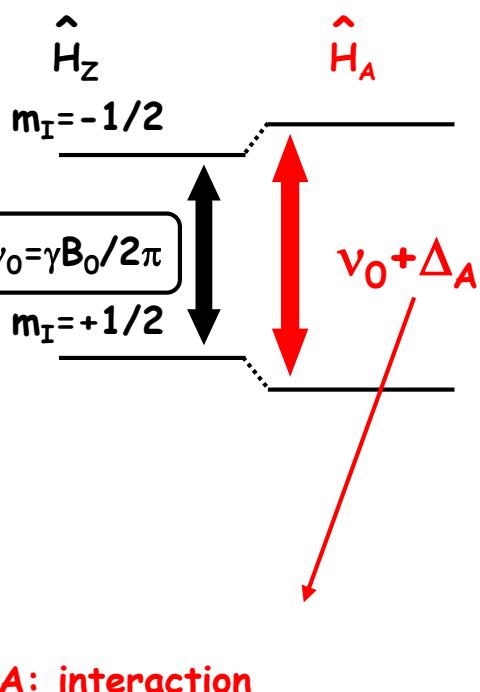


NMR and quantum mechanics

Towards quantum mechanics:

$$\hat{H}_z = -\gamma B_0 \hat{I}_z$$

$$\hat{H}_{RF} = -\gamma B_1 (\cos \omega_{ref} t) \hat{I}_x$$



10^{18} spins: density matrix

$$\rho_{ml} = \sum_q p^{(q)} c_m^{(q)} c_l^{(q)*} = \overline{c_m c_l^*}.$$

$$\rho(t) = \begin{bmatrix} \text{---} \\ \text{---} \\ \text{---} \end{bmatrix} |$$

$$\frac{\partial}{\partial t} \rho(t) = -\frac{i}{\hbar} [H(t), \rho(t)]$$

Liouville-von Neumann equation

$$\langle A \rangle = \sum_{l,m=-j}^j \rho_{ml} A_{lm} = \text{Tr}(\rho)(A)$$

observable
ex: I_y

Interactions in NMR

2

dipole-dipole
(dipolar)
(INT)

INT: internal

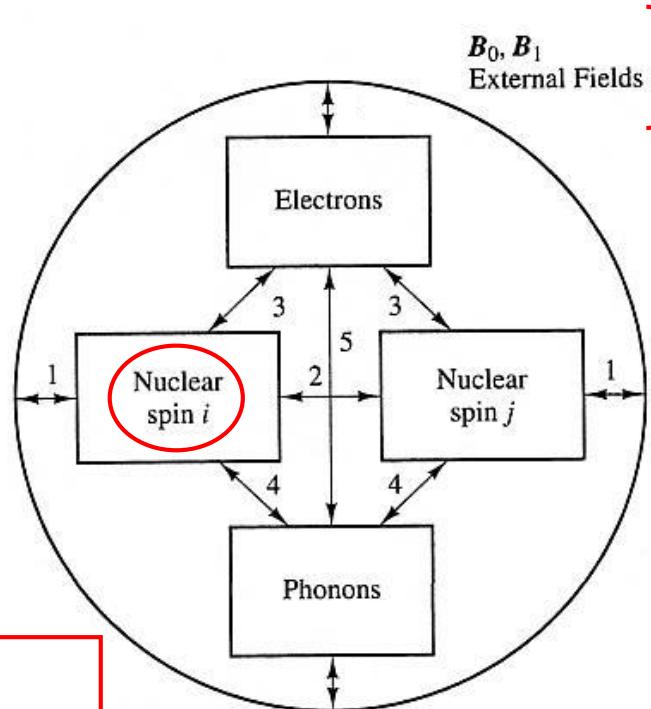
EXT: external

3

electron-nucleus
(chemical shift,
quadrupolar, J, Knight
shift)
(INT)

« spies » for
structure elucidation

$$\hat{H} = \underbrace{\hat{H}_{\text{ext}}}_{(\hat{H}_0 + \hat{H}_{\text{RF}})} + \underbrace{\hat{H}_{\text{int}}}_{(\hat{H}_D + \hat{H}_{\text{CS}} + \hat{H}_Q \dots)}$$



1

(EXT)

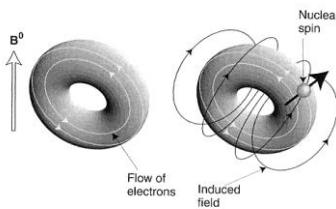
manipulation of the
quantum states

4, 5

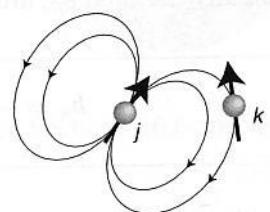
relaxation

back to equilibrium

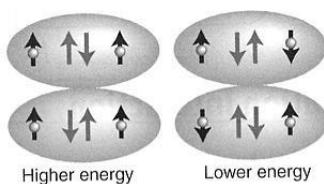
Fundamental interactions for chemists



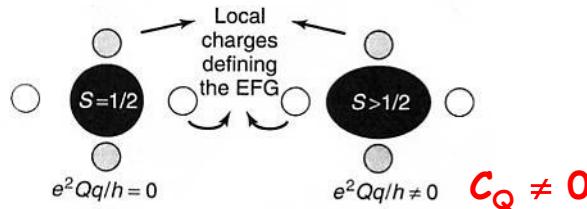
chemical shift :
 δ



dipolar interaction : D



indirect coupling: J



quadrupolar interaction ($I > \frac{1}{2}$)

Levitt, Spin dynamics, 2002.

Frydman, Encyclopedia of NMR, supp. Vol., 263.

mathematical treatment

$$\hat{\mathcal{H}}_{\text{int}} = \hbar \hat{\mathbf{I}} \cdot \mathbf{A} \cdot \hat{\mathbf{X}} = \hbar (\hat{I}_x \quad \hat{I}_y \quad \hat{I}_z) \begin{pmatrix} A_{xx} & A_{xy} & A_{xz} \\ A_{yx} & A_{yy} & A_{yz} \\ A_{zx} & A_{zy} & A_{zz} \end{pmatrix} \begin{pmatrix} \hat{X}_x \\ \hat{X}_y \\ \hat{X}_z \end{pmatrix}$$

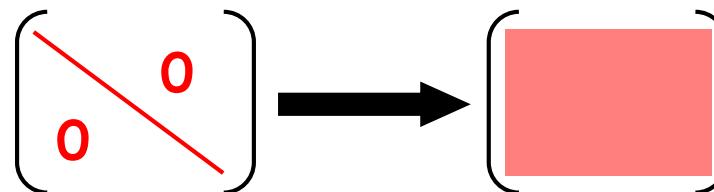
(CS, D, Q...)

nuclear spin operator

the interaction: a
second rank tensor
(symmetrical)

another spin
operator or B_0 ...

anisotropy : why ?

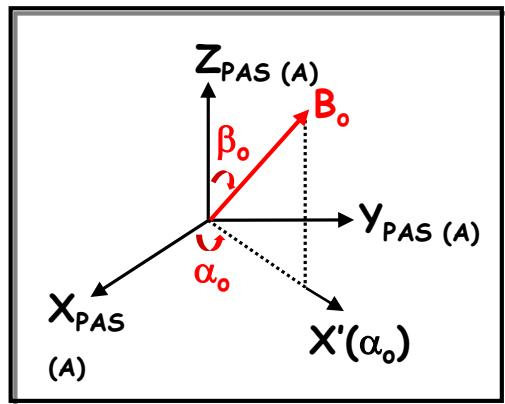


diagonal in the PAS
(Principal Axes System)

LAB

Principal values A_{ii} - Ellipsoid representation

For each interaction A (CS, D, Q...)



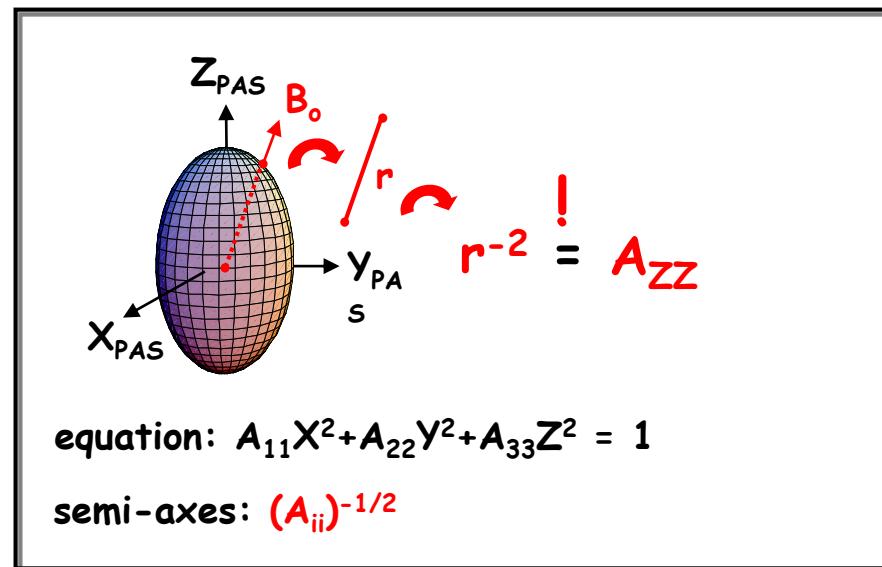
$$\begin{pmatrix} A_{11} & 0 \\ 0 & A_{22} \\ 0 & A_{33} \end{pmatrix}$$

... at the level of the nucleus ...



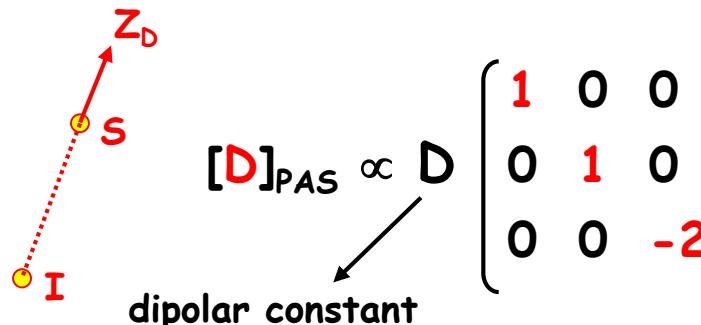
$$\begin{pmatrix} A_{xx} & A_{xy} & A_{xz} \\ A_{yx} & A_{yy} & A_{yz} \\ A_{zx} & A_{zy} & A_{zz} \end{pmatrix}_{LAB} = \begin{pmatrix} f(\alpha_0, \beta_0) \\ \\ \end{pmatrix}_{LAB}$$

« first order » perturbation



the trace $\text{Tr}A = \sum A_{ii}$ ou $A_{\text{iso}} = 1/3 \text{ Tr}A$

ex: null trace : D, Q



Interactions in solution state NMR

...a degenerated case : all interactions are averaged to their isotropic values...

Remember :

$$\text{Tr}(CS) \neq 0$$

$$\text{Tr}(J) \neq 0 !$$

$$\text{Tr}(D) = \text{Tr}(Q) = 0$$



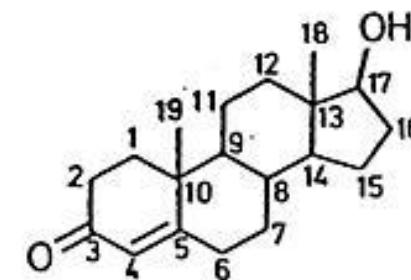
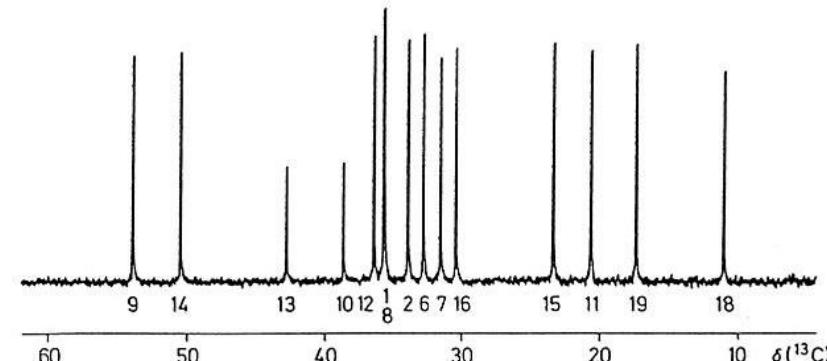
CS : "position of the lines", "fingerprints"

J : "multiplets", connectivity

(D and Q have a direct impact on relaxation...)

HIGH RESOLUTION NMR

$^{13}\text{C}-\{\text{H}\}$: a steroid



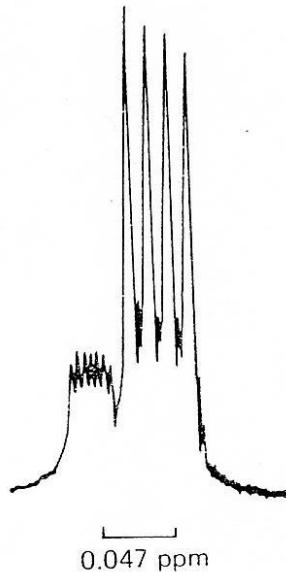
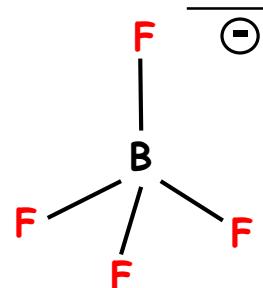
J couplings and multiplets

a spin J coupled to n spins I...



($2nI + 1$) expected lines

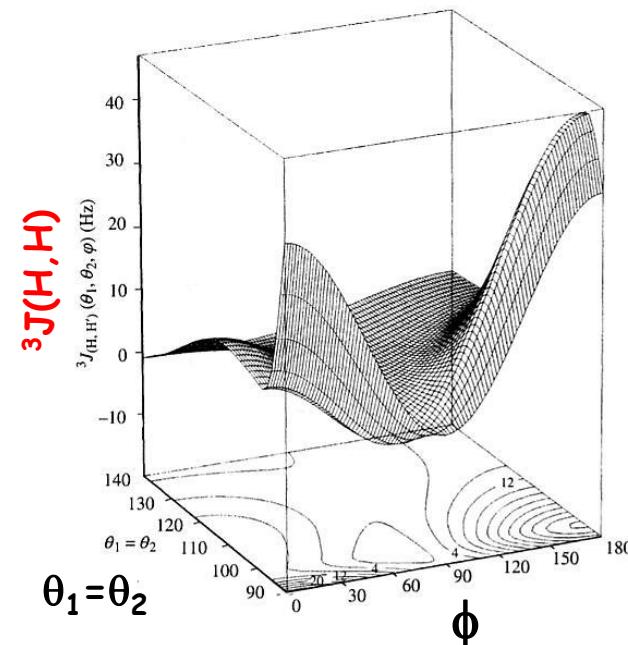
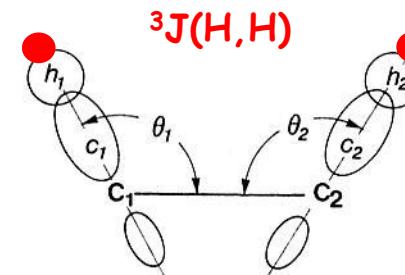
ex: ^{19}F : BF_4^-



^{10}B ($I=3$): 7 lines (nat. ab. 20%)

^{11}B ($I=3/2$) : 4 lines (nat. ab. 80%)

a tool for structural characterization

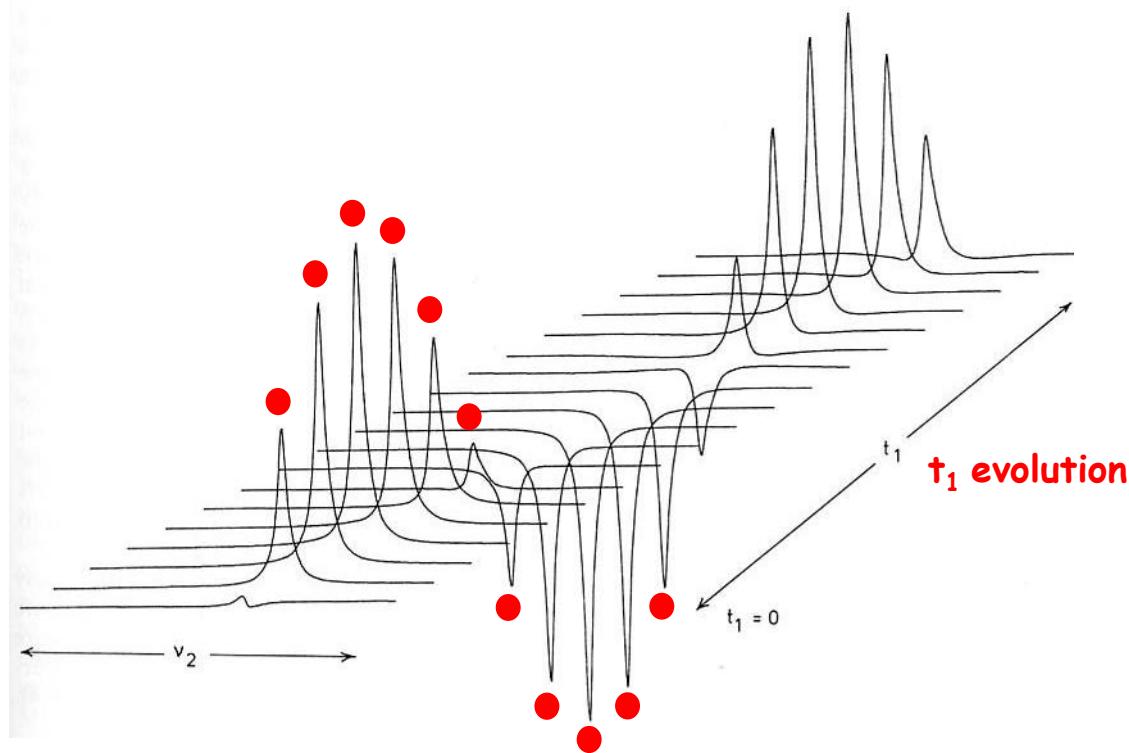


Barfield et al., J. Am. Chem. Soc., 1992,
114, 1574.

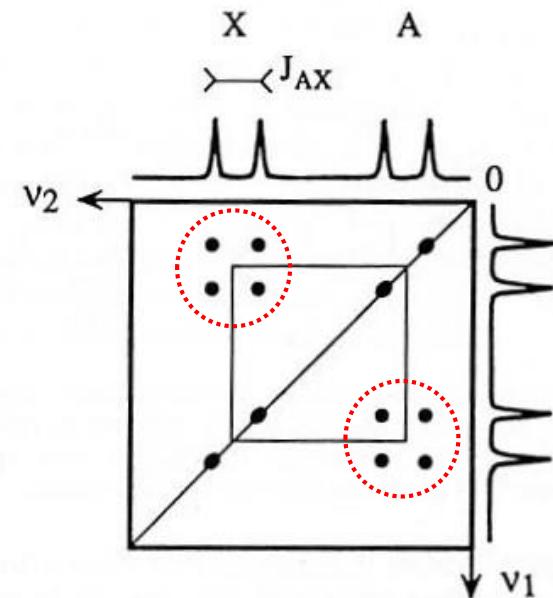
Combining dimensions...



J.Jeener ~ 1971



A-X system

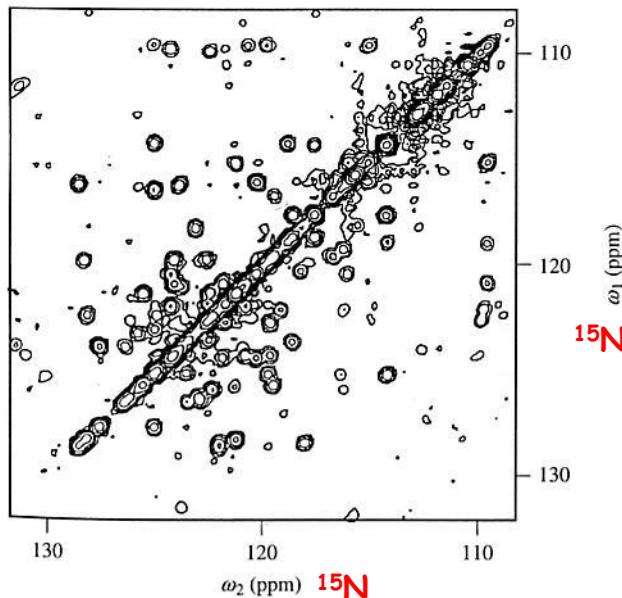
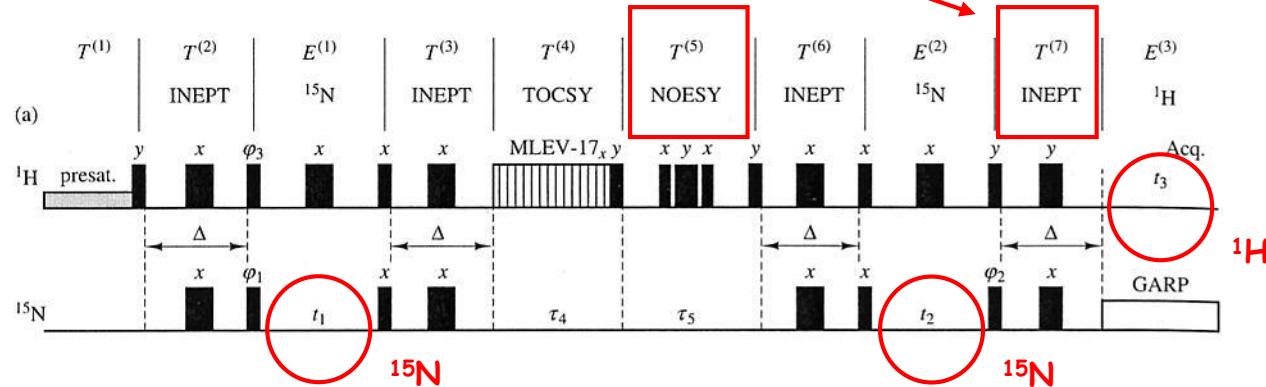


Canet, La RMN: concepts et méthodes, 1991.

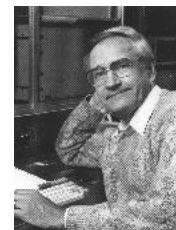
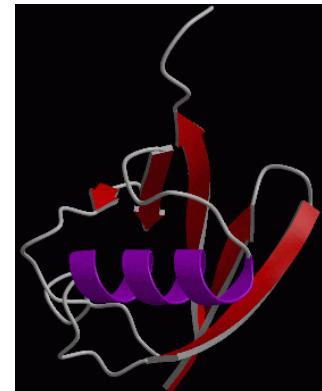
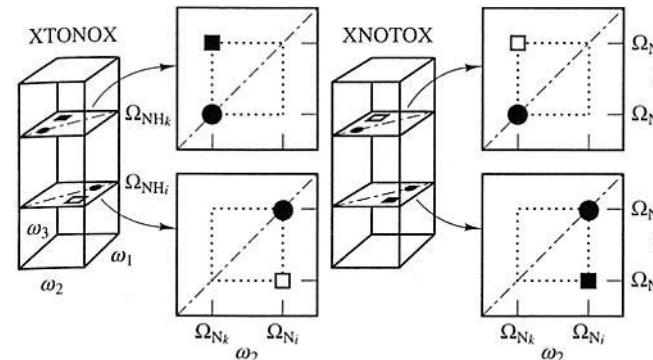
Multidimensional NMR: solution state

CS and J: selection, transfer, edition, correlation... (COSY, INEPT, HETCOR...)

D: relaxation... (NOESY...)



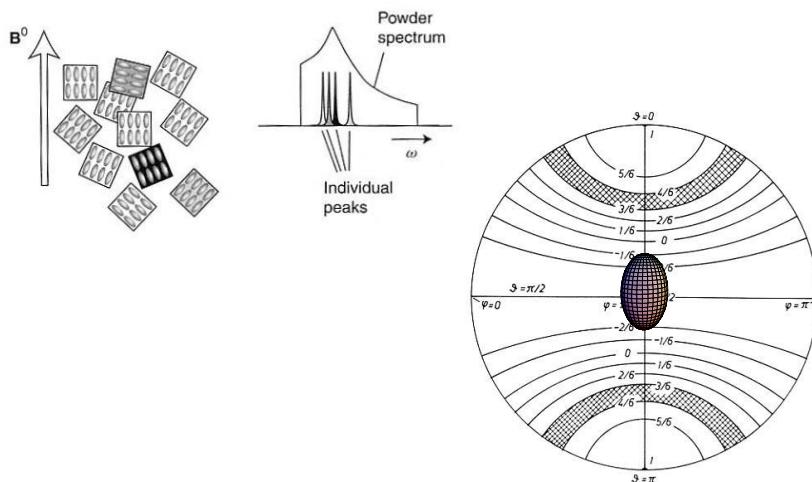
99% ^{15}N -human ubiquitine



Ernst, Encyclopedia of NMR, 1996, 3130.

When powders are available !

... how to build a CSA lineshape ?



ex : $\delta_{11}=\delta_{22}=\delta_{\perp}$ and $\delta_{33}=\delta_{//}$

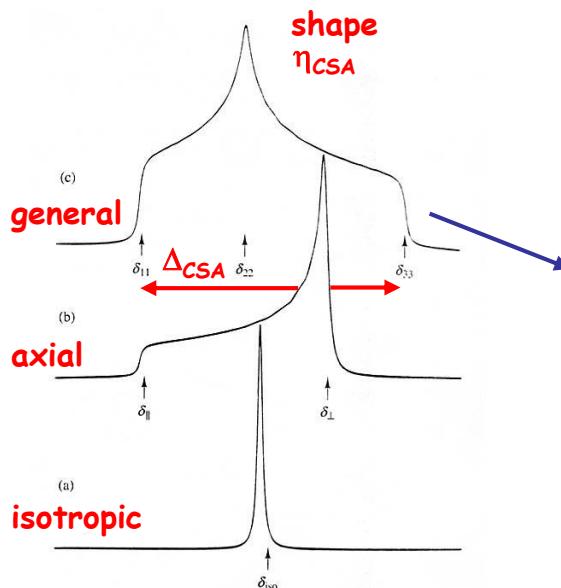
$$r^{-2} = \delta_{zz} = (\delta_{\perp} \sin^2 \beta_0 + \delta_{//} \cos^2 \beta_0)$$

Ellipsoid of revolution !

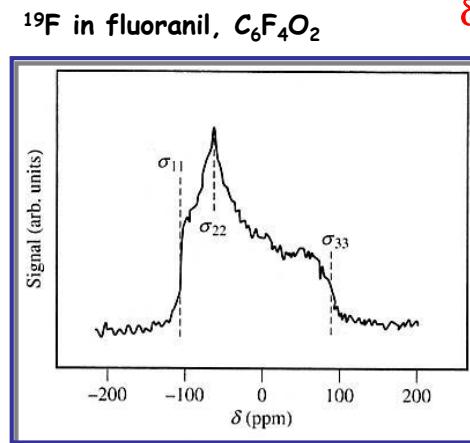
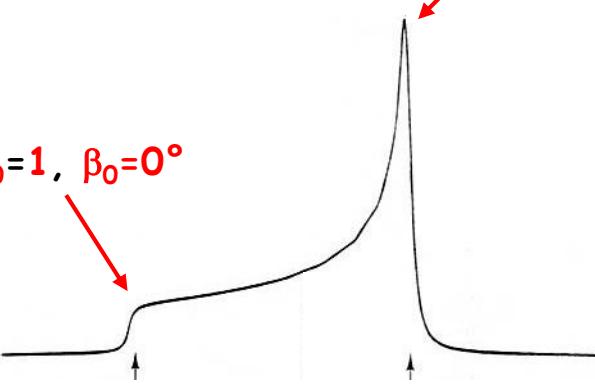
$$\cos^2 \beta_0 = 0, \beta_0 = 90^\circ$$

Levitt, Spin dynamics, 2002.

Haeberlen, High resolution NMR in solids, selective averaging, 1976.



$$\cos^2 \beta_0 = 1, \beta_0 = 0^\circ$$

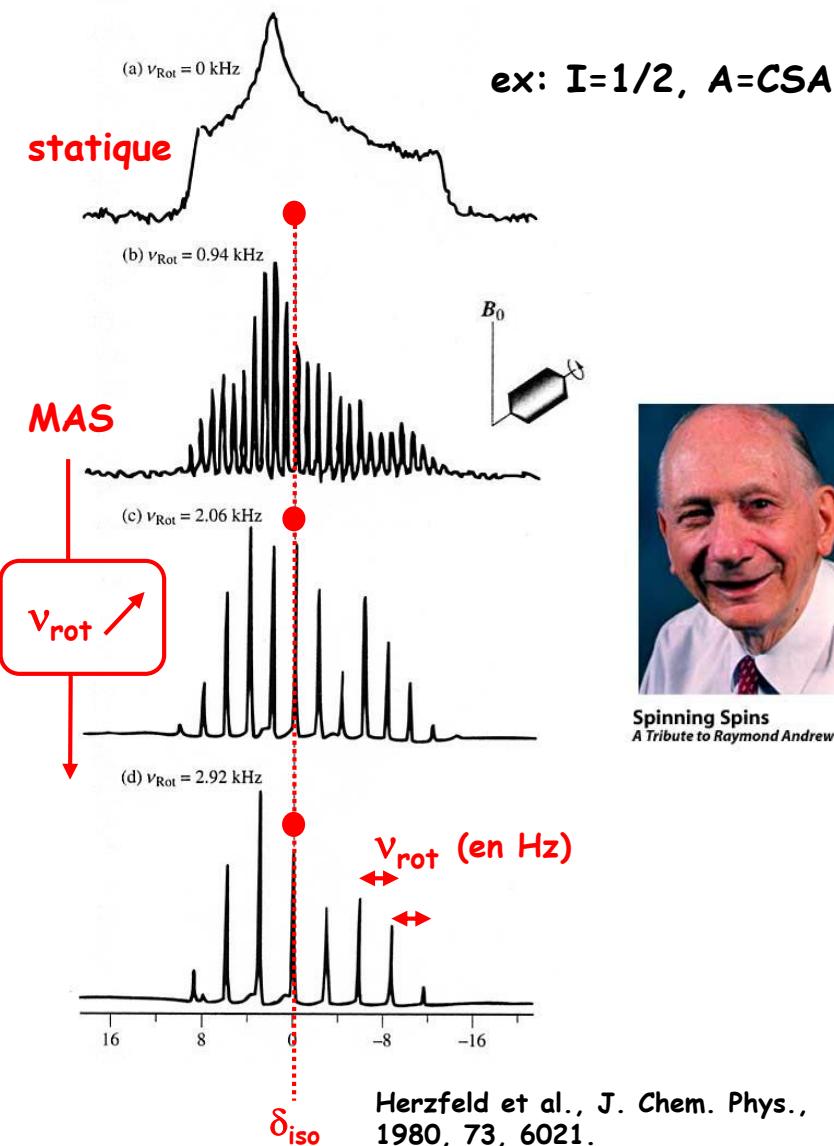


shape: elliptic integrals

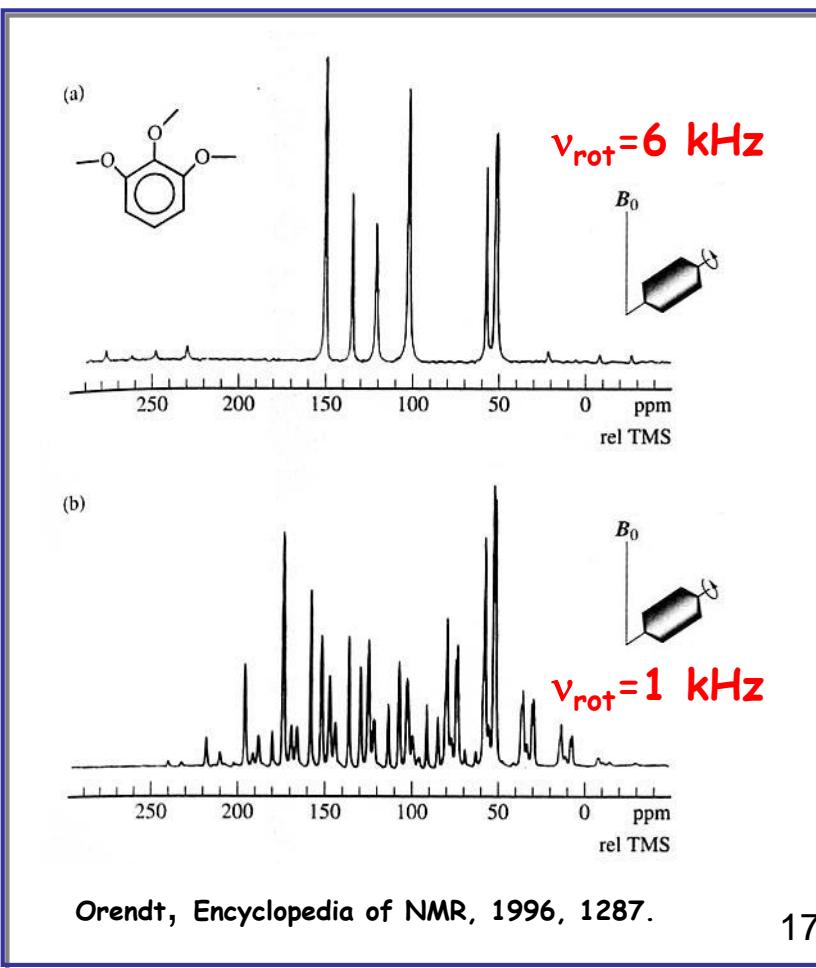
$$K(m) = \int_0^{\pi/2} d\varphi (1 - m \sin^2 \varphi)^{-1/2}$$

Magic Angle Spinning (MAS)

^{31}P : dipalmitoylphosphatidylcholine



"explosion" of the spectrum in sharp
rotation spinning sidebands



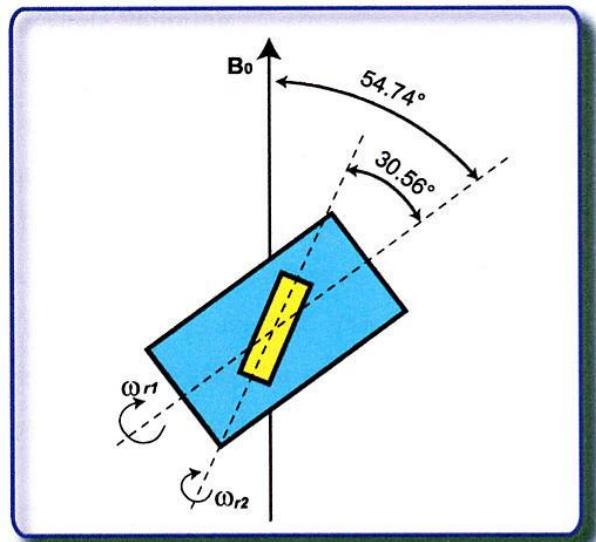
Quadrupolar nuclei and macroscopic rotations

MAS: « one degree of freedom" (1959)

we invent a new experiment involving 2 angles of reorientation !

DOR experiment (DOuble Rotation)

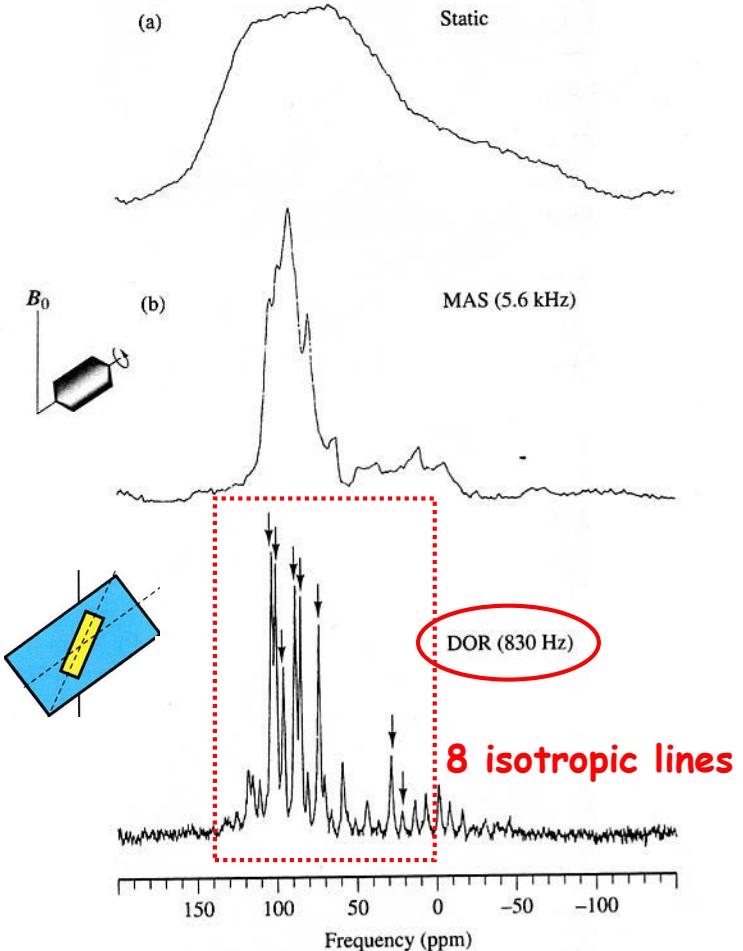
(Samoson, Pines, 1988)



Ziarelli, PhD thesis.

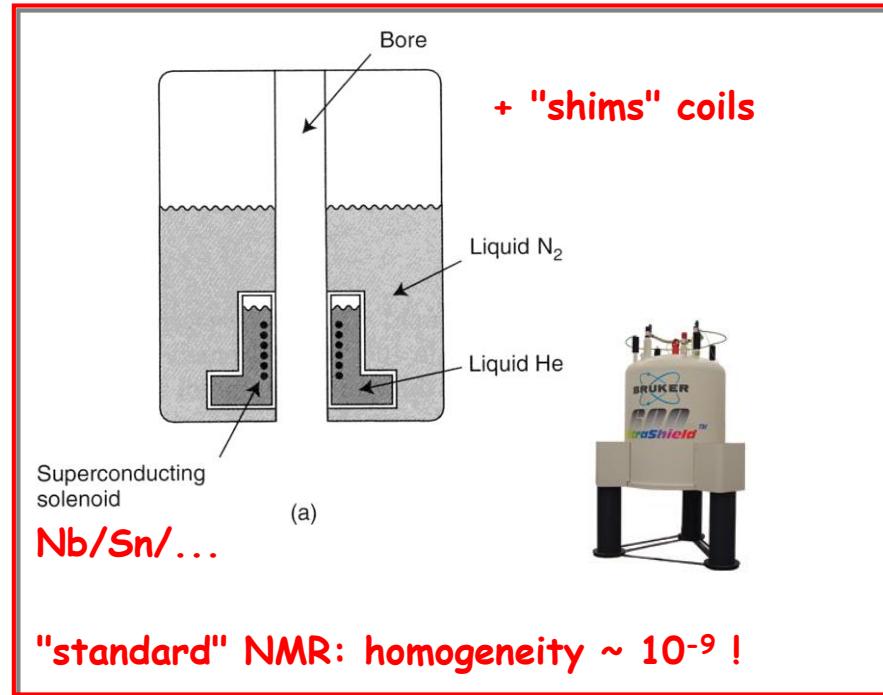
1D experiment

^{17}O ($I=5/2$) : CaSiO_3 wollastonite: 9 ^{17}O sites

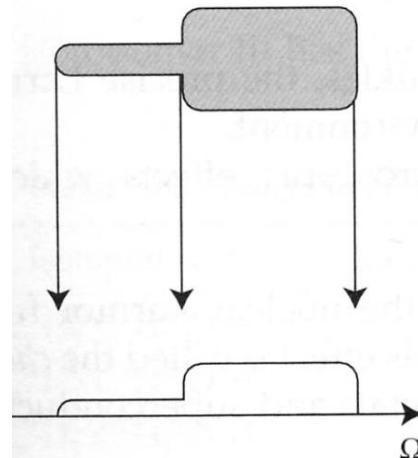
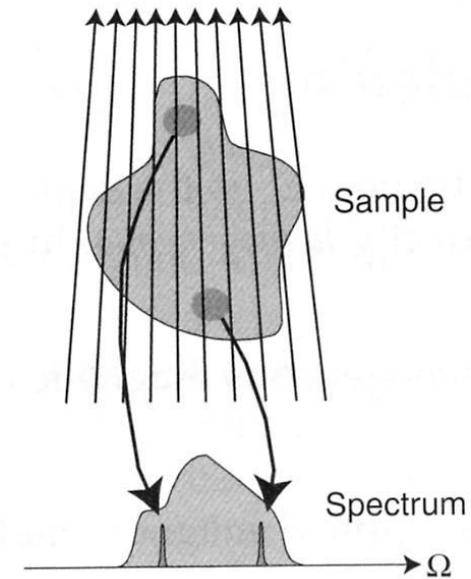


Wu, Encyclopedia of NMR, 1996, 1749.

B_0 : homogeneous or not ?



Inhomogeneous magnetic field: $B_0(r)$



~ 1972

!! → imaging

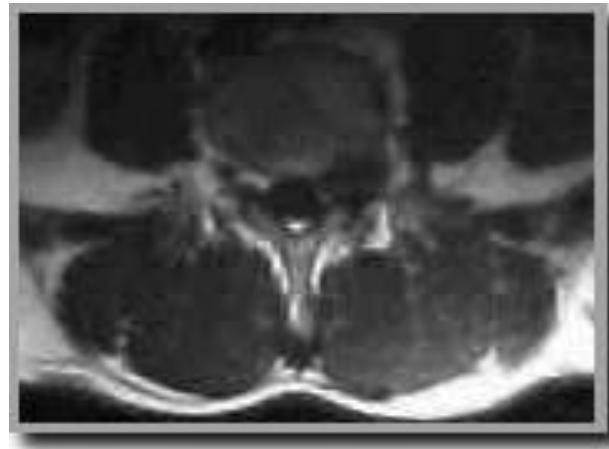
NMR imaging



brain



lungs



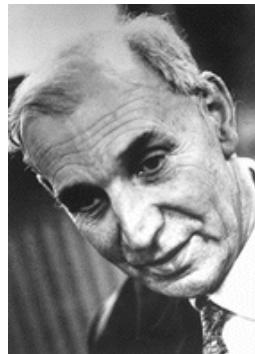
slipped disk



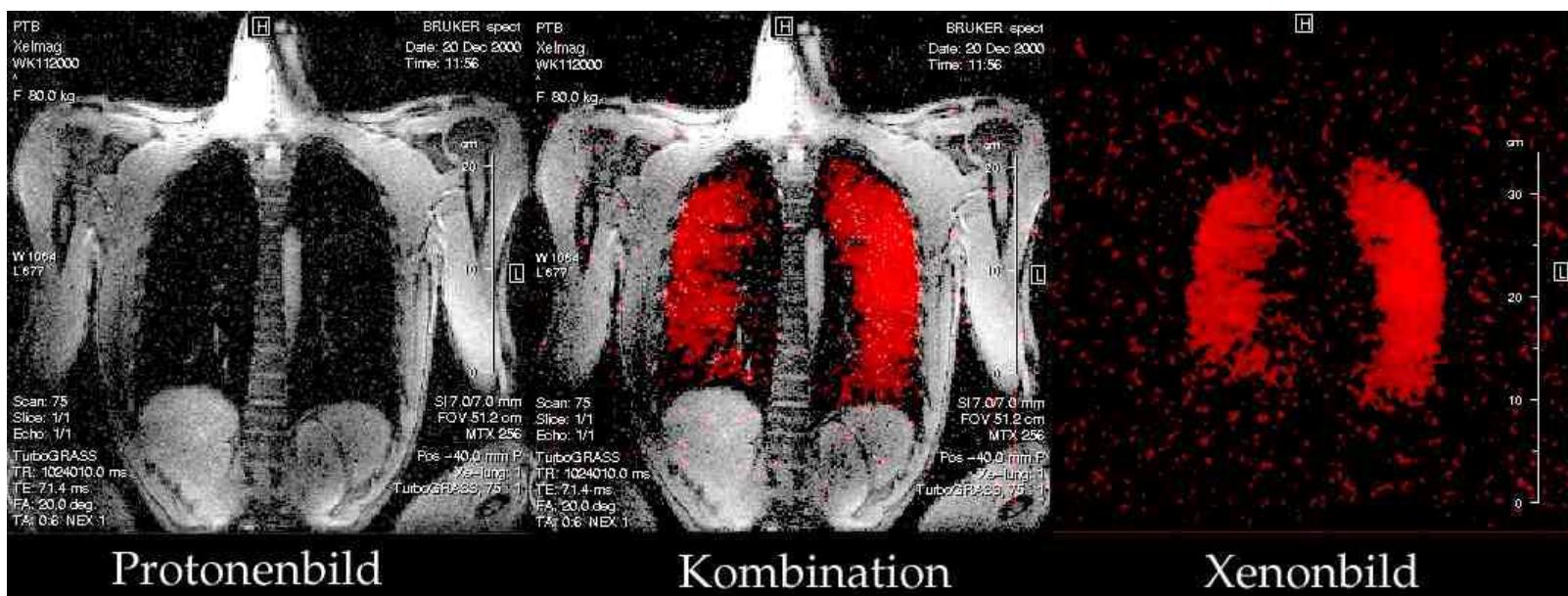
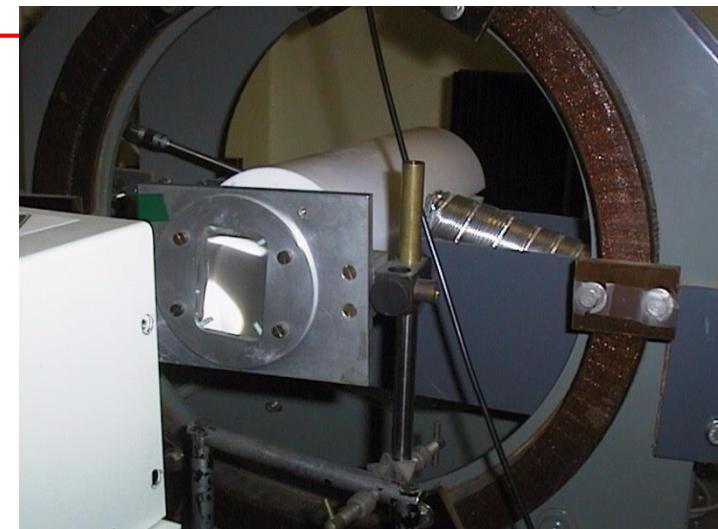
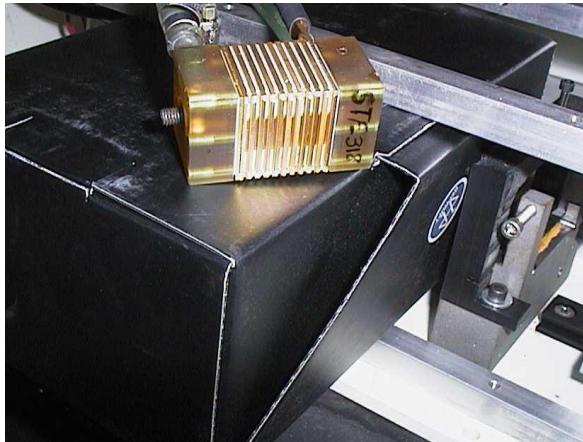
horizontal magnet: $B_0 \approx 2T$

Imaging by hyperpolarization

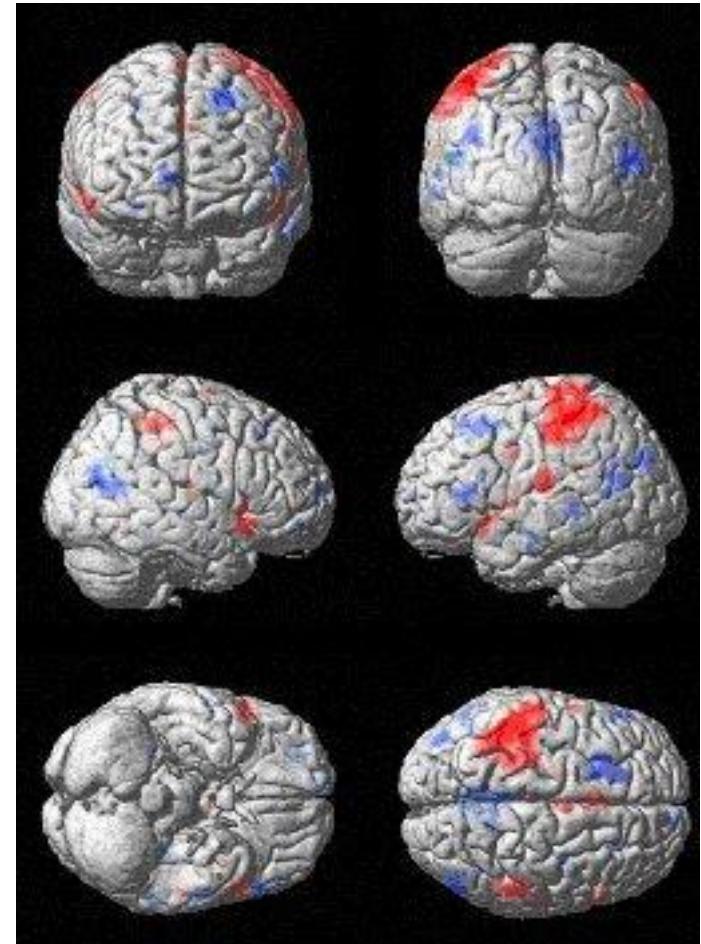
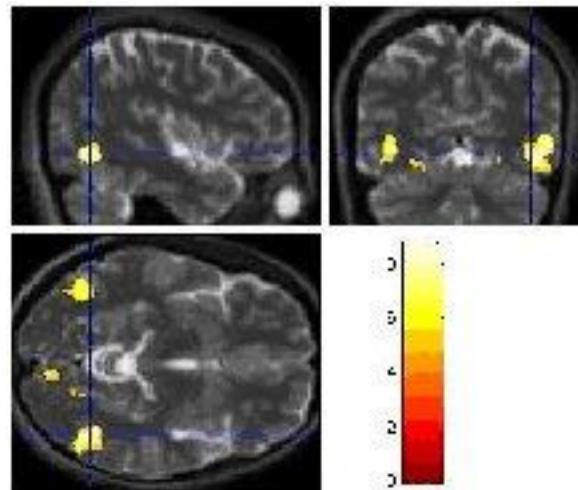
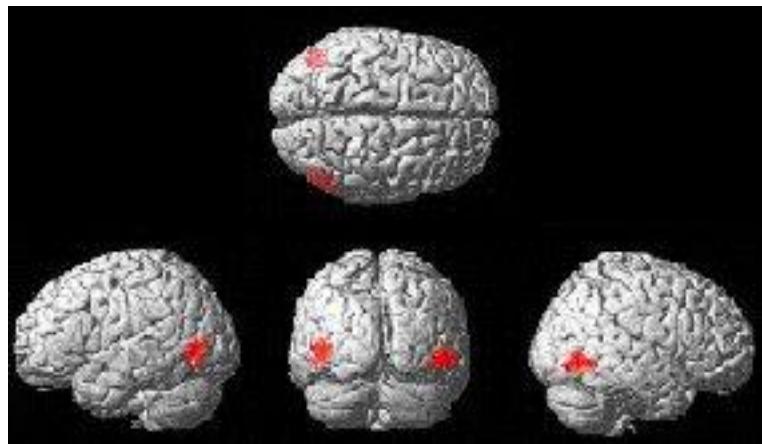
^3He , ^{17}O , ^{129}Xe ...



A. Kastler,
Physics, 1966



Functional imaging



Answers to stimuli: brain activity!

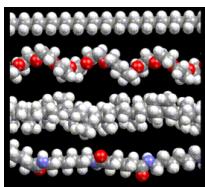
High Resolution Solid State NMR

Christian BONHOMME, Professor

Laboratoire de Chimie de la Matière Condensée
UMR CNRS 7574 - Sorbonne Université, Paris

Some key experiments in solid state NMR

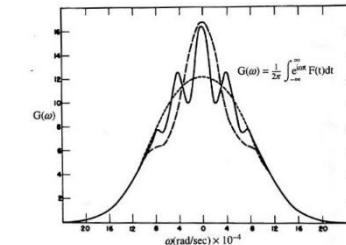
1) NMR of paraffin, 1946 (Purcell, Torrey, Pound)



Reprinted from THE PHYSICAL REVIEW, Vol. 69, Nos. 1 and 2, 37-38, January 1 and 15, 1946
Printed in U. S. A.

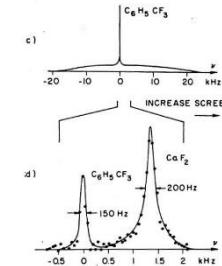
Resonance Absorption by Nuclear Magnetic Moments in a Solid

E. M. PURCELL, H. C. TORREY, AND R. V. POUND*
Radiation Laboratory, Massachusetts Institute of Technology,
Cambridge, Massachusetts
December 24, 1945

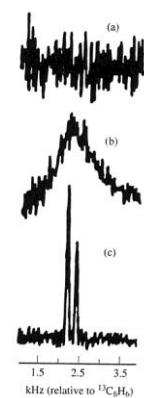


2) Magic Angle Spinning, MAS, 1959 (Andrew, Bradbury, Eades and Lowe)

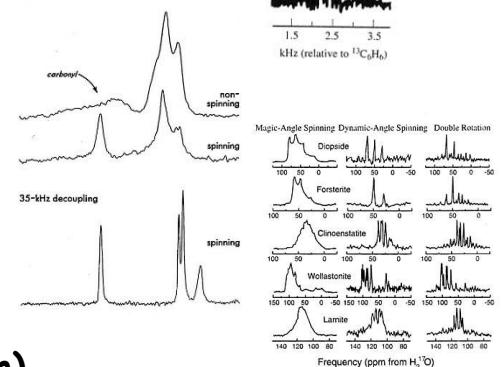
3) WAHUA sequence, 1968 (Waugh, Huber, Haeberlen)



4) Cross Polarization (CP), 1962 (Hartmann, Hahn)



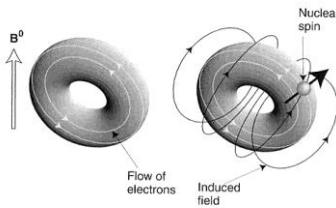
5) Indirect observation of dilute spins, 1972 (Pines, Gibby, Waugh)



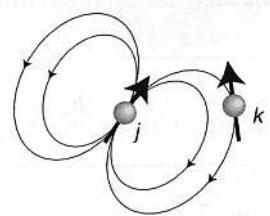
6) The CP MAS experiment, 1976 (Schaefer, Stejskal)

7) Quadrupolar nuclei: high resolution, 1988, 1995 (Virlet, Llor, Pines, Frydman)

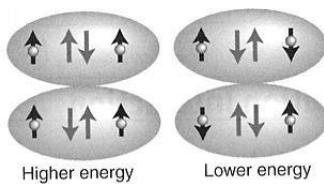
Fundamental interactions for chemists



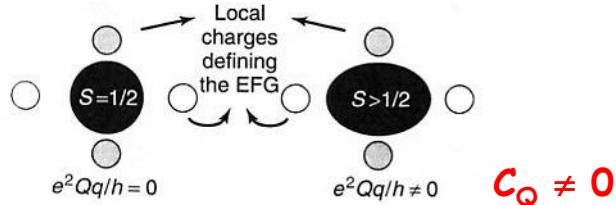
chemical shift :
 δ



dipolar interaction : D



indirect coupling: J



quadrupolar interaction ($I > \frac{1}{2}$)

Levitt, Spin dynamics, 2002.

Frydman, Encyclopedia of NMR, supp. Vol., 263.

mathematical treatment

$$\hat{\mathcal{H}}_{\text{int}} = \hbar \hat{\mathbf{I}} \cdot \mathbf{A} \cdot \hat{\mathbf{X}} = \hbar (\hat{I}_x \quad \hat{I}_y \quad \hat{I}_z) \begin{pmatrix} A_{xx} & A_{xy} & A_{xz} \\ A_{yx} & A_{yy} & A_{yz} \\ A_{zx} & A_{zy} & A_{zz} \end{pmatrix} \begin{pmatrix} \hat{X}_x \\ \hat{X}_y \\ \hat{X}_z \end{pmatrix}$$

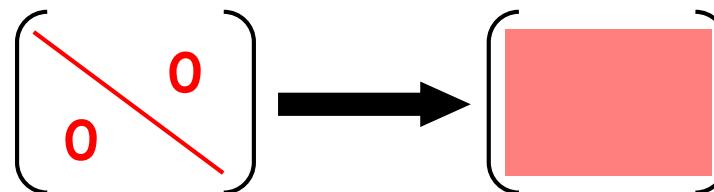
(CS, D, Q...)

nuclear spin operator

the interaction: a
second rank tensor
(symmetrical)

another spin
operator or B_0 ...

anisotropy : why ?

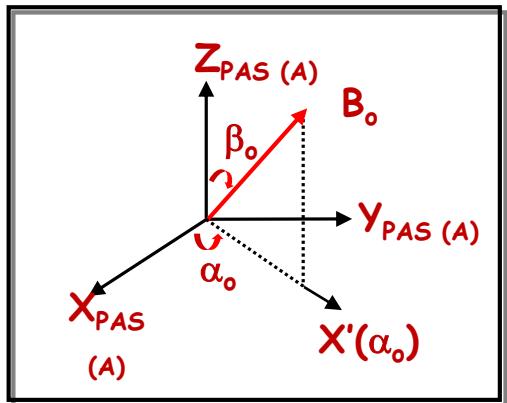


diagonal in the PAS
(Principal Axes System)

LAB

Principal values A_{ii} - Ellipsoid representation

For each interaction A (CS, D, Q...)

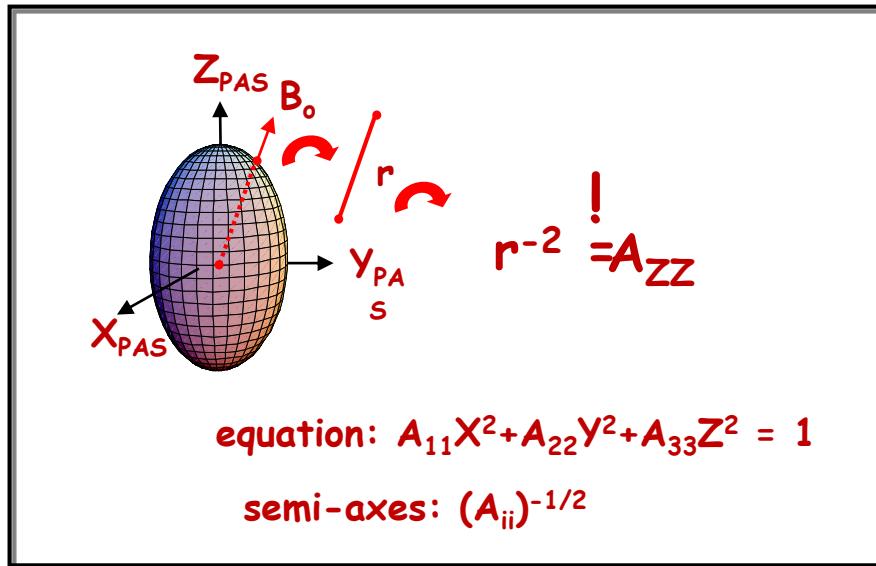


$$\begin{pmatrix} A_{11} & A_{22} & 0 \\ 0 & A_{33} & 0 \\ 0 & 0 & 0 \end{pmatrix}$$

... at the level of the nucleus ...

$$\begin{pmatrix} A_{xx} & A_{xy} & A_{xz} \\ A_{yx} & A_{yy} & A_{yz} \\ A_{zx} & A_{zy} & A_{zz} \end{pmatrix}_{LAB} = \begin{pmatrix} f(\alpha_0, \beta_0) \\ f(\alpha_0, \beta_0) \\ f(\alpha_0, \beta_0) \end{pmatrix}_{LAB}$$

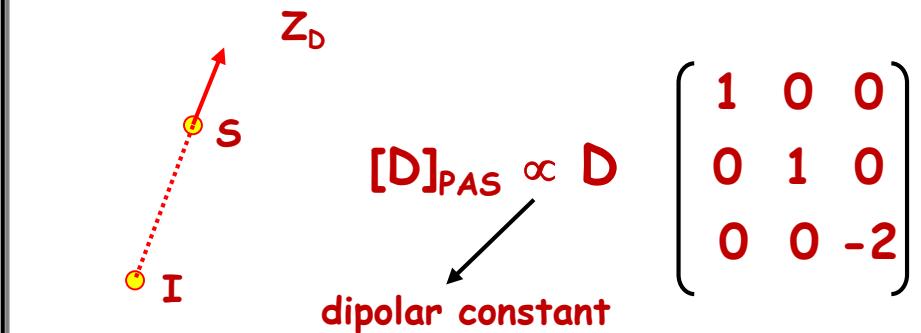
« first order » perturbation



equation: $A_{11}X^2 + A_{22}Y^2 + A_{33}Z^2 = 1$
semi-axes: $(A_{ii})^{-1/2}$

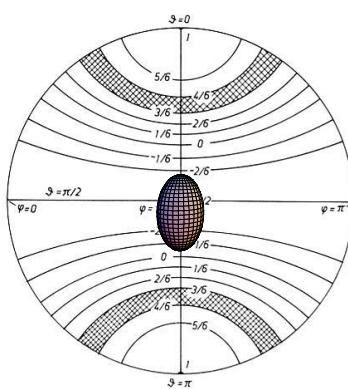
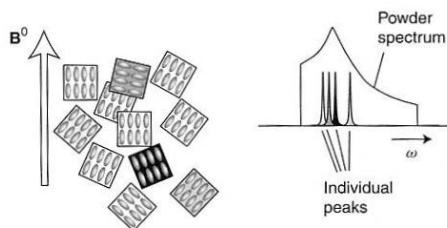
the trace $\text{Tr}A = \sum A_{ii}$ ou $A_{\text{iso}} = 1/3 \text{ Tr}A$

ex: null trace : D, Q



$$\begin{pmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & -2 \end{pmatrix}$$

When powders are available !



...how to build a CSA lineshape ?

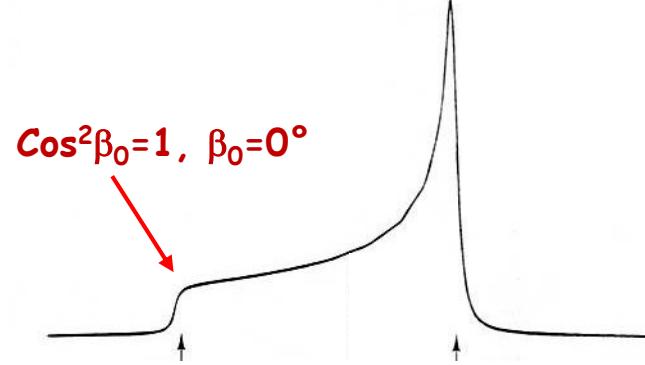
ex : $\delta_{11}=\delta_{22}=\delta_{\perp}$ and $\delta_{33}=\delta_{//}$

$$r^{-2} = \delta_{ZZ} =$$

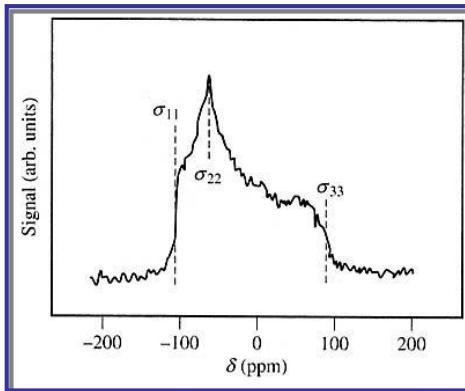
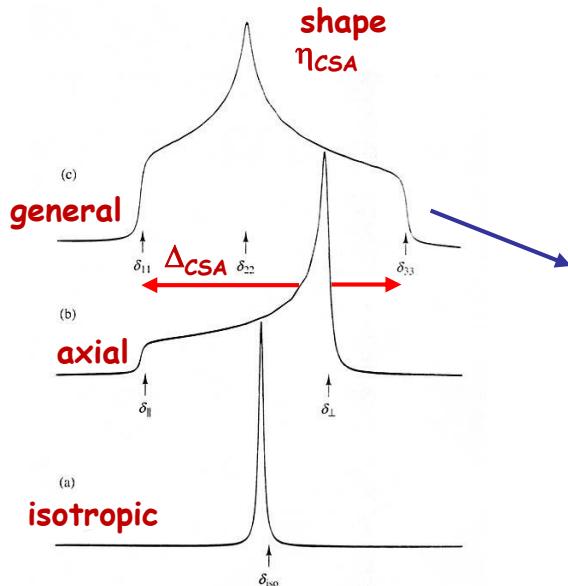
$$(\delta_{\perp} \sin^2 \beta_0 + \delta_{//} \cos^2 \beta_0)$$

Ellipsoid of revolution !

$$\cos^2 \beta_0 = 0, \beta_0 = 90^\circ$$



Levitt, Spin dynamics, 2002.
Haeberlen, High resolution NMR in solids, selective averaging, 1976.

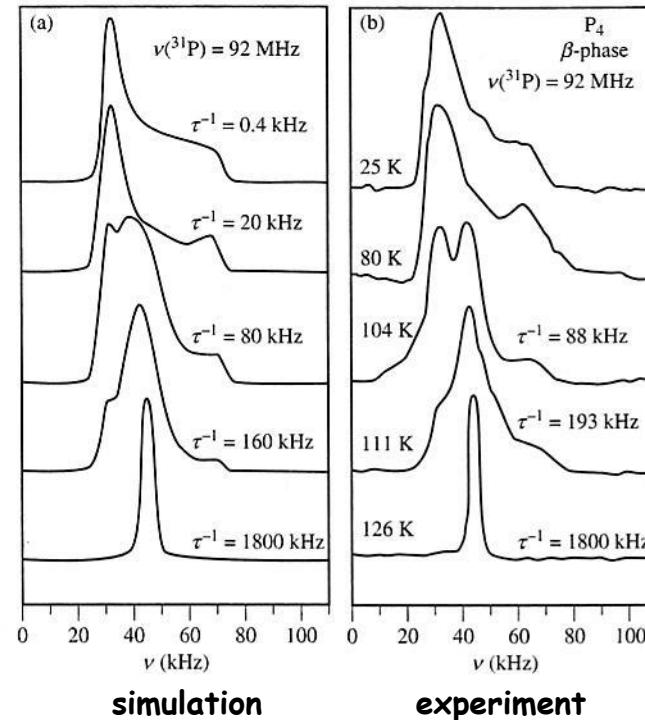
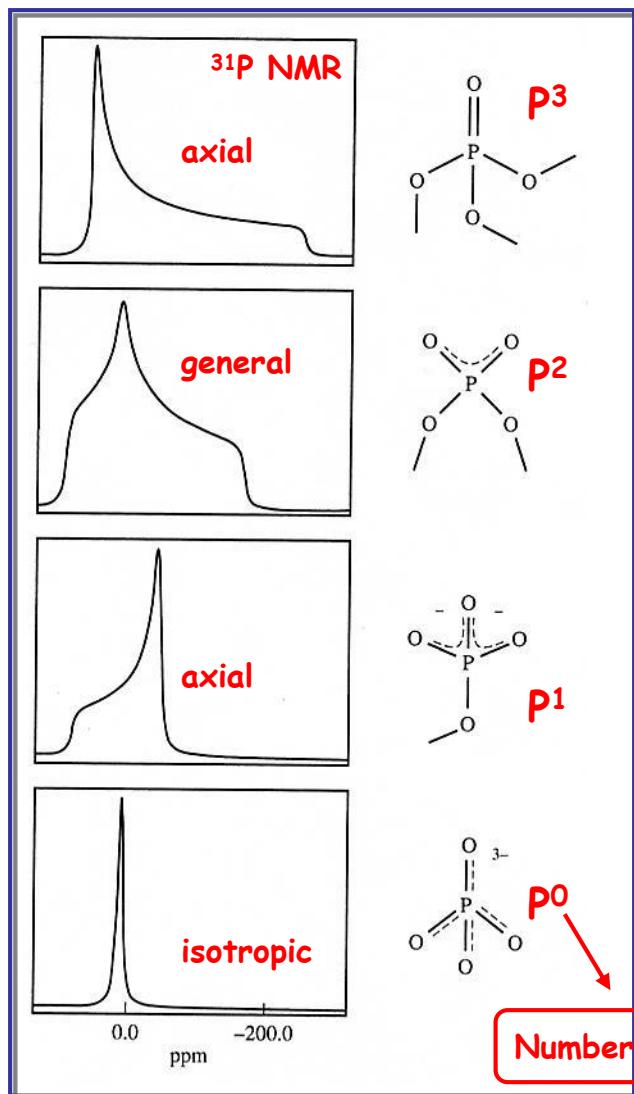


shape: elliptic integrals

$$K(m) = \int_0^{\pi/2} d\phi (1 - m \sin^2 \phi)^{-1/2}$$

Mehring et al., J. Chem. Phys., 1971, 59, 746.

Local symmetry and molecular dynamics



Random jumps : tetrahedron
 (τ^{-1}) (white phosphorus, β phase)

- ◆ Role of dynamics
- ◆ Average : $\tau^{-1} \gg \Delta_A$
- ...NMR in time domain...

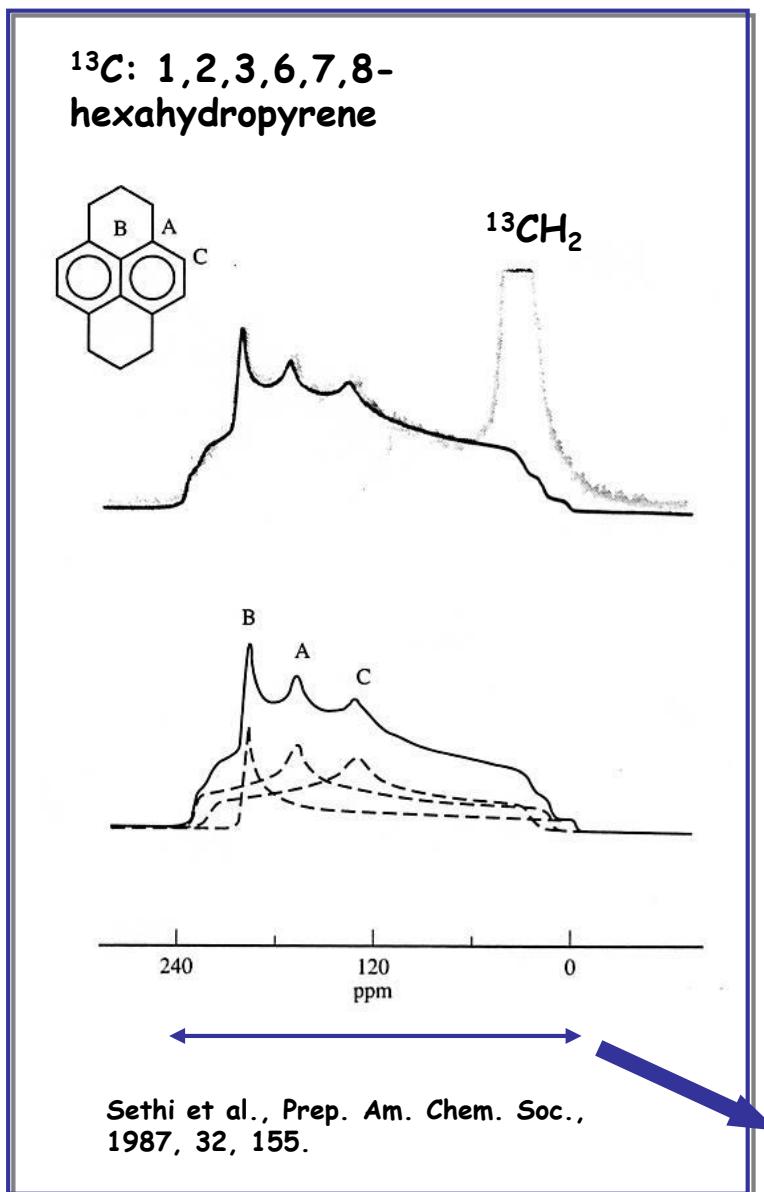
Back to ellipsoids !...

Eckert, Prog. Nucl. Magn. Reson.,
 1992, 24, 159.

Spiess et al., Chem. Phys., 1974, 6,
 226.

Resolution in solid state NMR ?

an example...



All crystallographically equivalent nuclei participate to the same lineshape

All interactions broaden the lines

◆ CSA: it depends...

$$\dots \propto B_0$$

◆ D: up to ~ 30 kHz !

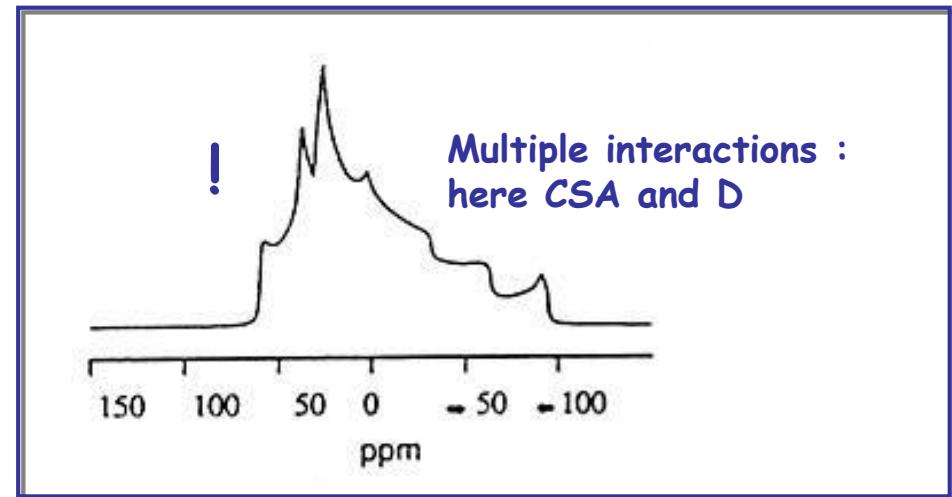
$$\dots \dots \dots \text{ind. } B_0$$

◆ Q: up to MHz !

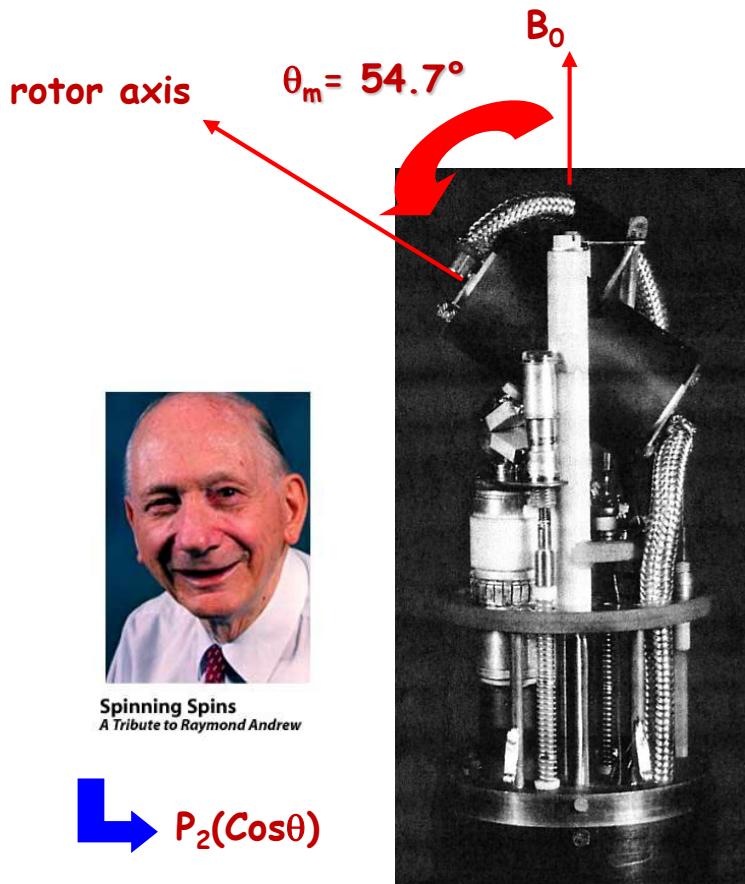
$$\left\{ \begin{array}{l} \text{ind. } B_0. \text{ (1}^{\text{st}}\text{)} \\ 1/B_0 \text{ (2}^{\text{nd}}\text{)} \end{array} \right.$$

◆ J: few 100^s Hz

$$\dots \dots \dots \text{ind. } B_0$$

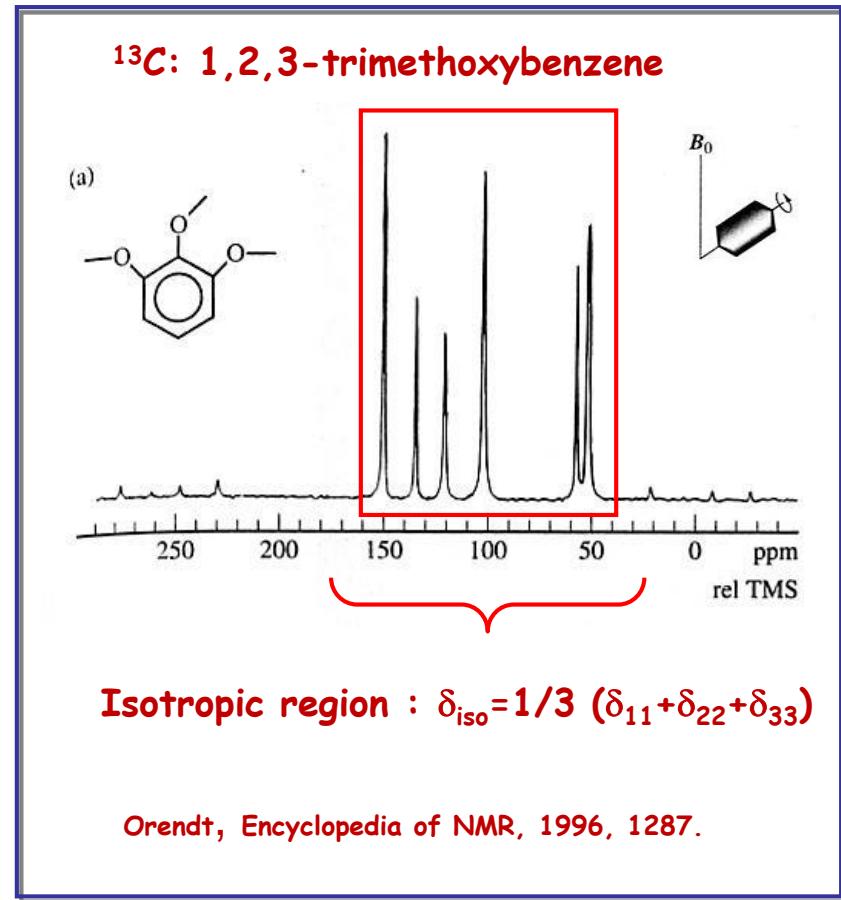
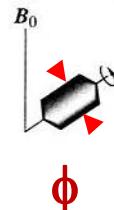


Broadening over the whole ^{13}C chemical shift range !



Doty, Encyclopedia of NMR, 1996, 4477.

- $\phi: 7\text{mm} \longrightarrow \dots 6 \text{ kHz}$
- $\phi: 4\text{mm} \longrightarrow \dots 15 \text{ kHz}$
- $\phi: 2.5\text{mm} \longrightarrow \dots 35 \text{ kHz}$
- $\phi: 1\text{mm} \longrightarrow \dots 100 \text{ kHz}$



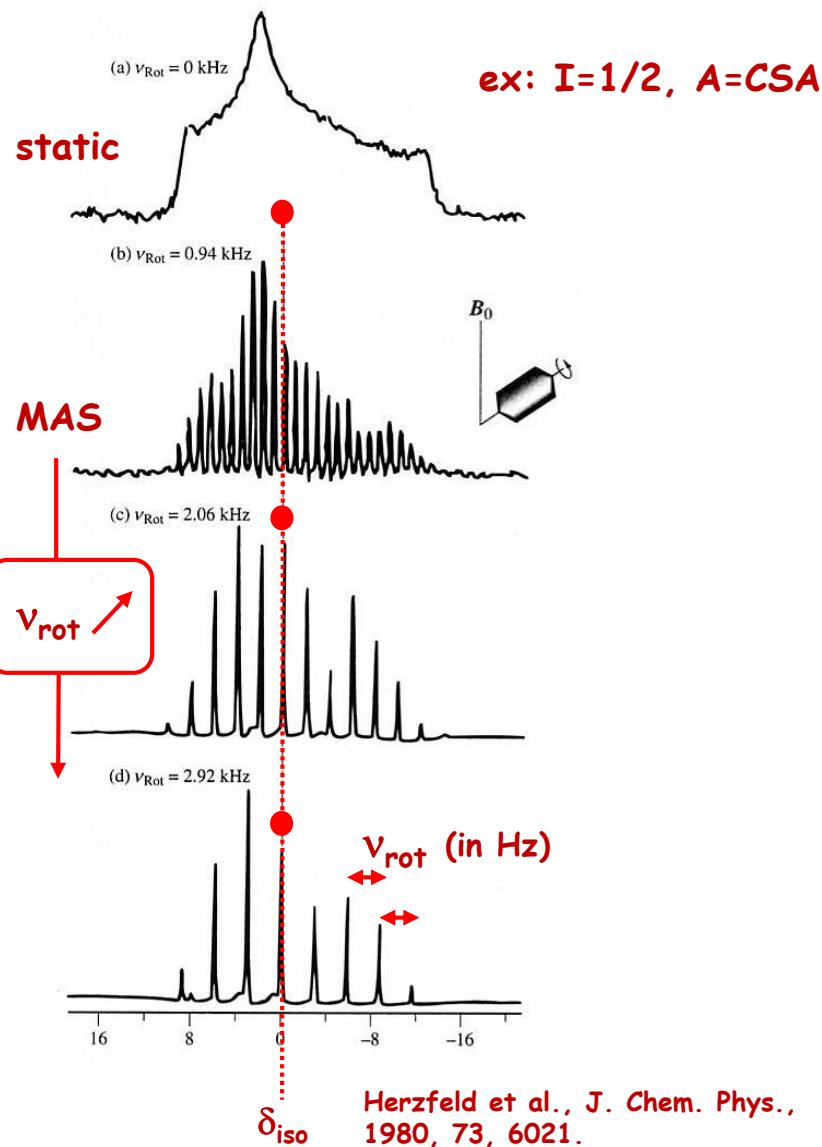
MAS at « infinite » frequency

$v_{\text{rot}} > \Delta_A$ ($A = \text{CSA}, \text{D}, \text{Q} \dots$)

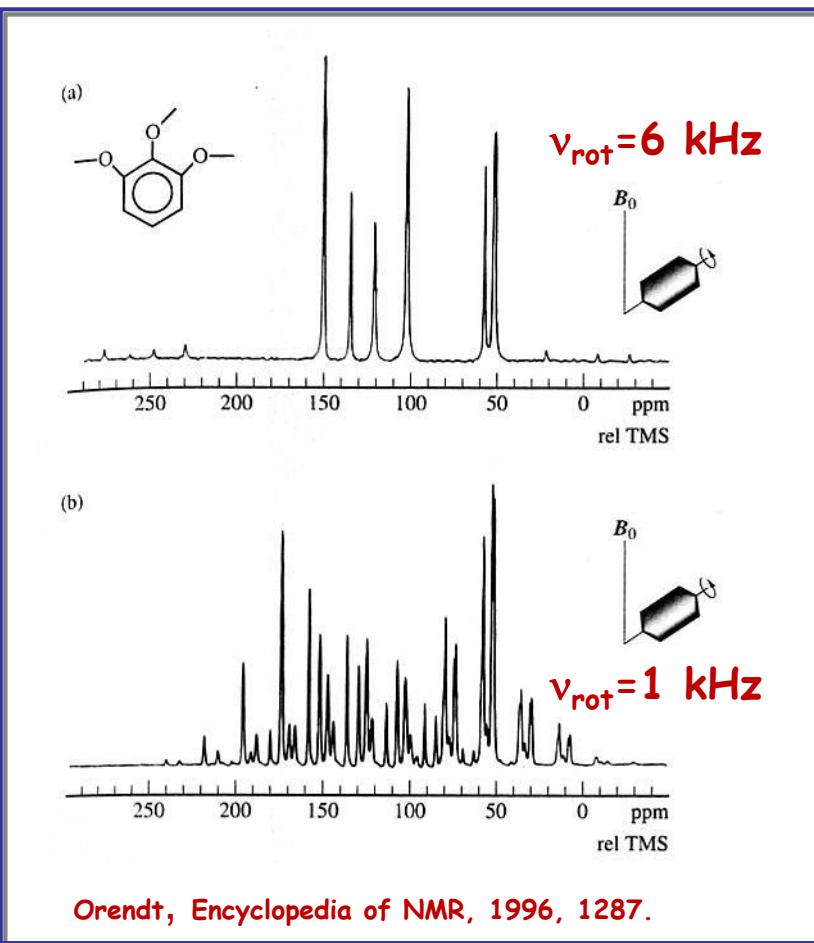
question: is it actually possible ?...

MAS at finite frequency

^{31}P : dipalmitoylphosphatidylcholine



"explosion" of the spectrum in sharp spinning sidebands

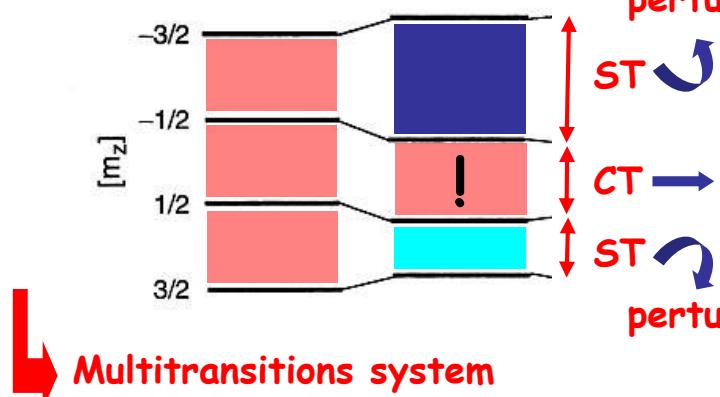


Quadrupolar interaction: first order effects

$I > \frac{1}{2}$ (^{27}Al , ^{23}Na , ^{17}O ...)

ex: $I=3/2$

Zeeman interaction First-order effect

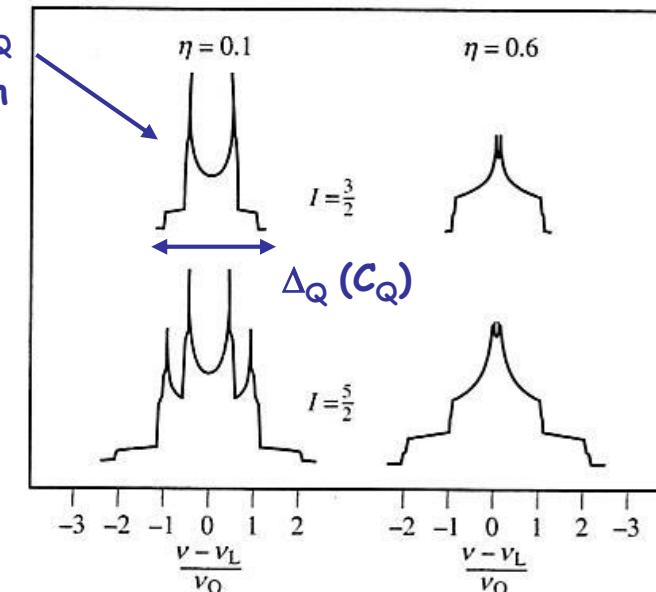


↳ Multitransitions system

CT: central transition

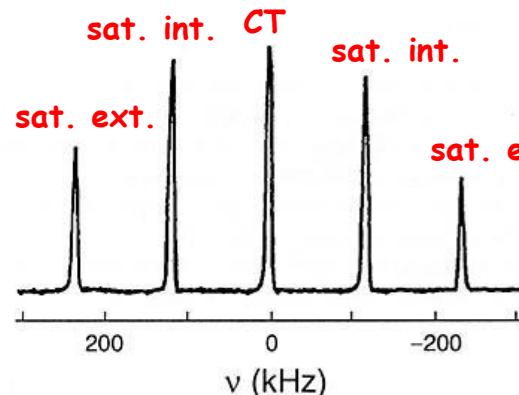
ST: satellite transitions

$$\text{shape } \eta_Q \\ C_Q = e^2 q Q / h$$



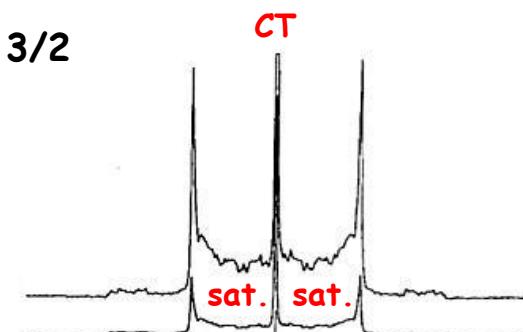
Freude et al., NMR Basic
Princ. Prog., 1993, 29, 25.

$I = 5/2$



^{27}Al in $\alpha\text{-Al}_2\text{O}_3$
single crystal

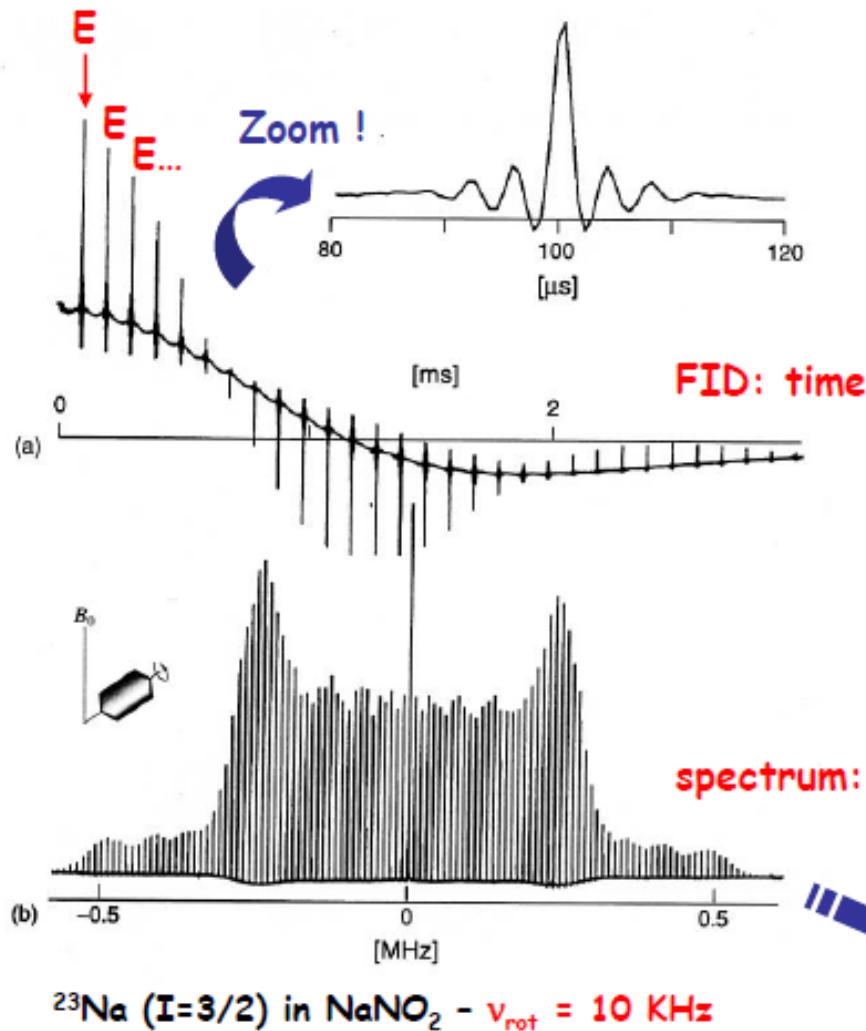
$I = 3/2$



^{23}Na in NaNO_3
powder

Man, Encyclopedia of analytical chemistry, 2000, 12229.

MAS and quadrupolar nuclei (1st order)



to set the magic angle...

K^{81}Br ($I=3/2$)

K^{127}I ($I=5/2$)

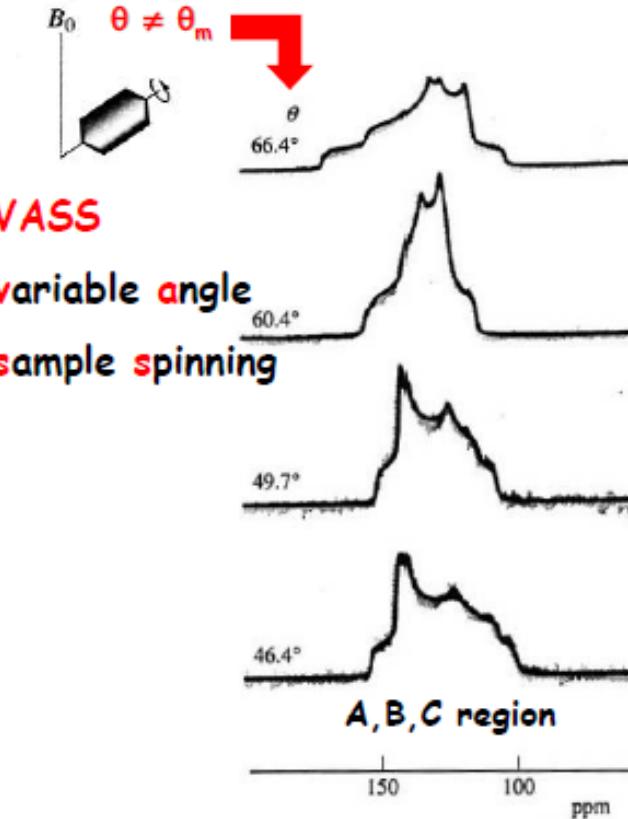
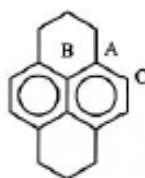
and maximize the number
of echoes in the time
domain (1 scan !)...

(but is it crucial ?...)

a huge number of SSB :
why ?!

Off MAS experiments

^{13}C : 1,2,3,6,7,8-hexahydronaphthalene

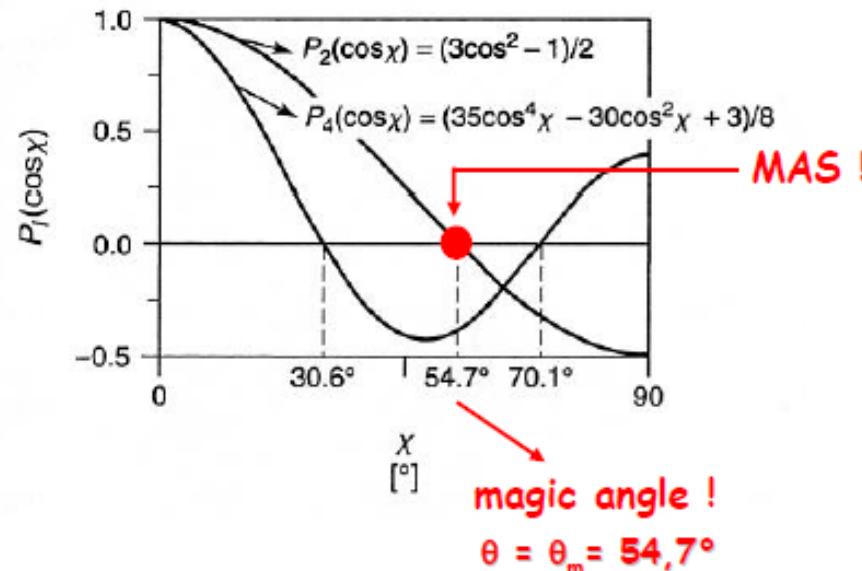


Sethi et al., Prep. Am. Chem. Soc., 1987, 32, 155.

fast rotation at θ : scaling factor of Δ_A

$$\Delta_A P_2(\cos\theta) = \Delta_A \frac{1}{2} (3\cos^2\theta - 1)$$

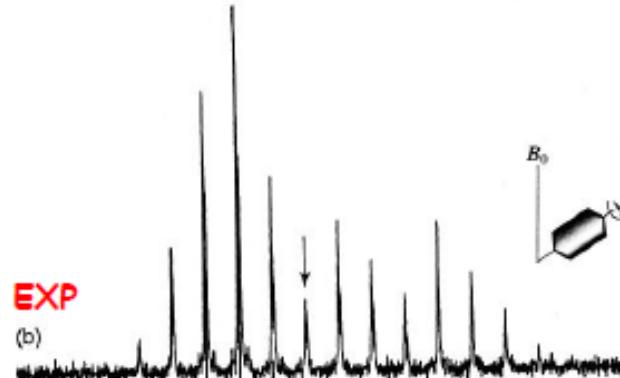
Legendre polynomial !



Information to extract from the set of spinning sidebands

ex: CS interaction $I = 1/2$

^{119}Sn MAS in: tin sulfate

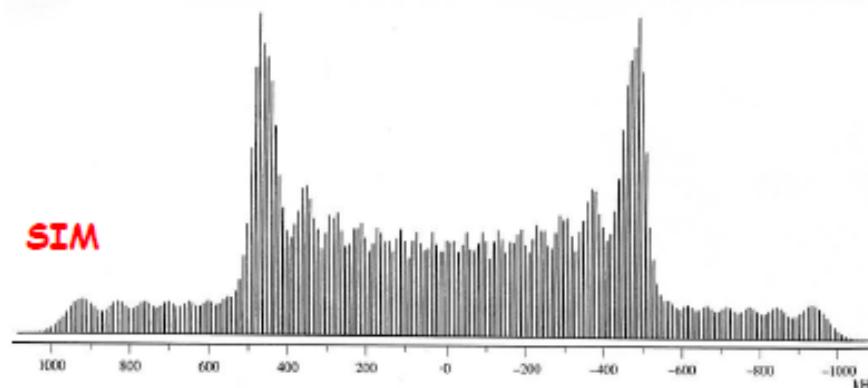
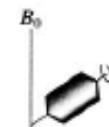


SIM

(a)

extracted data:
 δ_{iso} , η_{CSA} , Δ_{CSA}

ex: Q interaction $I = 3/2$



extracted data:

δ_{iso} , η_Q , C_Q

Jakobsen, Encyclopedia of NMR, 1996, 2374.

DM2011

SIMulation programs:
DM2003 by Massiot
SIMPSON by Bak et al. } WEB !

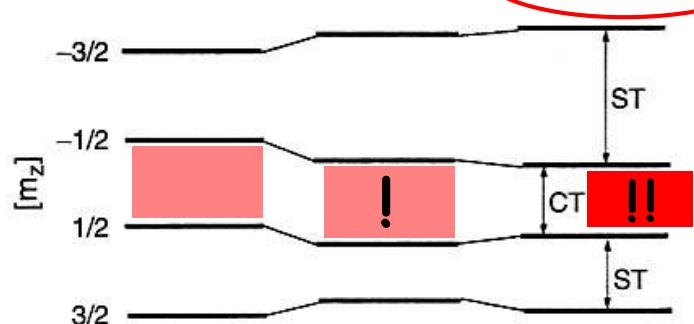
Harris et al., Encyclopedia of NMR, supp. Vol., 145.

Strong quadrupolar nuclei

C_Q : 3 to 15 MHz...

$I=3/2$

Zeeman interaction First-order effect Second-order effect



$H_Q \sim H_{\text{Zeeman}}$: second order perturbations



All transitions (ST and CT) are perturbed

Mathematical treatment...?

$$w_{-1/2,1/2}^{(2)\text{static}} = -\frac{1}{6w_L} \left[\frac{3e^2qQ}{2I(2I-1)\hbar} \right]^2 \left\{ I(I+1) - \frac{3}{4} \right\} \times \{A(\alpha, \eta) \cos^4 \beta + B(\alpha, \eta) \cos^2 \beta + C(\alpha, \eta)\}$$

$$A(\alpha, \eta) = -\frac{27}{8} + \frac{9}{4}\eta \cos 2\alpha - \frac{3}{8}(\eta \cos 2\alpha)^2$$

$$B(\alpha, \eta) = \frac{30}{8} - \frac{1}{2}\eta^2 - 2\eta \cos 2\alpha + \frac{3}{4}(\eta \cos 2\alpha)^2$$

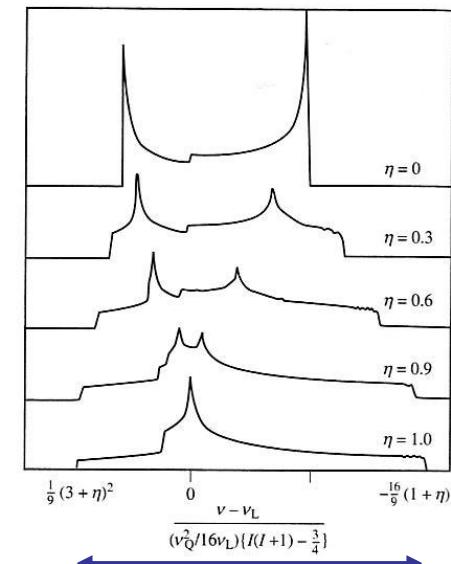
$$C(\alpha, \eta) = -\frac{3}{8} + \frac{1}{3}\eta^2 - \frac{1}{4}\eta \cos 2\alpha - \frac{3}{8}(\eta \cos 2\alpha)^2$$

Man, Encyclopedia of analytical chemistry, 2000, 12229.

shape : η_Q

$$\Delta \sim C_Q^2 / v_L$$

idea : B_0



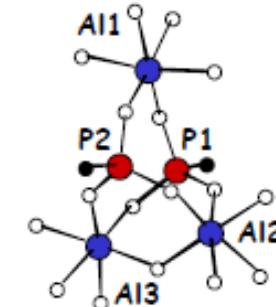
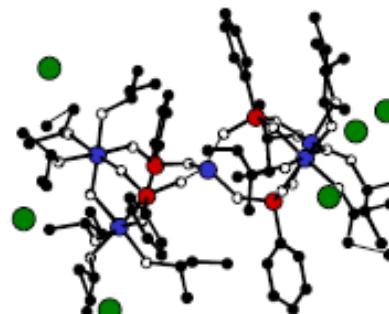
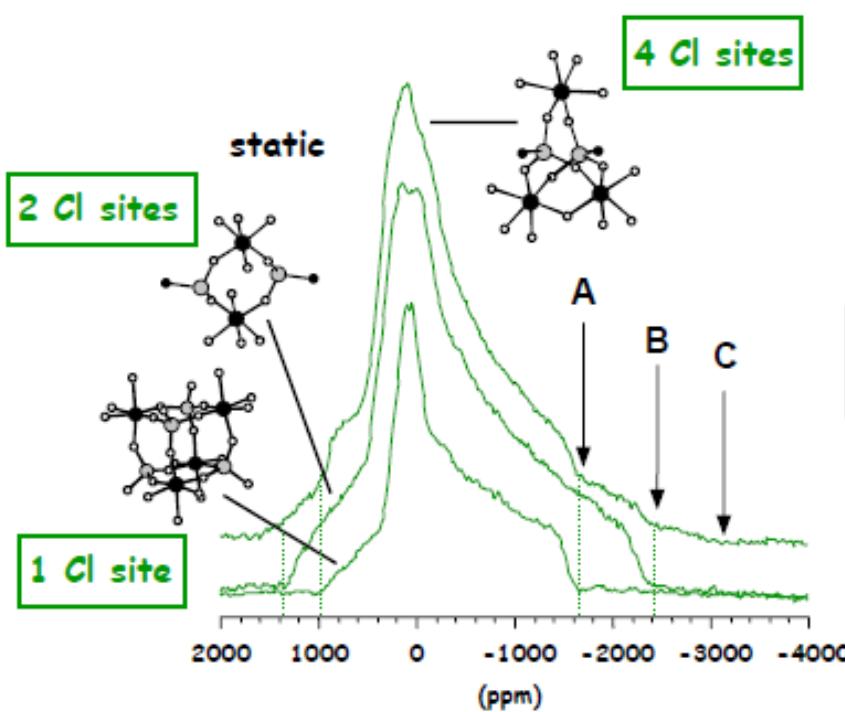
Freude et al., NMR Basic

Princ. Prog., 1993, 29, 27.

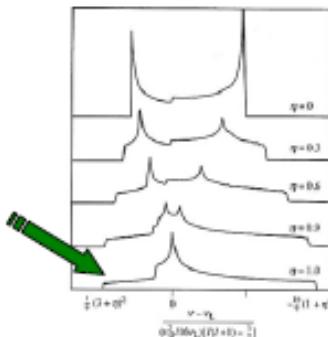
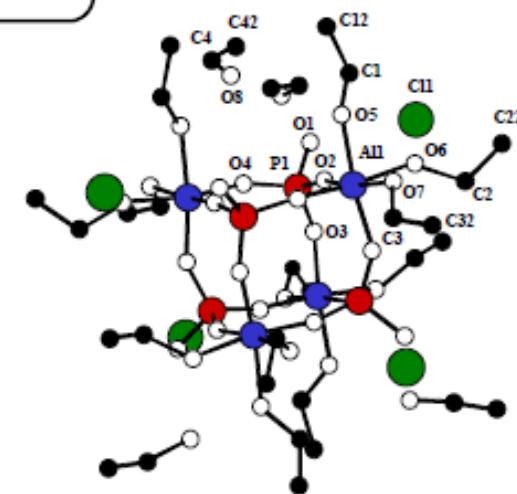
Δ

Examples: ^{35}Cl solid state NMR

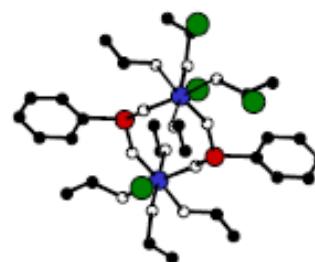
^{35}Cl : $I=3/2$



Al-O-P clusters
Cl⁻ as counter ions



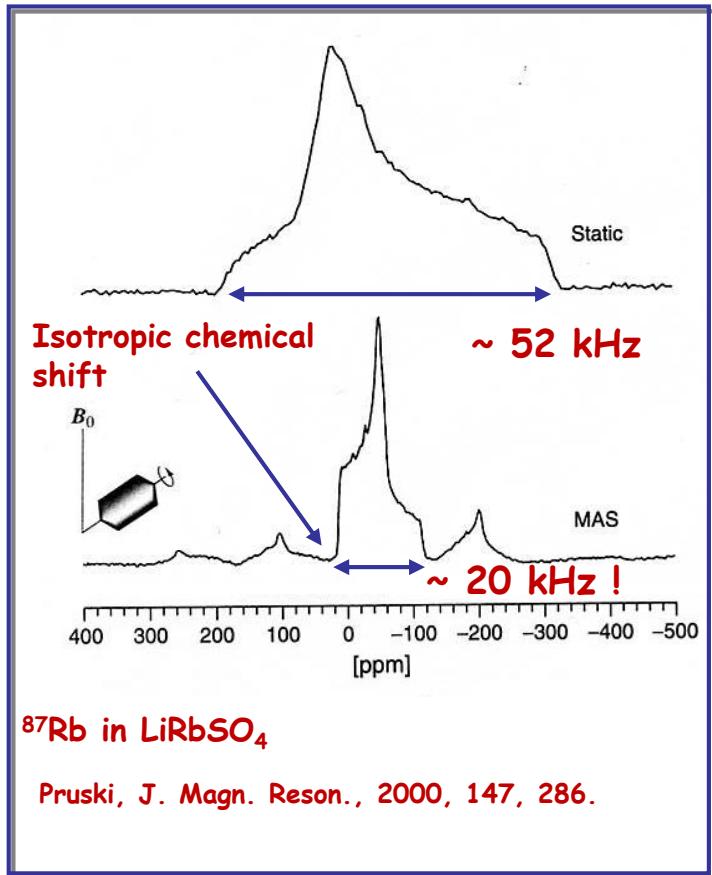
$$\begin{aligned} \eta I_Q &\sim 1 \\ C_Q &\sim 6-8 \text{ MHz} \end{aligned}$$



Azaïs et al., Solid State NMR,
2003, 23, 14.

Quadrupolar nuclei (2nd order) and MAS rotation

theorem: MAS has an effect... But the second order broadening effect is only partially averaged !



even at « infinite » MAS frequency !

WHY ? (without any calculation)

MAS rotation: efficient for

◆ ellipsoids

◆ $\text{Cos}^2(\alpha_0, \beta_0)$

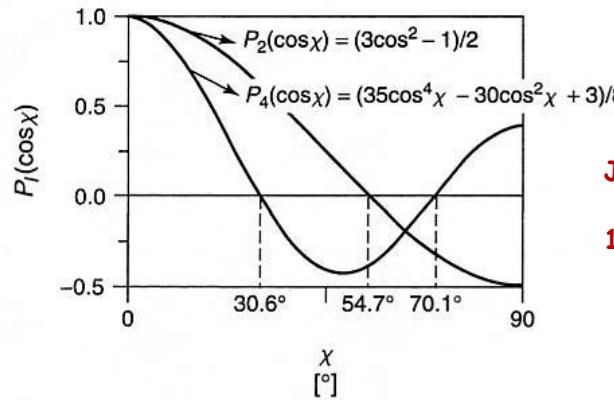
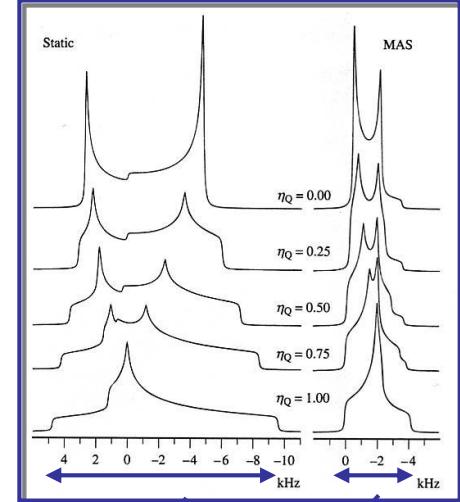
◆ $P_2(\text{Cos}\theta)$

... But not for:

◆ quartics

◆ $\text{Cos}^4(\alpha_0, \beta_0), \text{Cos}^2(\alpha_0, \beta_0)$

◆ $P_4(\text{Cos}\theta), P_2(\text{Cos}\theta)$



Jakobsen, Encyclopedia of NMR,
1996, 2371.

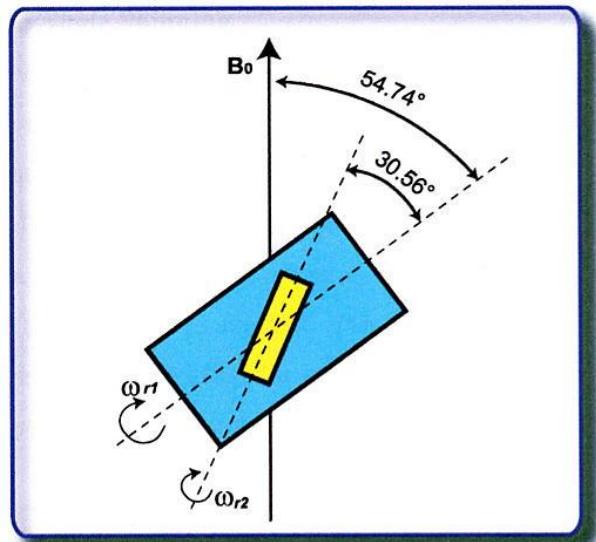
? $P_4(\text{Cos}\theta) = P_2(\text{Cos}\theta) = 0$? ...NO !!

Quadrupolar nuclei and macroscopic reorientations

MAS: "one unique degree of freedom" (1959)

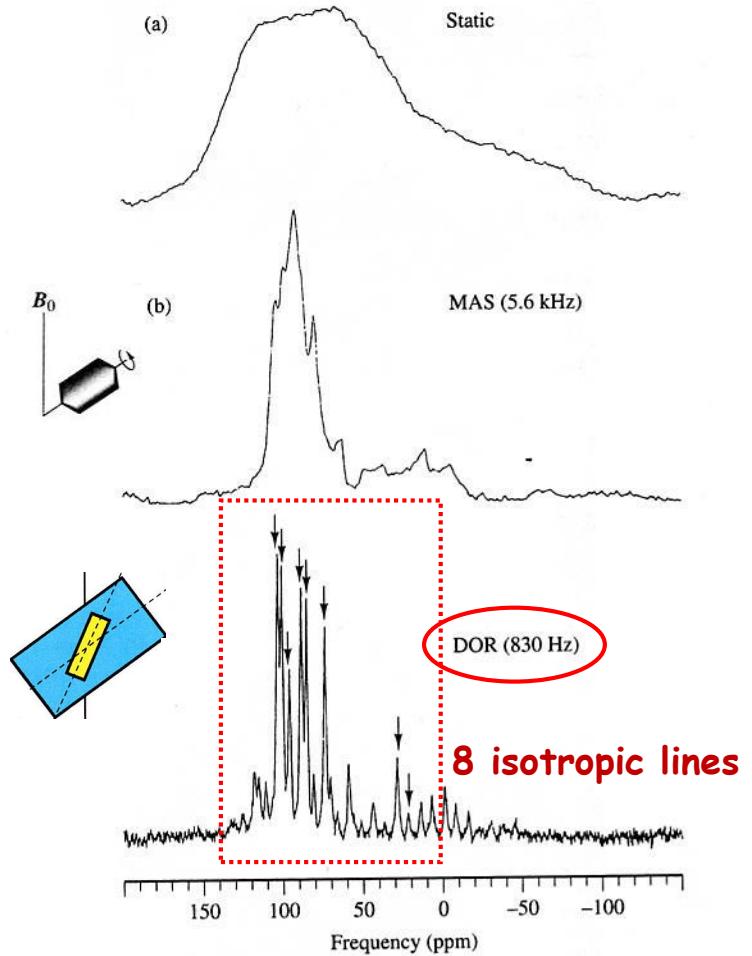
Let us invent an experiment with 2 angles of reorientation !

DOR experiment (DOuble Rotation)
(Samoson, Pines, 1988)



1D experiment

^{17}O ($I=5/2$) : CaSiO_3 wollastonite: 9 sites (^{17}O)



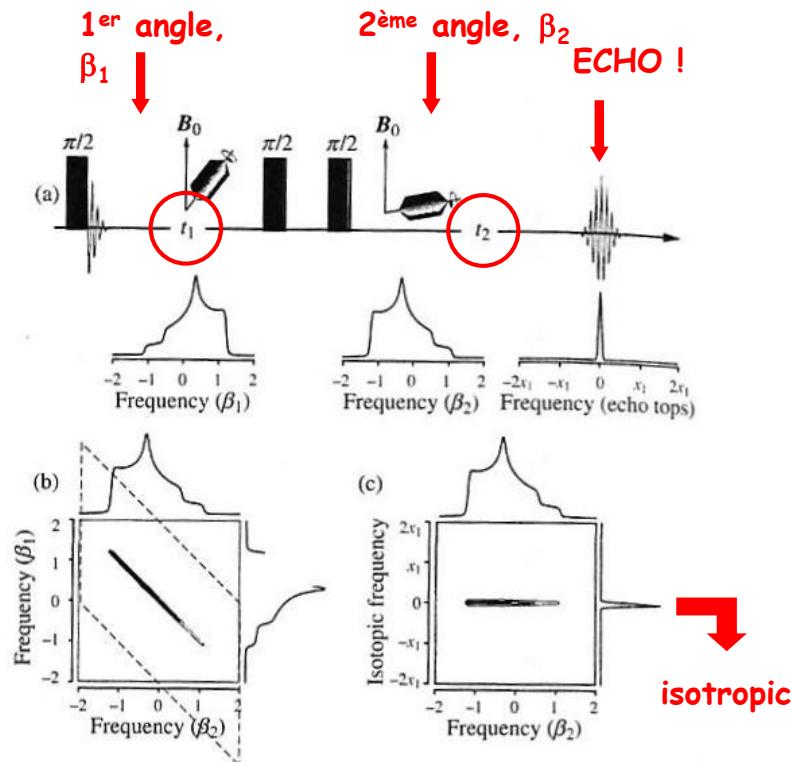
Wu, Encyclopedia of NMR, 1996, 1749.

DAS approach: Dynamic Angle Spinning

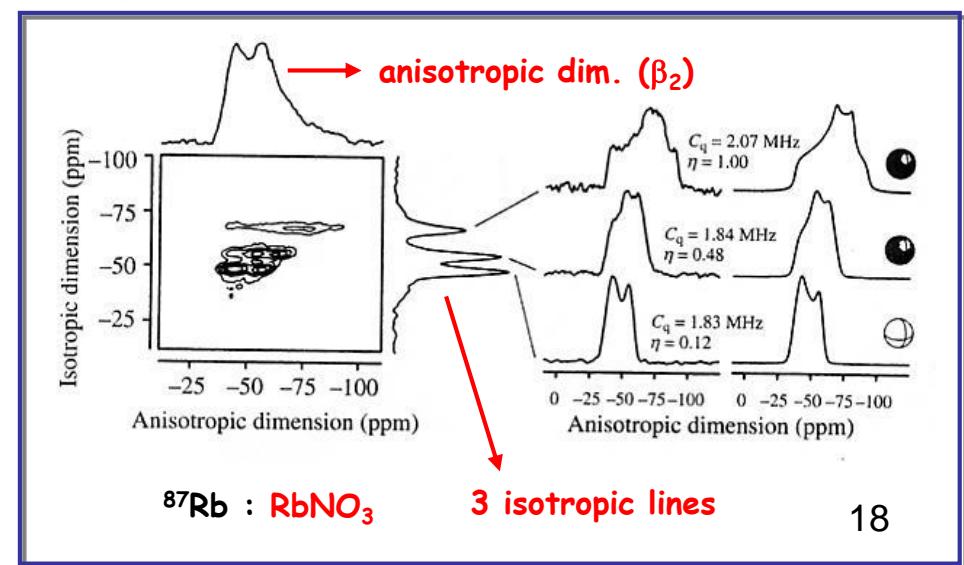
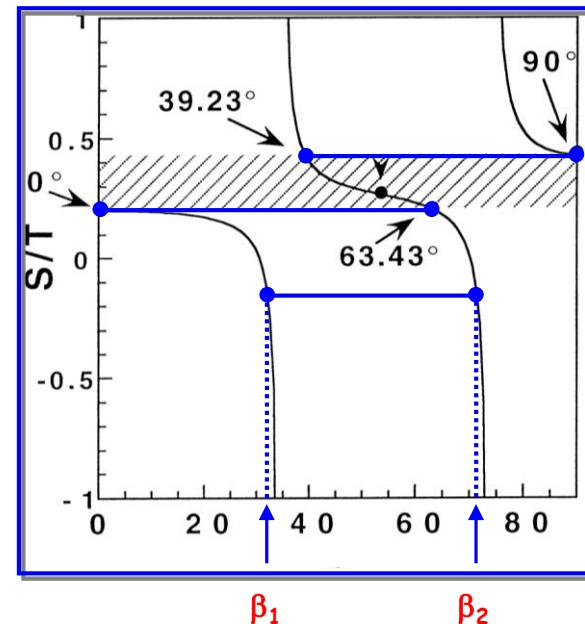
Another way to involve 2 angles

DAS experiment (Dynamic Angle Spinning)

(Llor, Virlet, 1988)



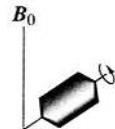
DAS angles pairs



Rotation around a unique angle: MQ-MAS !

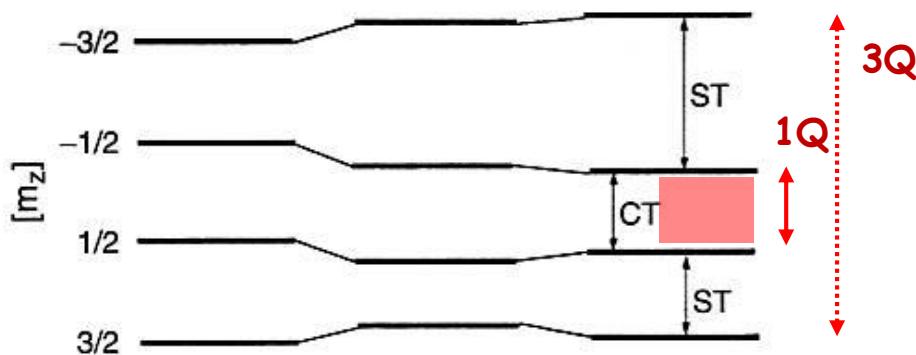
DAS and DOR: 1 transition (CT) et 2 angles...

MQ-MAS (Multiple Quantum MAS)
(Frydman, 1995)

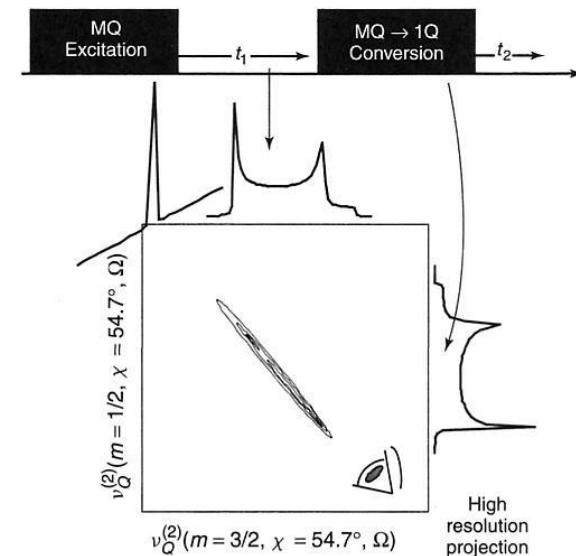


2 transitions (CT/MQ) and 1 angle (MAS) !

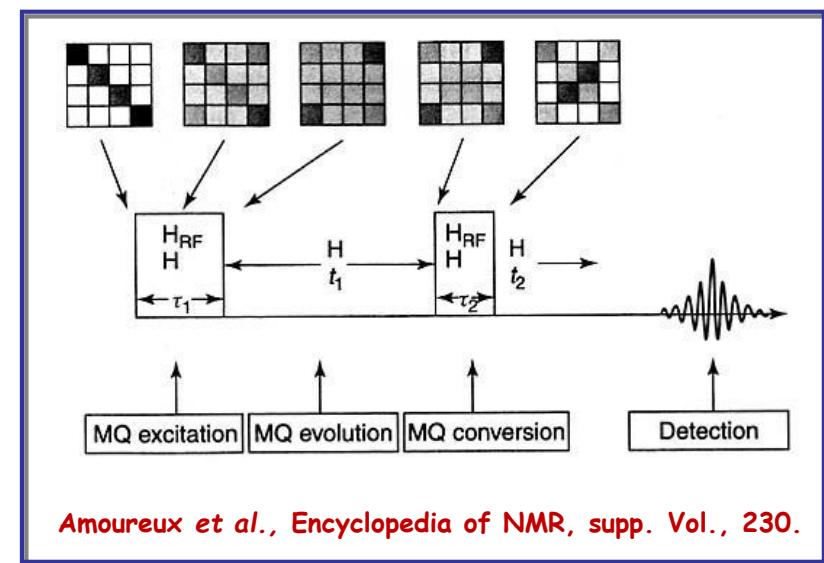
Zeeman interaction First-order effect Second-order effect



idea: 1Q and 3Q correlation to give ... an ECHO !



Frydman, Encyclopedia of NMR, supp. Vol., 265.

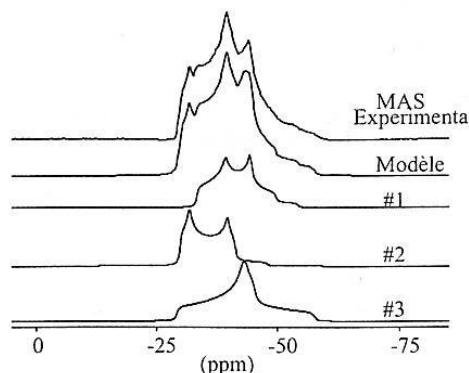
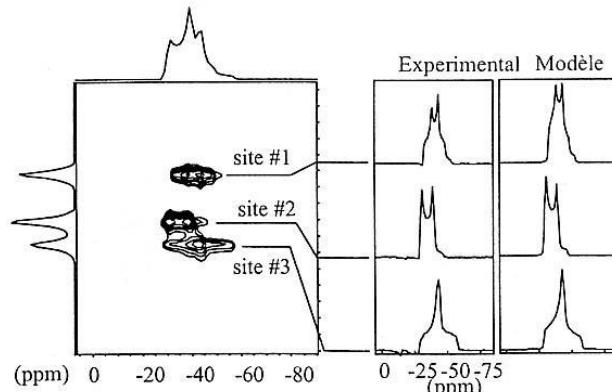


Amoureaux et al., Encyclopedia of NMR, supp. Vol., 230.

MQ-MAS: examples

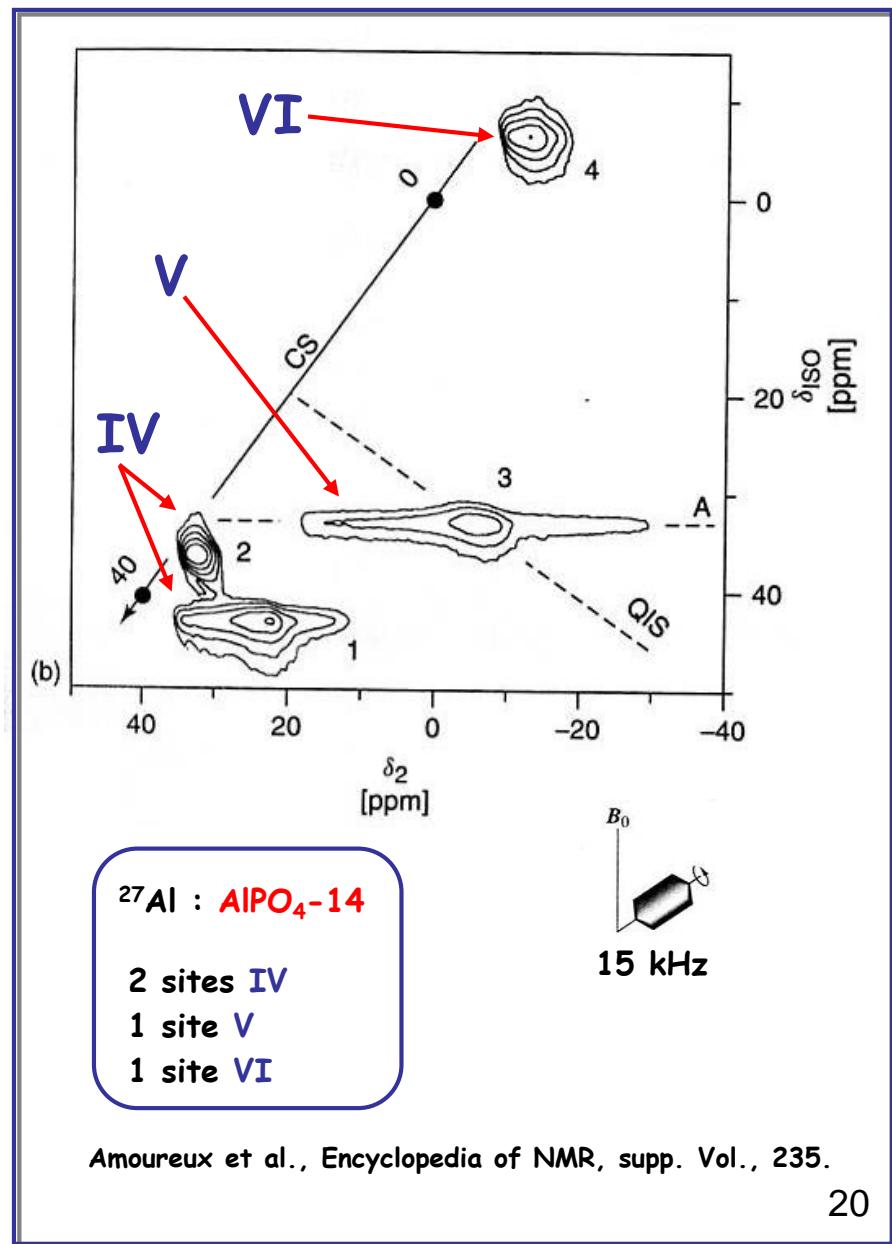
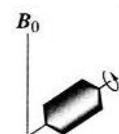
DAS and DOR: demanding techniques

MQ-MAS: much easier (...)



$^{87}\text{Rb} : \text{RbNO}_3$

Massiot, Ecole RMN des Houches, 1997.



Amoureaux et al., Encyclopedia of NMR, supp. Vol., 235.

^1H solid state NMR: another challenge

^1H : strongly coupled by the homonuclear dipolar interaction !

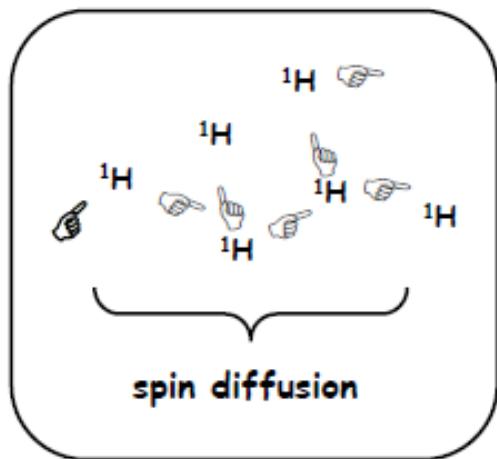
remember:

$$D_{II} \sim \gamma^2 / r_{II}^3$$

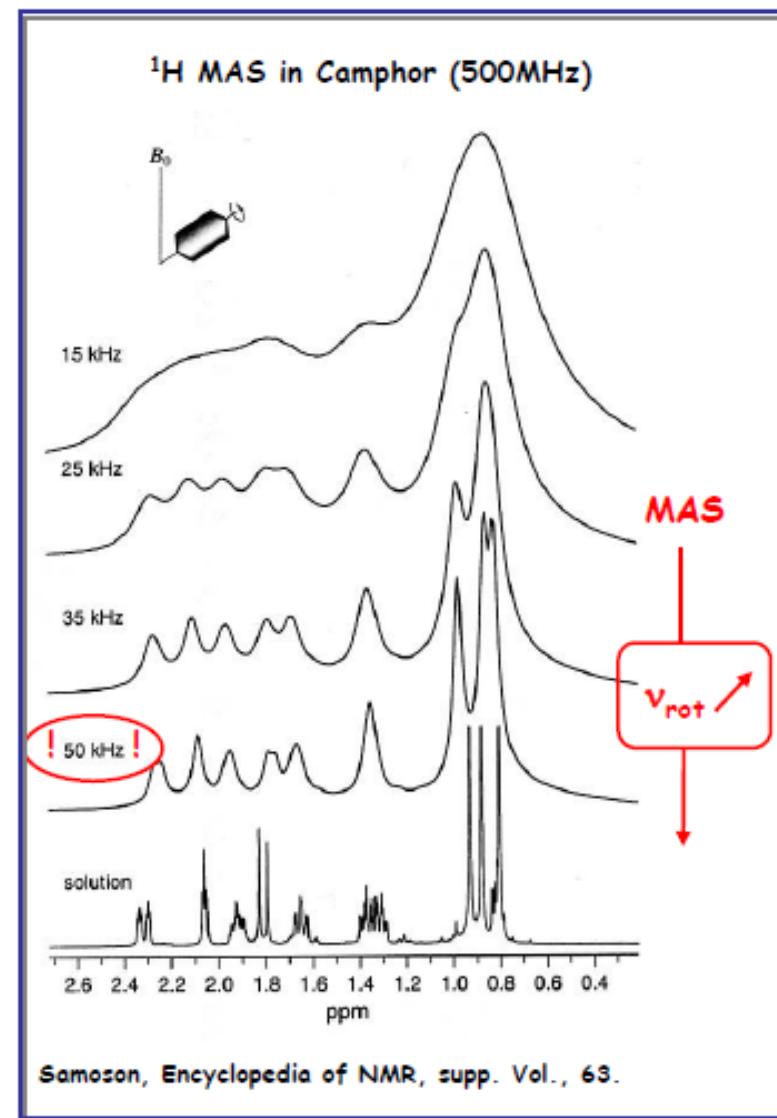
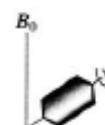
high for ^1H

up to 30 kHz...

rather small

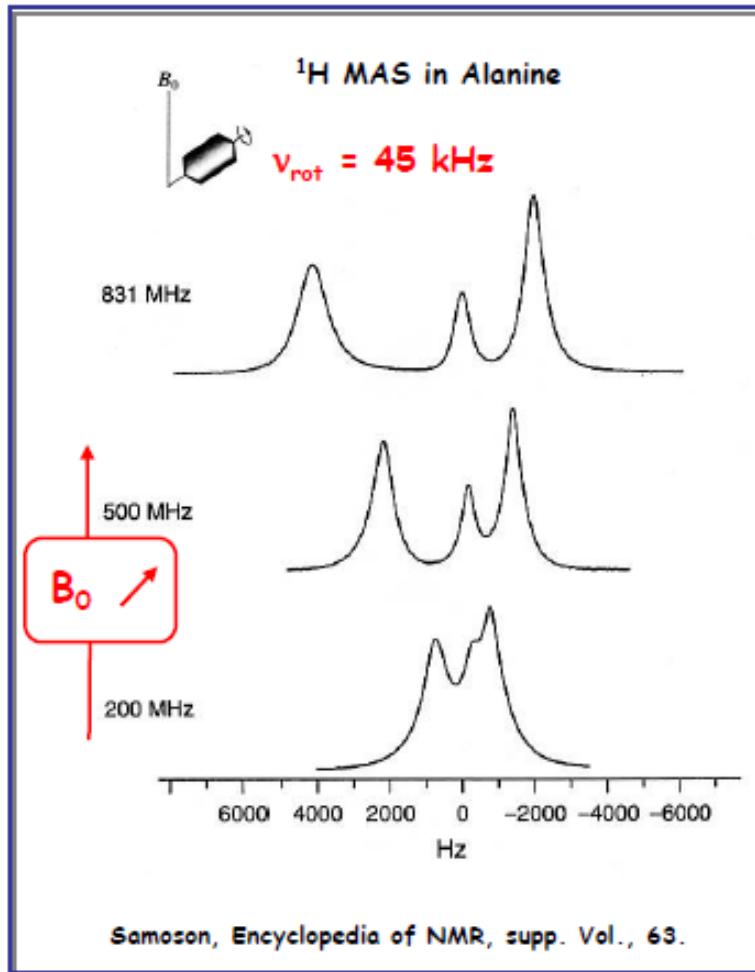


question: is the MAS reorientation efficient ?

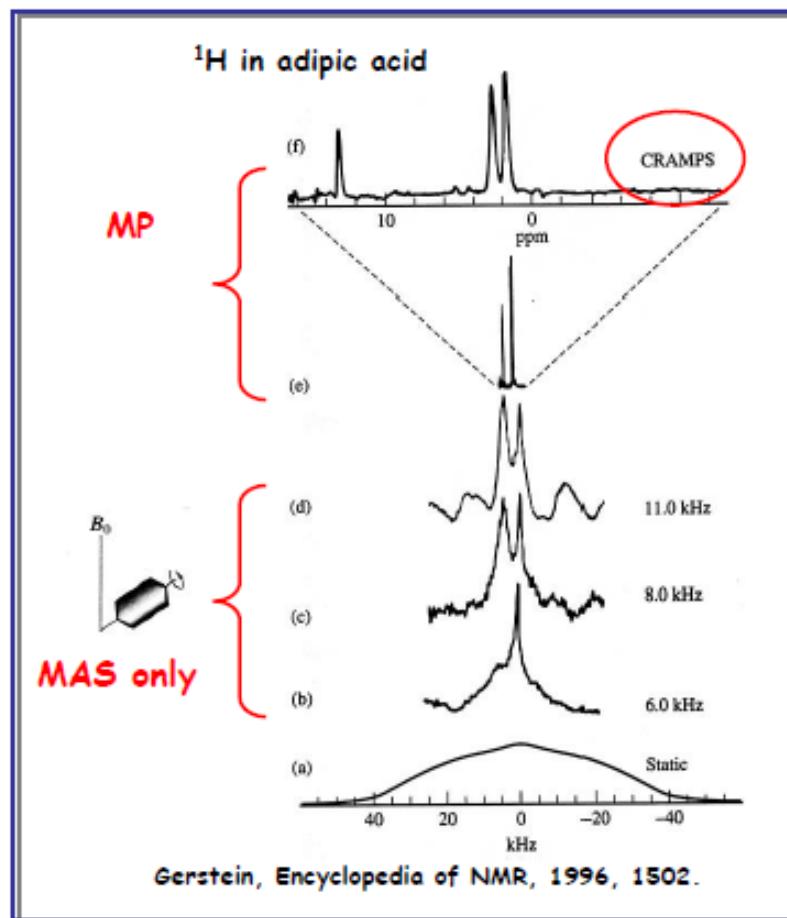


Two different approaches

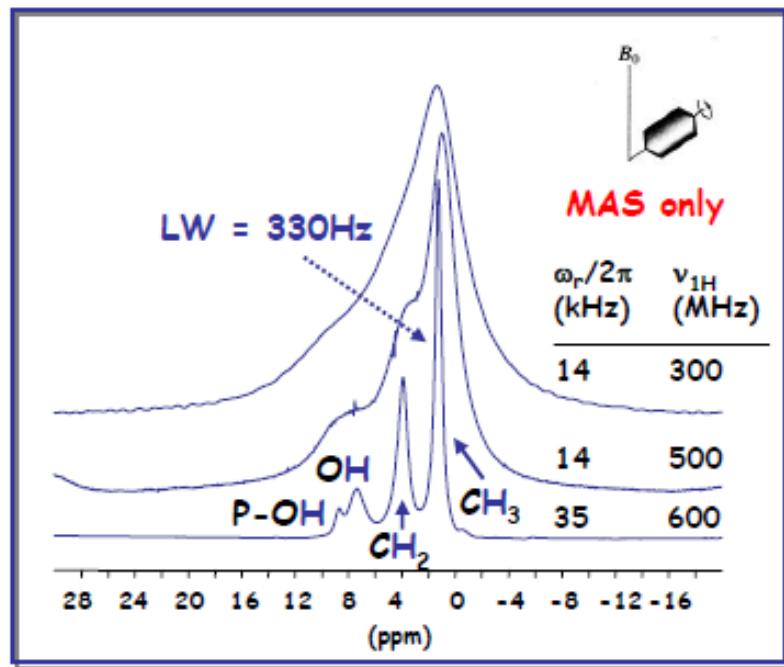
first idea: highest B_0 and highest ν_{rot} !



second idea: rotations in spin space !



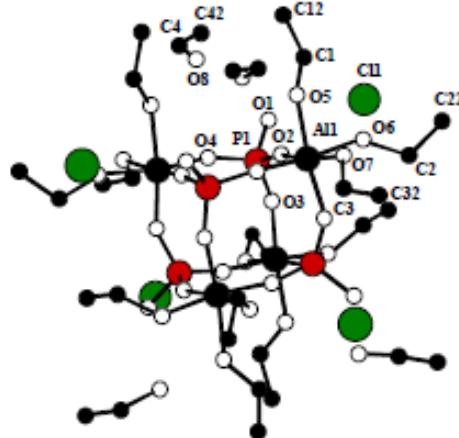
Comparison of techniques



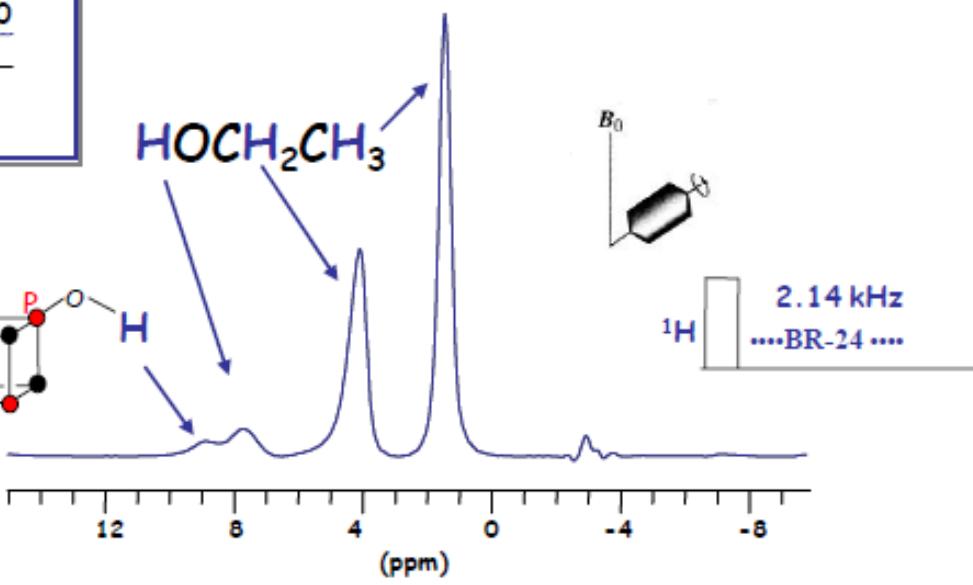
... what is the limit of resolution
in 1H solid state NMR !?

don't forget the role of molecular motion !....

CRAMPS: Combined Rotation And
Multiple Pulses Spectroscopy



CRAMPS

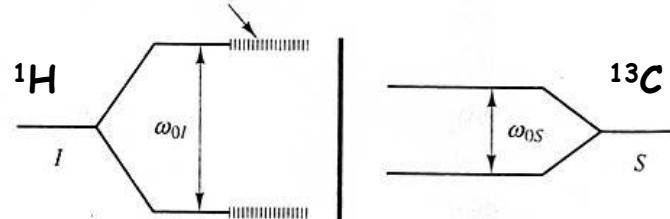


Azaïs, PhD thesis.

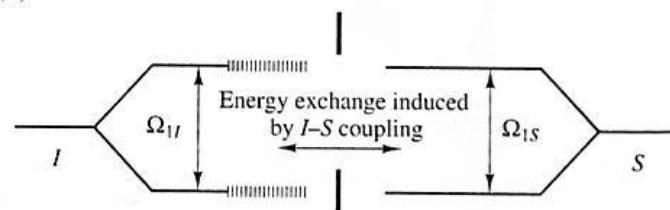
Cross Polarization (CP): a simplified description

question: is it possible to transfer magnetization from ^1H to ^{13}C ?

- (a) Broadening of energy levels due to $I-I$ dipolar couplings



(b)



Engelke, Encyclopedia of NMR, 1996, 1530.

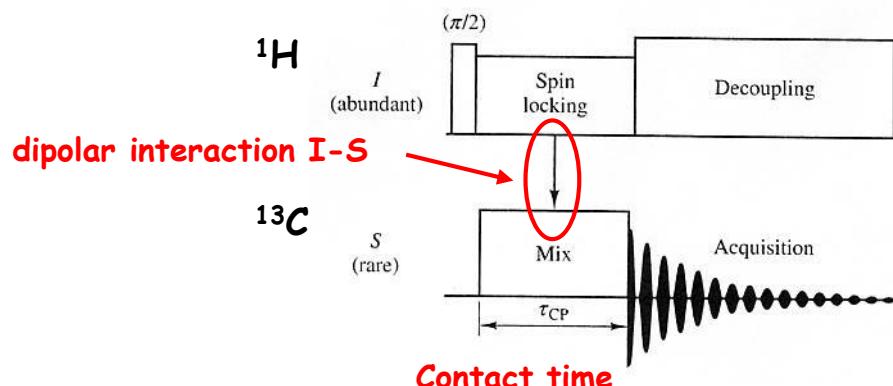
Hartmann and Hahn (1962):

NO in the LAB frame mais YES in the rotating frame

$$\Omega_{1I} = \gamma_I B_{1I} = \Omega_{1S} = \gamma_S B_{1S}$$

Hartmann-Hahn condition on $B_1(\text{RF})$ fields

the most popular sequence

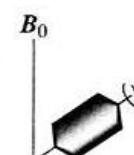


advantages:

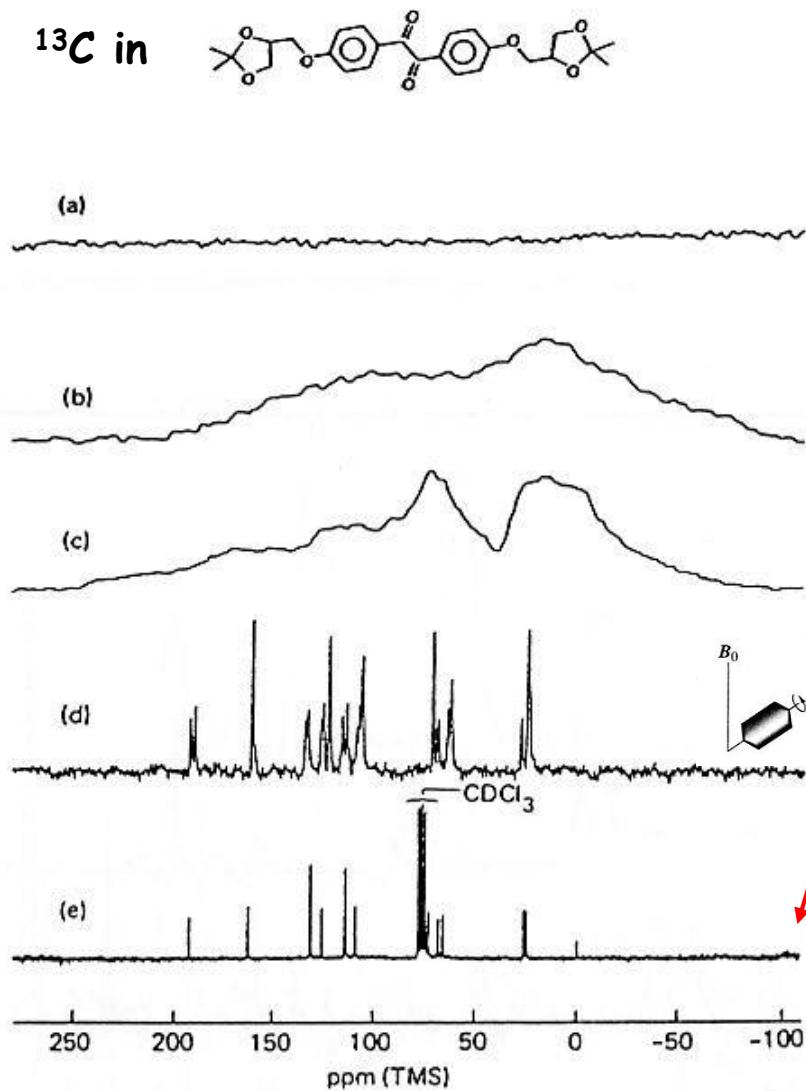
- ◆ gain: $M_s (\gamma_{1\text{H}} / |\gamma_{13\text{C}}|) \rightarrow 4 \text{ for } ^{13}\text{C}$
10 for ^{15}N !
- ◆ $\tau_{CP} \sim ms$!
- ◆ $T_1(^1\text{H}) \ll T_1(^{13}\text{C})$
- ◆ ^{13}C FID with ^1H decoupling

How to manage the ^{13}C CSA interaction ?

idea...



The CP MAS experiment



(a) solid (solution state conditions)

(b) CP (low power decoupling)

(c) CP (high power decoupling)

(d) CP MAS (high power decoupling)

(e) solution (low power decoupling)

$$\Omega_{1I} = \Omega_{1S} \pm n \Omega_{\text{rot}}$$

with $n = 1, 2$

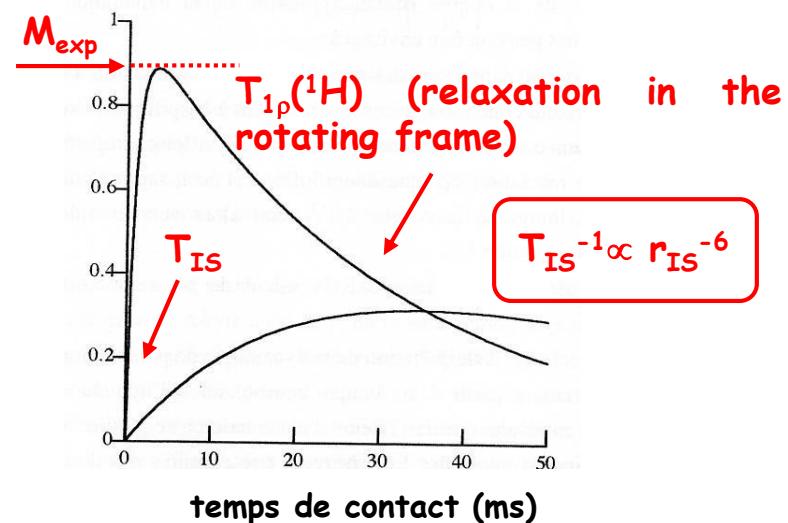
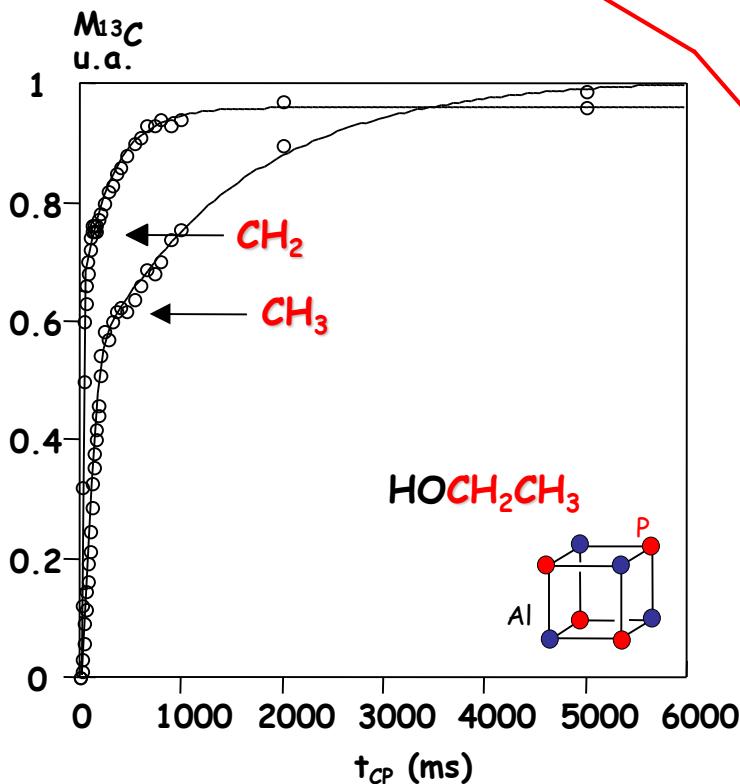
modified Hartmann-Hahn conditions

CP dynamics: quantitative approach

exponential build-up of the magnetization

$$M_S^{CP}(t_c) = \frac{\gamma_1}{\gamma_s} M_s \frac{1}{1-\lambda} \left[\exp\left(-\frac{t_c}{T_{1p}^I}\right) - \exp\left(-\frac{t_c}{T_{IS}}\right) \right]$$

gain ! loss ! gain !



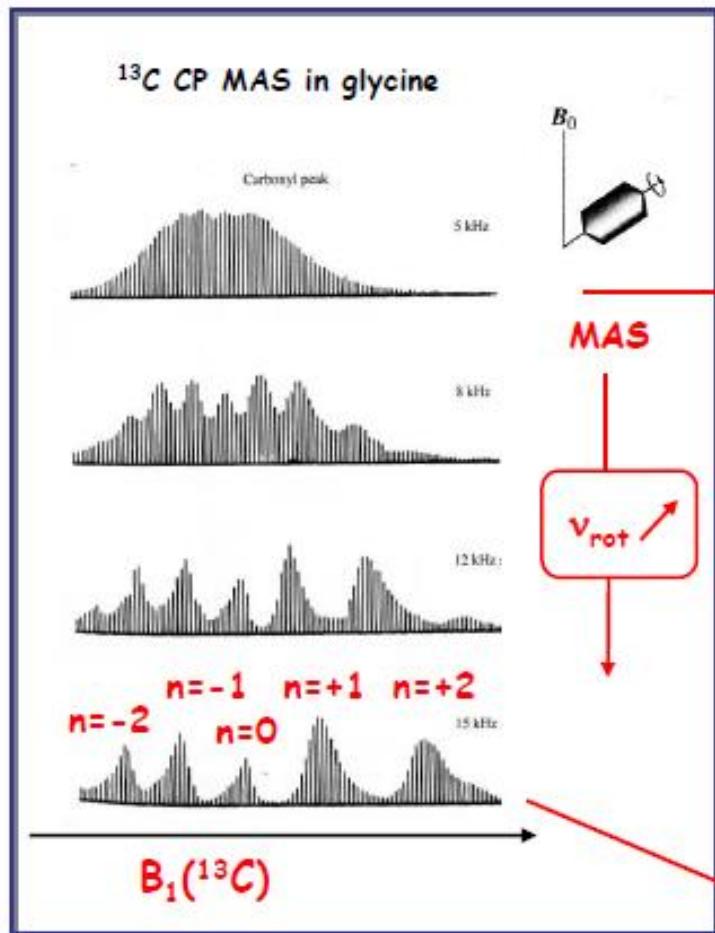
$T_{CH}(^{13}\text{CH}_2) < T_{CH}(^{13}\text{CH}) \ll T_{CH}(^{13}\text{C}_{\text{quat}})$

$^{13}\text{C}_{\text{quat}}$ are underestimated at short contact time

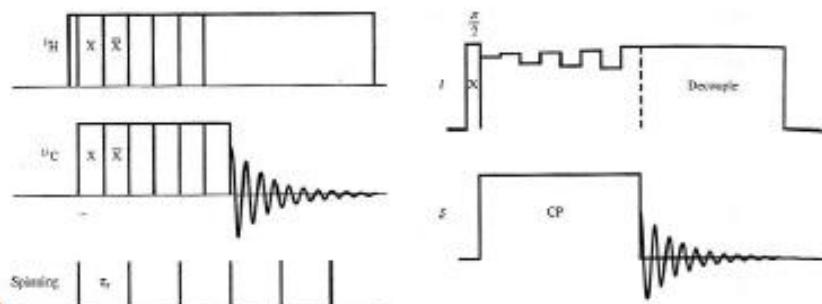
$\lambda = T_{CH}/T_{1p}(^1\text{H})$ influence M_{exp}

Care must be taken with materials involving several $T_{1p}(^1\text{H})$!

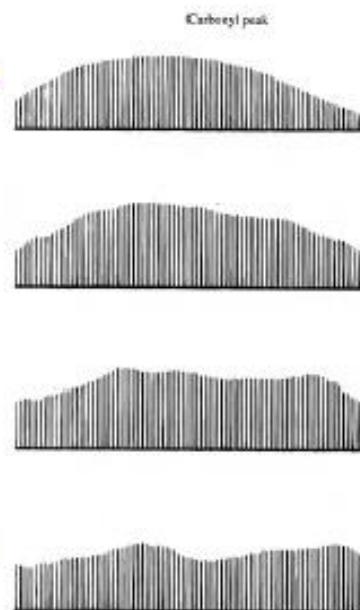
CP and MAS



solutions...



variable amplitude
during the contact



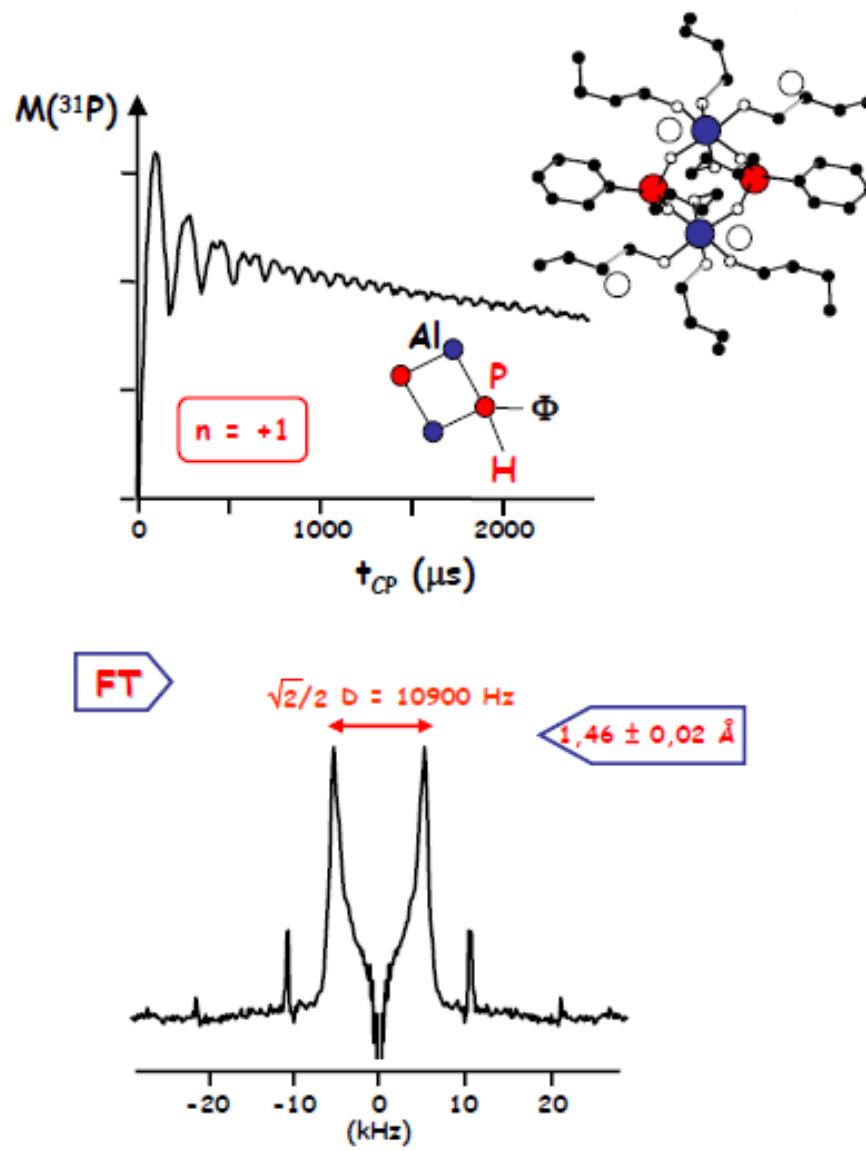
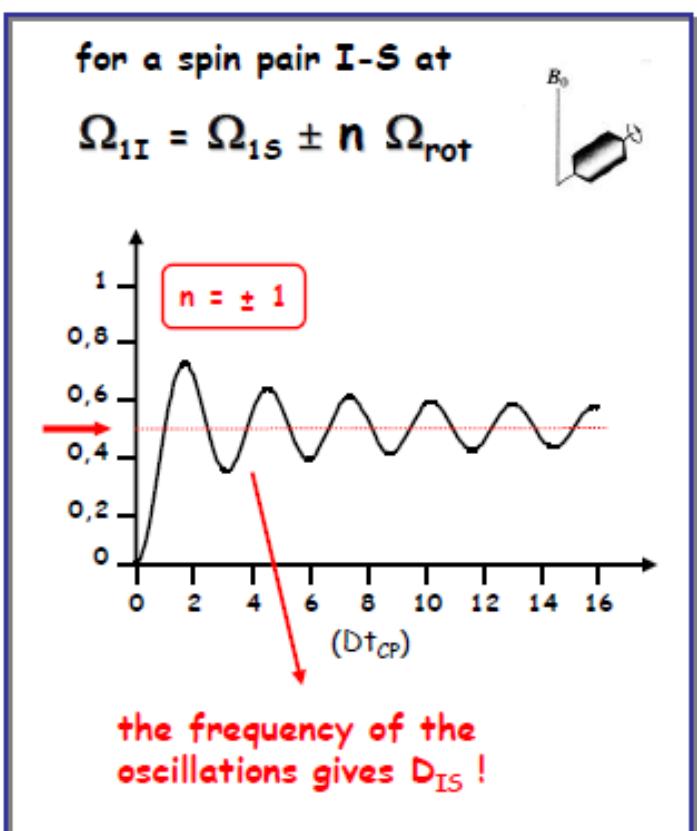
Burum, Encyclopedia of NMR, 1996, 1539.

Distance measurement by CP MAS experiments

why distances ?...

remember: $D_{IS} \sim \gamma_I \gamma_S / r_{IS}^3$

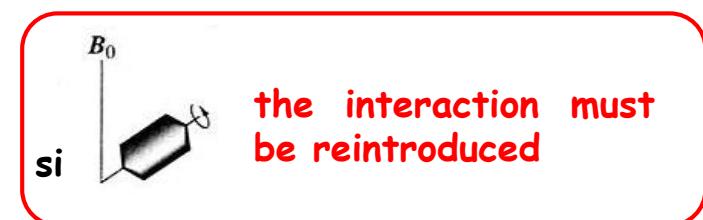
CP transfer !



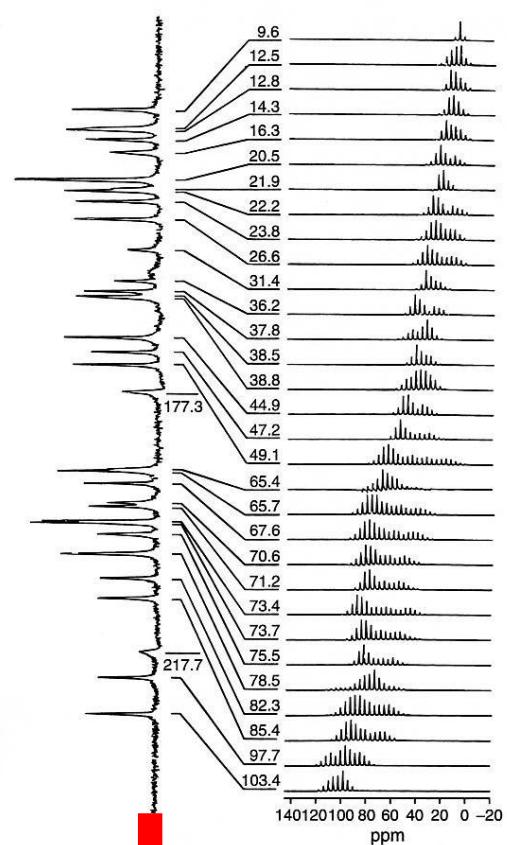
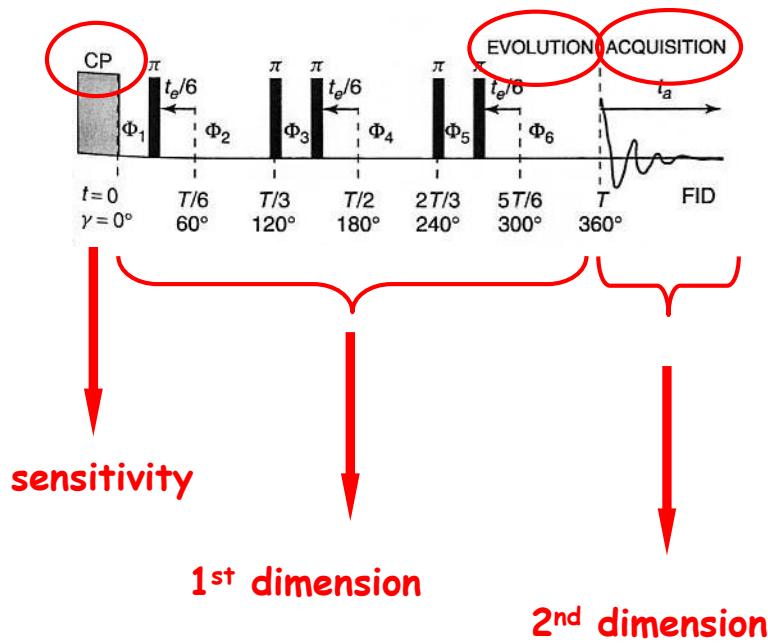
Azaïs et al., Inorg. Chem., 2002, 41, 981.

2D experiments: isotropic / anisotropic data

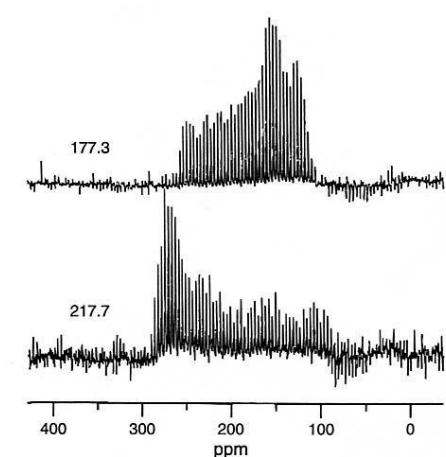
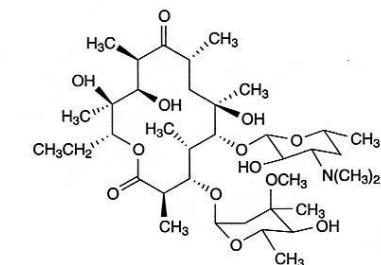
general idea: 2D corrélations between isotropic δ and anisotropies



ex: δ_{iso} vs Δ_{CSA} - Magic Angle Turning



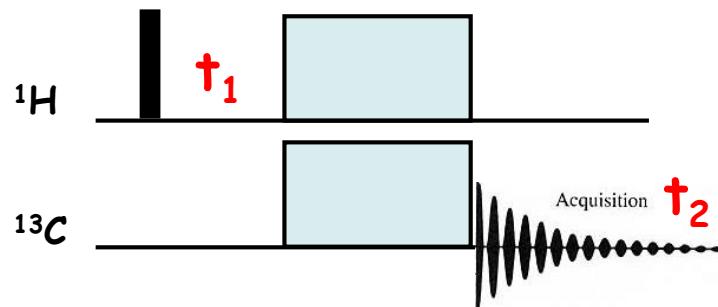
^{13}C : erythromicine A



Alderman et al., Mol. Phys., 1998, 95, 113.

Heteronuclear 2D HETCOR correlation

general idea: evolution of ^1H evolves during t_1 and is transferred to ^{13}C by CP.

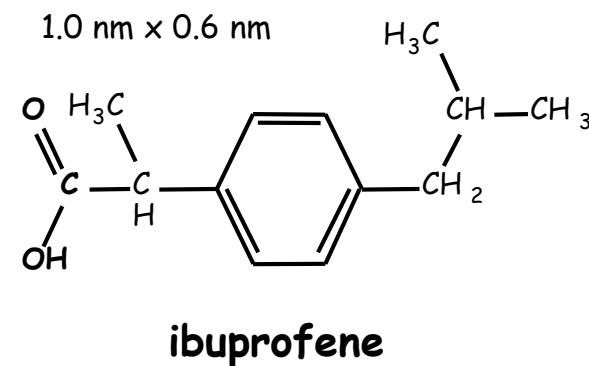
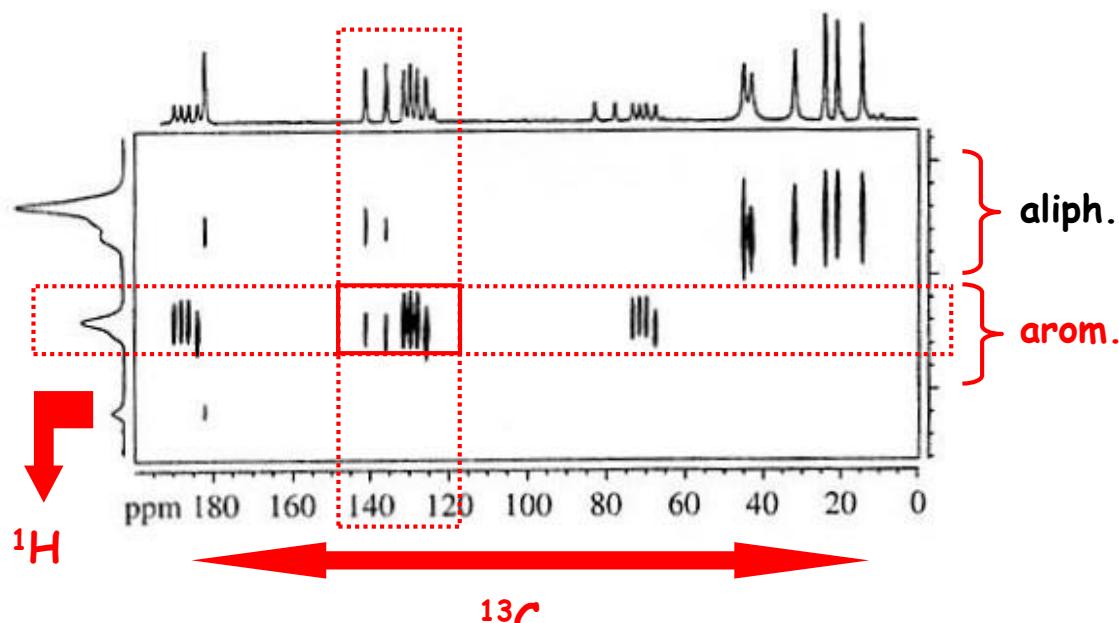


dipolar correlation peaks



short distances

$^{13}\text{C}-^1\text{H}$ HETCOR for ibuprofen



Burum, Encyclopedia of NMR, 1996, 1542.



Solid State Nuclear Magnetic Resonance: Applications to Materials



Christian BONHOMME

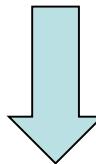
christian.bonhomme@upmc.fr

Laboratoire de Chimie de la Matière Condensée de Paris

UMR CNRS 7574 - Sorbonne Université, Paris, France

Solid state NMR basics

Applications

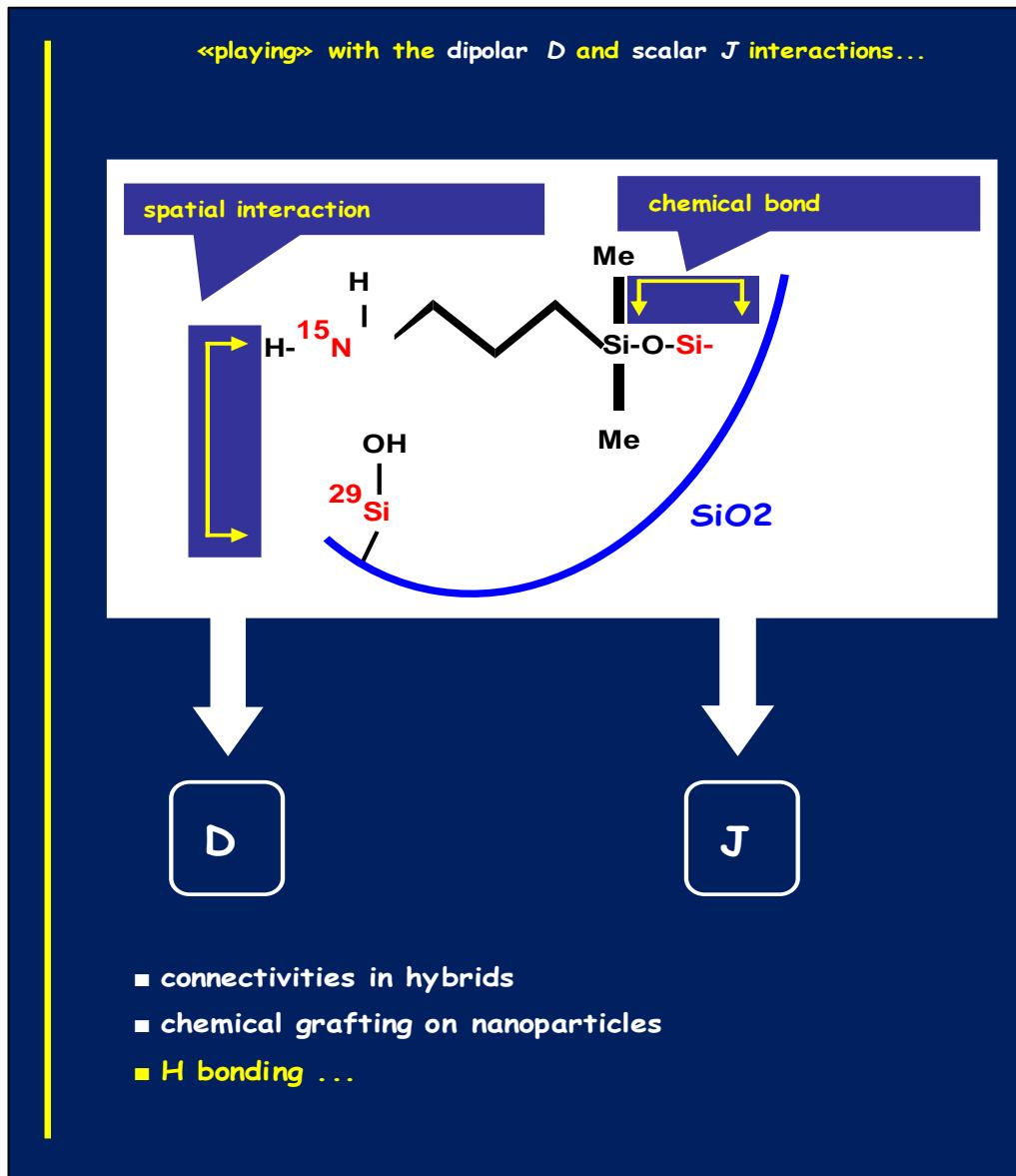


- OI hybrids
- interfaces
- calcium phosphates

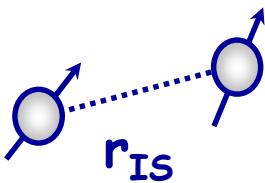
Sensitivity

◆ Organic Inorganic Hybrid Materials

- Part I



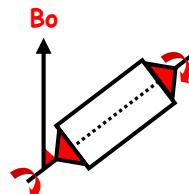
◆ DIPOLAR INTERACTION D



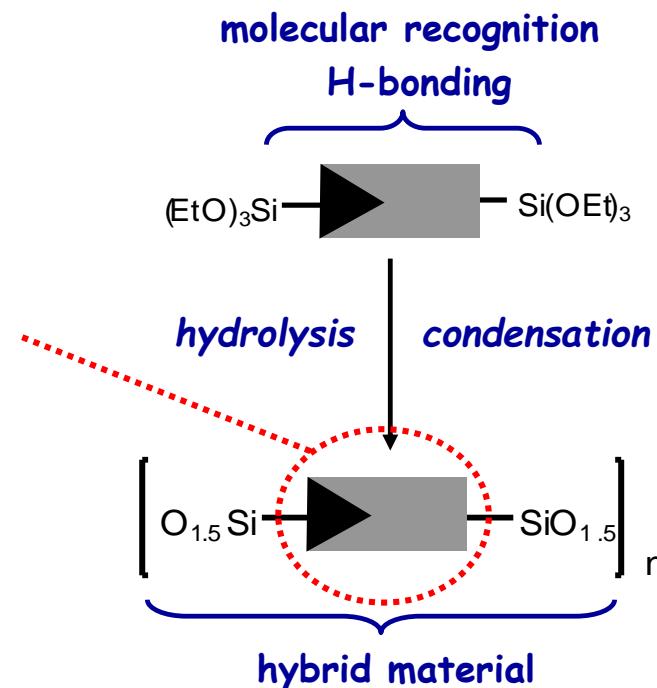
$$D \propto \frac{1}{r_{IS}^3}$$



CEMHTI,
Orléans

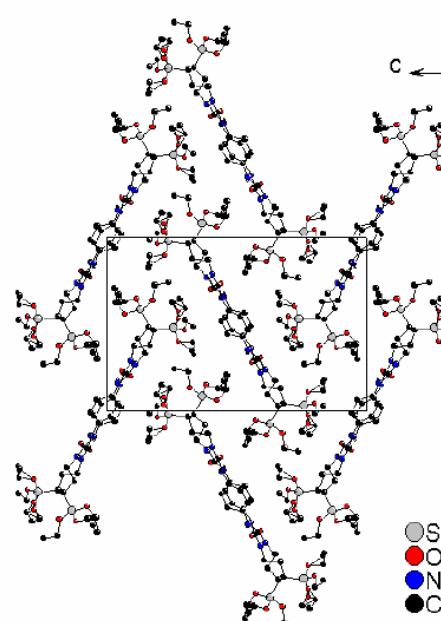
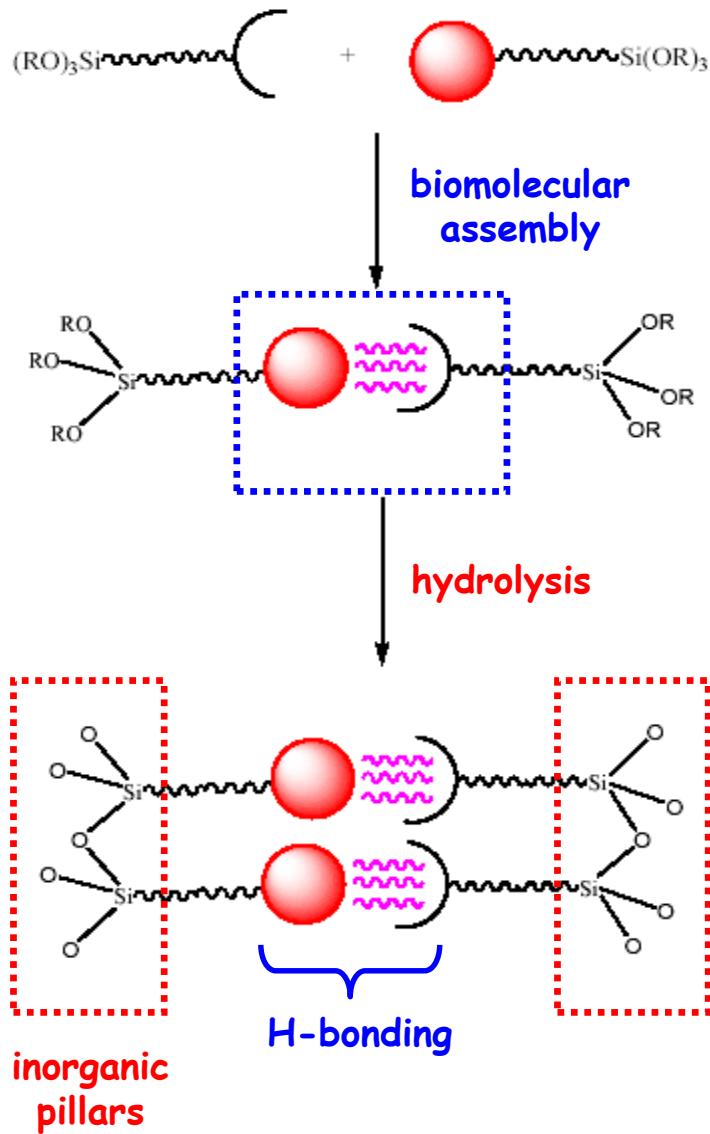


fast-, ultra fast MAS

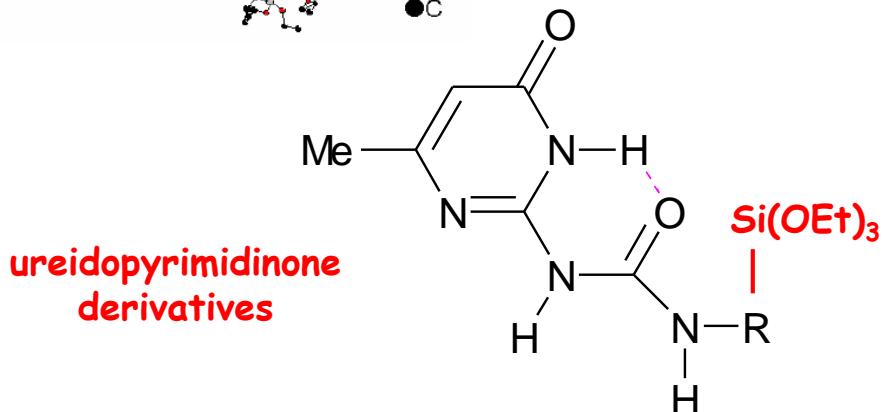


- $^{1}\text{H}-^{1}\text{H}$ dipolar interaction
- ureidopyrimidinone models
- bio-inspired materials

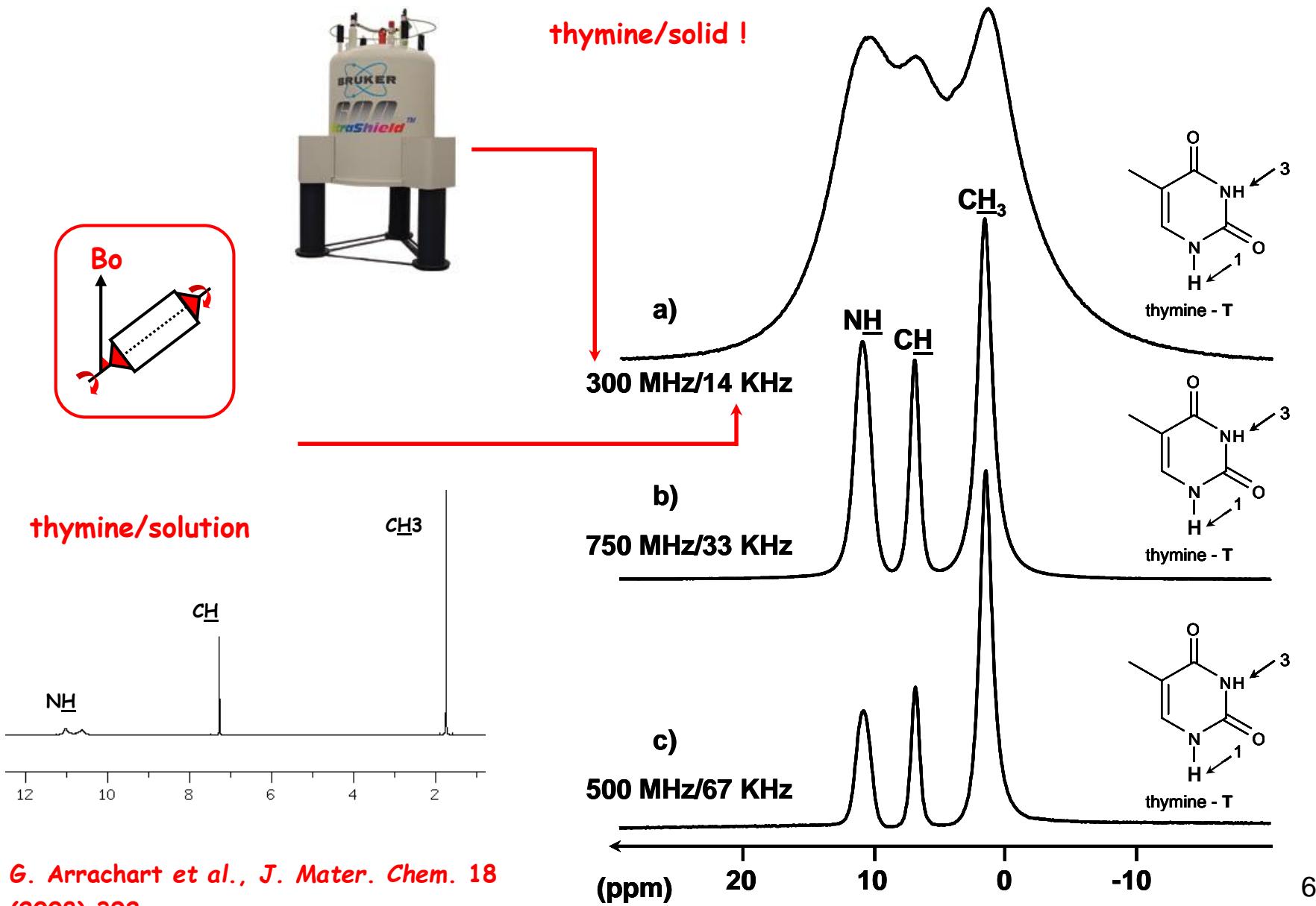
Ureidopyrimidinone based systems



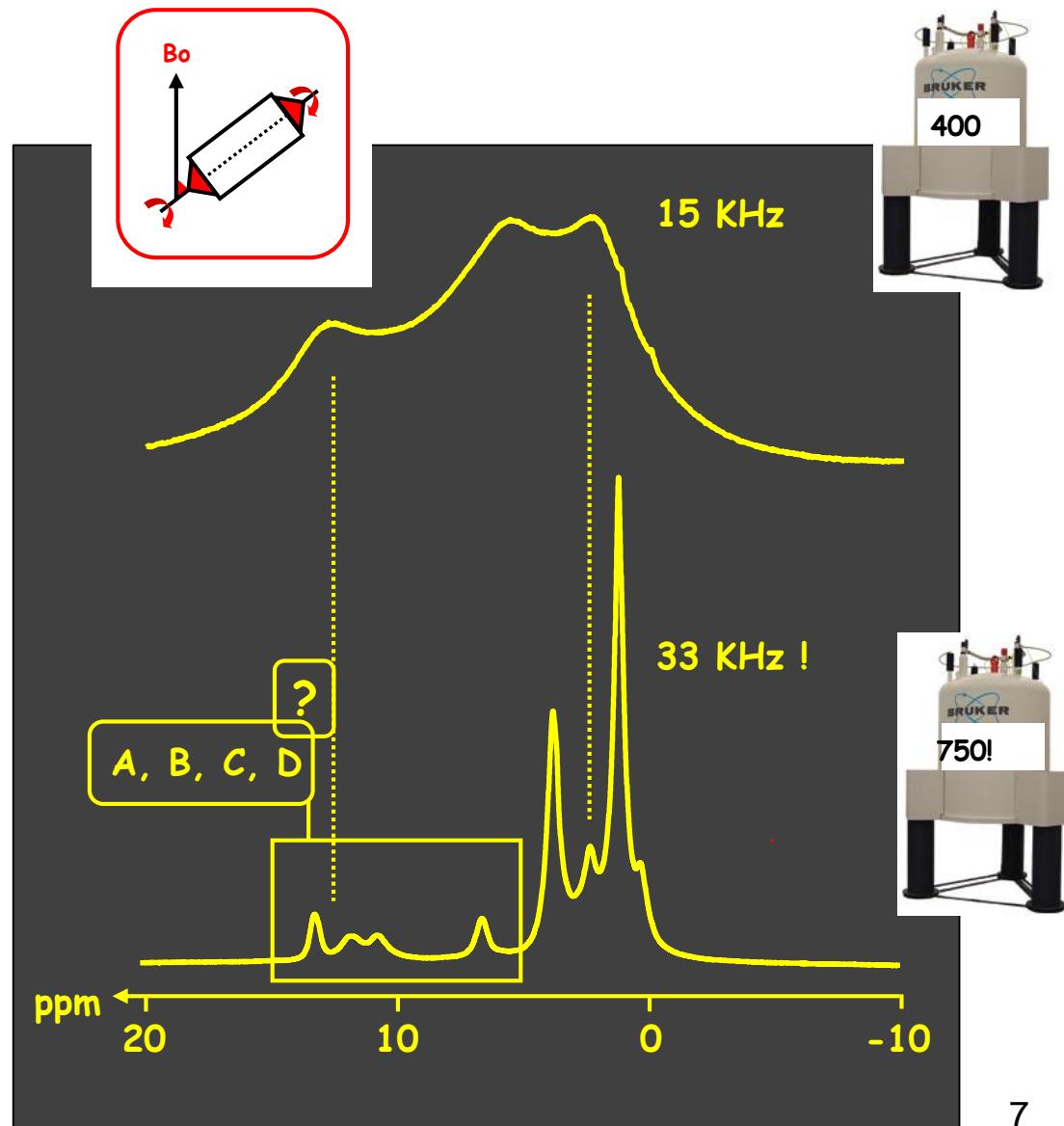
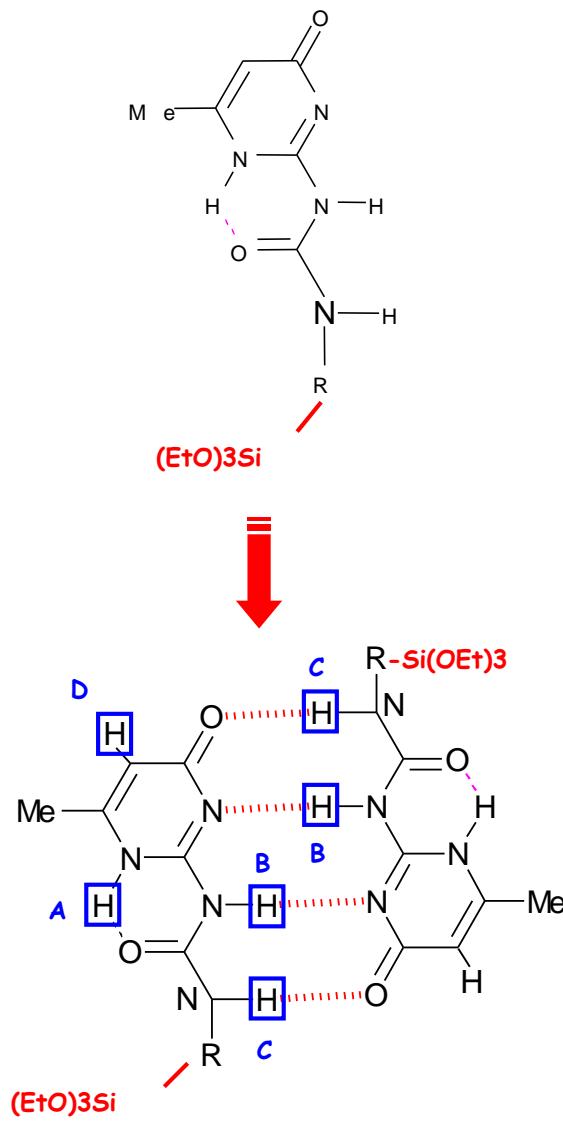
XRD of precursors



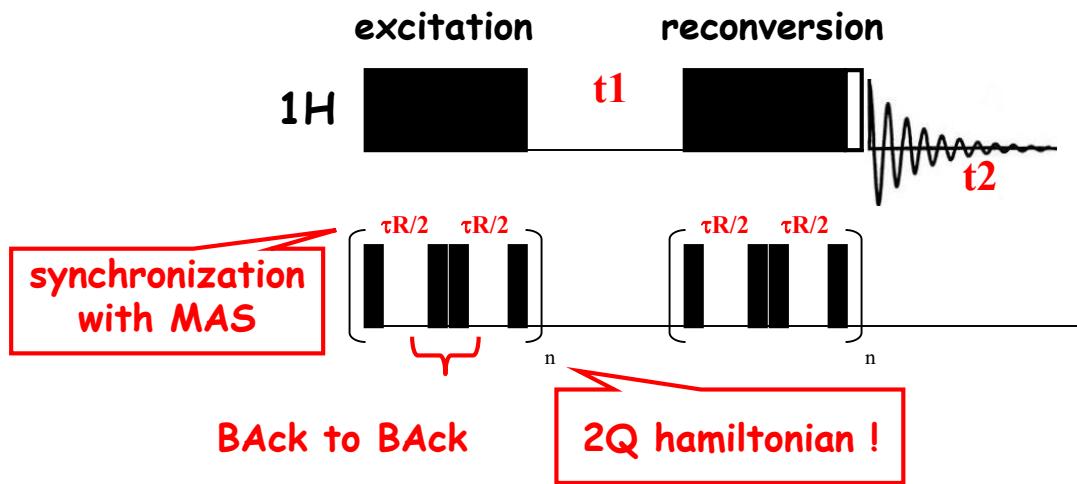
¹H high resolution solid state NMR



Ureidopyrimidinones: ^1H high resolution solid state NMR



Spatial connectivities: DQ ^1H fast MAS spectroscopy

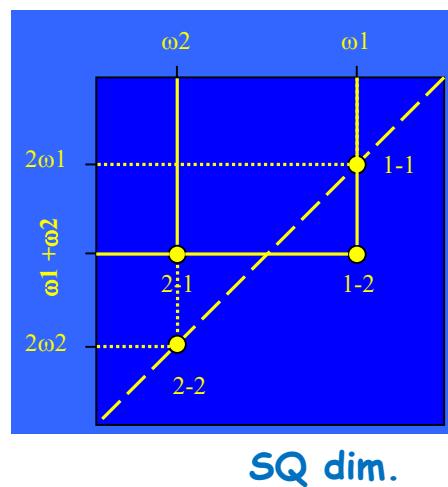


$$\text{DHH} \propto 1/r^3$$

${}^1\text{H}$ ${}^1\text{H}$
 r

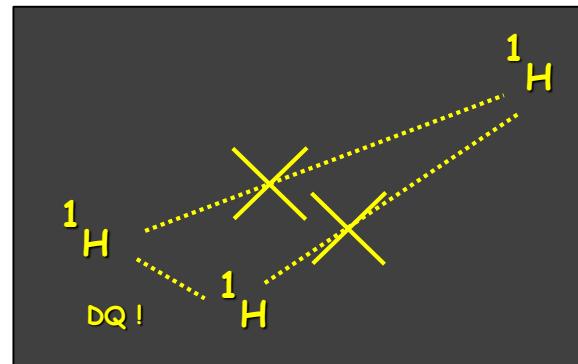
$I=1/2$ $I=1/2$

DQ
 $\langle + + | \leftrightarrow | - - \rangle$

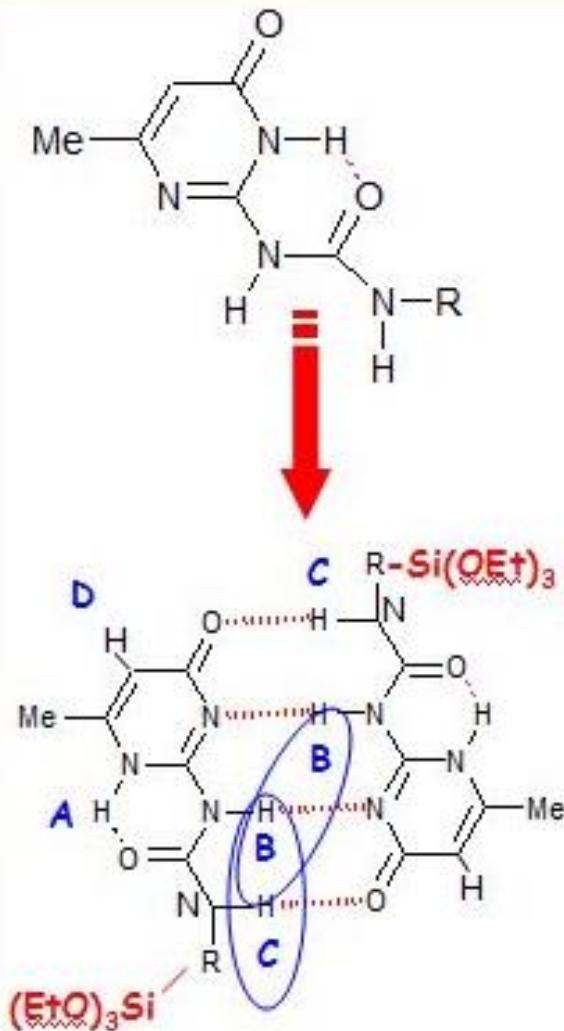


$\delta_{\text{iso.}}$: very fast MAS, very high B_0 !

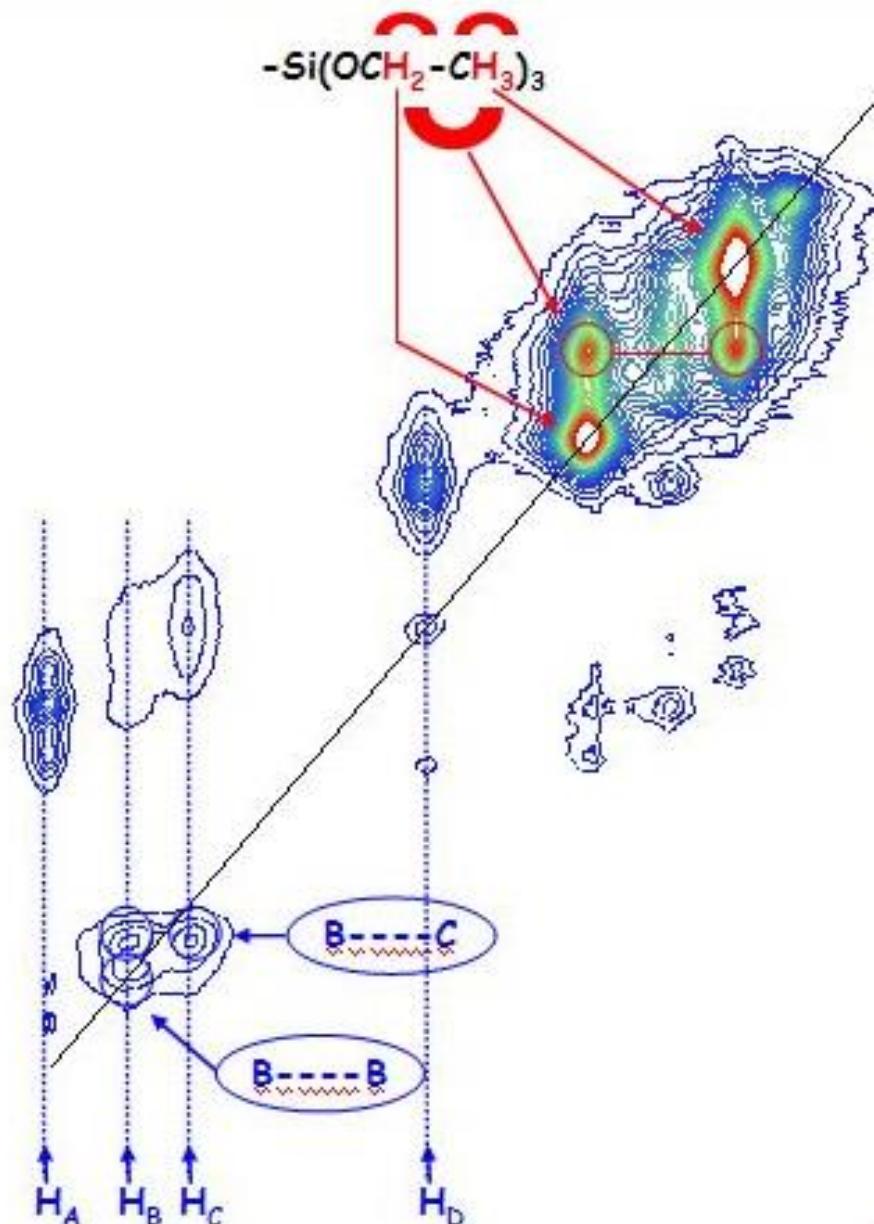
selectivity



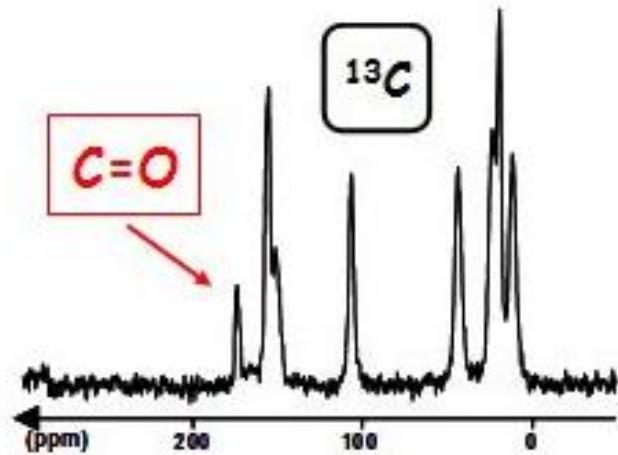
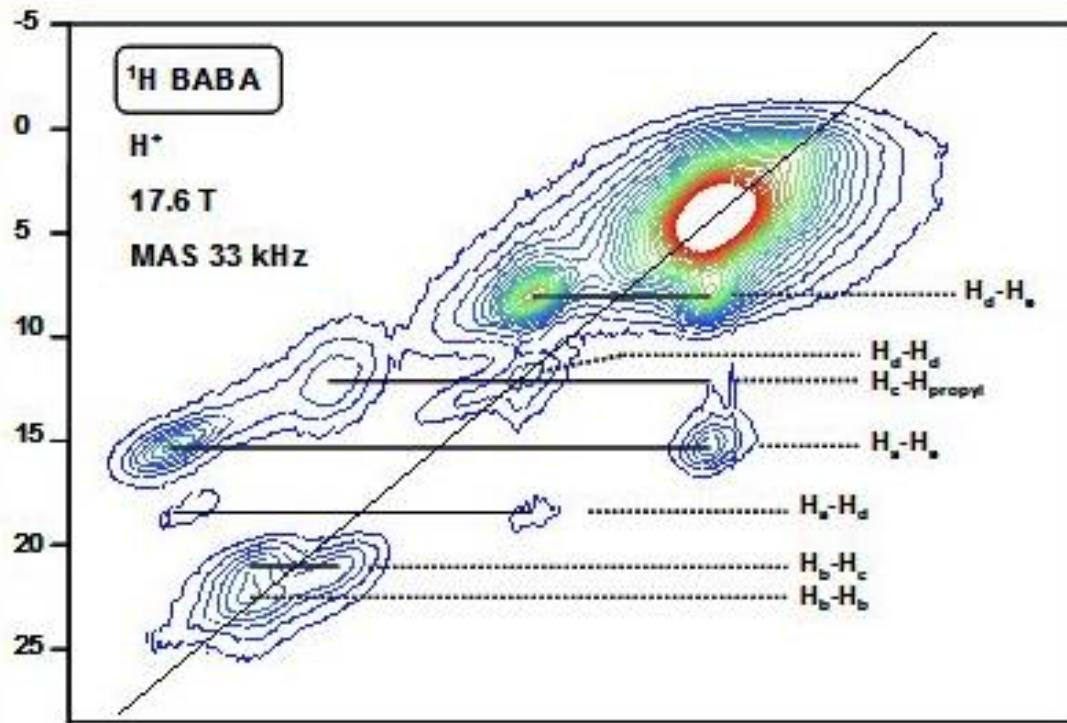
Application to ureidopyrimidinone precursors



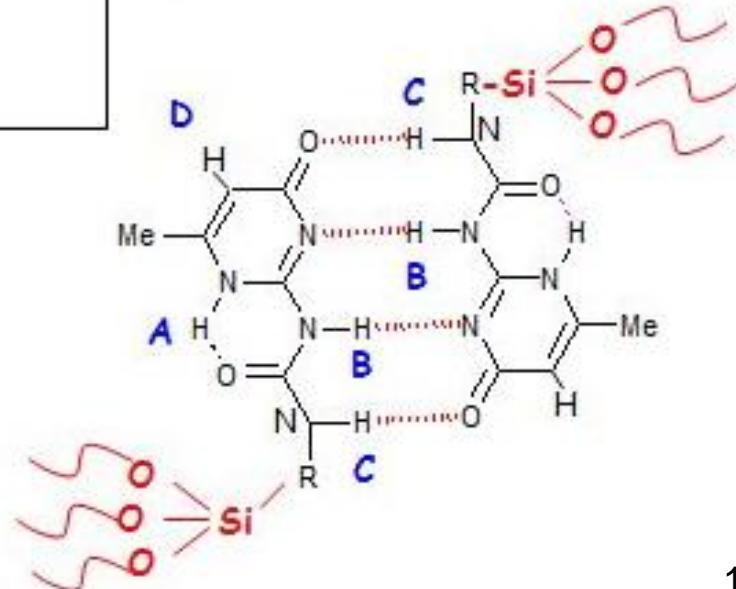
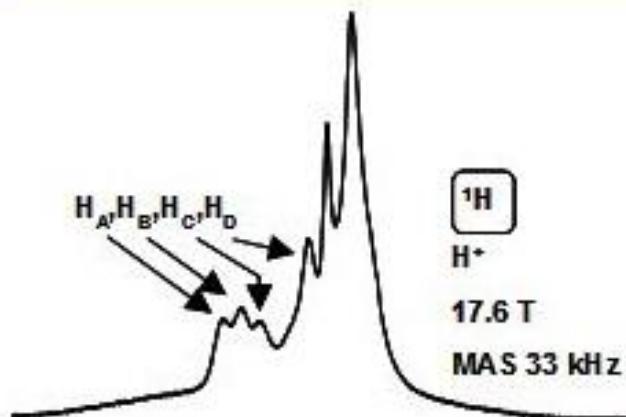
C. Bonhomme et al. Chem. Eur. J. (2008)



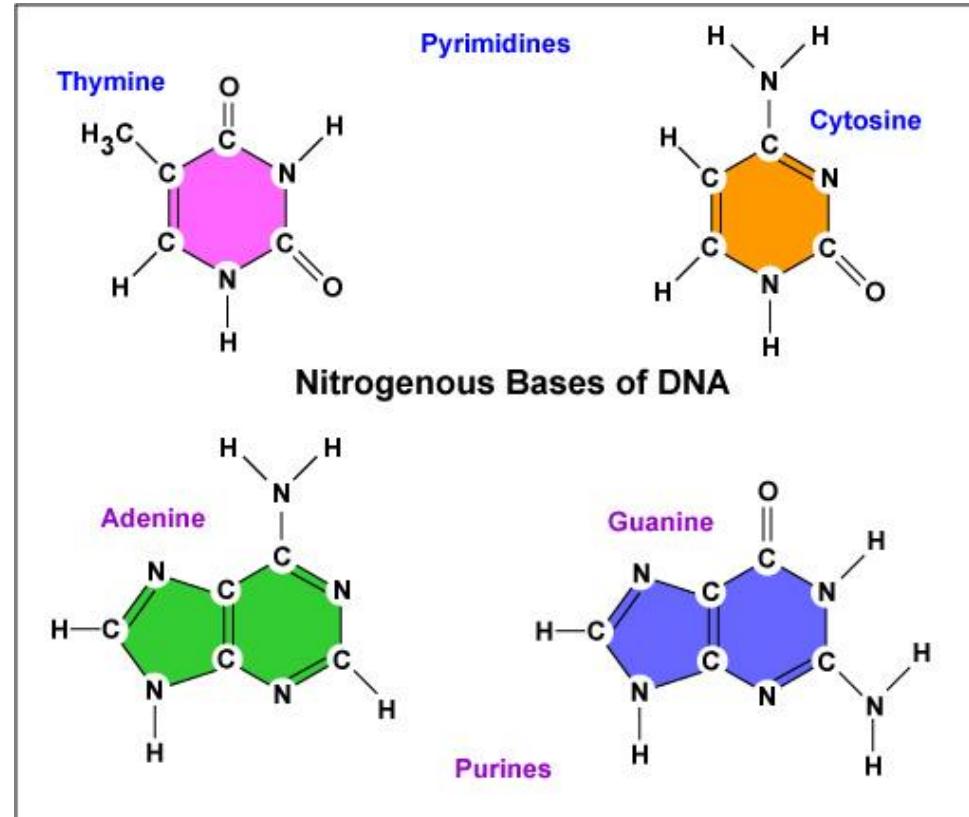
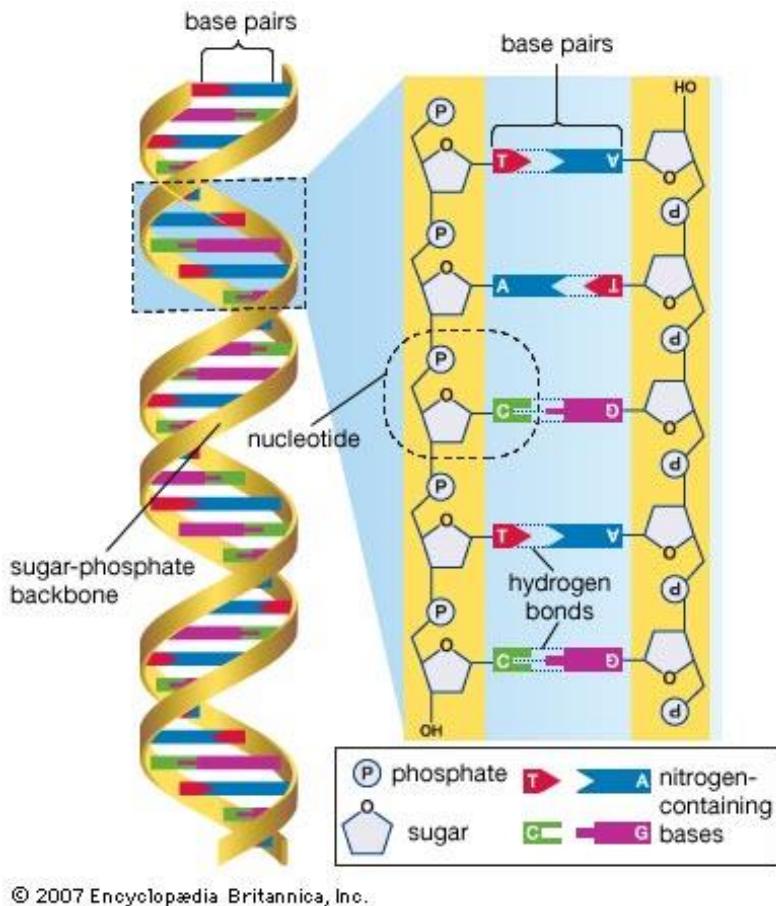
Application to ureidopyrimidinone derived materials: hybrid silica



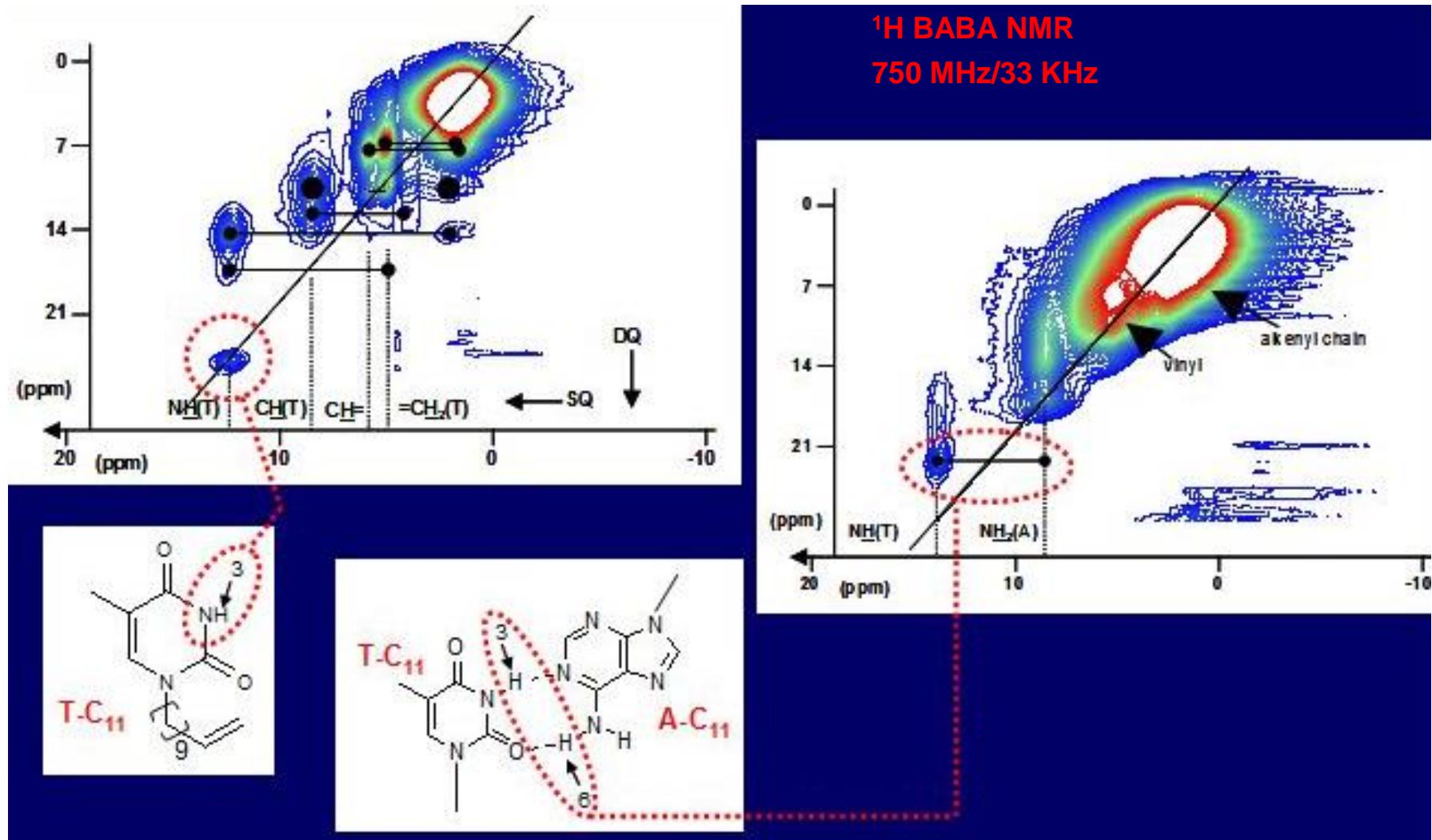
hydrolysis-condensation



Motivations



Towards bio-inspired materials: adenine (A) and thymine (T) derivatives

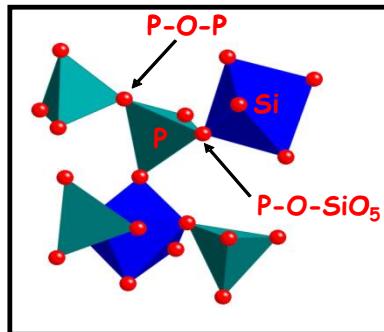




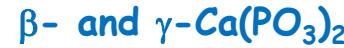
monoclinic 1

monoclinic 2

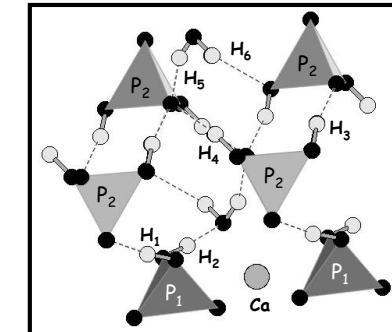
...



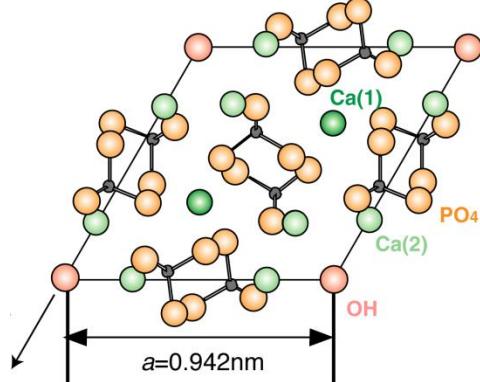
■ silicophosphates



...

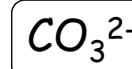
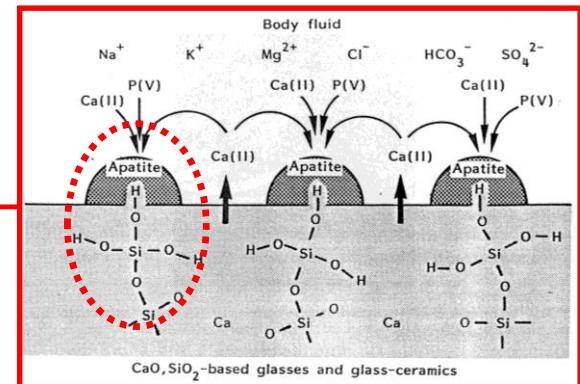


■ calcium phosphates

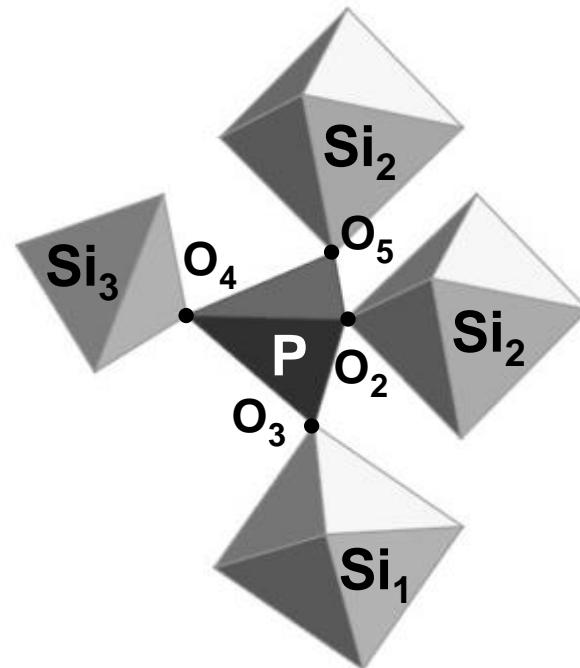
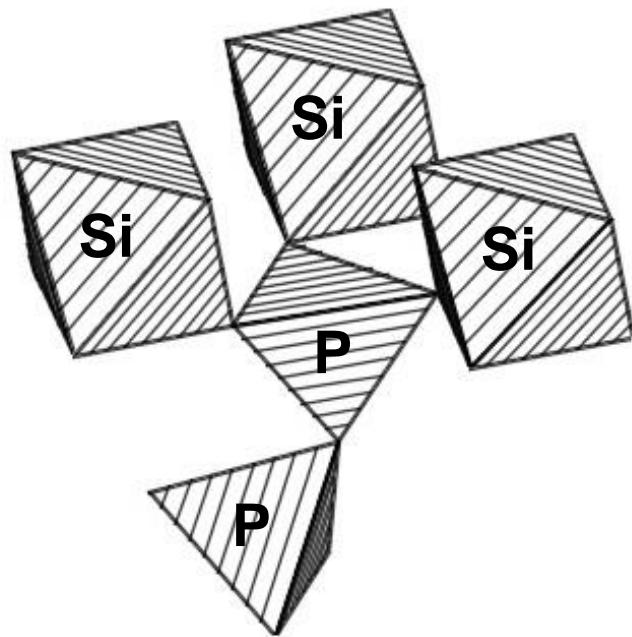


■ hydroxyapatite (HAp)

biocompatibility and
bioactivity of complex
systems

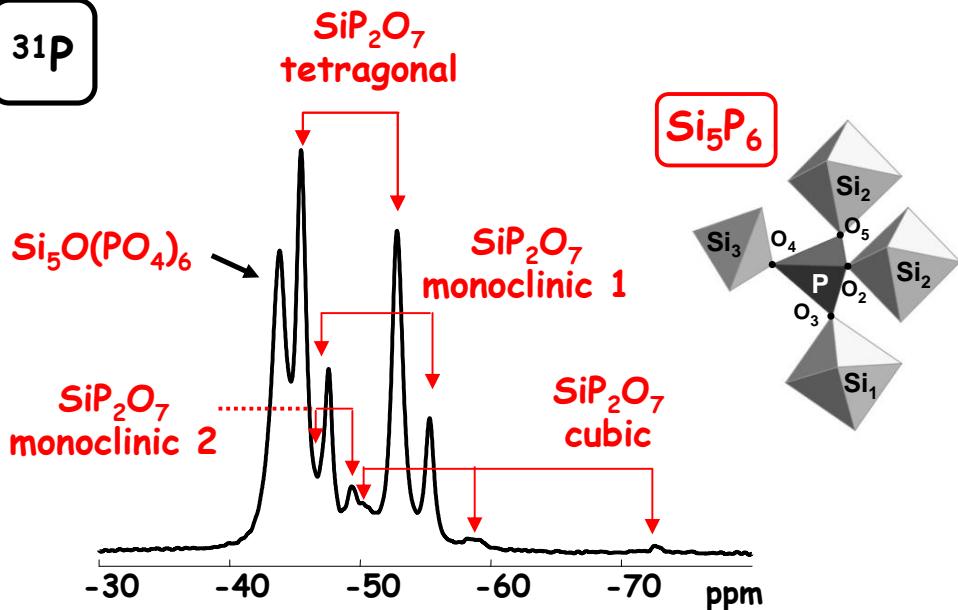


◆ silicophosphates

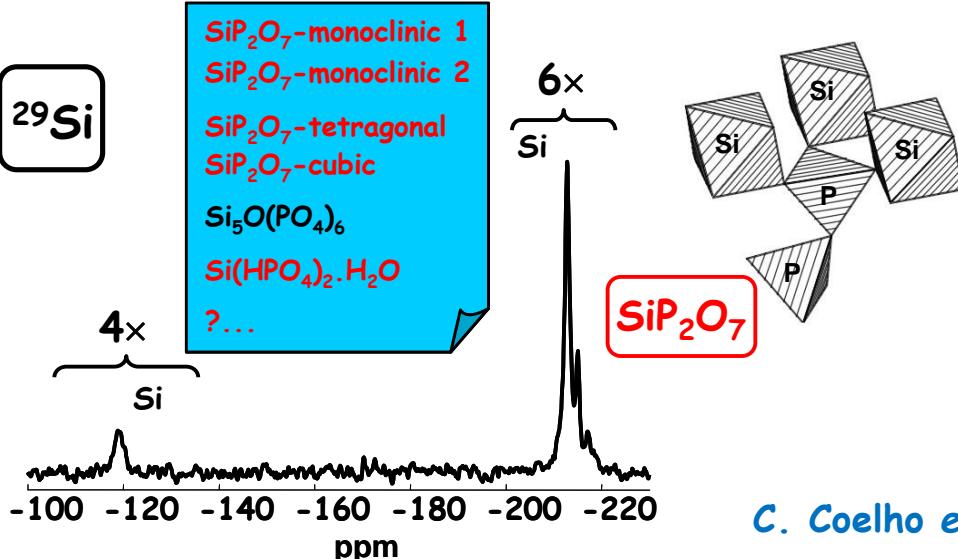


Crystalline silicophosphates: $\text{Si}_5\text{O}(\text{PO}_4)_6$ and SiP_2O_7 polymorphs

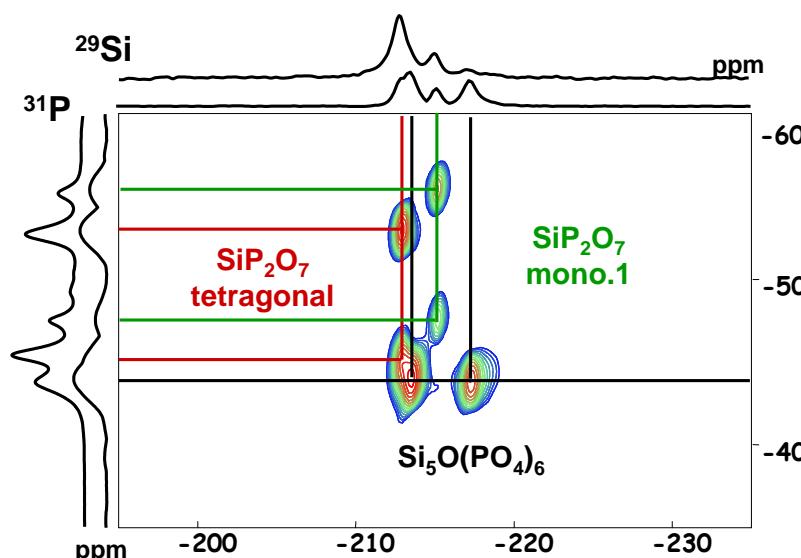
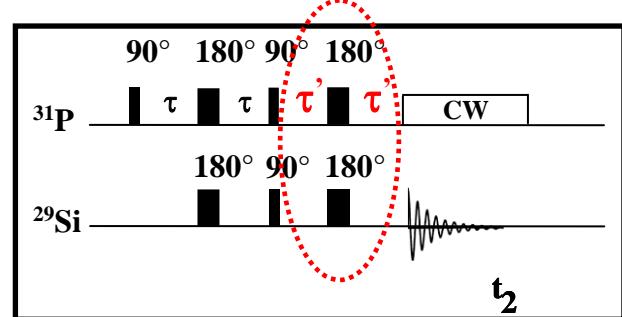
^{31}P



^{29}Si



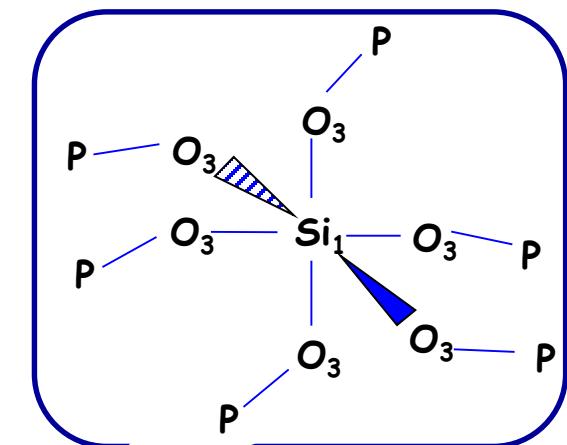
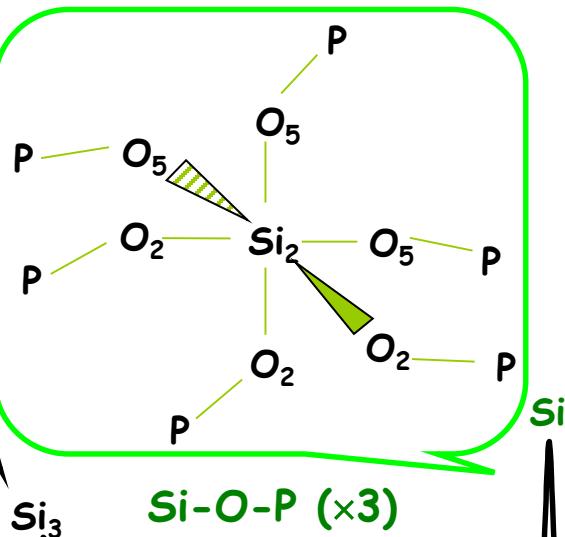
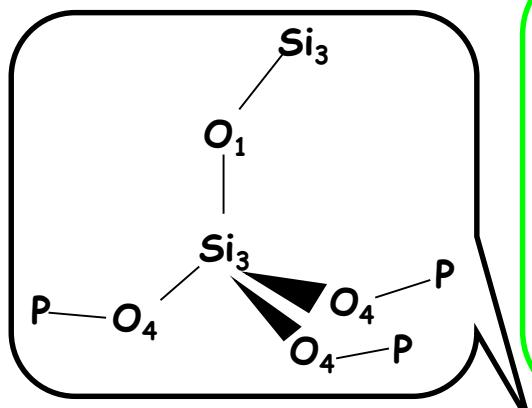
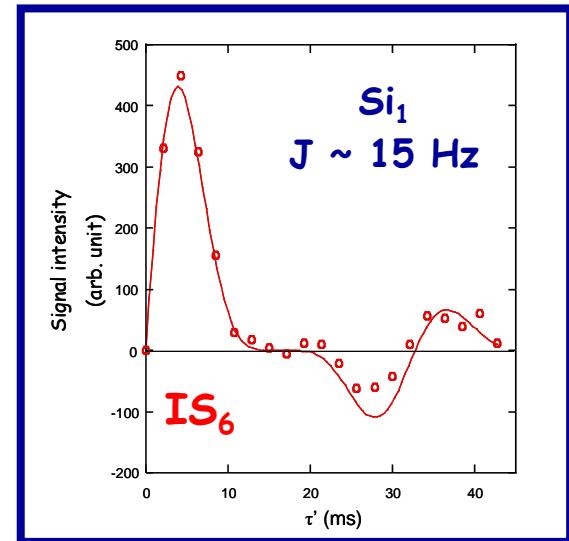
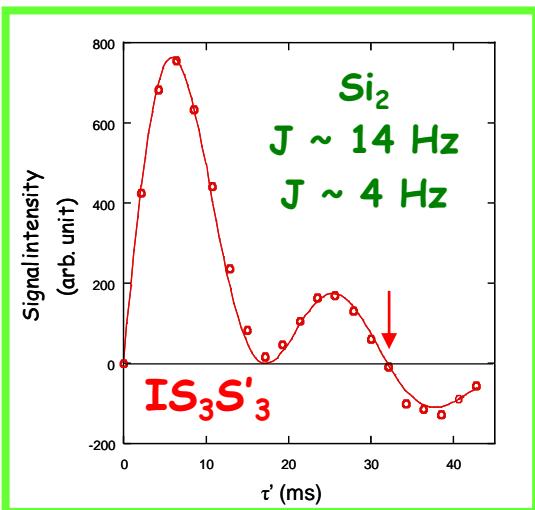
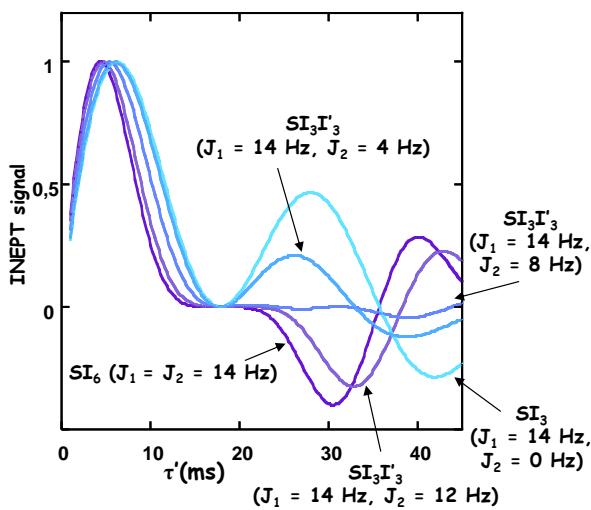
$^{31}\text{P} \rightarrow ^{29}\text{Si}$ MAS-J-INEPT



C. Coelho et al., Inorg. Chem. 46 (2007) 1379.

C. Coelho et al., J. Magn. Reson., 179 (2006) 106.

$^{31}\text{P} \rightarrow ^{29}\text{Si}$ MAS-J-INEPT curves: $\text{Si}_5\text{O}(\text{PO}_4)_6$



C. Coelho et al., Inorg. Chem. 46 (2007) 1379.

-100 -200 (ppm)

First principles calculations: the GIPAW approach

Pickard, Mauri, Phys. Rev. B (2001)

DFT
periodic systems

all-electron hamiltonians

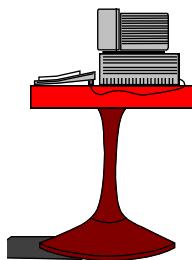
evaluation of $j^{(1)}(r')$ using pseudopotentials

$$B_{in}^{(1)}(r) = 1/c \int d^3r' j^{(1)}(r') \times \frac{r-r'}{|r-r'|^3}$$



EFG

J



IDRIS

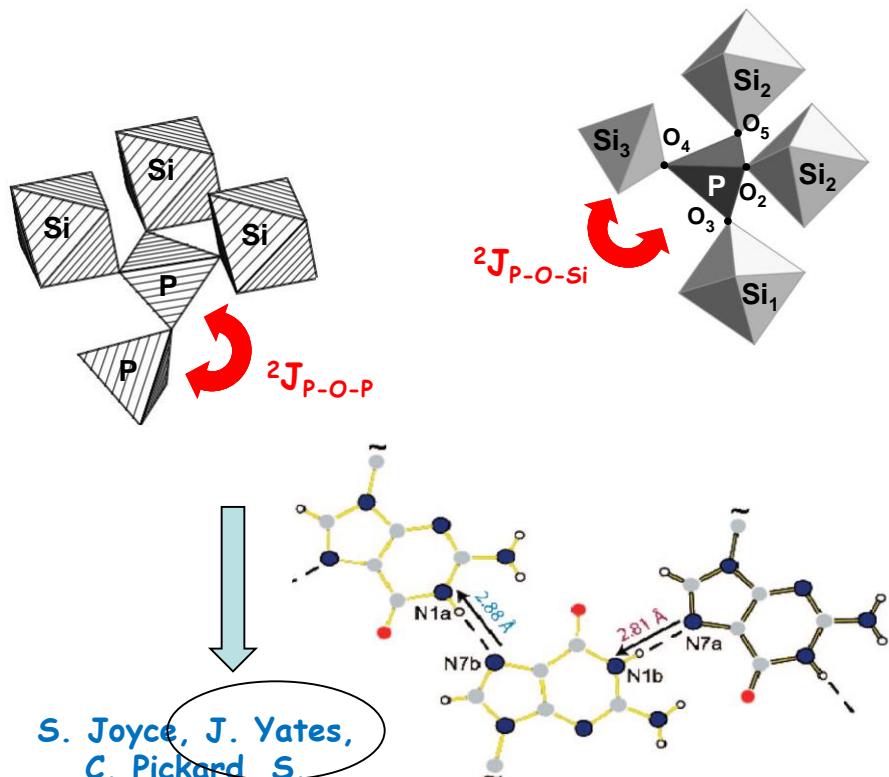
Gervais et al., Magn. Reson. Chem. 42 (2004) 445.

Gervais et al., J. Phys. Chem. A 109 (2005)
6960.

Bonhomme et al., J. Magn. Reson. 2007,
Chem. Mater. 2008, J. Am. Chem. Soc.
2009, 2010

GIPAW

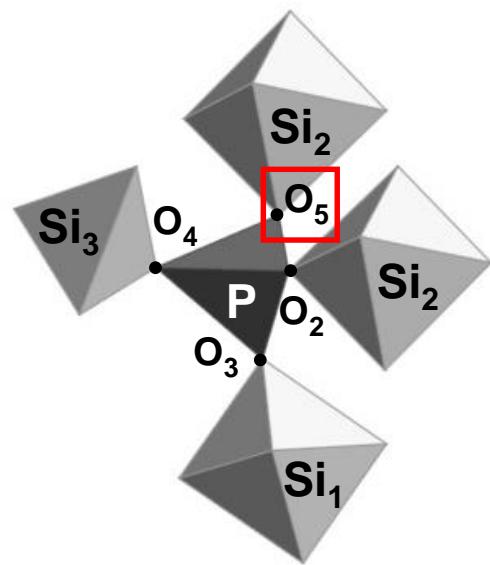
inorganic and organic
derivatives...



S. Joyce, J. Yates,
C. Pickard, S.
Brown, J. Am.
Chem. Soc. (2008)

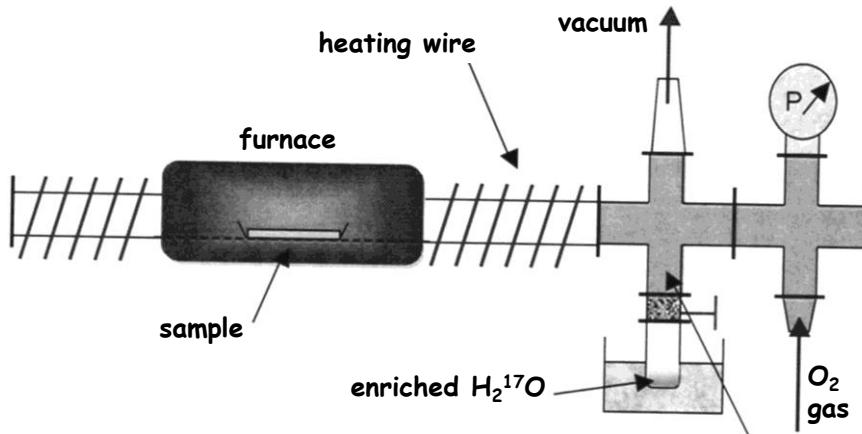
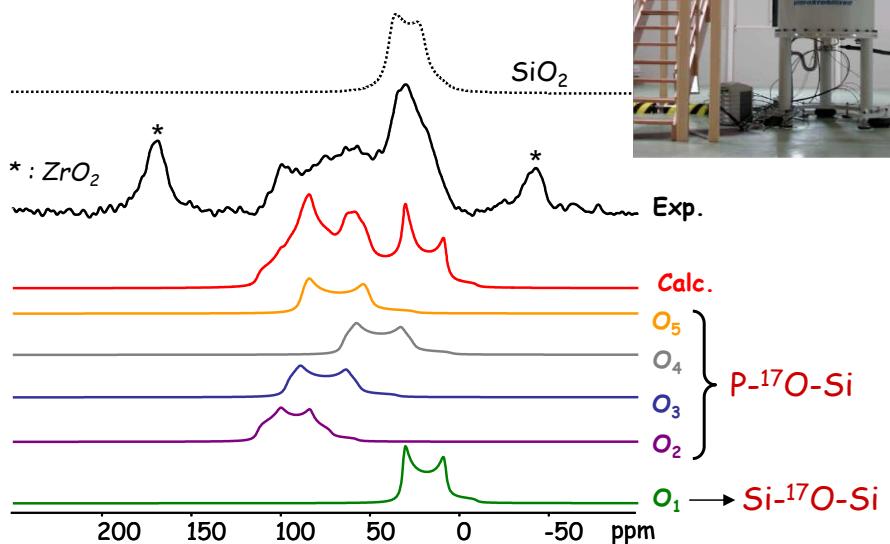
$J_{(N-N)}$

^{17}O data

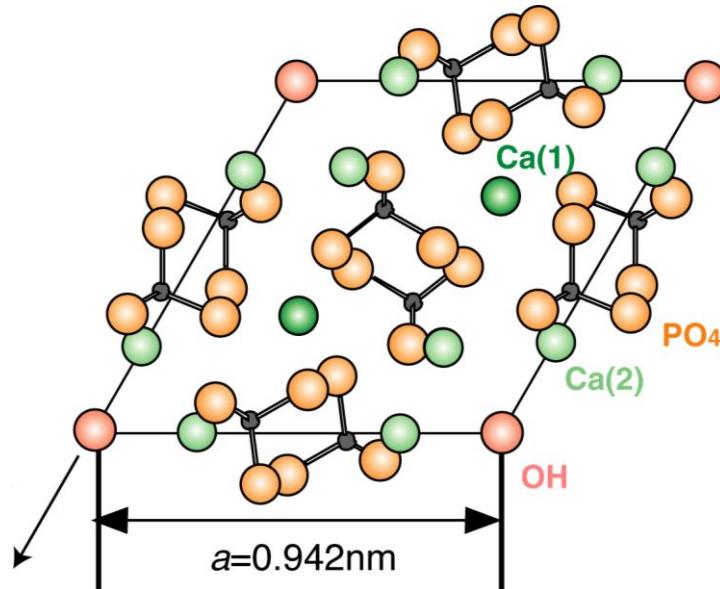
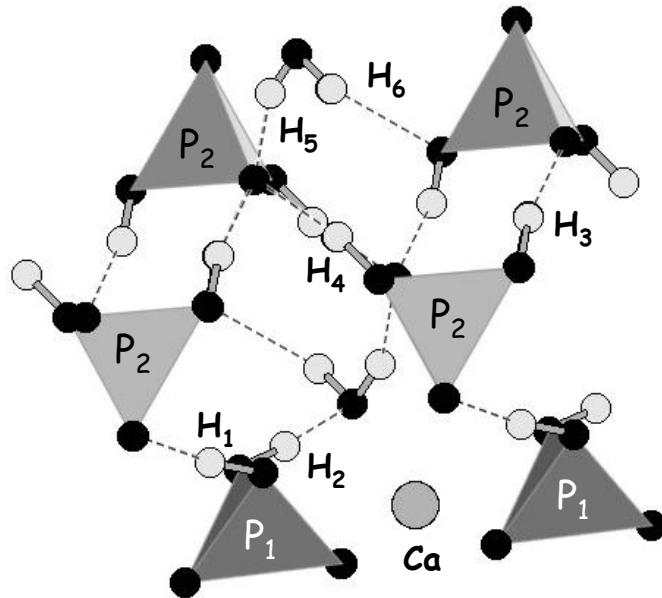


PAW
Q

N.A. ^{17}O MAS experiment



◆ Calcium phosphates and HAp structures



Biocompatible calcium phosphates

Brushite, $\text{CaHPO}_4 \cdot 2\text{H}_2\text{O}$

MCPM, $\text{Ca}(\text{H}_2\text{PO}_4)_2 \cdot \text{H}_2\text{O}$

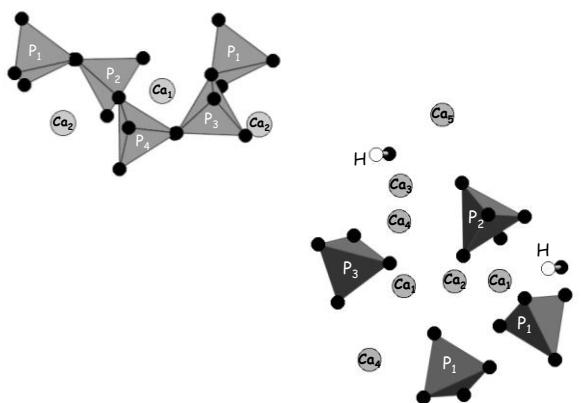
β - and γ - $\text{Ca}(\text{PO}_3)_2$

$\text{Ca}_4\text{P}_2\text{O}_9$

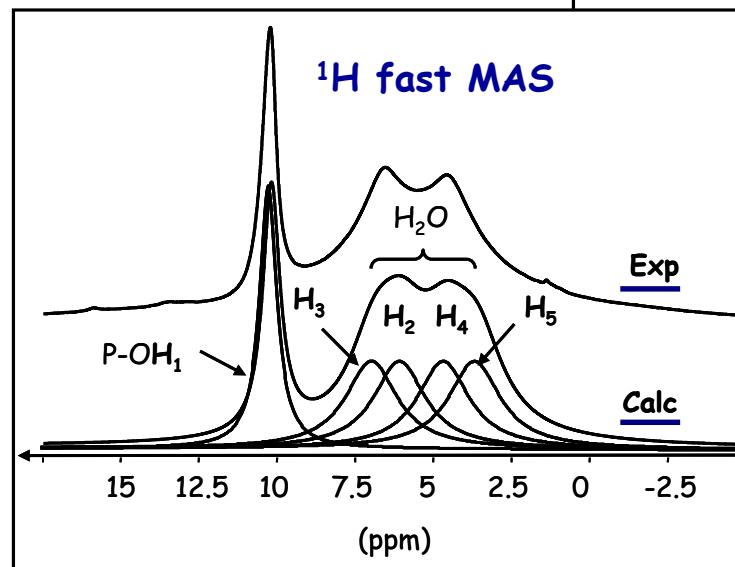
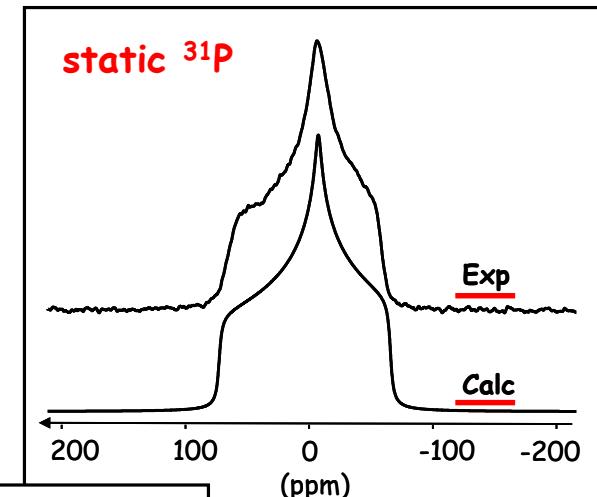
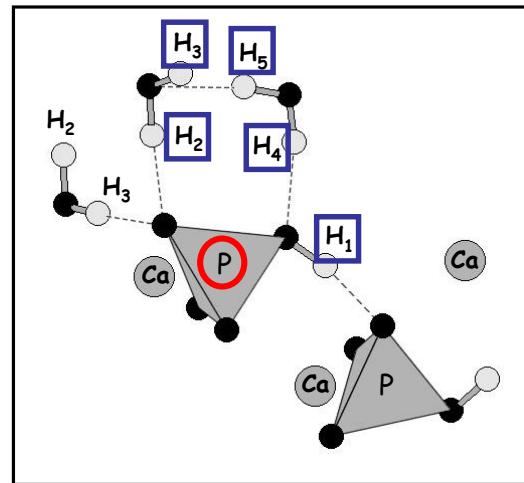
$\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$ (HAp)

...

hydrated, dehydrated,
and hydroxylated
structures



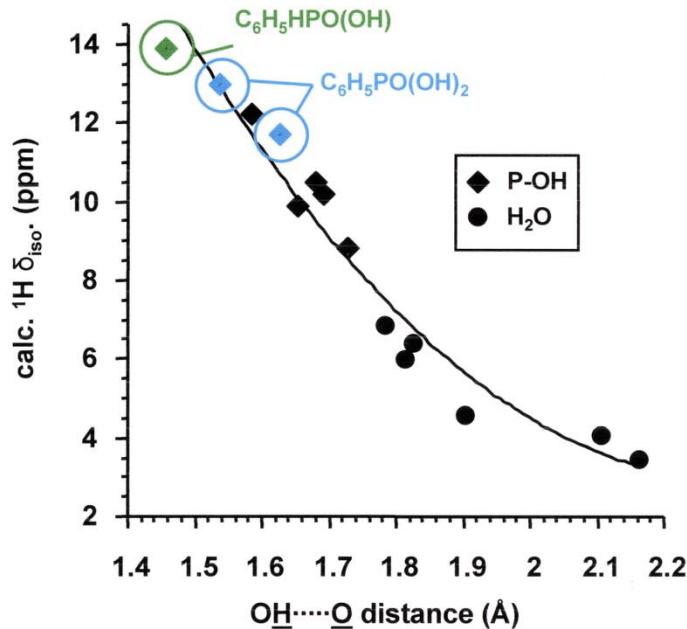
Brushite: the GIPAW approach (^{31}P , ^1H)



B. Alonso, D.
Massiot, CEMHTI,
Orléans, France

More from ^1H GIPAW data: H-bonding and CSA tensors

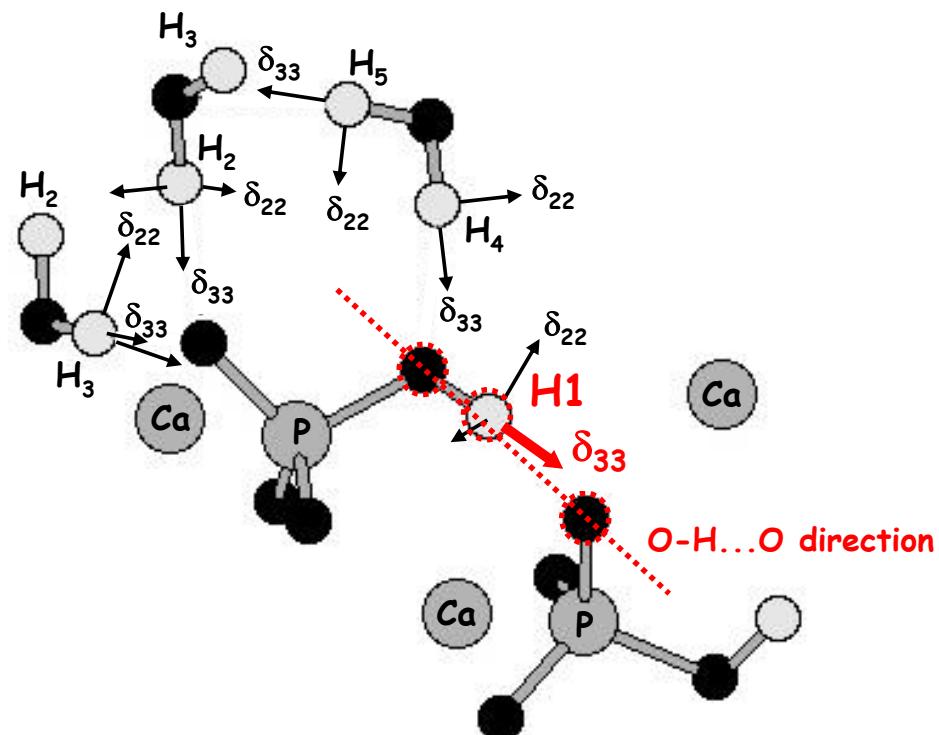
^1H isotropic chemical shifts



H-bonding in calcium phosphates and phosphonic acids

C. Gervais et al., J. Magn. Reson. 187 (2007) 181.

Brushite: $\text{CaHPO}_4 \cdot 2\text{H}_2\text{O}$

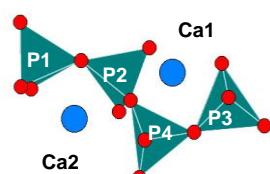


^1H CSA tensors and orientations

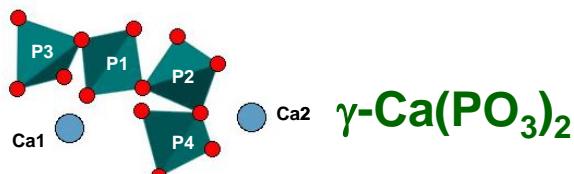
F. Pourpoint et al., Appl. Magn. Reson. 32 (2007) 435.

β - and γ -Ca(PO₃)₂

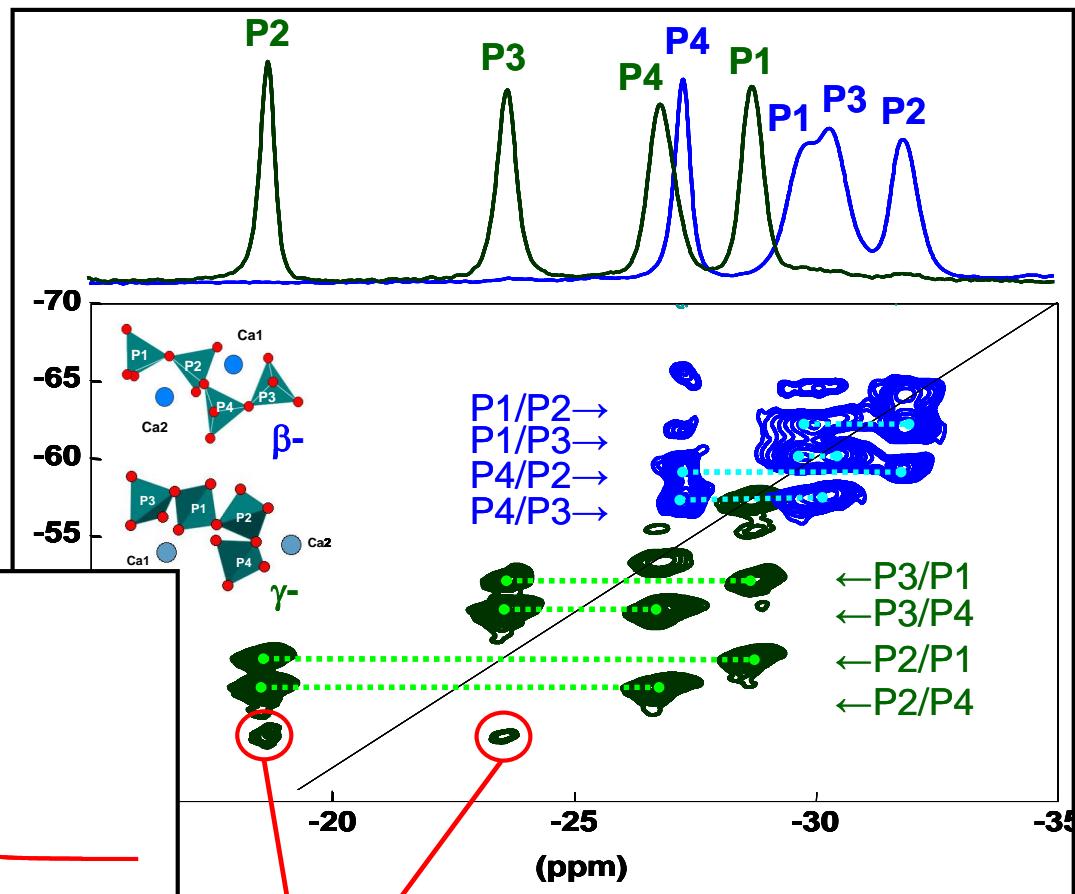
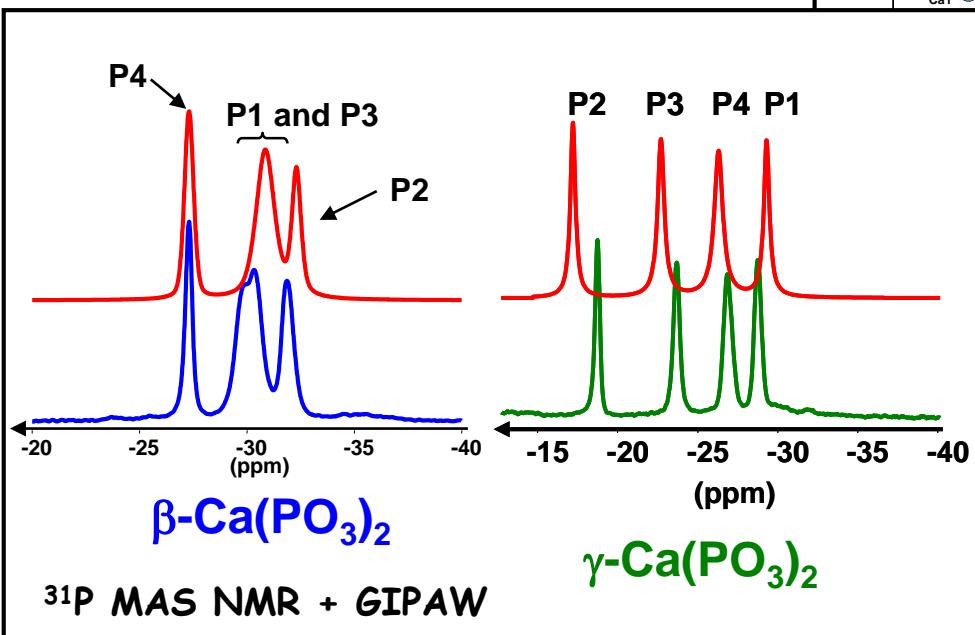
2D ³¹P MAS-J-INADEQUATE



β -Ca(PO₃)₂



γ -Ca(PO₃)₂



$4J_{(P-O-Ca-O-P)}$ up to 3.4 Hz

^{43}Ca NMR spectroscopy

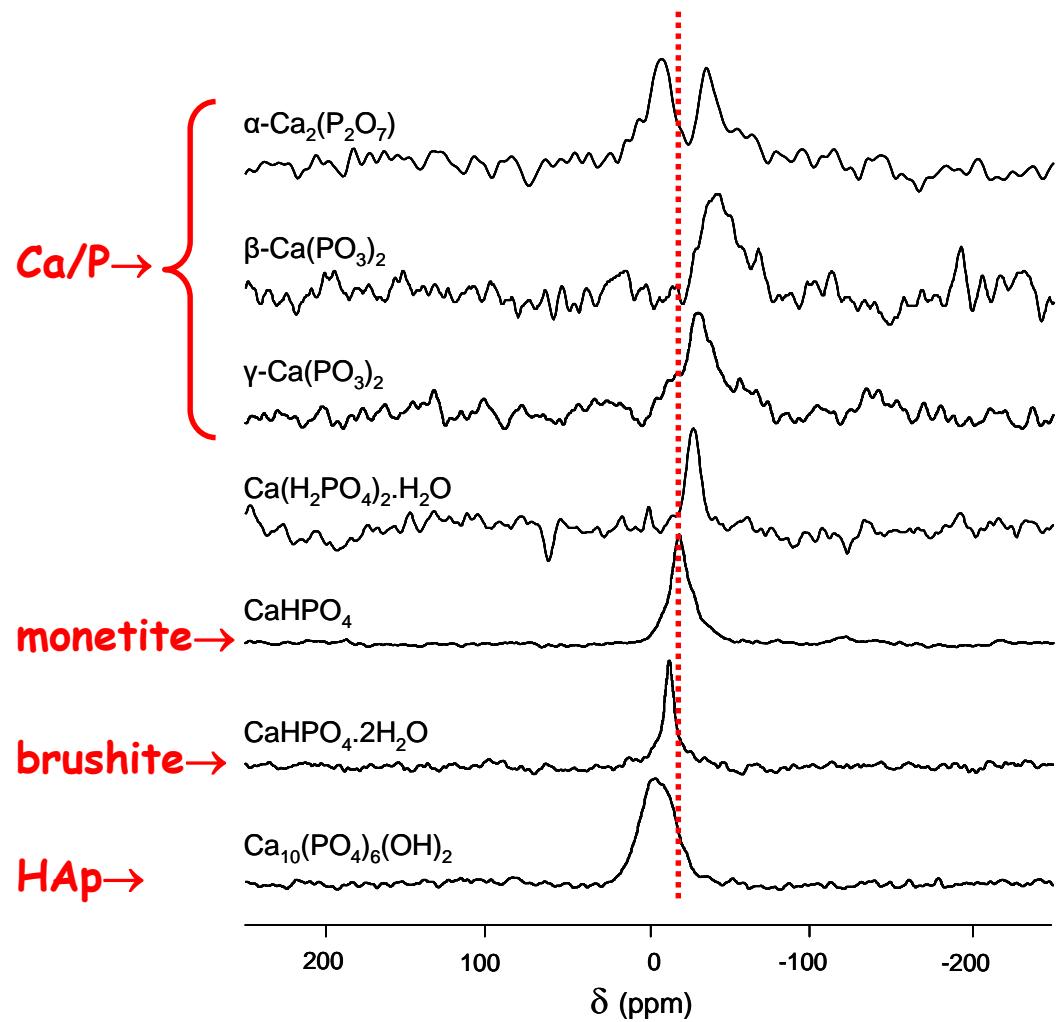
Chemical Physics Letters

New perspectives on calcium environments in inorganic materials containing calcium–oxygen bonds: A combined computational–experimental ^{43}Ca NMR approach

Christel Gervais, Danielle Laurencin, Alan Wong, Frédérique Pourpoint, John Labram, Bleddyn Woodward, Andrew P. Howes, Kevin J. Pike, Ray Dupree, Francesco Mauri, Christian Bonhomme, Mark E. Smith

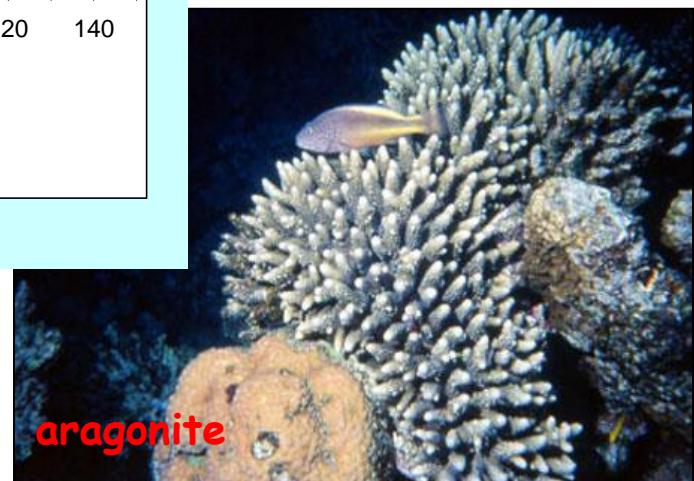
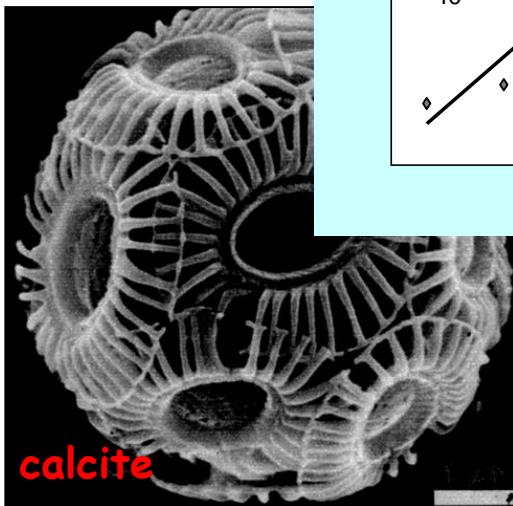
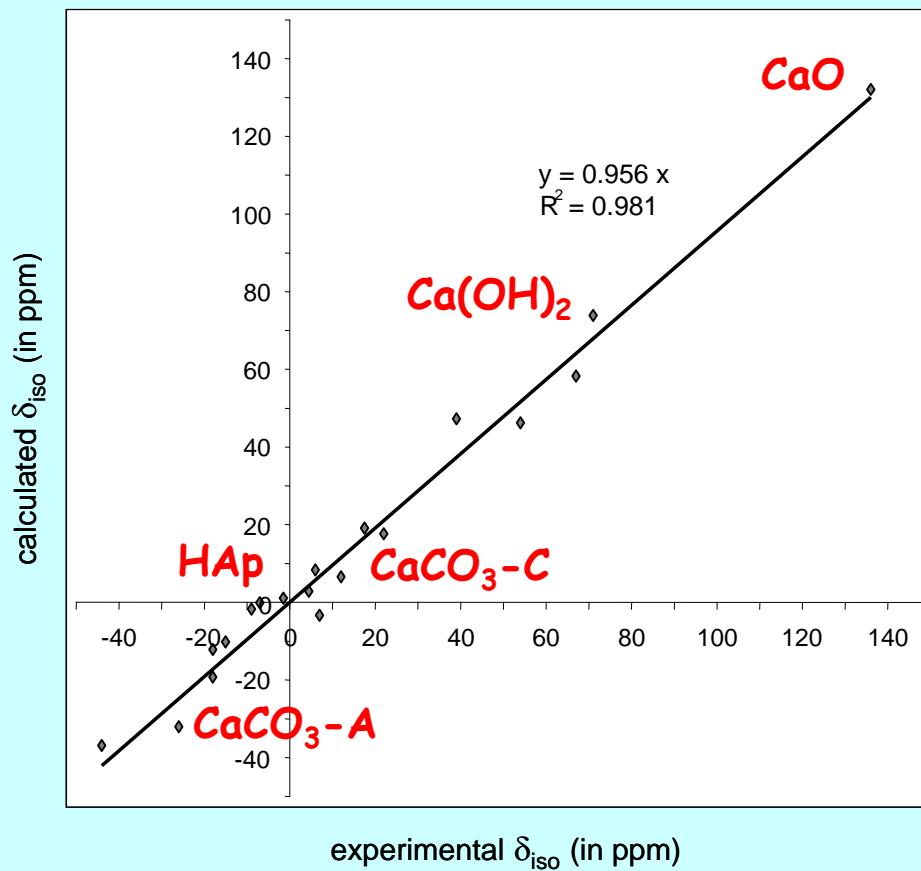
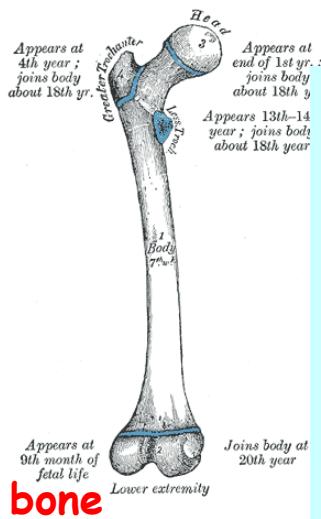
The potential of a combined experimental–computational ^{43}Ca solid state NMR approach for the structural analysis of inorganic compounds is presented.

large volume MAS rotor ($\varnothing 9.5\text{mm}$)
multi field experiments (up to 18.8 T)



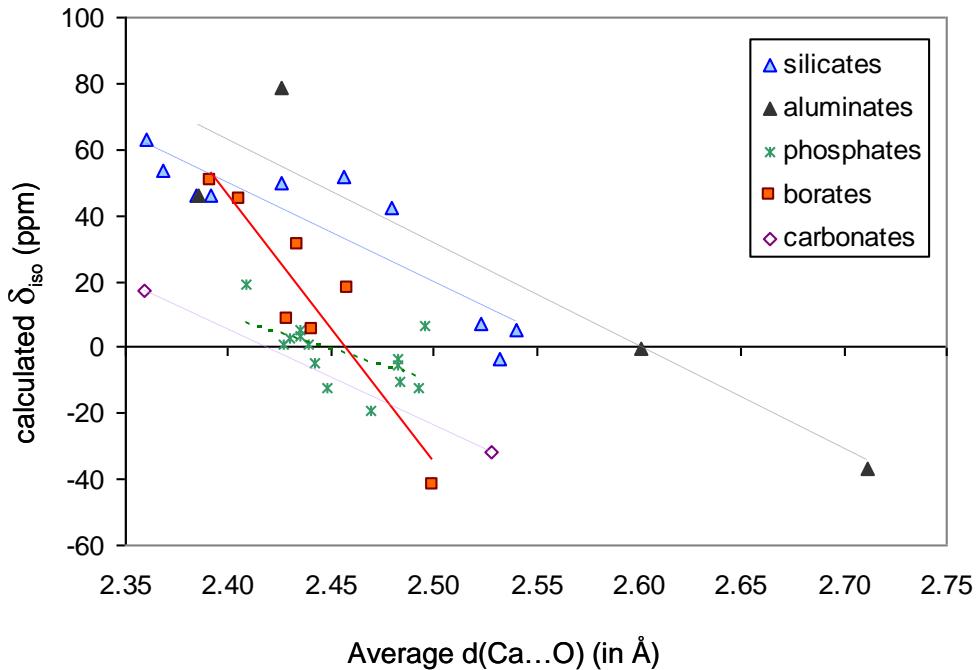
natural abundance (0.14 %) ^{43}Ca MAS NMR

^{43}Ca CSA/Q GIPAW calculations



see also: Bryce et al., J. Am. Chem. Soc. 130 (2008) 9282

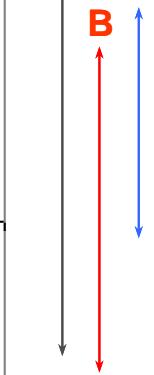
^{43}Ca NMR: a structural tool



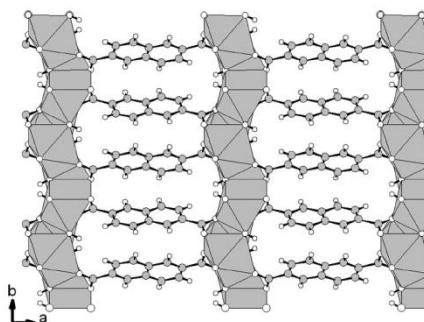
Al

Si

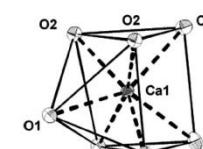
P



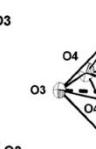
Ca based Metal Organic Frameworks



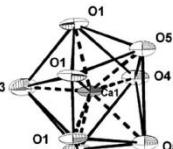
drugs



(a)

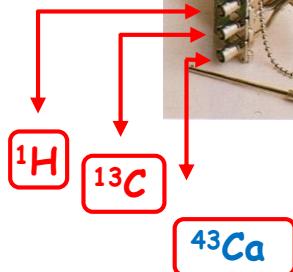
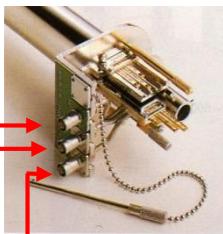


(b)



(c)

C. Volkringer et al., Cryst. Growth & Design 8 (2008) 685



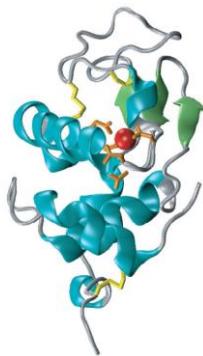
D. Laurencin et al., J. Am. Chem. Soc. 130 (2008) 2412

D. Laurencin, C. Bonhomme et al. J. Am. Chem. Soc. (2009)

Towards $^1\text{H}/^{13}\text{C}/^{43}\text{Ca}$ triple resonance experiments

some structural key questions:

Acharya, 1991, Bushmarina, 2005

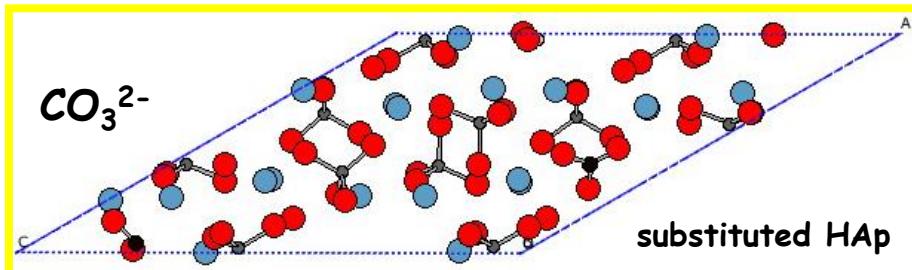


CALcium MODULated proteIN
 α -Lactalbumin

calcium ligands in human and baboon α -lacs

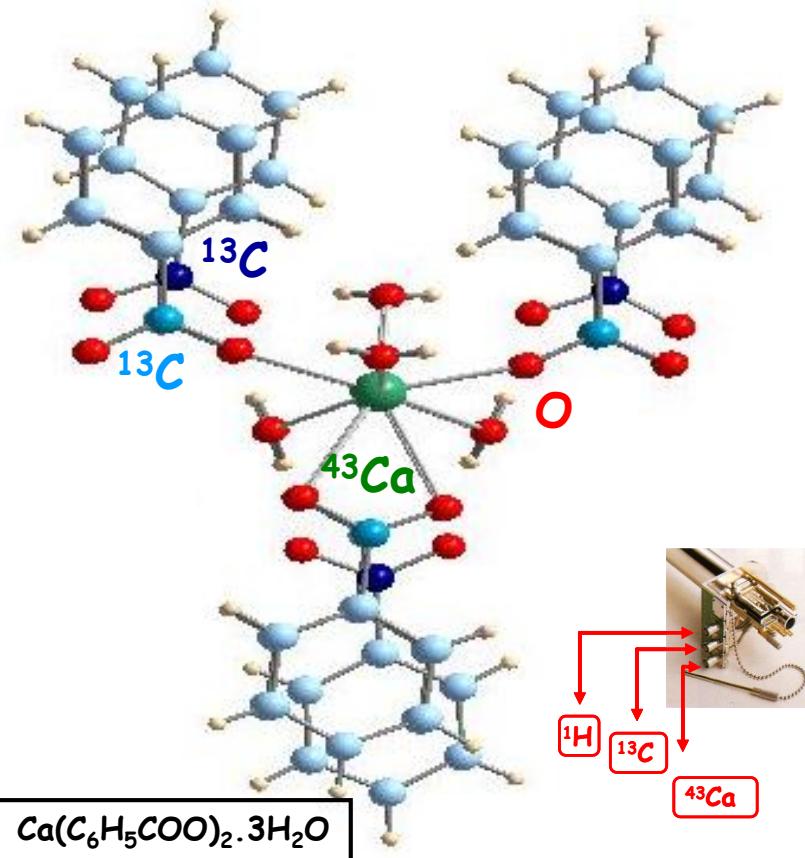
Residues	Group	Distance to Ca (\AA)		B-value (\AA^2)	
		Human	Baboon	Human	Baboon
Lys79	Carbonyl O	2.3	2.2	8.4	13.9
Asp82	Carboxylate OD1	2.4	2.4	11.0	11.7
Asp84	Carbonyl O	2.2	2.3	5.8	12.6
Asp87	Carboxylate OD1	2.4	2.3	10.4	7.9
Asp88	Carboxylate OD1	2.4	2.3	7.3	18.8
	Water O	2.3	2.4	19.1	16.1
	Water O	2.5	2.6	7.9	21.0

■ Ca/protein interactions



■ carbonated HAp

calcium benzoate: a model compound
(^{43}Ca : 60% ; ^{13}C : 100% or natural abundance)

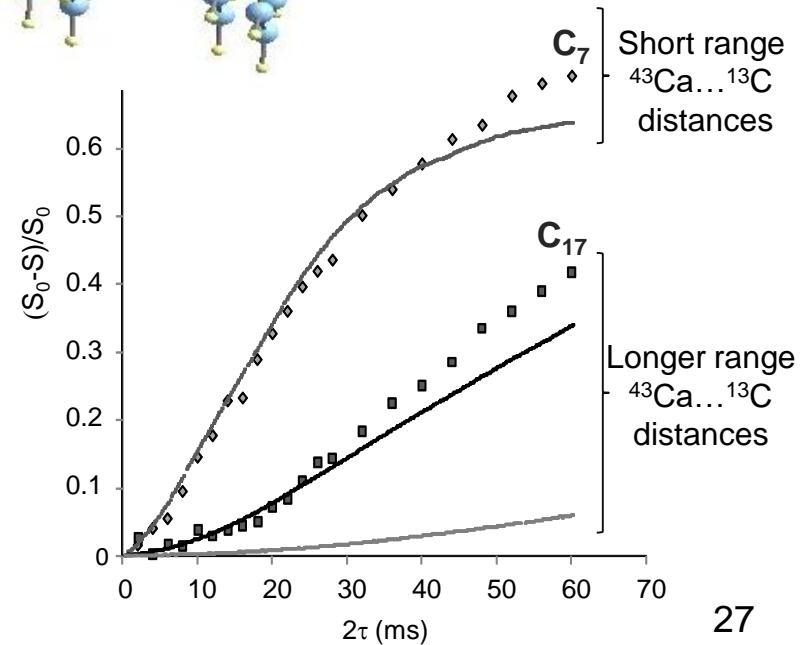
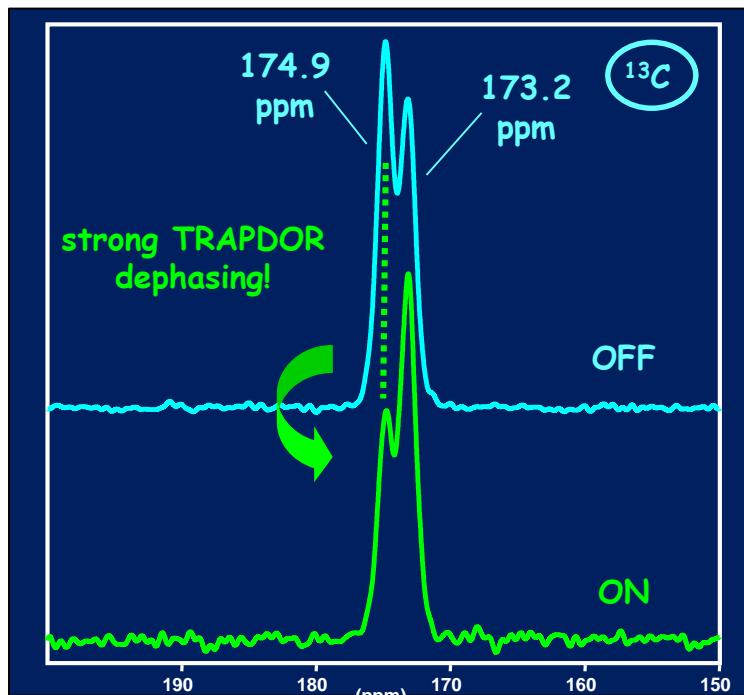
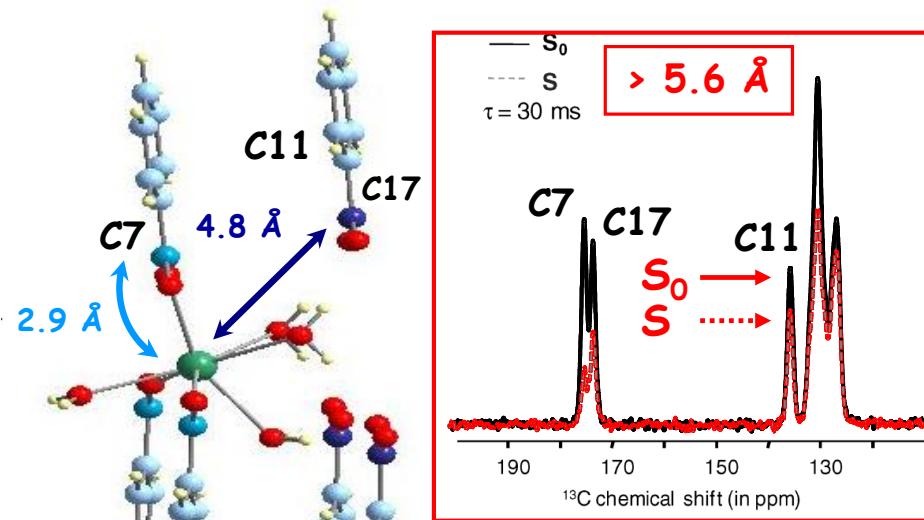
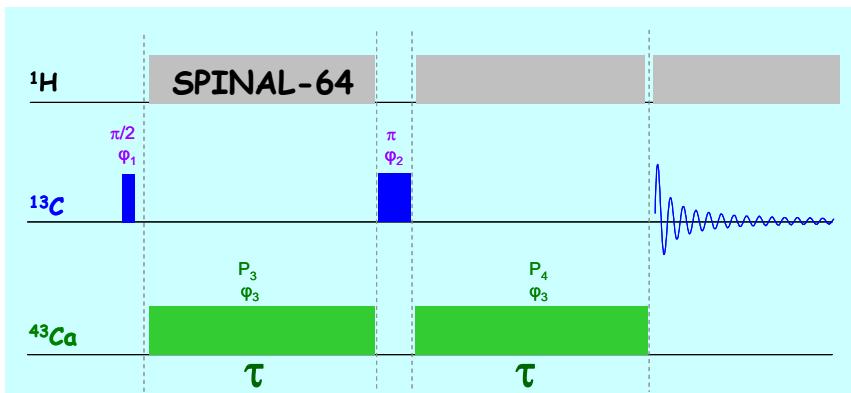


D. Laurencin, C. Bonhomme et al., J. Am. Chem. Soc. 2009

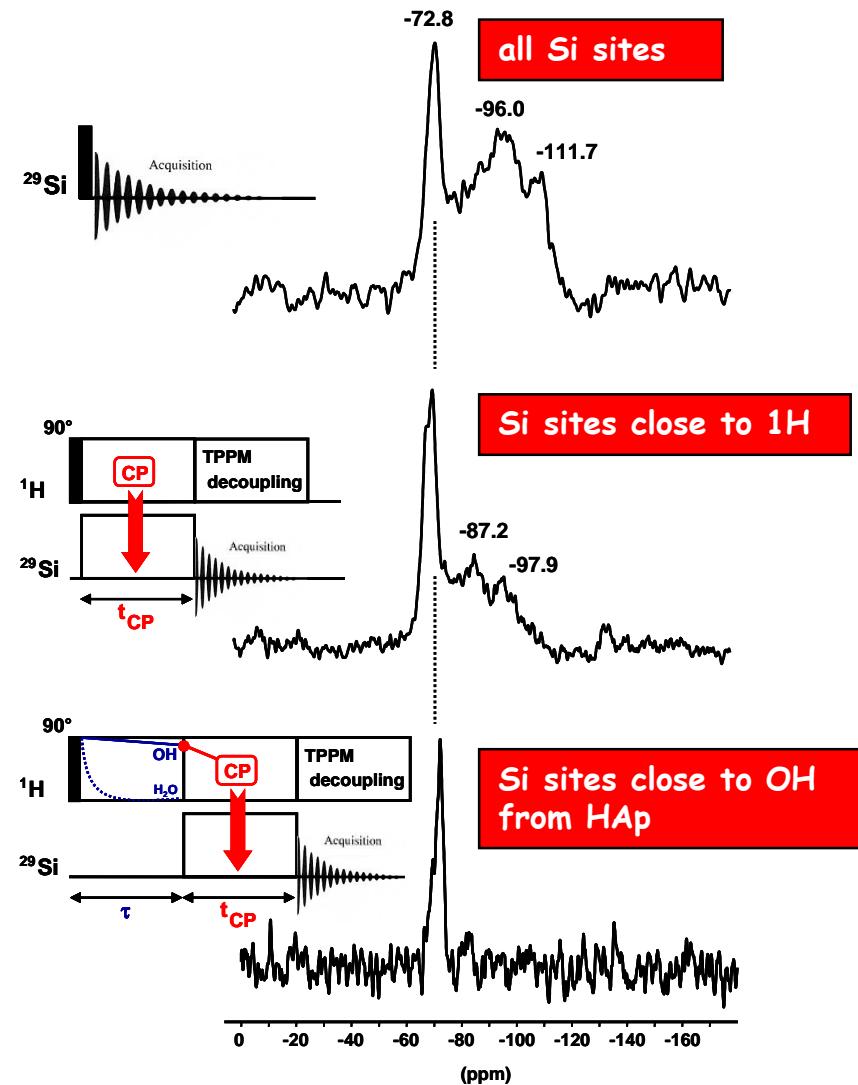
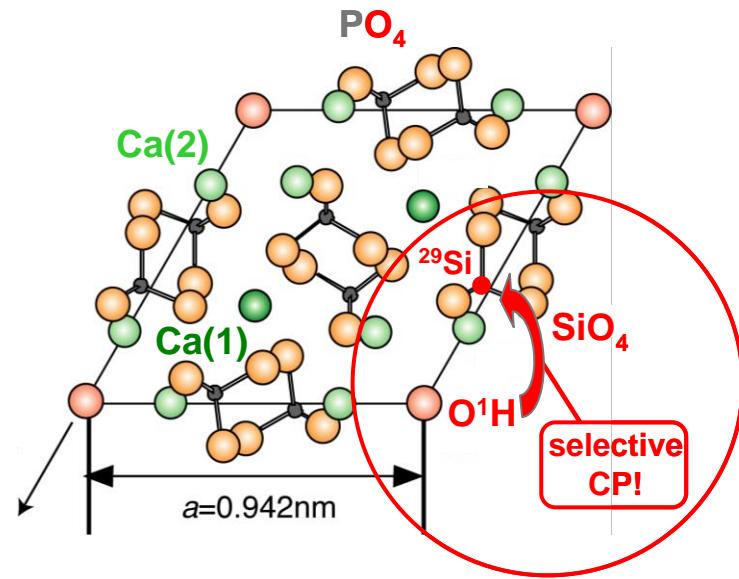
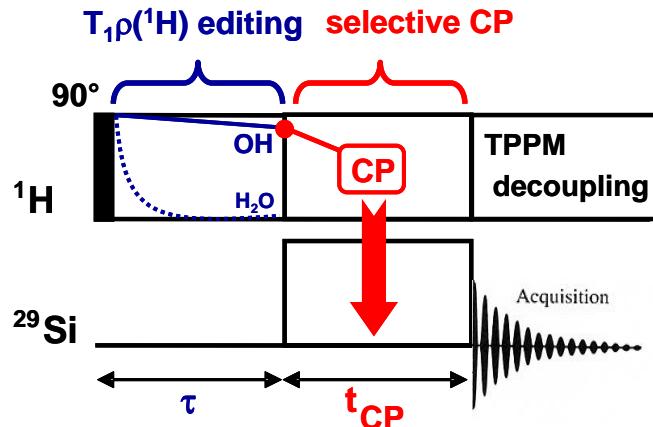
Dipolar MAS experiments: ^{43}Ca - ^{13}C proximities

■ $^{13}\text{C}/^{43}\text{Ca}$ TRAPDOR experiments

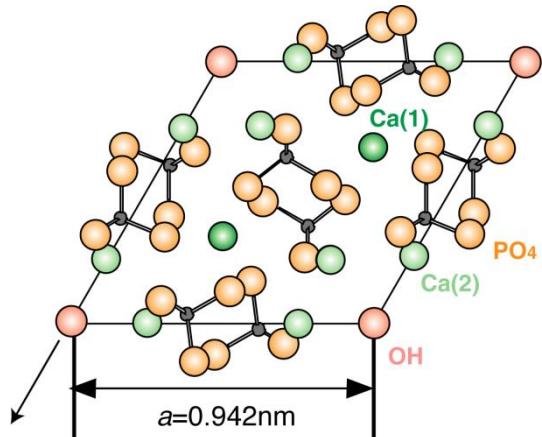
TRANSfer of Population in DOuble Resonance



Silicate substituted HAp



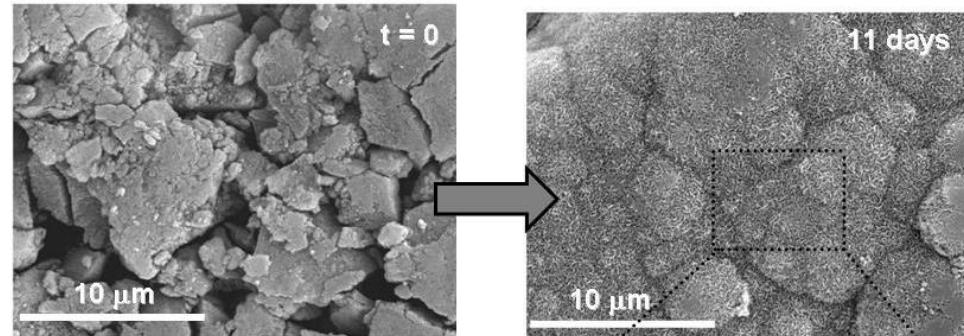
Substituted HAp structures



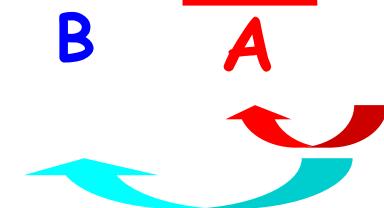
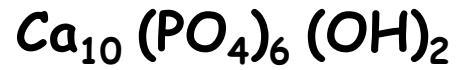
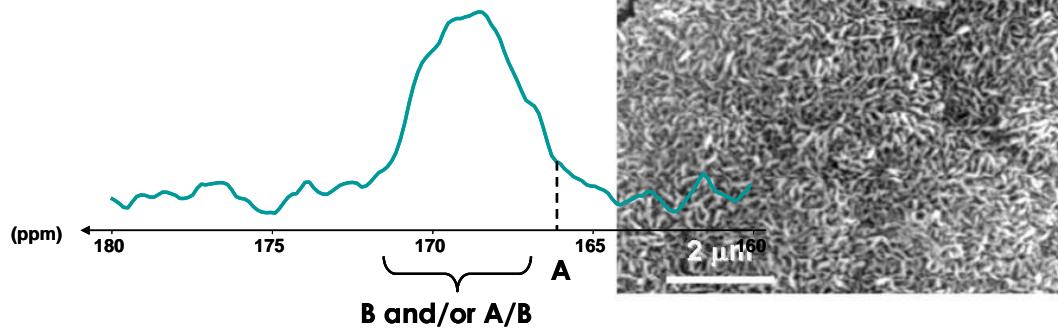
$\text{Mg}^{2+}, \text{Zn}^{2+}, \text{Na}^+, \text{K}^+ \dots$

$\text{SO}_4^{2-}, \text{CO}_3^{2-} \dots$

$\text{CO}_3^{2-}, \text{F}^-, \text{Cl}^- \dots$



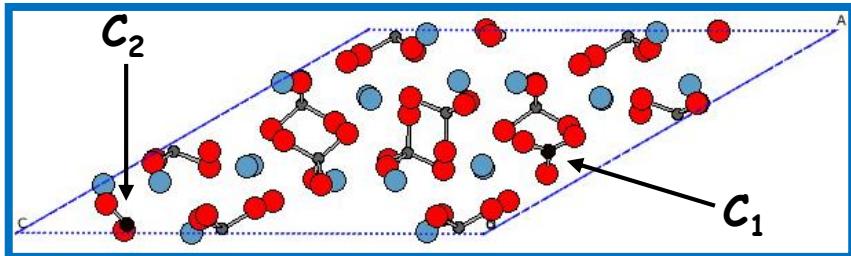
^{13}C enriched SBF



SiO_4^{4-}

CO_3^{2-}

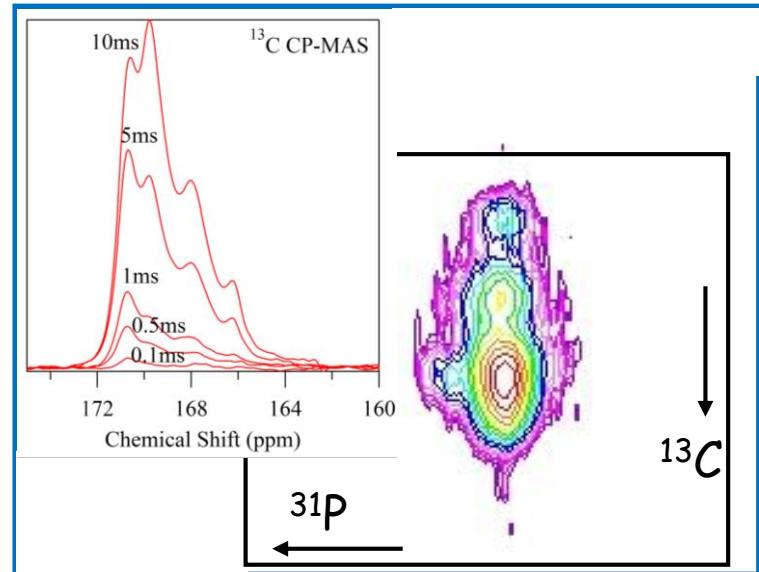
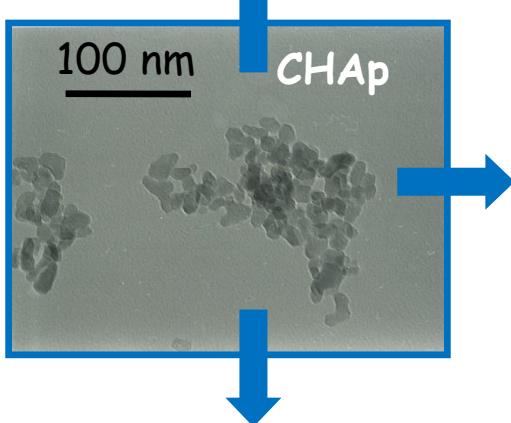
Models, 2D NMR, *ab initio* calculations: a combined approach



Astala et al., Chem. Mater. 2005

Peroos et al., Biomat. 2006

■ Modelling



■ 1D, 2D NMR experiments

	δ (ppm)		δ (ppm)		δ (ppm)
P1	2.1		P7	1.9	
P2	0.1		P8	2.1	
P3	2.1		P9	1.8	
P4	3.3		P10	4.0	
P5	1.1		P11	3.3	
P6	1.5				

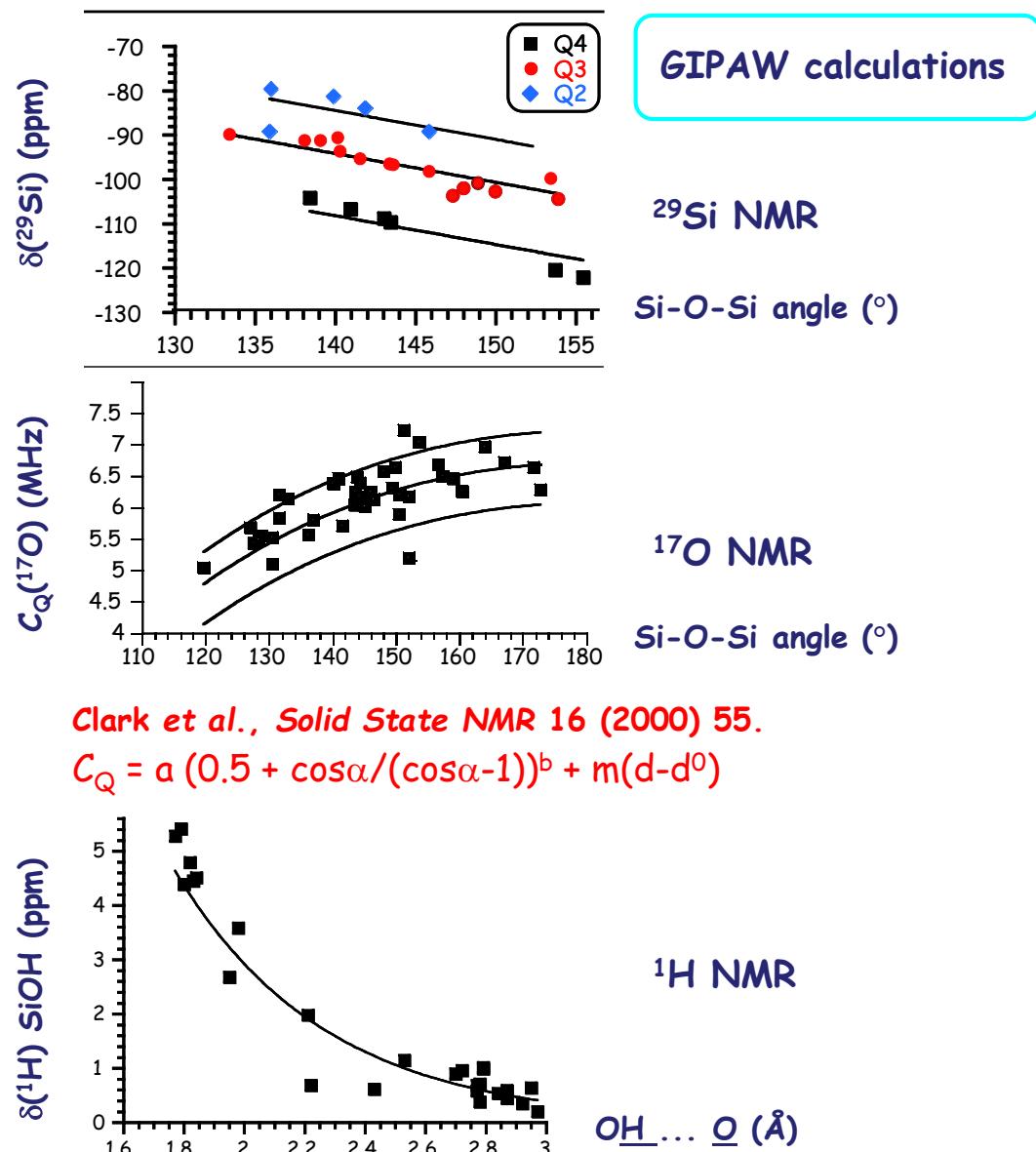
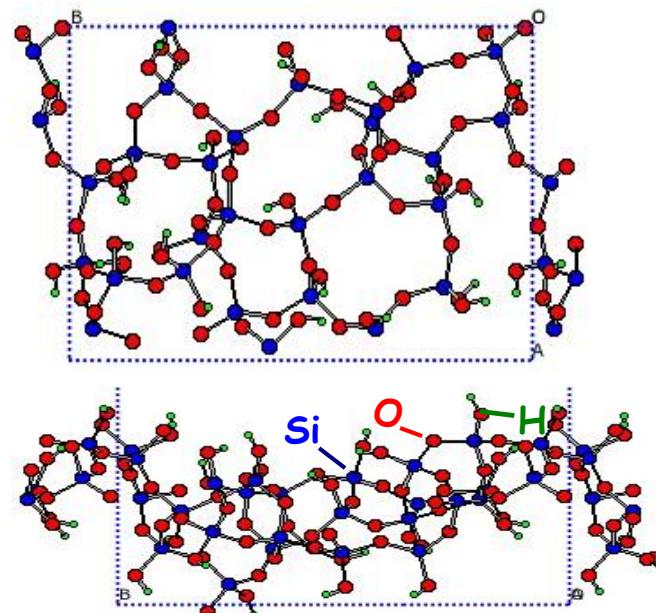
The last three columns of the table are highlighted with red borders: C1 (166.7 ppm), C2 (165.7 ppm), H1 (1.1 ppm), H2 (1.1 ppm), and H3 (-0.7 ppm).

distribution of A-, B-
and A/B sites...

■ first principles calculations

Hydroxylated silica surface: towards interfaces

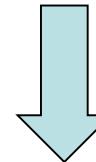
b-initio
VASP
Vienna package simulation



Solid state NMR basics

Applications

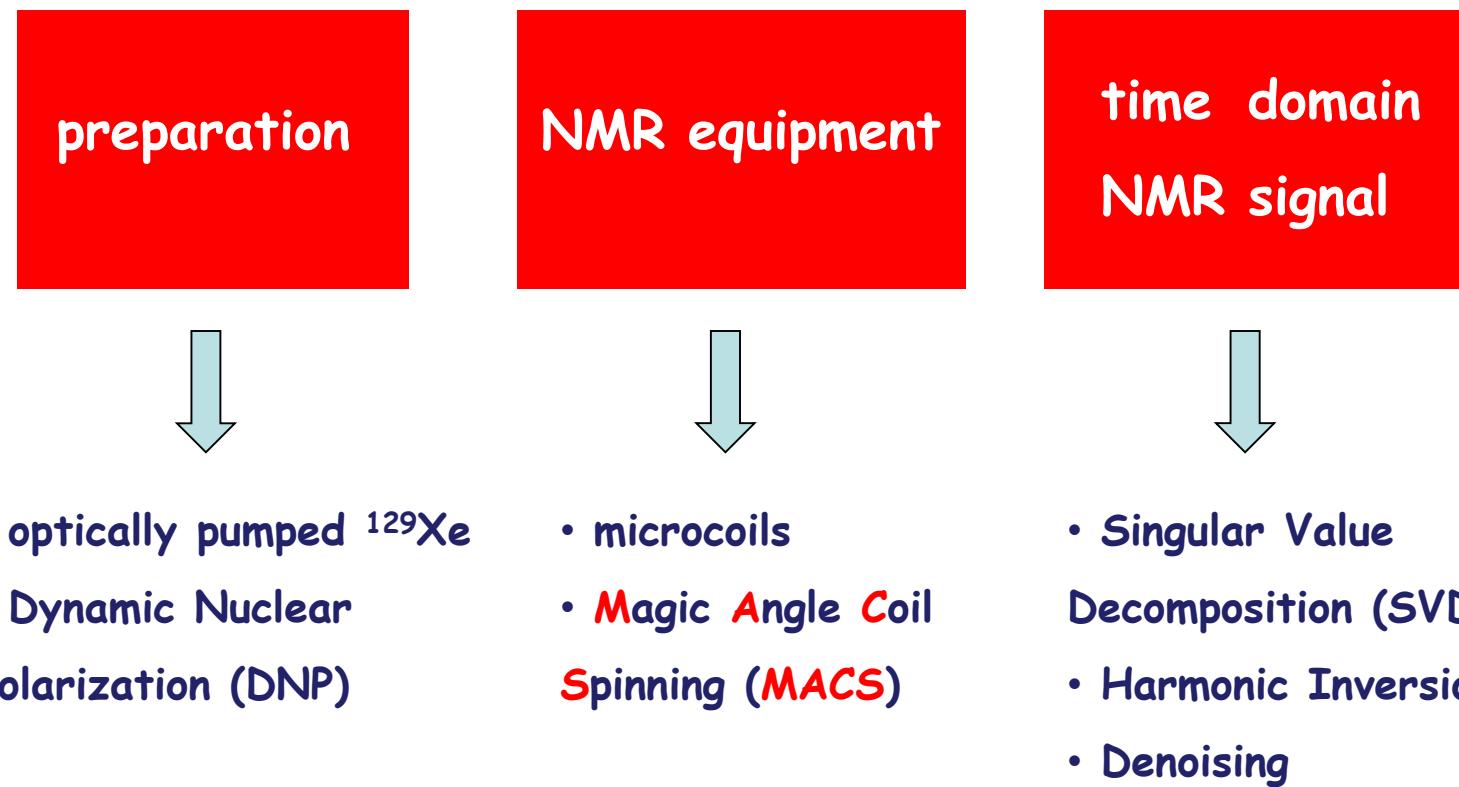
Sensitivity



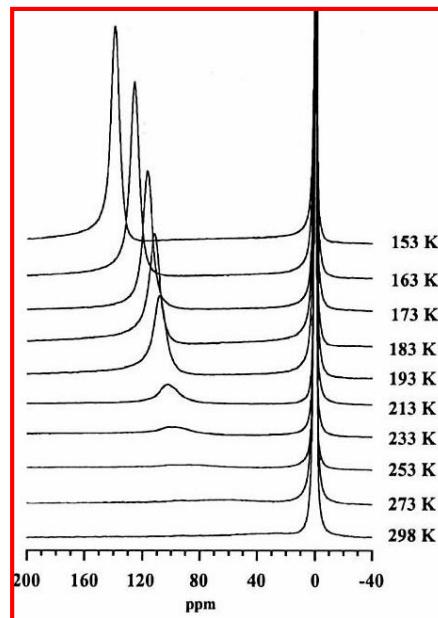
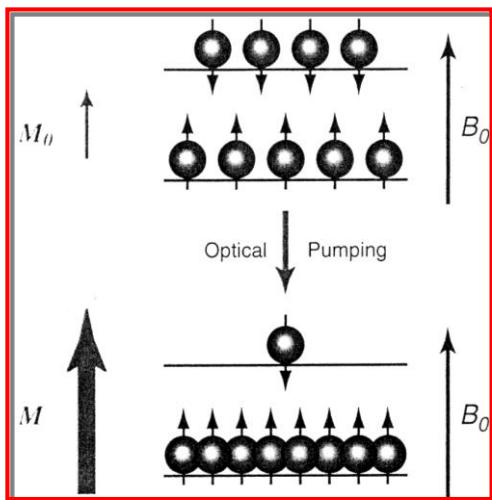
- the key question
- the future!

Increasing the sensitivity in solid state NMR

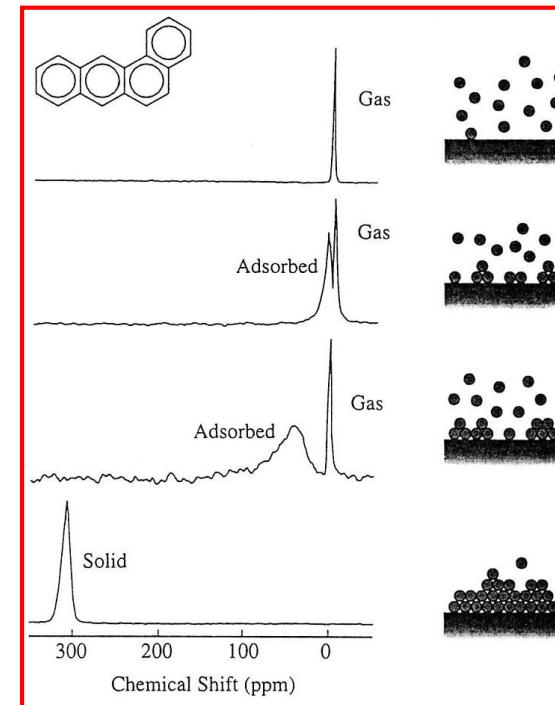
"... the sensitivity of conventional NMR techniques is fundamentally limited by the ordinarily low spin polarization achievable in even the strongest NMR magnets..." in:
B. M. Goodson, *J. Magn. Reson.* 155 (2002) 157.



Surfaces and interfaces seen by hyperpolarized ^{129}Xe



Nossov (2002)



Raftery, Pines (1991)

A simple formalism for the analysis of NMR in the presence of exchange

By M. GOLDMAN
CEA/DSM/DRECAM/Service de Physique de l'Etat Condensé, CE Saclay,
F-91191 Gif-sur-Yvette Cedex, France

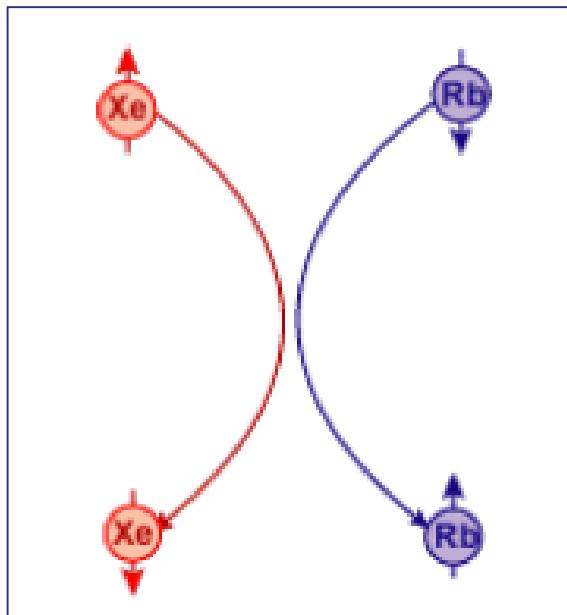
(Received 15 May 1995; accepted 7 June 1995)

A theoretical model is presented for the analysis of NMR in the liquid state, when the NMR parameters are modulated by chemical exchange. This model, based on the sudden jump approximation, describes directly the free induction decay signals, without analysing first their derivative. This formalism, which is extremely simple, applies in particular to the modulation of indirect interactions by intermolecular exchange, a case whose usual description is rather complicated.

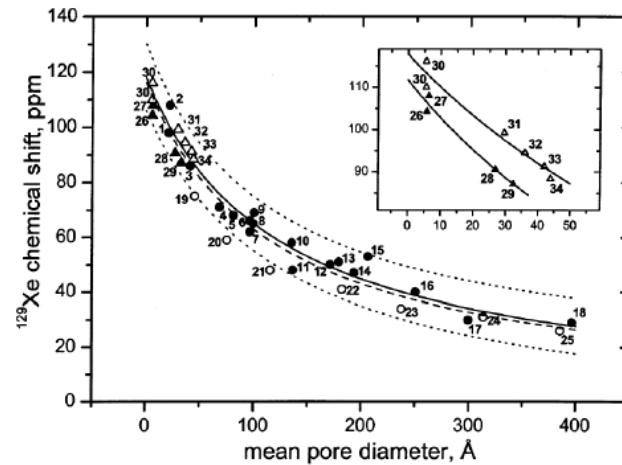
Goldman (1995)

Probing surfaces with ^{129}Xe NMR

- xenon: inert gas
- possible isotope: ^{129}Xe (spin $1/2$, 26.4%)
- possibility to enhance its magnetization through optical pumping up to 25 000 times!!



$\delta(^{129}\text{Xe})$ in various mesoporous silica



- Gels de silice ; ○ Vycor/CPG ; ▲ Organo-silicates

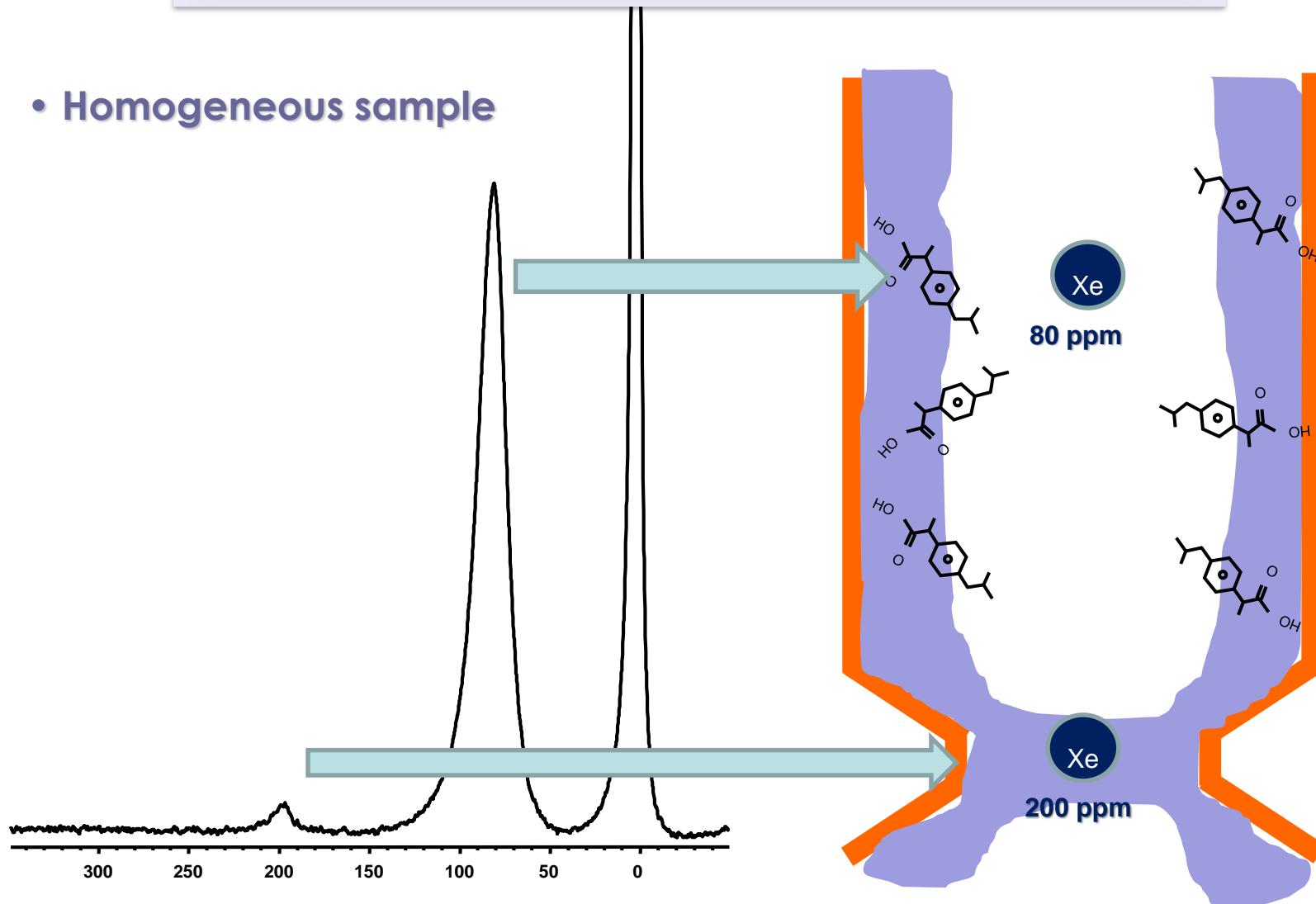
Terskikh et al., Langmuir 18 (2002) 5653

- $\delta(^{129}\text{Xe})$ related to \emptyset

courtesy of Dr. T. Azaïs, LCMCP

Ibuprofen loaded MCM-41 100 \AA

- Homogeneous sample

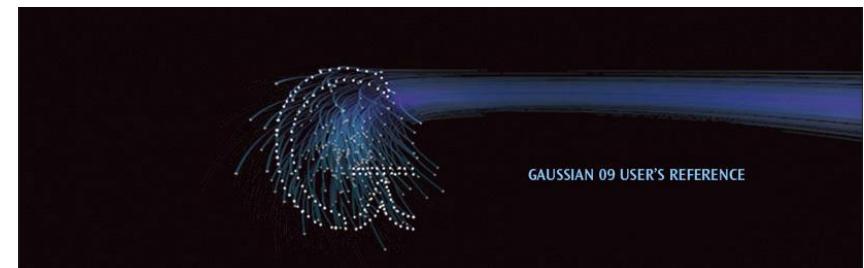
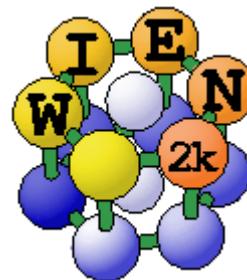


courtesy of Dr. T. Azaïs, LCMCP

To be done ...

- ◆ prediction/calculation of ^{129}Xe NMR shifts
- ◆ from a fundamental QM point of view: "dispersion forces" and DFT ?
- ◆ Hartree-Fock and post HF methods + GAUSSIAN -> chemical shifts
- ◆ a periodic approach by VASP ? GIPAW calculations ?
- ◆ what about the Xe (and other noble gases ...) pseudopotential ?
- ◆ another approach: ^{131}Xe ($I = 3/2$) -> calculation of EFG by Wien2k

- ◆ DYNAMICS ?

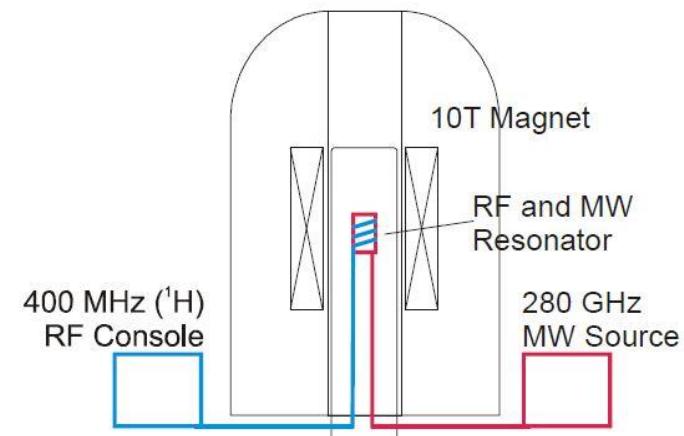


Dynamic Nuclear Polarization (DNP)

State-of-the-art

Dynamic nuclear polarization is based on the idea of transferring the at least 600 times higher electron spin polarization onto nuclear spins. This concept originally proposed by Overhauser in 1953 [Overhauser] was at first experimentally proven in metals by Slichter [Slichter] and later on also observed in liquids [Hauser, Müller-Warmuth]. Analyzing DNP experiments performed at low temperatures on solids with localized electron spins in order to obtain highly polarized targets revealed that in such solid-state systems other polarization transfer mechanism are effective: the so-called solid-state effect [Abragam], cross-effect [Hill] and thermal mixing [Goldman]. They refer to the dipolar coupling of the nuclear spin to one, two or more electron spins, respectively. All the investigated mechanisms predicted reduced transfer efficiencies at higher magnetic field values. This in combination with the lacking microwave technology to effectively excite e' field values above 1 T turned DNP into an 'endangered species' while N towards higher spectral resolution and higher magnetic fields [Wind, Yannon 20 years until breakthroughs of DNP at high magnetic fields were achieved SS-MAS-DNP for structural biology applications [Griffin] and later on the A the possibility to employ such polarized samples at very low temperatures for

High Frequency DNP Spectrometer



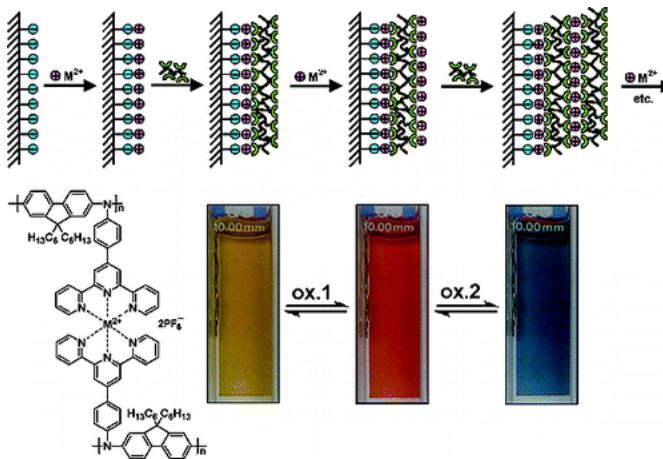
see for instance:

<http://www.postgenomicnmr.net/NMRLife/docs/DynamicNuclearPolarization.pdf>

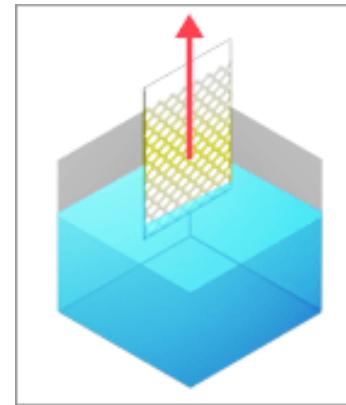
a revival:

- combination with MAS at low T -> Griffin's group (at MIT)
- applications to biosolids, inorganic materials, surfaces and grafted species ...

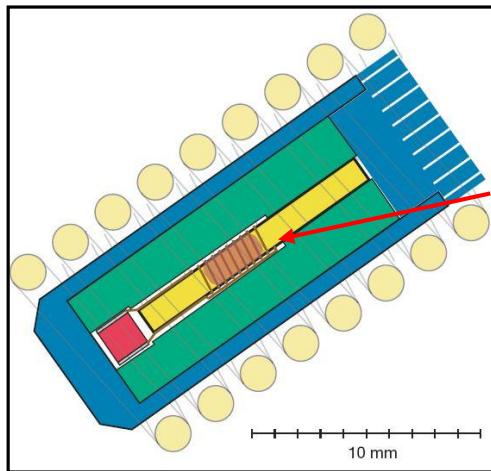
Towards μ g experiments: MACS NMR



Maier et al., Chem.
Mater. 2009



Magic Angle Coil Spinning



rotor at θ_m
 μ -coil
static coil

cea

D. Sakellariou et al., Nature, 447, 2007.

P. Aguiar et al., J. Magn. Reson., 200, 2009

potential applications:

■ films



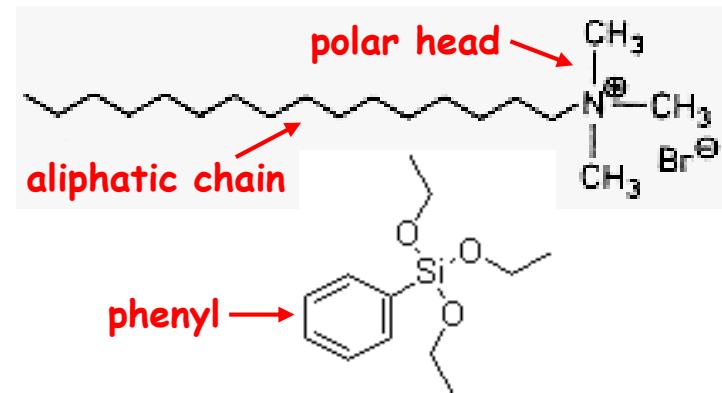
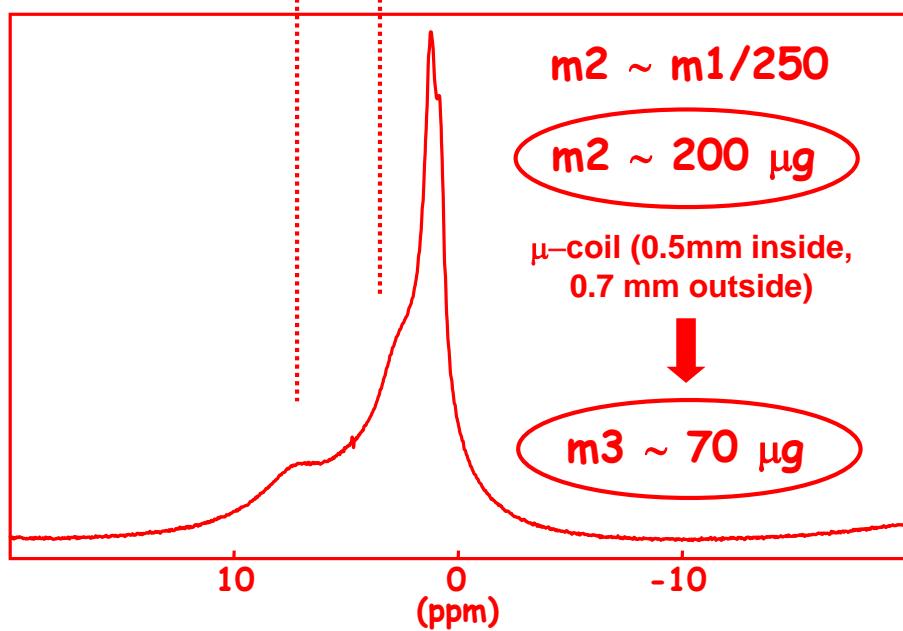
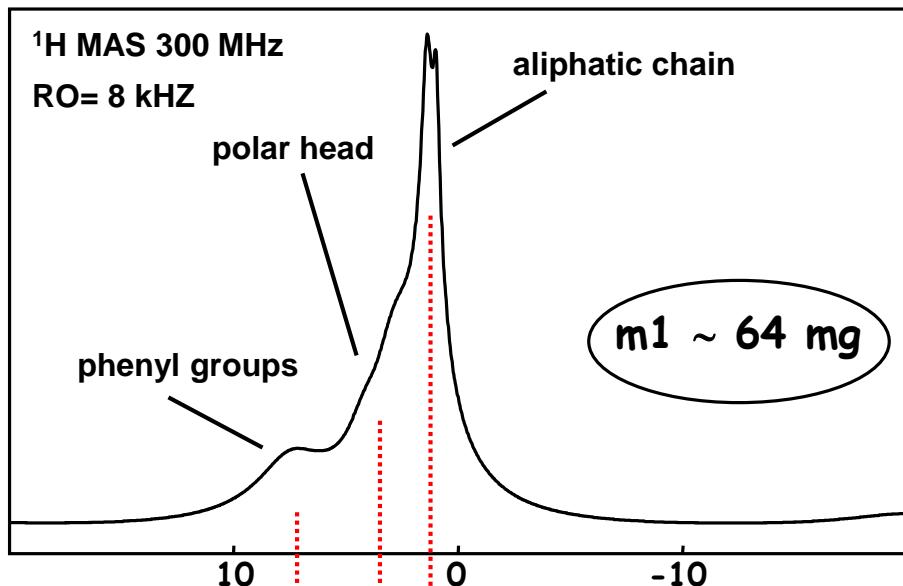
$S \sim 2 \text{ cm}^2$
 $\text{Th} \sim 300 \text{ nm}$

$m \sim 100 \mu\text{g}$

^1H (but also ^{29}Si , ^{13}C ...)

¹H MACS: mesoporous powder (CTAB / TEOS / PhSi(OEt)₃)

with P. Aguiar, D. Sakellariou - CEA - Saclay, France

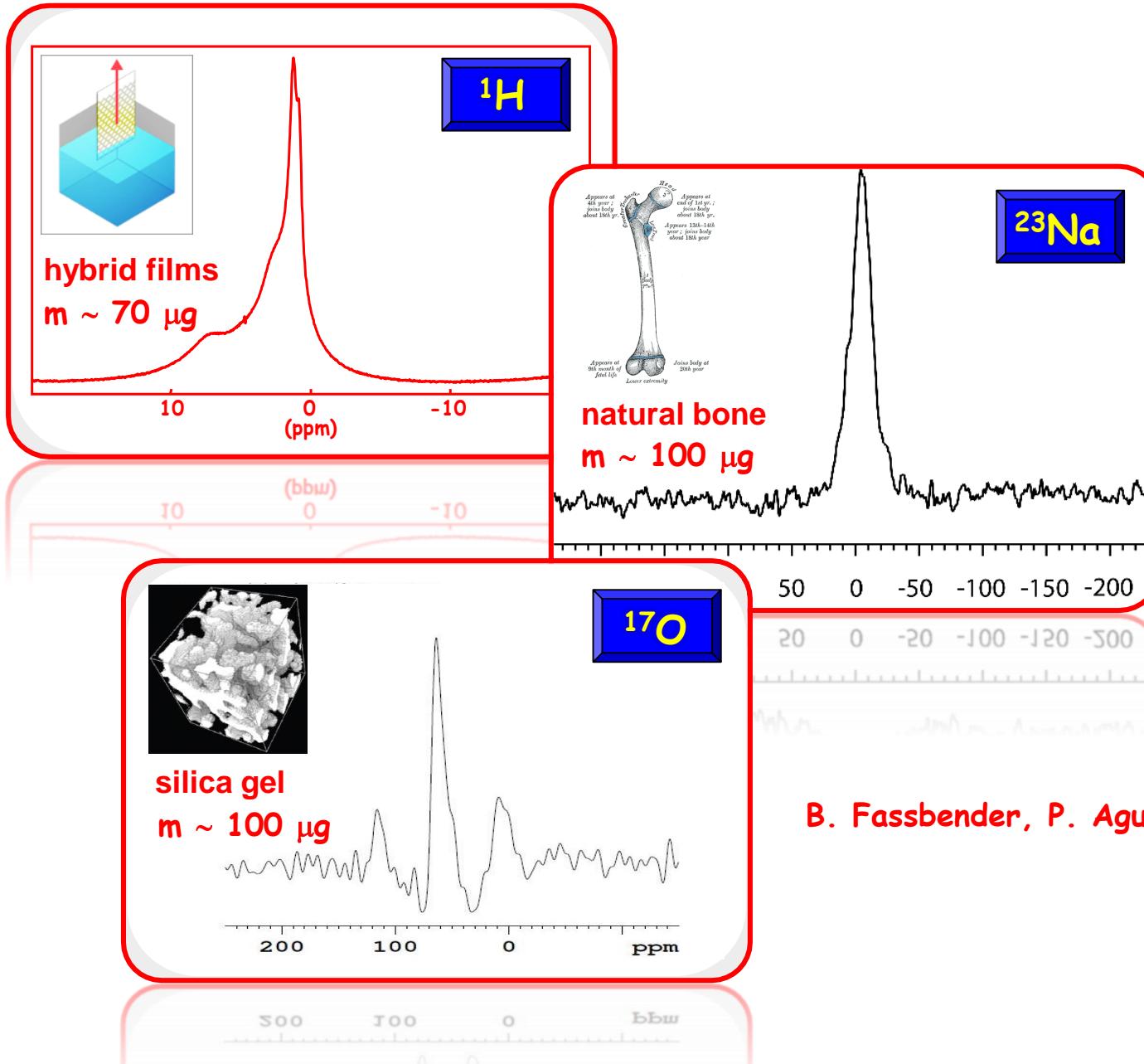


3.2mm (24 kHz)



■ expensive isotopes: ²⁹Si, ⁴³Ca, ²⁵Mg ...

Magic Angle Coil Spinning

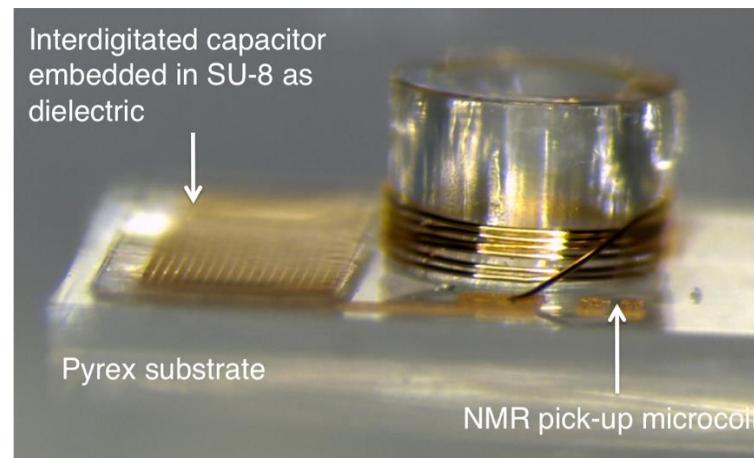


B. Fassbender, P. Aguiar, UPMC-CEA

MEMS approach

machine made micro-coils with reproducible enhancement factor (collaboration with Freiburg*, IMTEK, V. Badilita and CEA, D. Sakellariou)

MEMS compatible techniques



J. Micromech. Microeng. **20** (2010) 015021

K Kratt *et al*

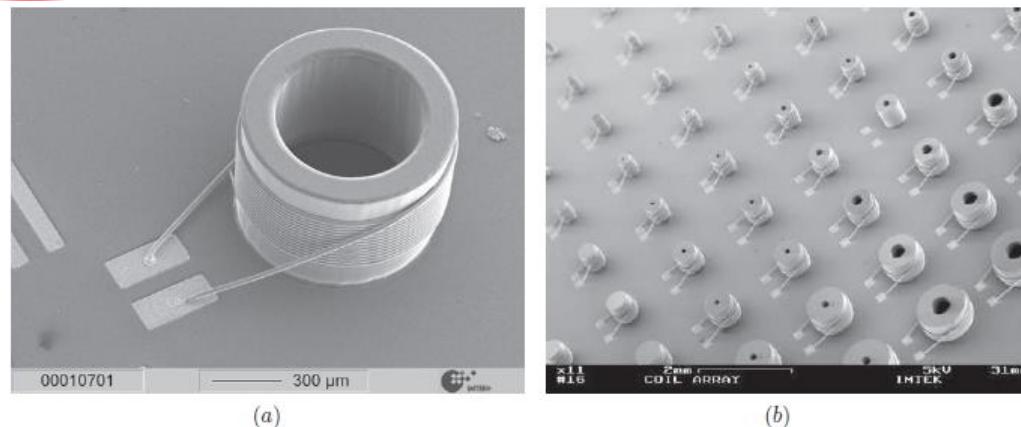
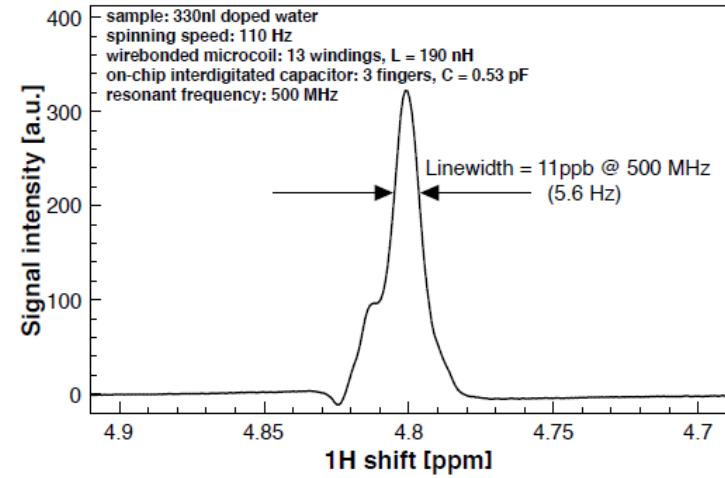
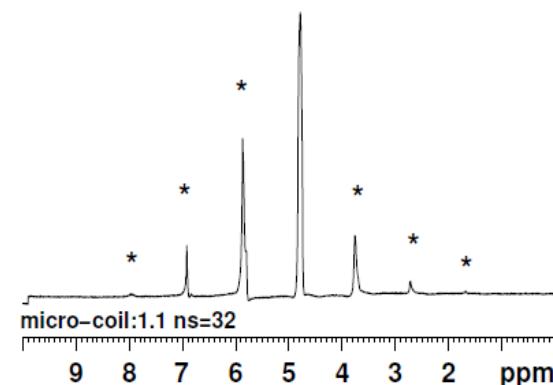
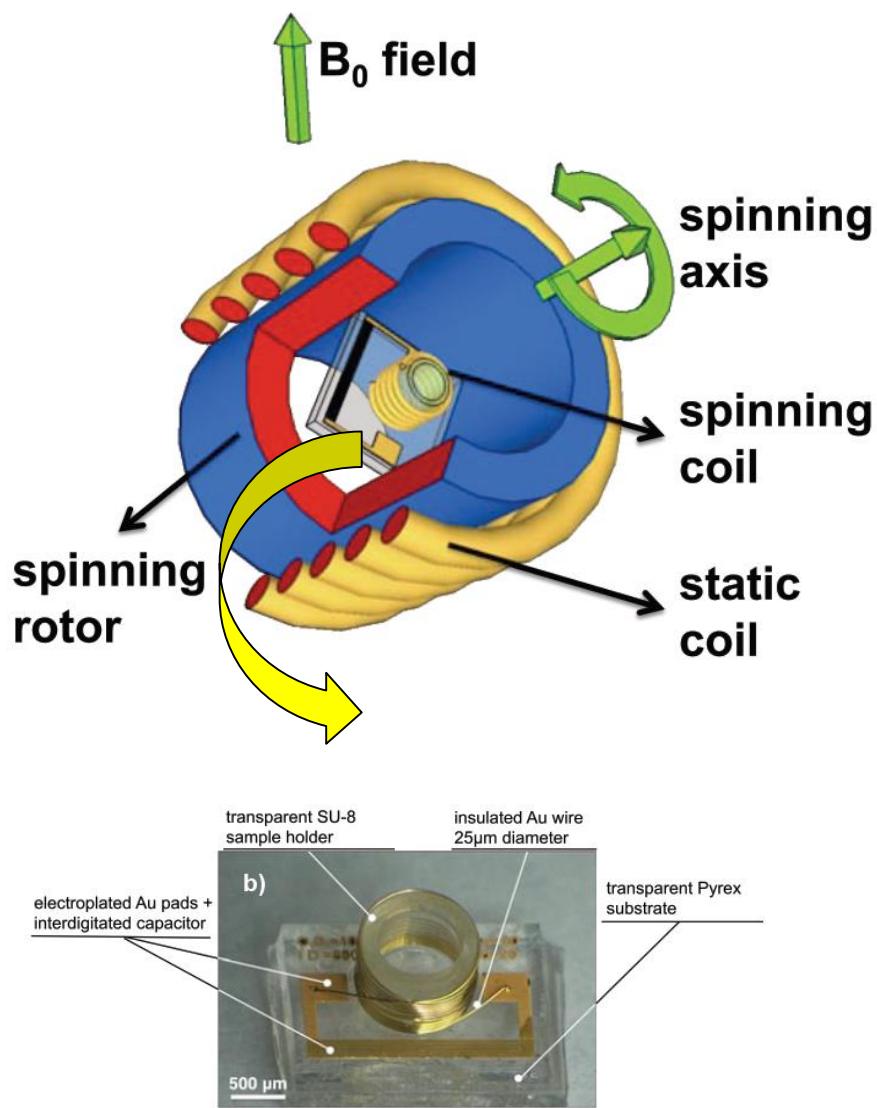


Figure 4. Micro coils made with a wire bonder on a silicon substrate around SU-8 posts. (a) Single coil and (b) array of coils.

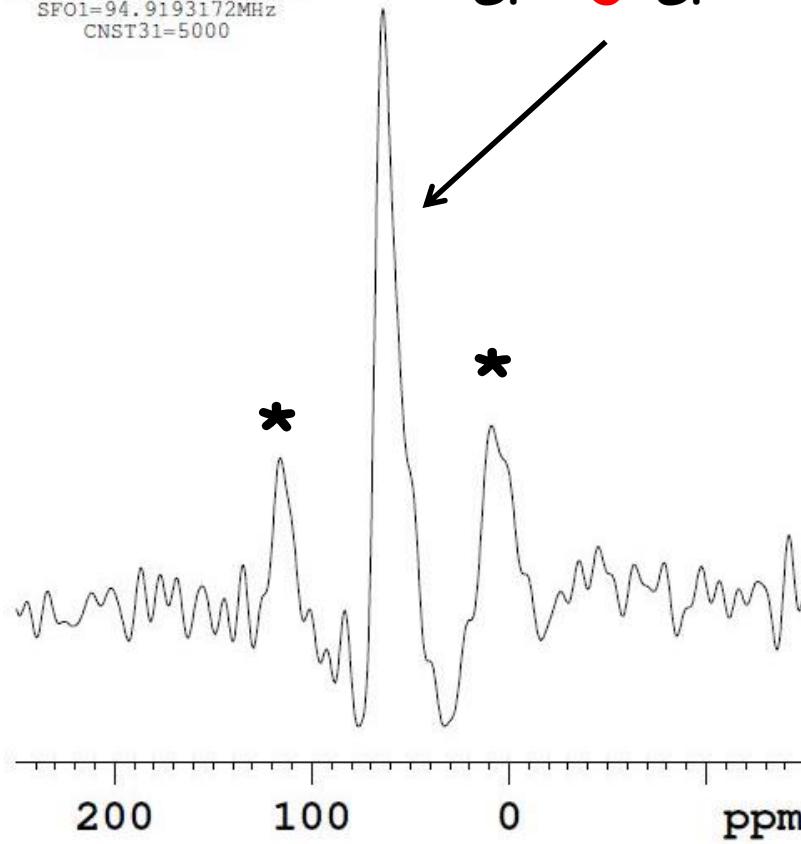
First results at 700 MHz (^1H)



The first ^{17}O MACS experiments - silicate hybrid gels

Micro-coil
16.3 T
5 kHz MAS
 $< 0.200 \text{ mg}$
40 min.

hahnecho-coil 1
SR=-10306.59Hz (0ppm H₂O)
O1= 3894.05Hz (150ppm)
pL1(2.3us)=15dB
d1= 300ms; dE= 30us; dw=5us
td=1k ns=8k rg=1440
SFO1=94.9193172MHz
CNST31=5000

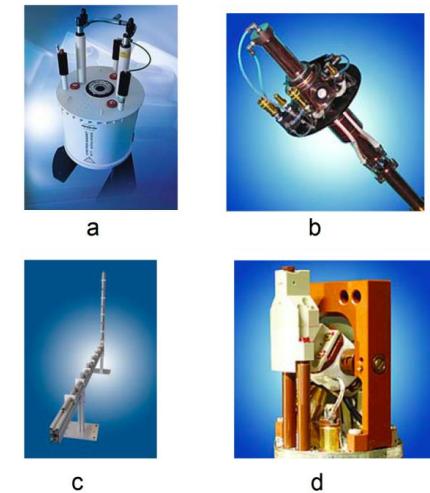


Birgit Fassbender
(LCMCP + CEA,
CNano'IDF grant)

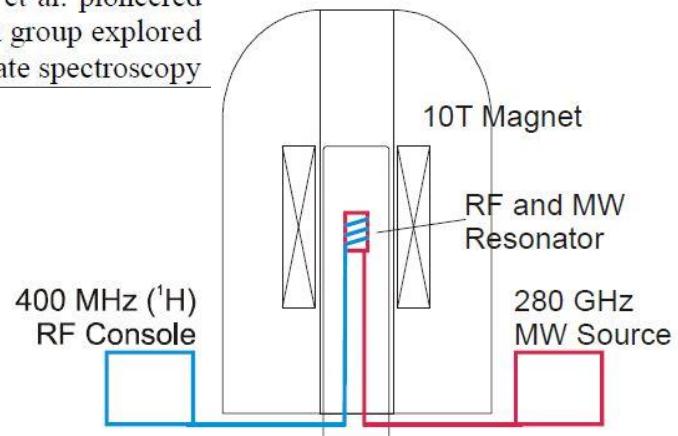
Dynamic Nuclear Polarization (DNP)

State-of-the-art

Dynamic nuclear polarization is based on the idea of transferring the at least 600 times higher electron spin polarization onto nuclear spins. This concept originally proposed by Overhauser in 1953 [Overhauser] was at first experimentally proven in metals by Slichter [Slichter] and later on also observed in liquids [Hauser, Müller-Warmuth]. Analyzing DNP experiments performed at low temperatures on solids with localized electron spins in order to obtain highly polarized targets revealed that in such solid-state systems other polarization transfer mechanism are effective: the so-called solid-state effect [Abragam], cross-effect [Hill] and thermal mixing [Goldman]. They refer to the dipolar coupling of the nuclear spin to one, two or more electron spins, respectively. All the investigated mechanisms predicted reduced transfer efficiencies at higher magnetic field values. This in combination with the lacking microwave technology to effectively excite electron spins at magnetic field values above 1 T turned DNP into an ‘endangered species’ while NMR was briskly moving towards higher spectral resolution and higher magnetic fields [Wind, Yannoni, Schäfer]. It took almost 20 years until breakthroughs of DNP at high magnetic fields were achieved: Griffin et al. pioneered SS-MAS-DNP for structural biology applications [Griffin] and later on the Amersham group explored the possibility to employ such polarized samples at very low temperatures for liquid state spectroscopy



<http://www.postgenomicnmr.net/NMRLife/docs/DynamicNuclearPolarization.pdf>

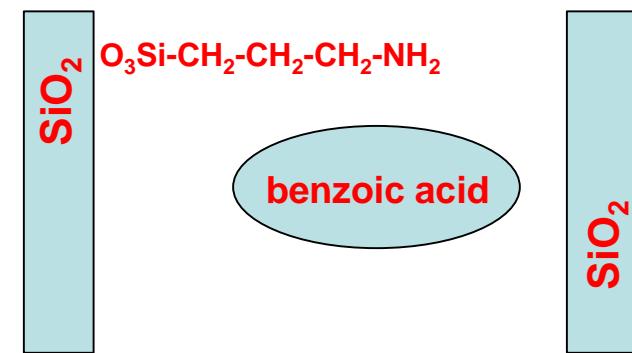
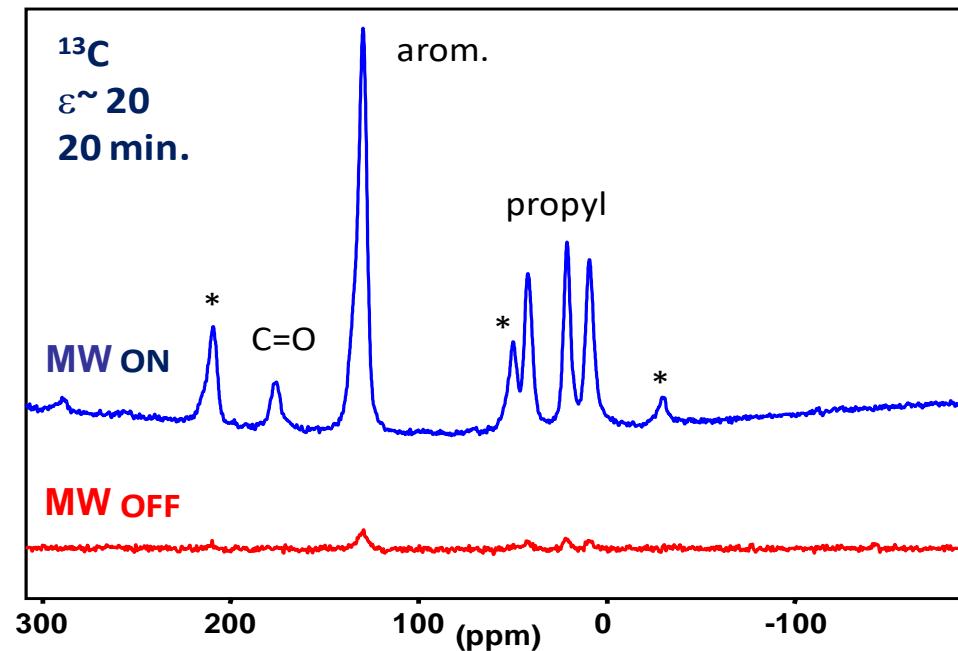
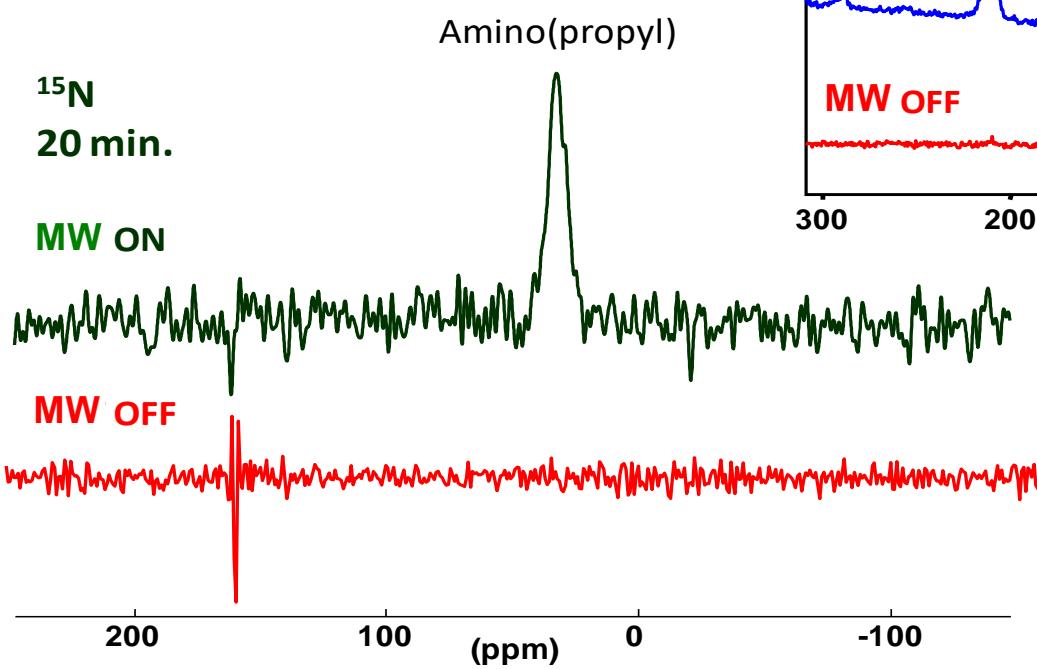
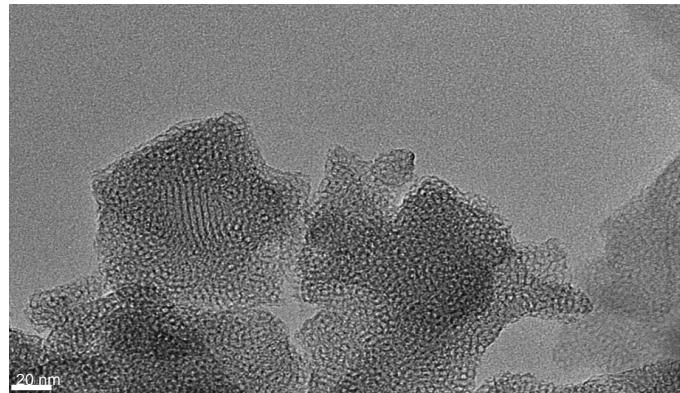


renewal:

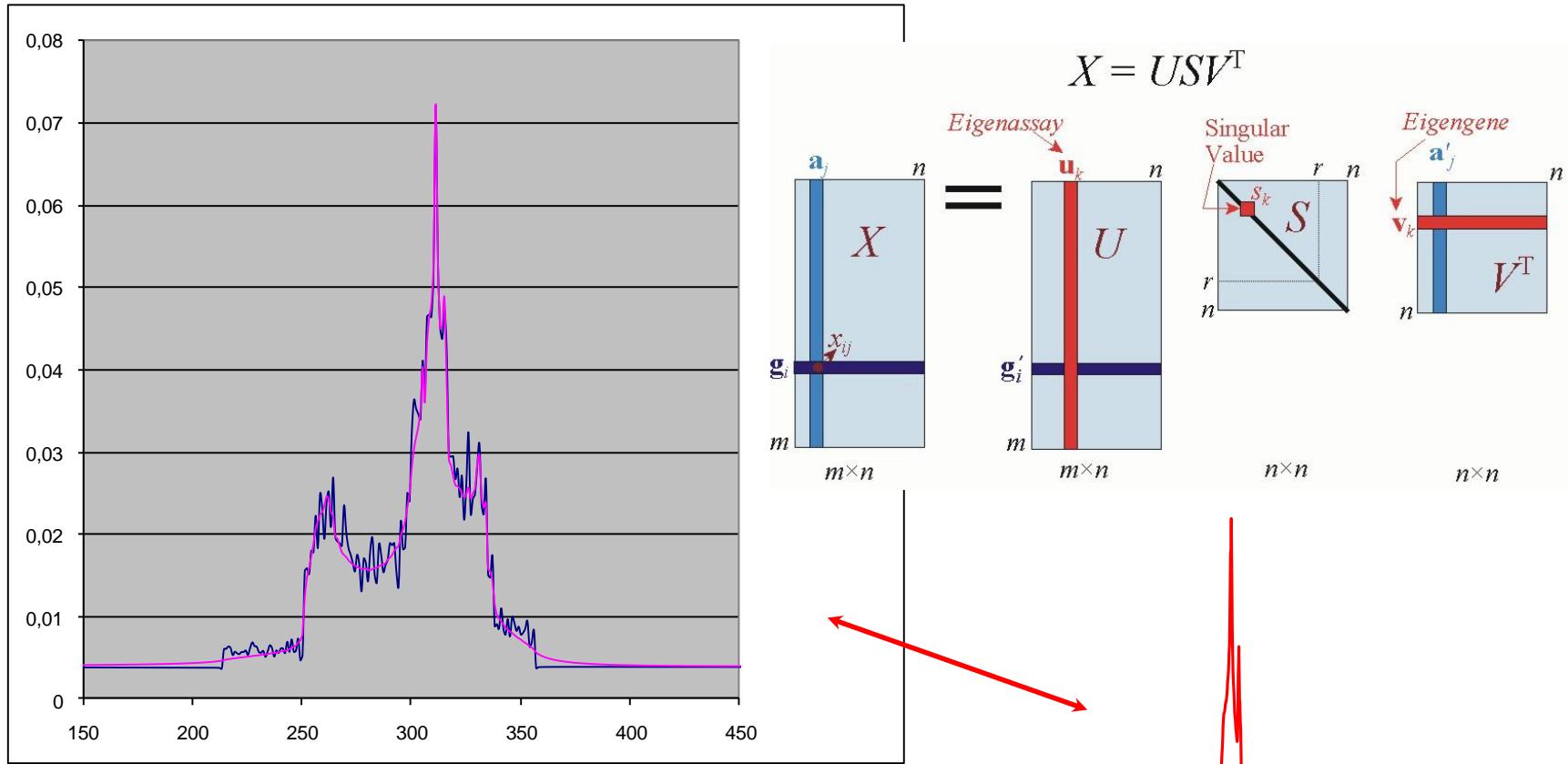
- high field, low T (90 K) MAS (10 kHz)
- applications : biosolids (B. Griffin, MIT), materials (Emsley, Bodenhausen)

Application to functionnalized MCM-41 structures

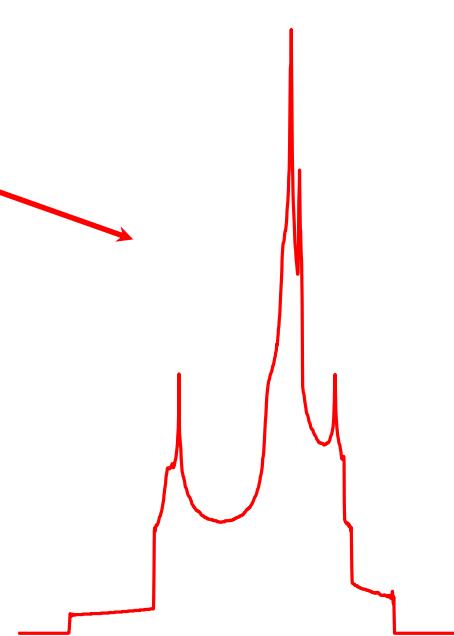
Coll.: M. Caporini, G. Bodenhausen, EPFL, Lausanne



Denoising of the FID: post processing of the data



- FID as a complex Hankel matrix
 - Singular Value Decomposition
 - Cadzow denoising
- Coll.: Pascal Man (UPMC)



Résonance Magnétique Nucléaire multidimensionnelle et multinucléaire en solution

Certaines illustrations sont extraites de :

Hore, Nuclear Magnetic Resonance, 1998.

Levitt, Spin dynamics, 2002.

Braun, 150 and more basic NMR experiments, 1998.

Derome, Modern NMR Techniques for Chemistry Research, 1991.

Christian BONHOMME, Professeur

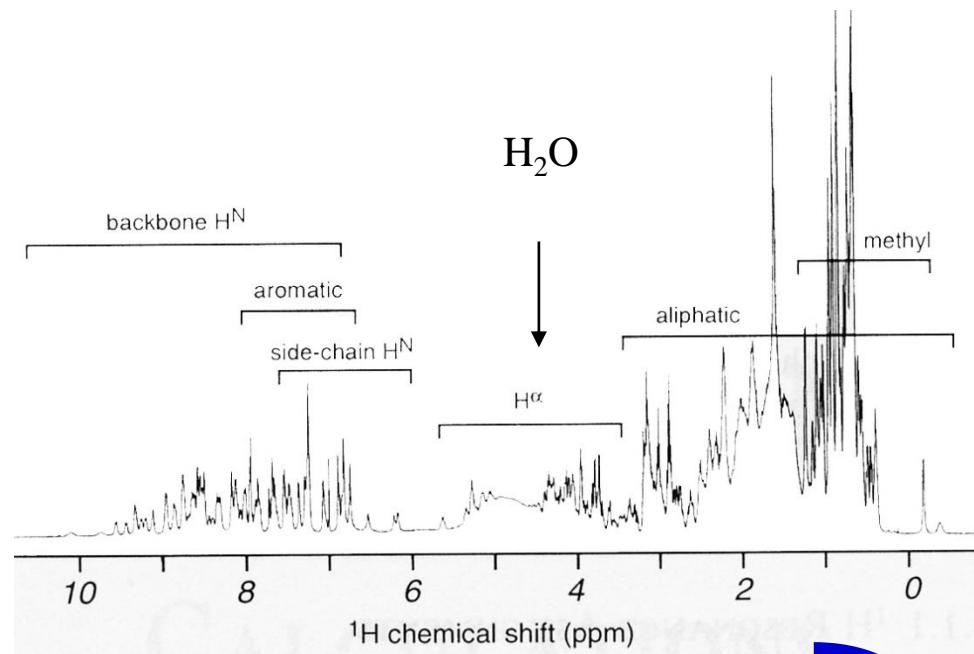
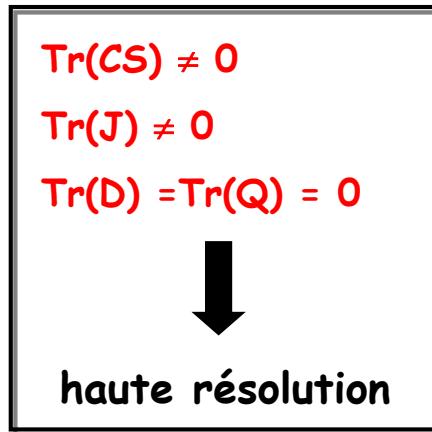
Laboratoire de Chimie de la Matière Condensée

UMR CNRS 7574 - Sorbonne Université, Paris

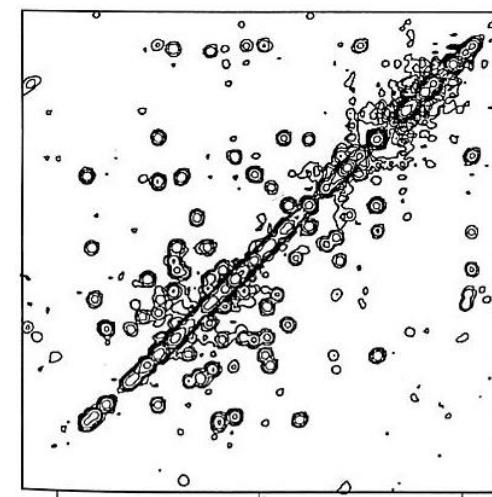
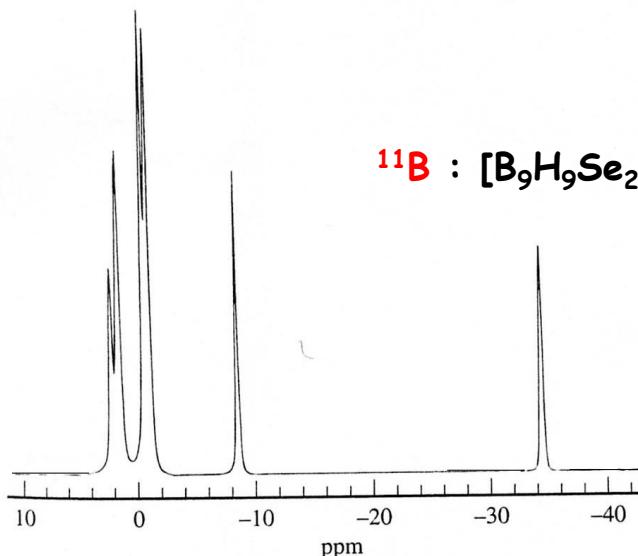
Rappels

toutes les interactions sont
moyennées à leur valeur isotrope...

ex : RMN ^1H en solution ♦ peptides, protéines



ex : RMN des noyaux quadripolaires



2D, 3D...

Déplacement chimique, couplage J - Relaxation

1D

multiplets

transferts
INEPT, DEPT

2D

J-resolved

effet Overhauser
nucléaire

NOE-diff.

CORrelation
SpectroscopY
COSY

HETeronuclear
CORrelation
HETCOR

NOESY

TOTAL Correlation
SpectroscopY
TOCSY

relaxation
couplage dipolaire

.....

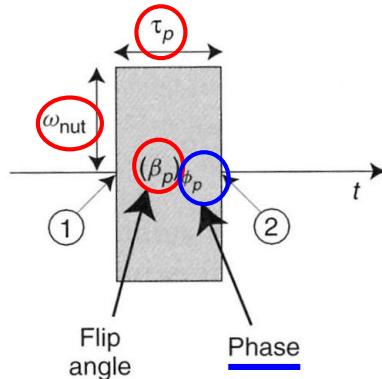
couplage J
"chemins de liaisons"



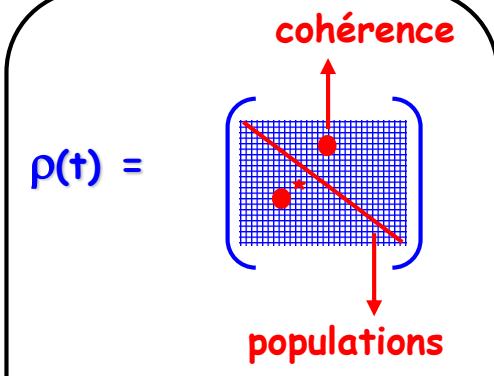
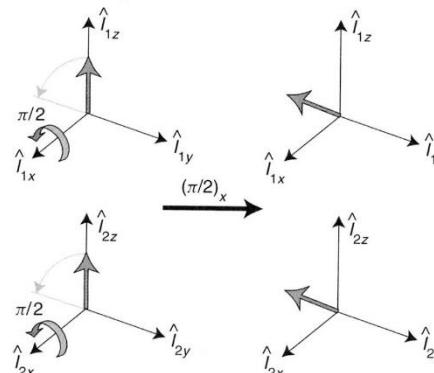
3D, 4D...

Approche quantique

"action des impulsions (\hat{I}_x, \hat{I}_y) sur les opérateurs de spin" \Rightarrow ROTATIONS



$$\beta_p(\text{rad.}) = \omega_{\text{nut}} \cdot \tau_p = \gamma B_1 \cdot \tau_p$$

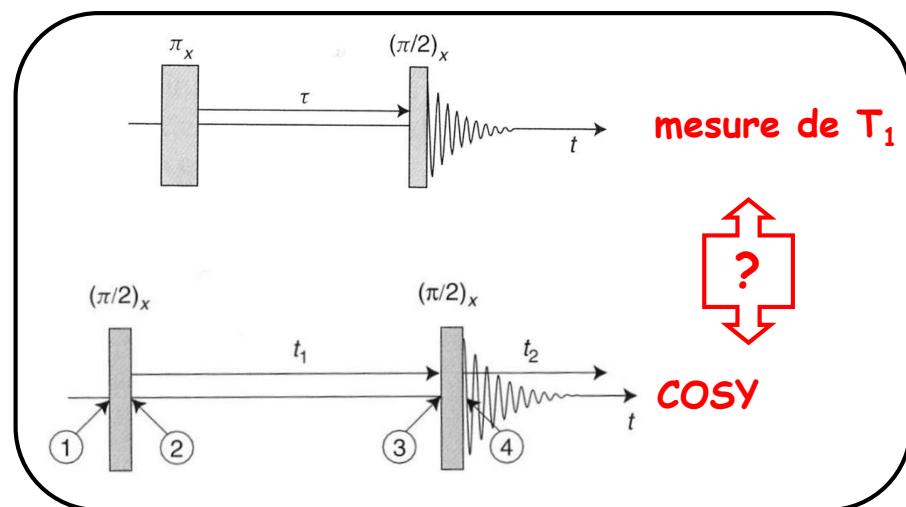


ordre de cohérence :

$\langle \alpha\beta \rangle \leftrightarrow \langle \beta\beta \rangle$: 1Q

$\langle \alpha\alpha \rangle \leftrightarrow \langle \beta\beta \rangle$: 2Q...

$$\frac{\partial}{\partial t} \rho(t) = -\frac{i}{\hbar} [H(t), \rho(t)]$$



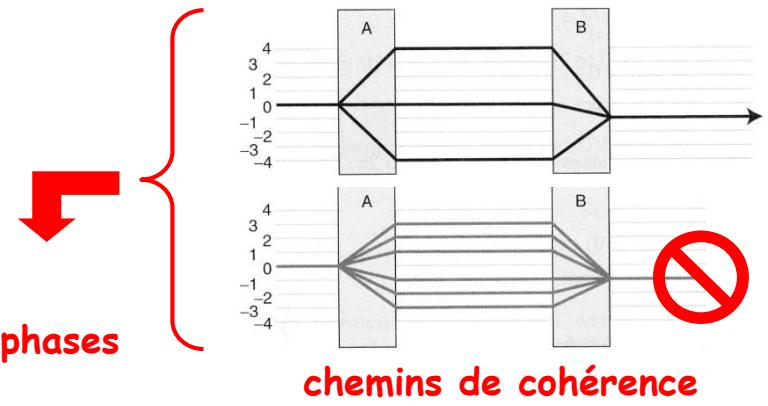
impulsions

récepteur

mesure de T_1



COSY



chemins de cohérence

Systèmes faiblement couplés : opérateurs produits (OP)

ex: système $AX \rightarrow 16$ OP

construction

$$2\hat{I}_{1x}\hat{I}_{2z} = 2 \frac{1}{2} \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix} \otimes \frac{1}{2} \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$$

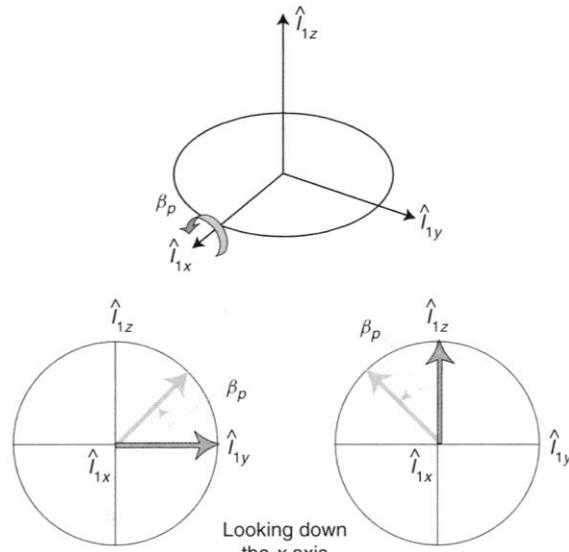
$$= \frac{1}{2} \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix} \otimes \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$$

$$= \frac{1}{2} \begin{pmatrix} 0 & 0 & 1 & 0 \\ 0 & 0 & 0 & -1 \\ 1 & 0 & 0 & 0 \\ 0 & -1 & 0 & 0 \end{pmatrix}$$

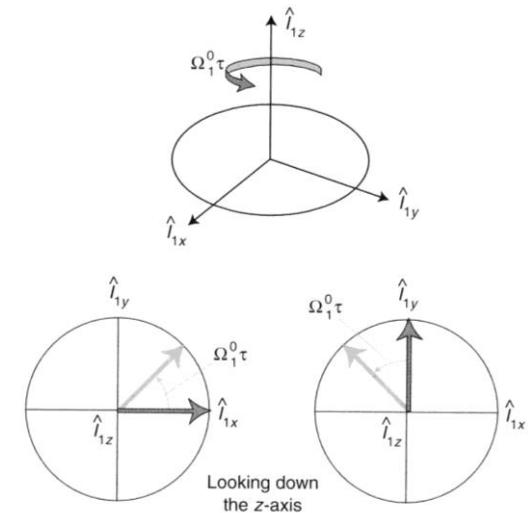
$$= \frac{1}{2} \begin{pmatrix} 0 & 0 & 1 & 0 \\ 0 & 0 & 0 & -1 \\ 1 & 0 & 0 & 0 \\ 0 & -1 & 0 & 0 \end{pmatrix}$$

etc...

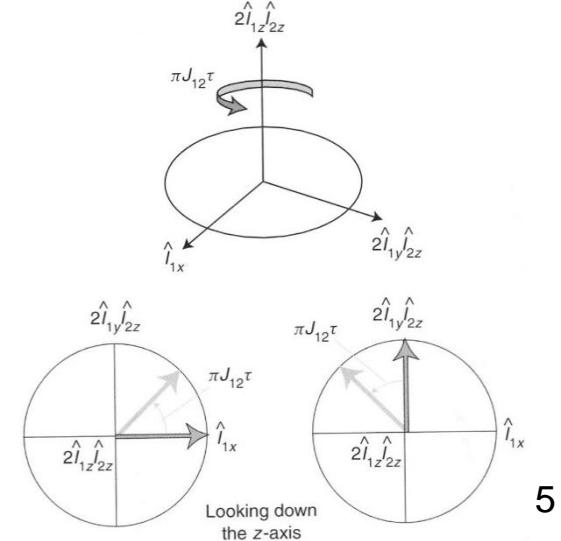
action des impulsions



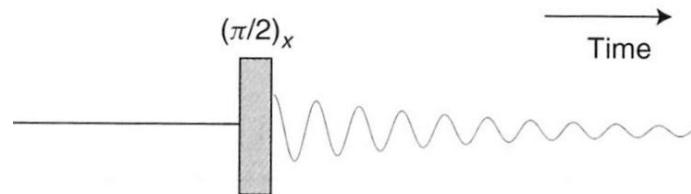
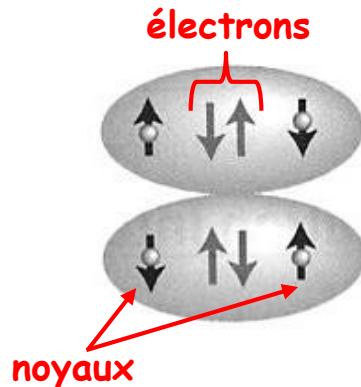
évolution sous δ



évolution sous J



Couplages J homonucléaires - Multiplets au 1^{er} ordre



rappel :

un spin J couplé à n spins I...



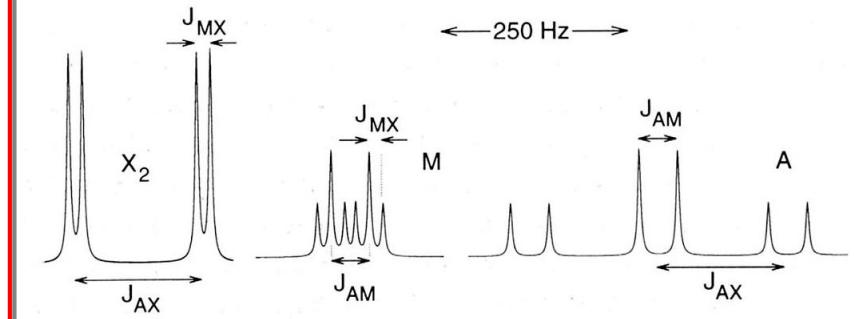
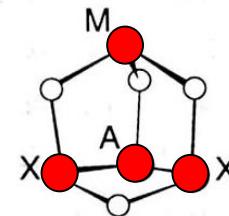
(2nI + 1) raies attendues

cas particulier (!) : $I = 1/2$



(n + 1) raies, binôme de Newton

$$I = 1/2$$

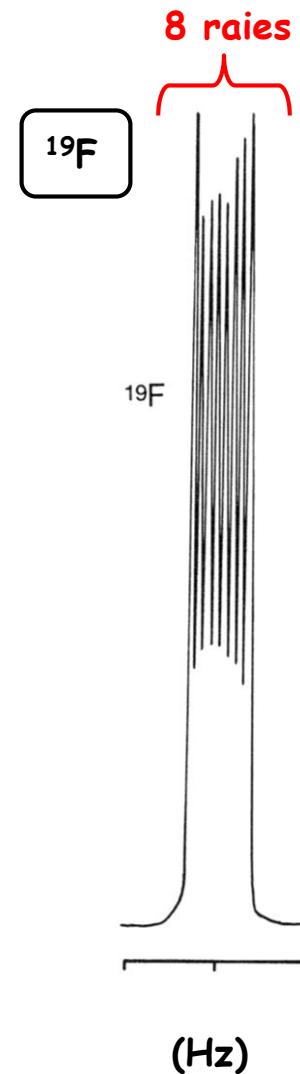
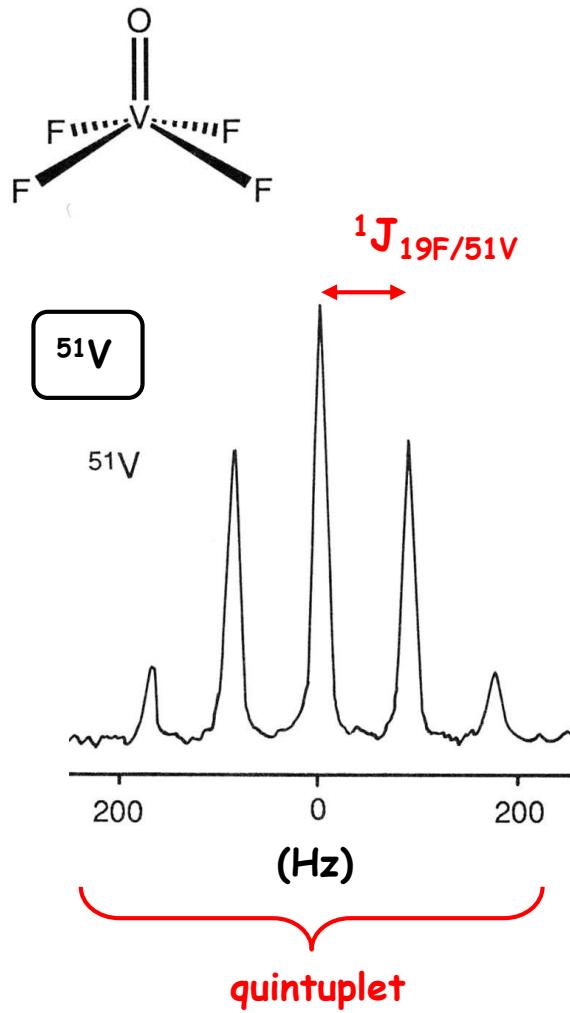


Couplages J : noyaux quadripolaires

-1

^{19}F : $I=1/2$ 100% D^P : 0,83

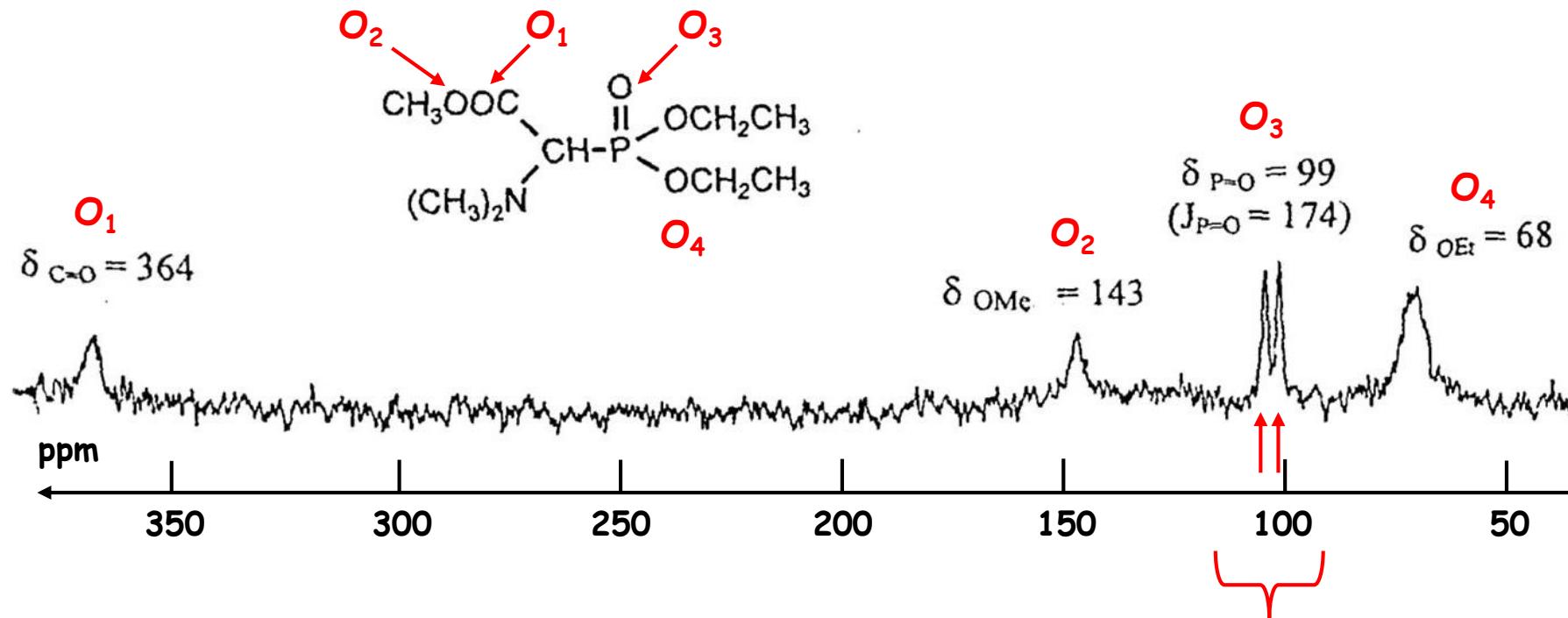
^{51}V : $I=7/2$ 99,8% D^P : 0,38



Couplages J : noyaux quadripolaires

-2

^{17}O : I=5/2 0.037% D^P : 1.08×10^{-5}



relaxation rapide

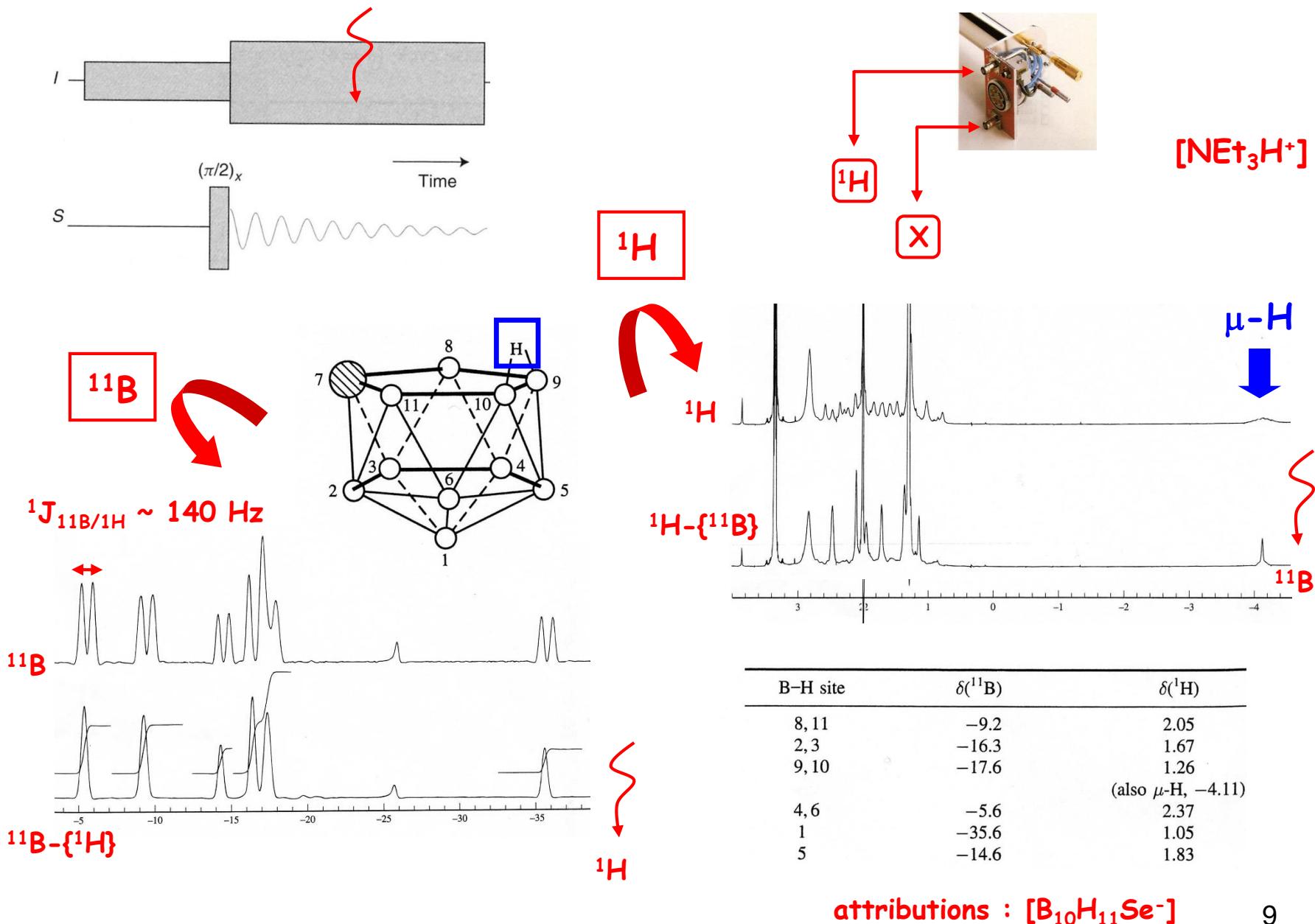
NS élevé

enrichissement isotopique possible
via H₂¹⁷O

doublet

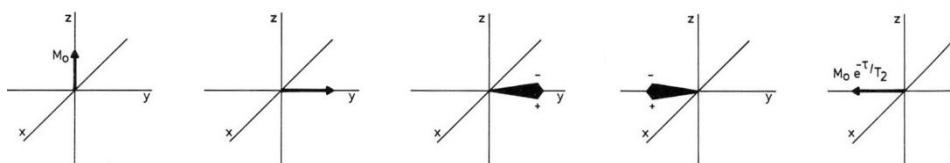
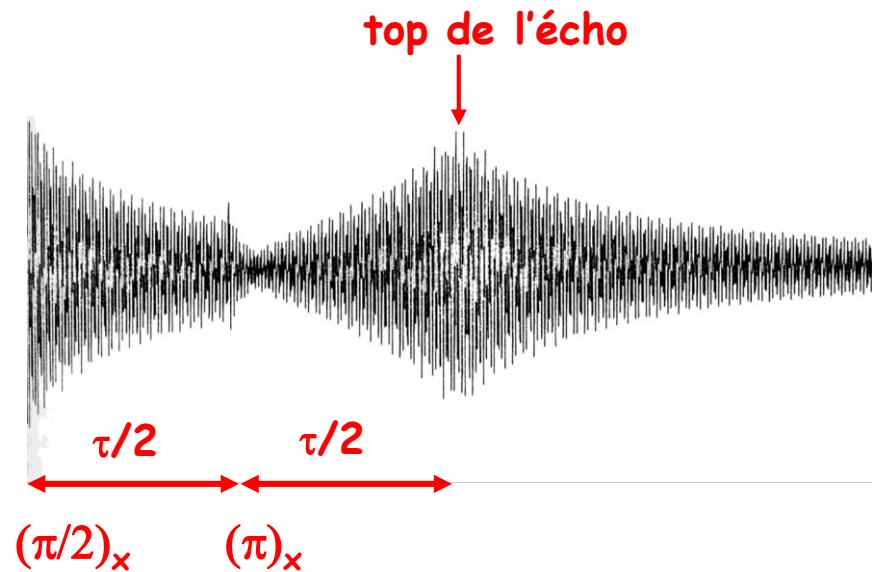
$^1\text{J}_{^{17}\text{O}/^{31}\text{P}} = 174 \text{ Hz}$

Simplification des spectres : découplage hétéronucléaire



Echos de spins

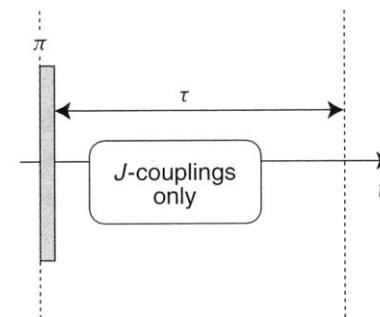
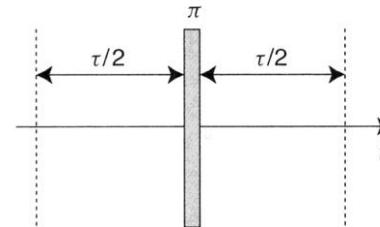
réponse du système de spins à 2 impulsions



refocalisation au temps $\tau/2$

après l'impulsion π

systèmes homonucléaires :
effet de l'écho sur δ et J



refocalisation du
déplacement chimique

couplage J uniquement...

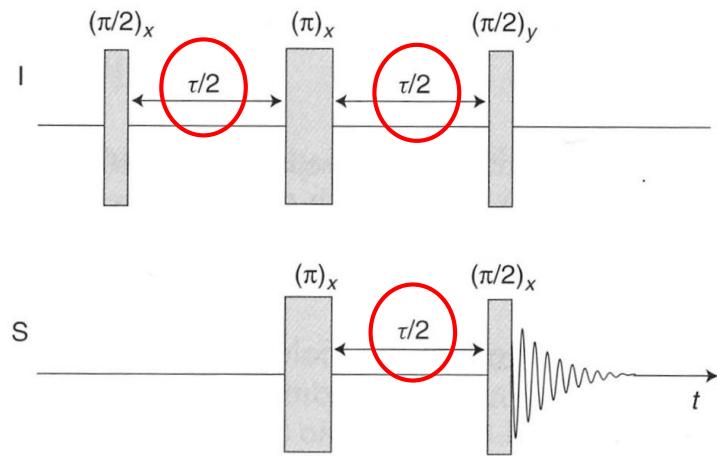
calculs quantiques simplifiés

efficacité accrue des séquences 10

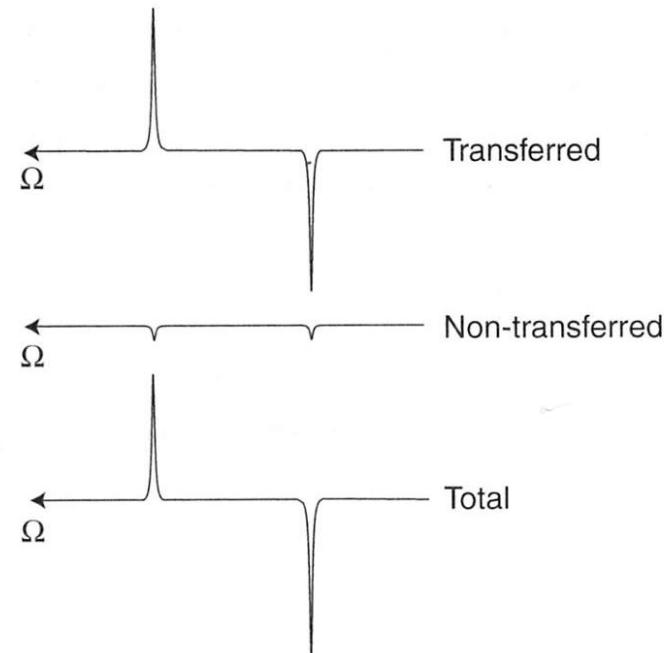
Transfert de polarisation : INEPT

Insensitive Nuclei Enhanced by Polarization Transfer

$$\text{signal/bruit} \propto |y|^{5/2} (B_0)^{3/2}$$



$$\tau = 1/(2|J_{IS}|)$$



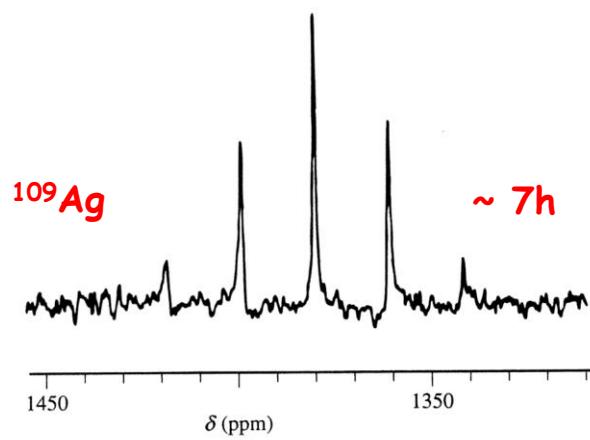
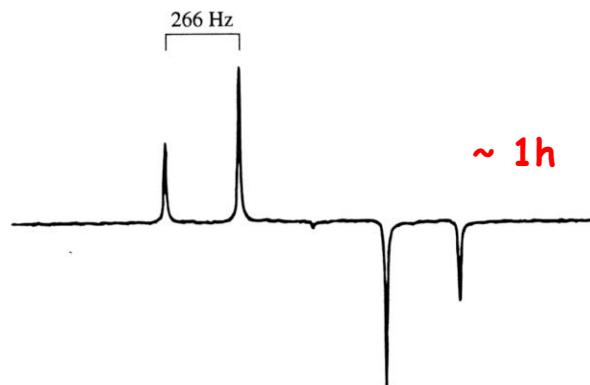
$$\text{gain}_{\text{INEPT}} \propto |\gamma(^1\text{H})/\gamma(X)|$$

ex : ≈ 10 pour ^{15}N !

$$\tau_1(^1\text{H}) < \tau_1(^{15}\text{N})$$

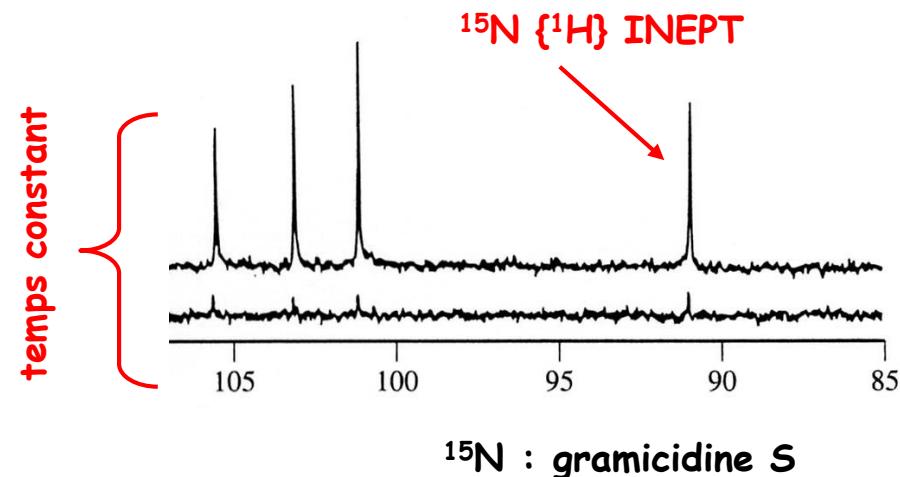
INEPT : exemples

^{109}Ag { ^{31}P } INEPT



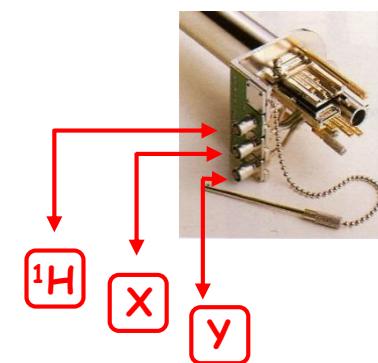
^{109}Ag : $[\text{Ag(dppe)}_2]\text{NO}_3$,
dppe=bisdiphenylphosphinoethane

^{15}N { ^1H } INEPT



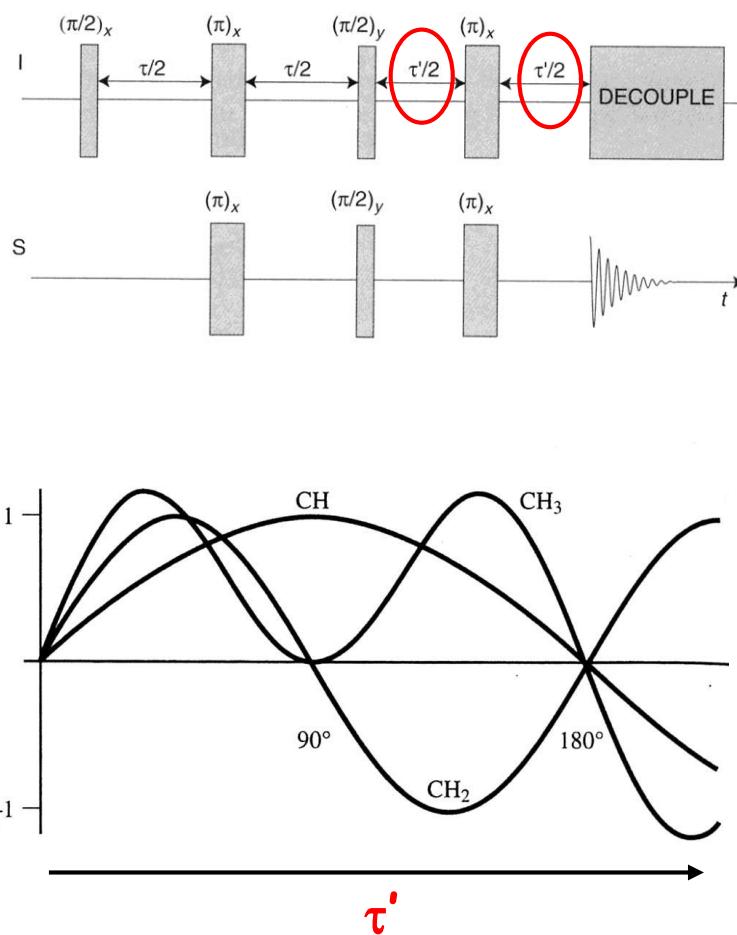
^{15}N : gramicidine S

INEPT en version refocalisée



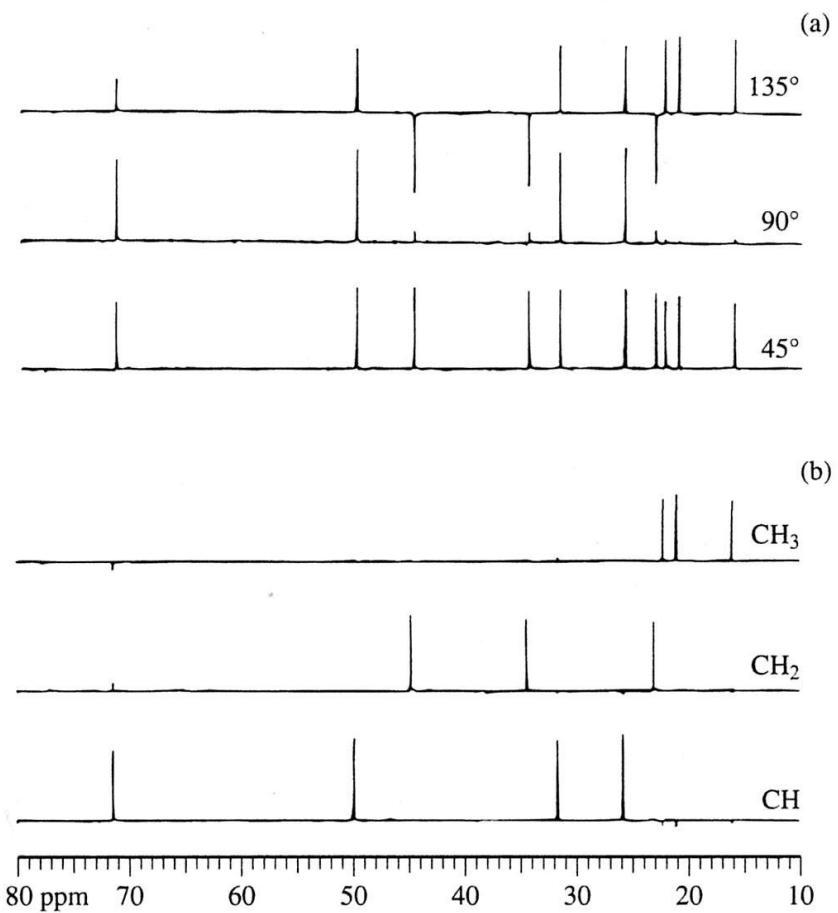
INEPT : édition spectrale

INEPT en version refocalisée



$$^1J_{IS} = 125 \text{ Hz}$$

$^{13}\text{C} \{^1\text{H}\}$ INEPT



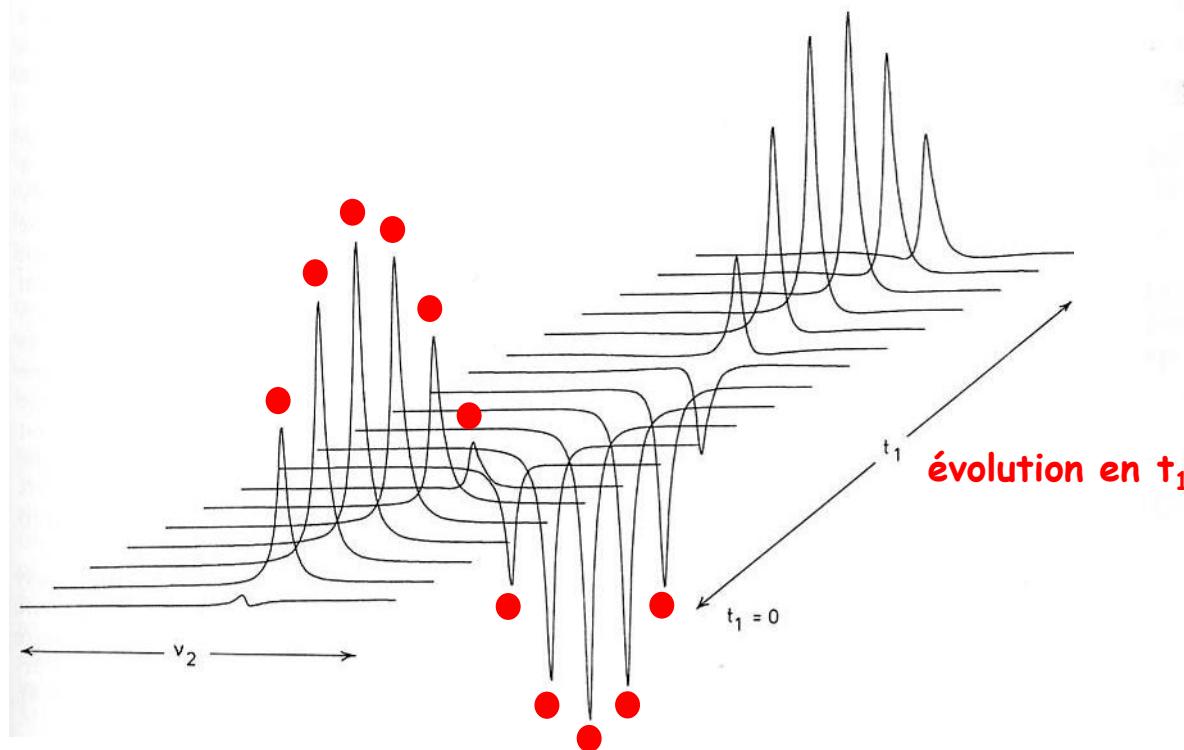
INEPT refocalisé : **menthol**
(2,4,6 ms)

Corrélations bidimensionnelles

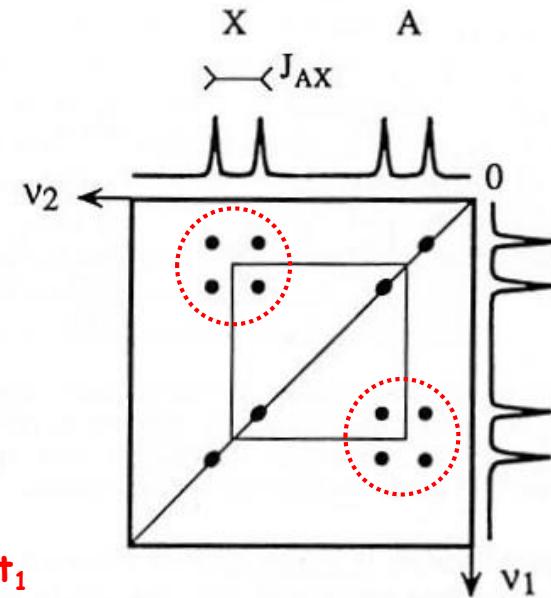
rappels :



J.Jeener ~ 1971

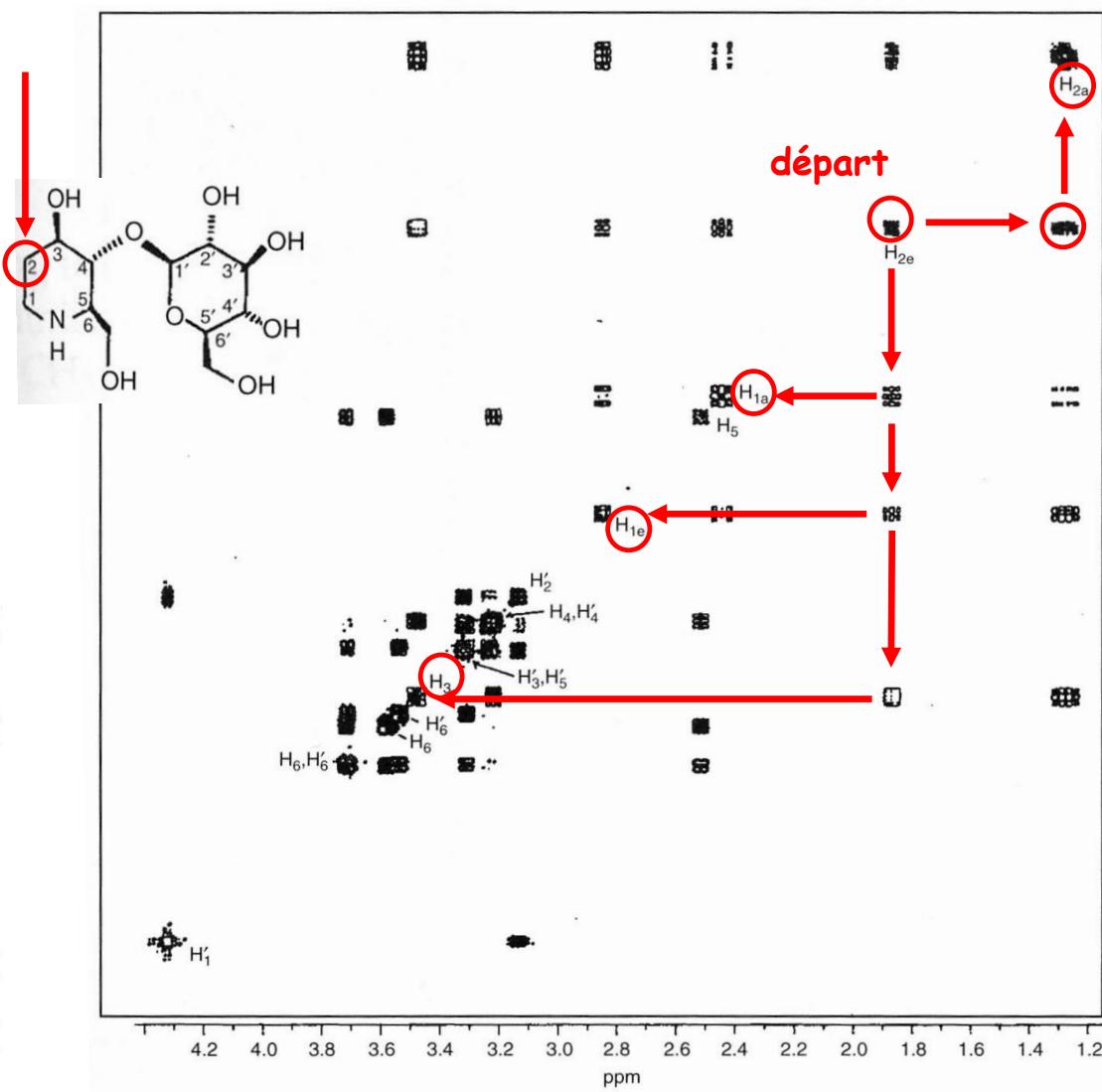
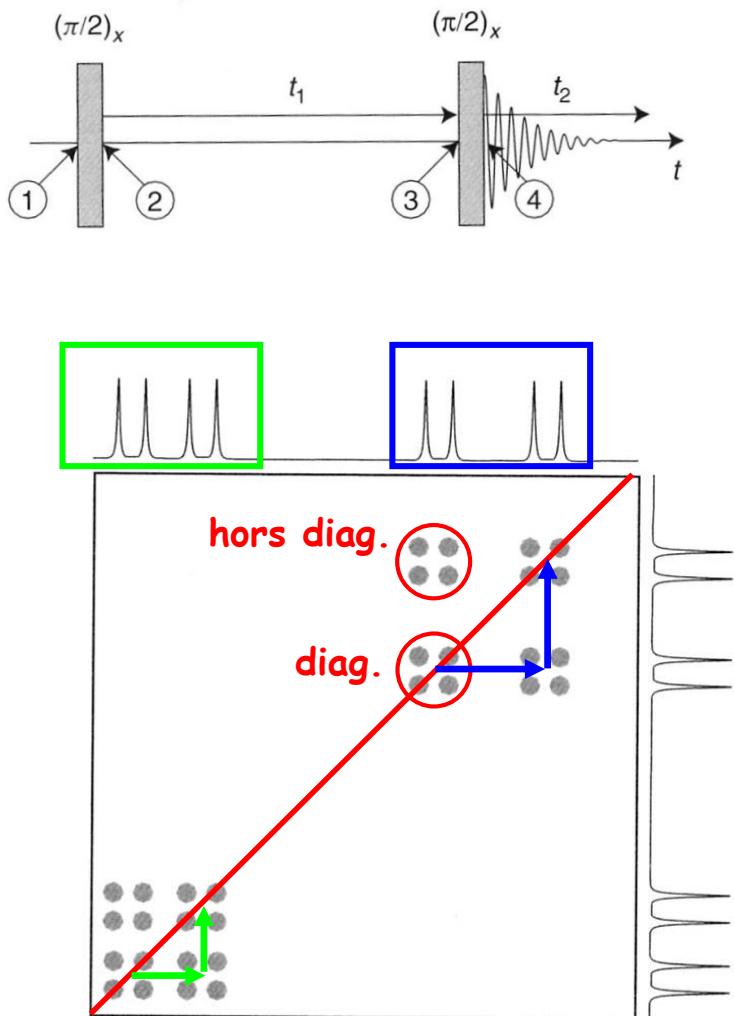


système A-X



Corrélation 2D homonucléaire : COSY

COrelation SpectroscopY

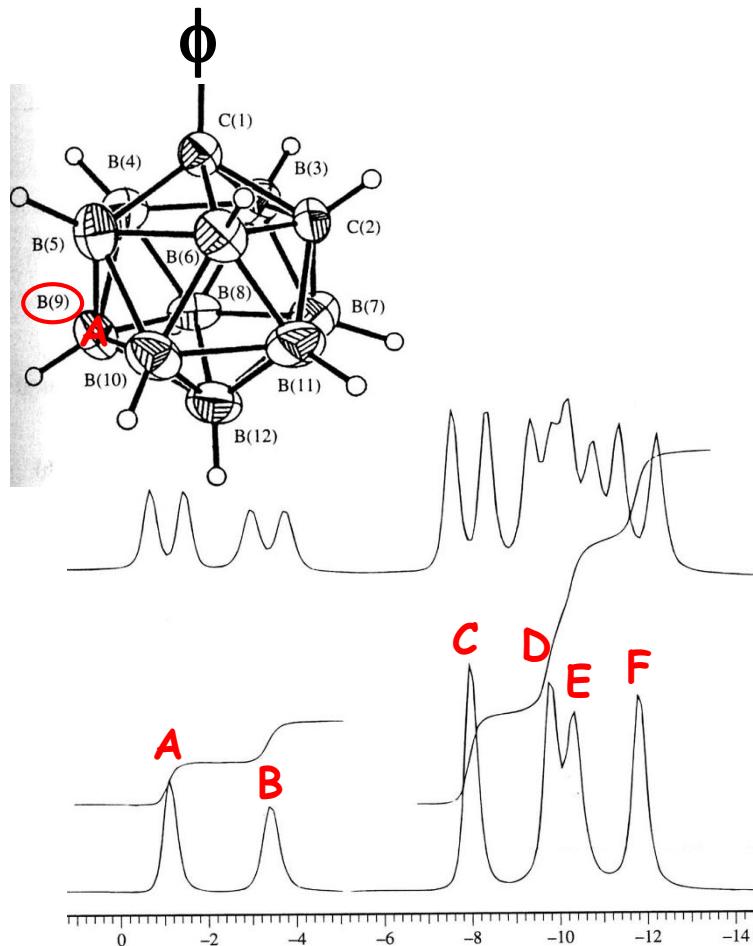


COSY $^1\text{H}/^1\text{H}$

Corrélation 2D homonucléaire : COSY quadripolaire

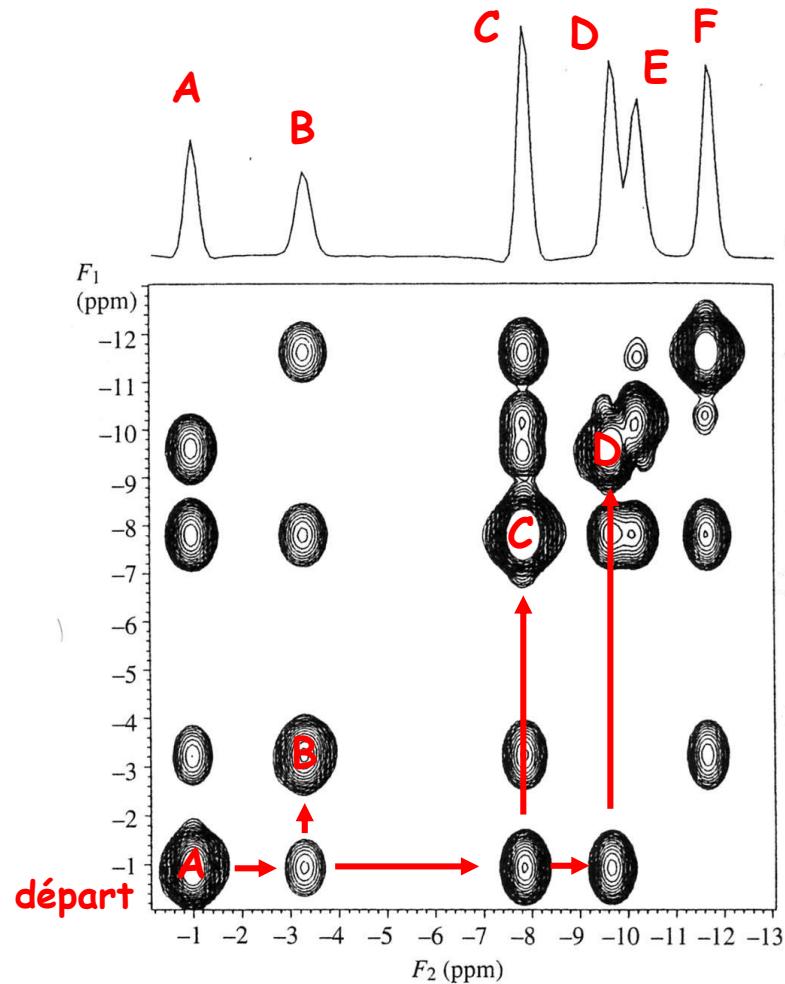
COSY $^{11}\text{B}/^{11}\text{B}$, déc.{ ^1H }

$^1\text{J}_{^{11}\text{B}/^{11}\text{B}}$ non observables en RMN 1D



$[1-\text{Ph}-1,2-\text{C}_2\text{B}_{10}\text{H}_{11}]$

plan de symétrie passant par :
C(1), C(2), B(9), B(12)

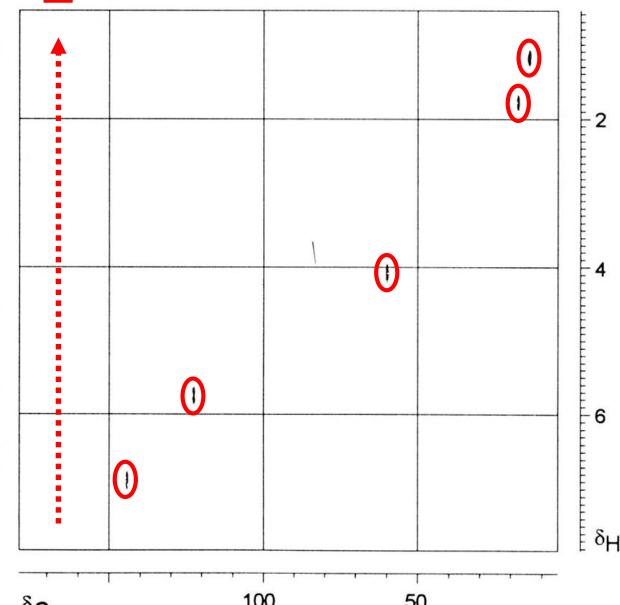
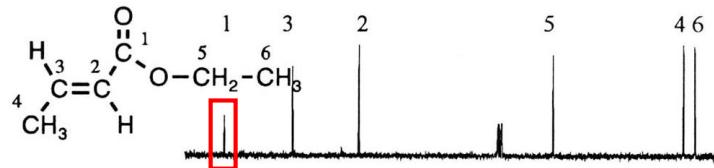
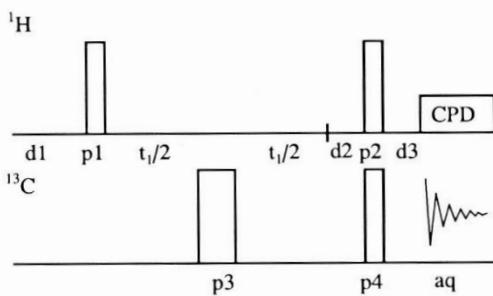


$\text{A} \equiv \text{B}(9)$ $\text{B} \equiv \text{B}(12)$ $\text{C} \equiv \text{B}(8,10)$

$\text{D} \equiv \text{B}(4,5)$ $\text{E} \equiv \text{B}(3,6)$ $\text{F} \equiv \text{B}(7,11)$

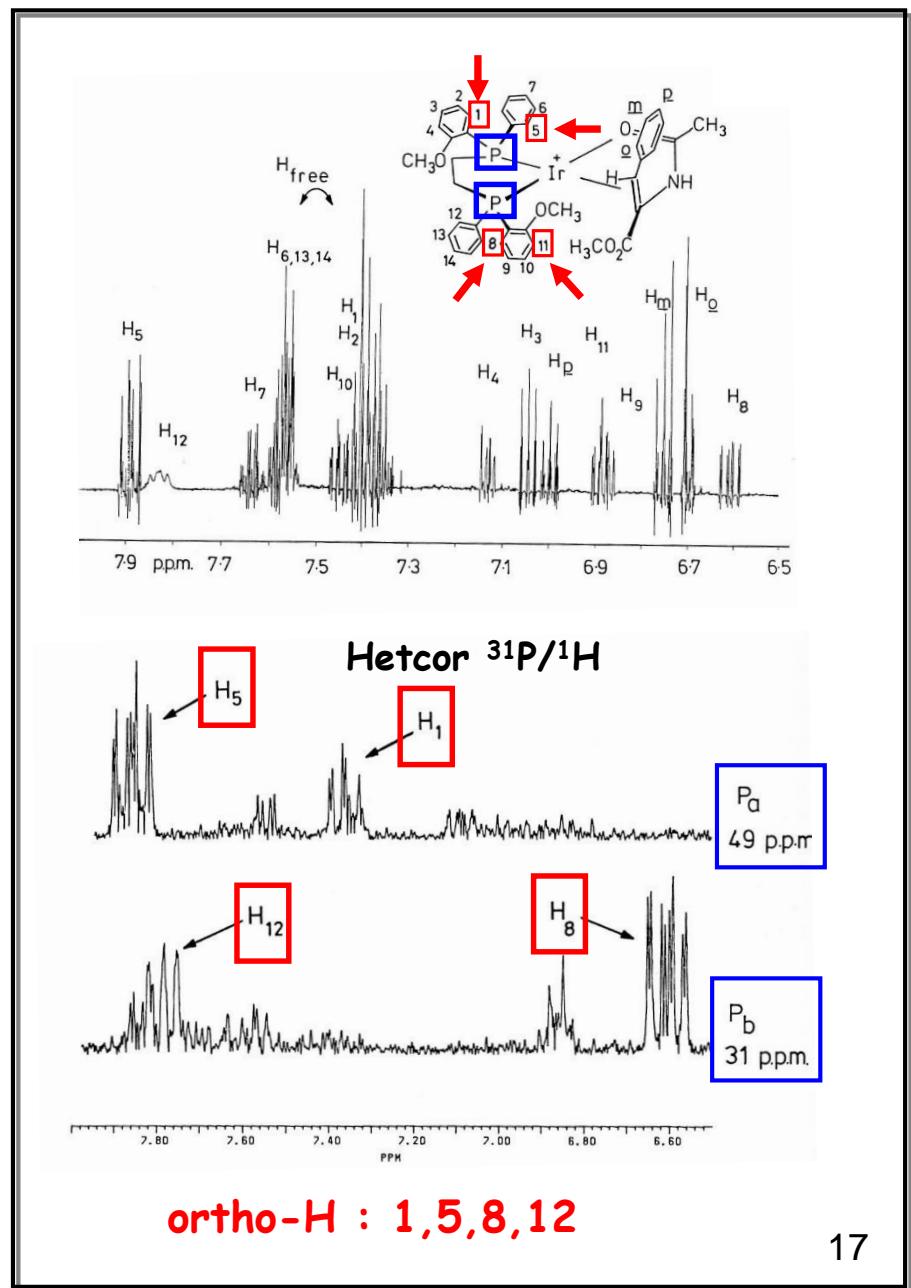
Transfert de polarisation hétéronucléaire : HETCOR

HETeronuclear CORrelation



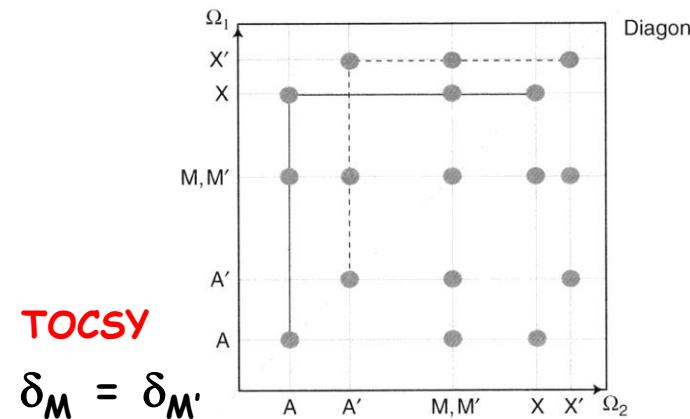
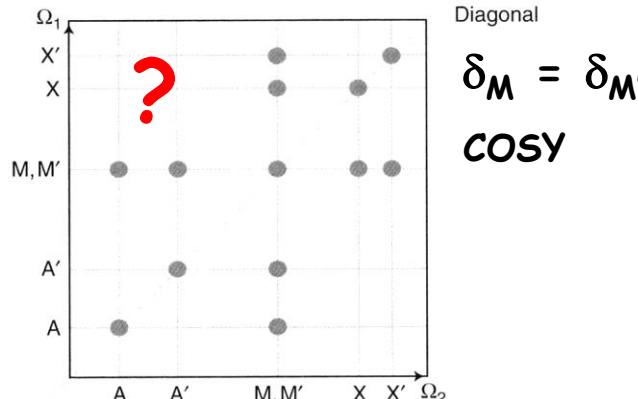
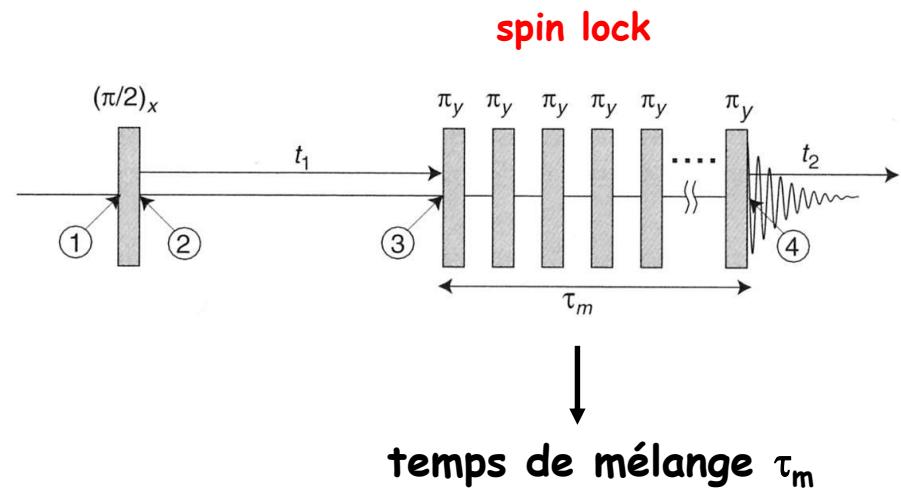
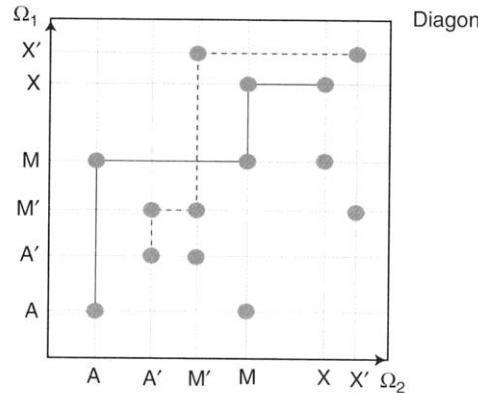
crotonate d'éthyle

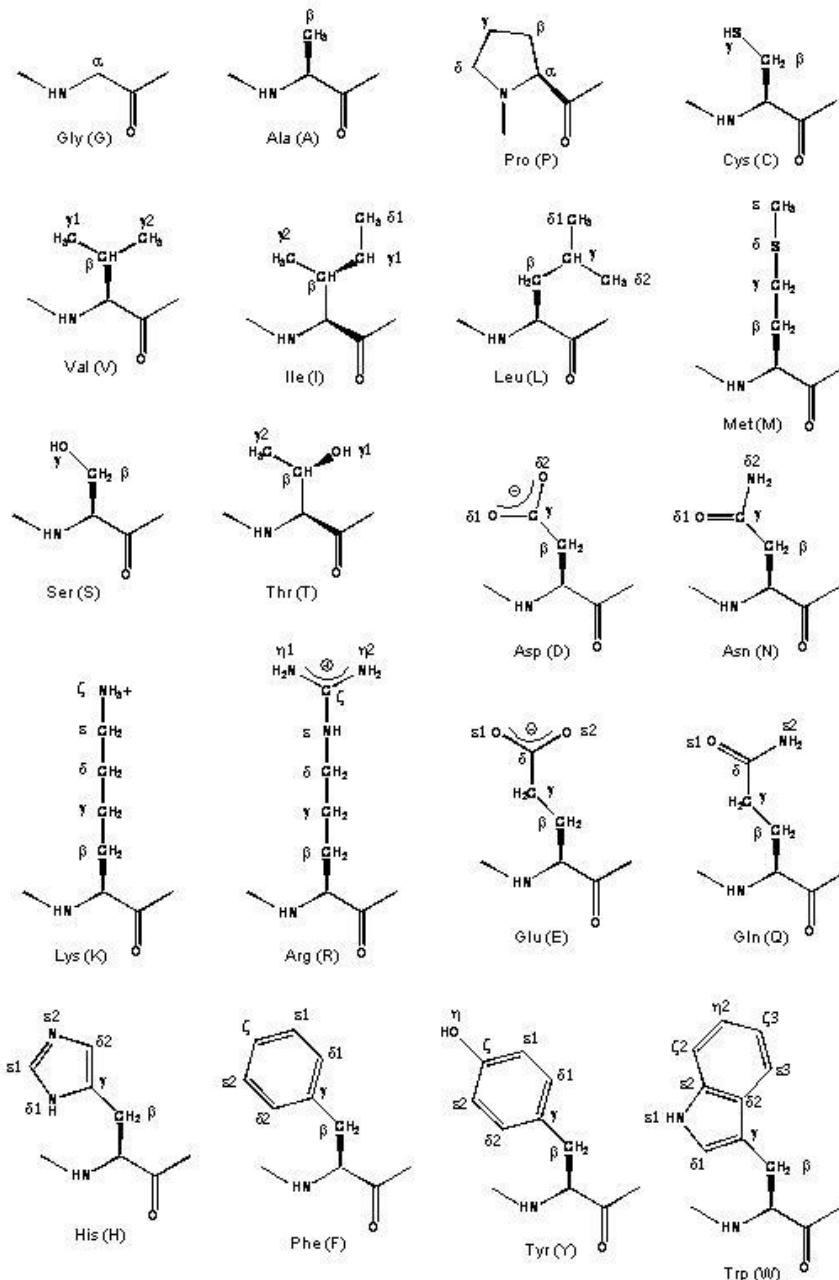
Hetcor $^{13}\text{C}/^1\text{H}$



Corrélations complètes : TOCSY

TOTal Correlation SpectroscopY

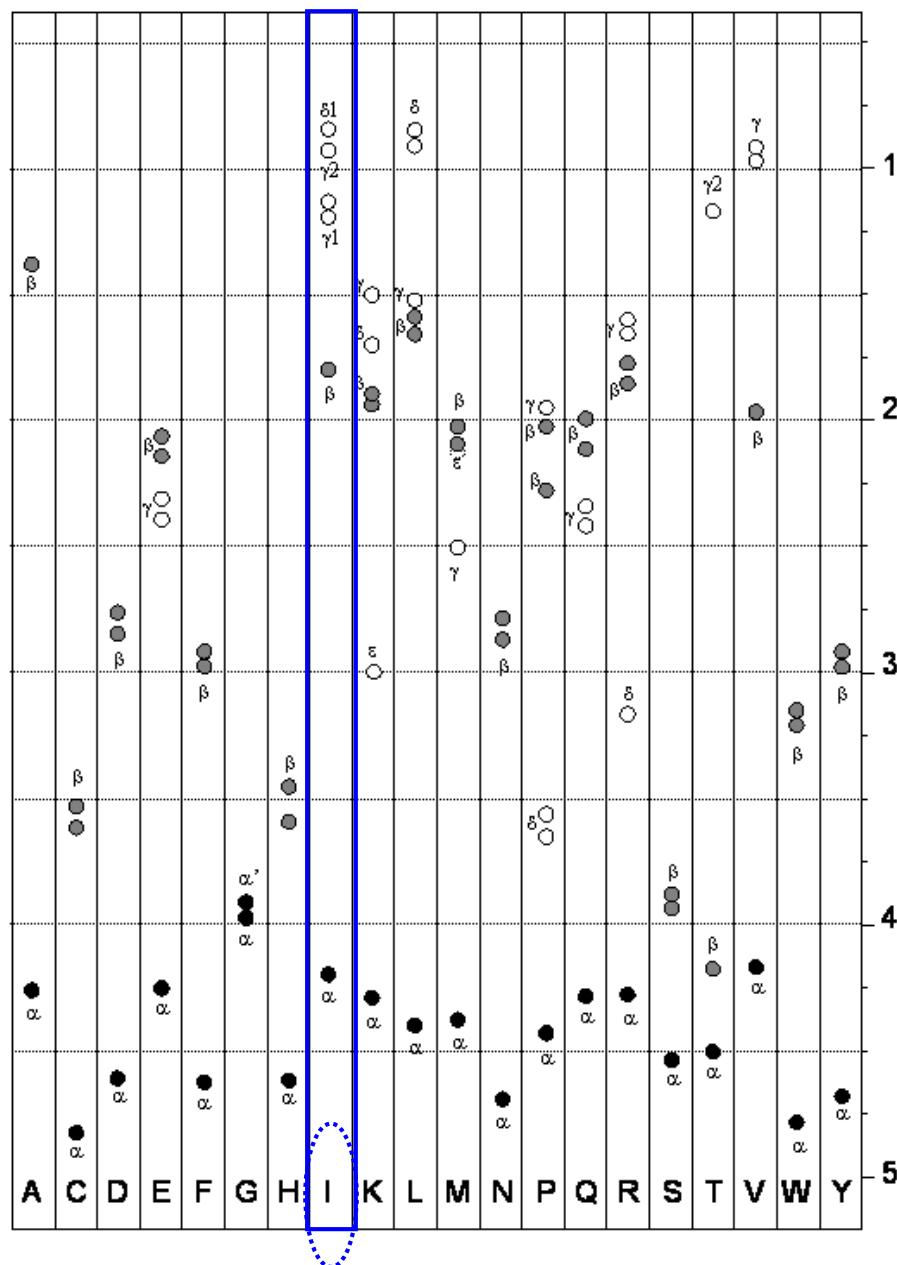




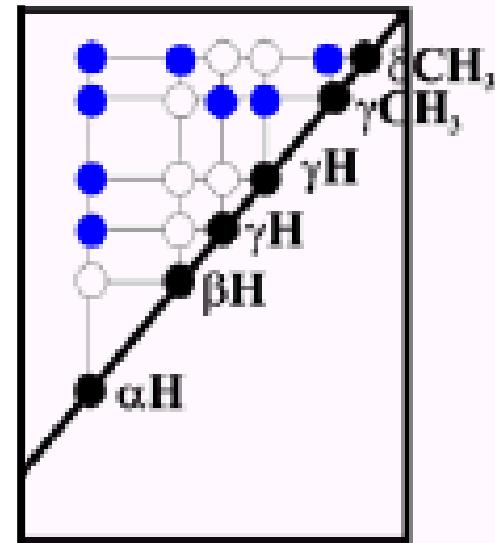
identification des systèmes de spins

**COSY
TOCSY**

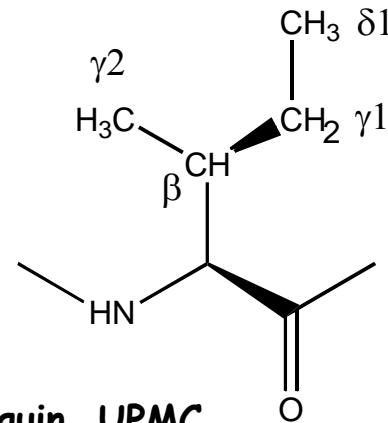
NH-H^α-Hlat



Ile (I)

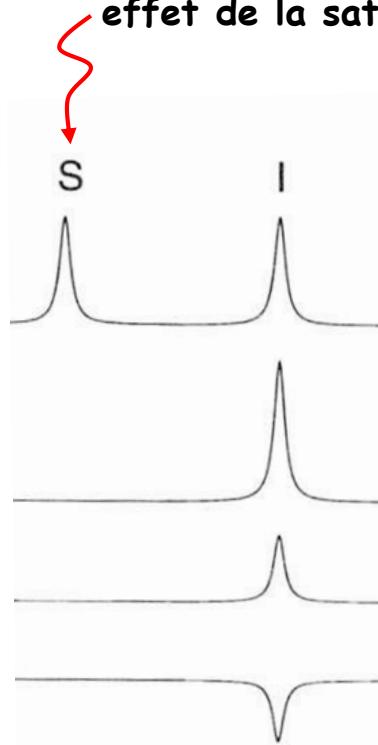


○ COSY
● TOCSY



Nuclear Overhauser Effect

effet de la saturation de S



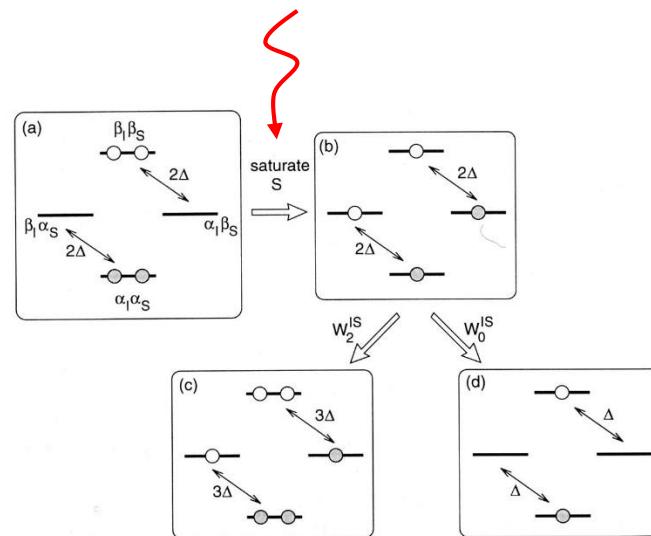
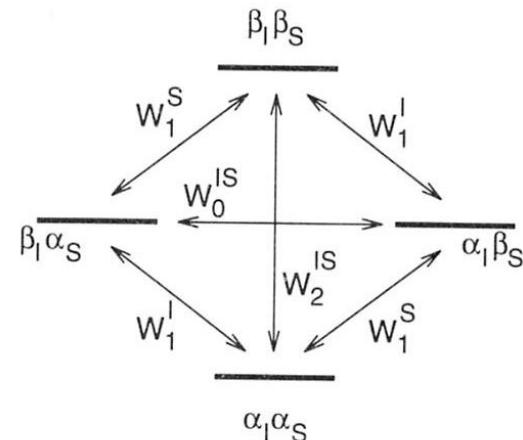
$$W \propto r^{-6}$$

W_0, W_1, W_2 dépendent de τ_c

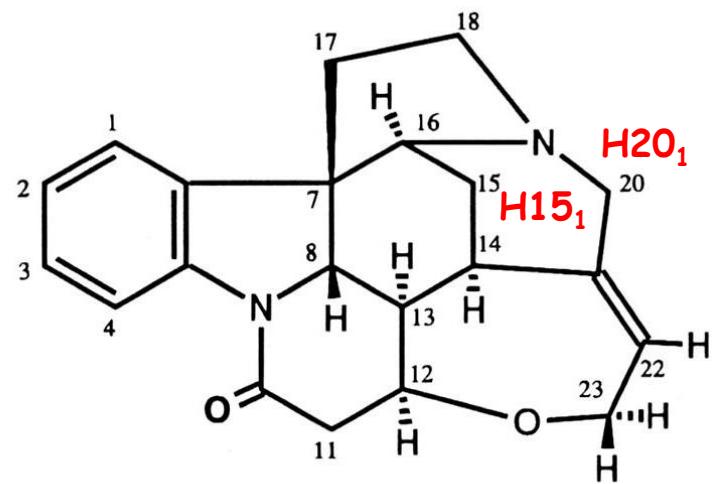
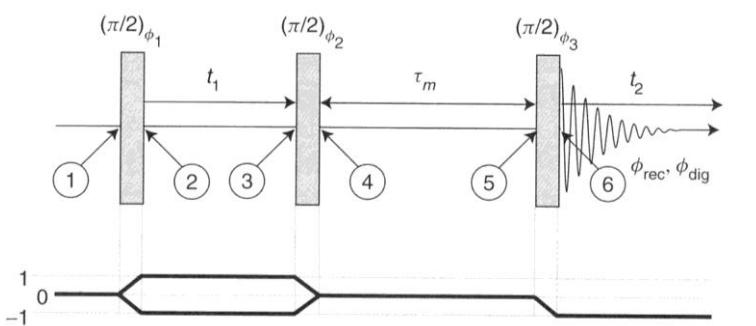
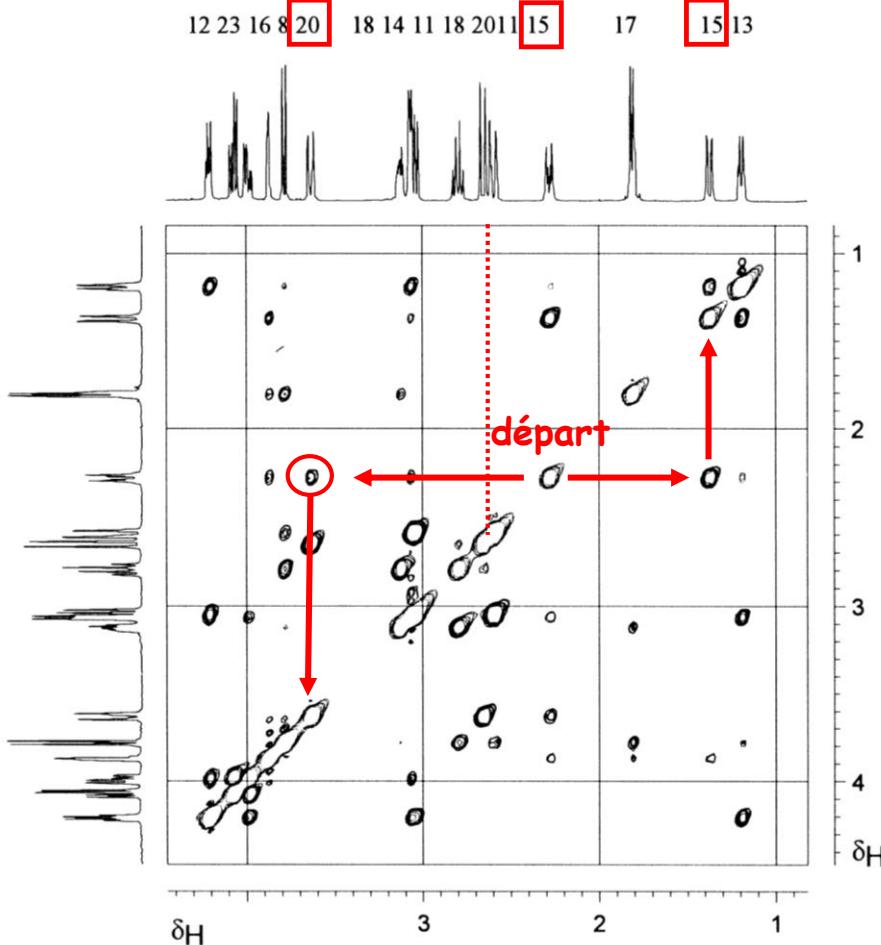
$$\eta_{\max} = 1/2 \gamma_S / \gamma_I$$

(paire, retrécissement extrême)

chemins de relaxation possibles pour un système IS (spins 1/2)



Nuclear Overhauser Effect Spectroscopy



Structure des protéines

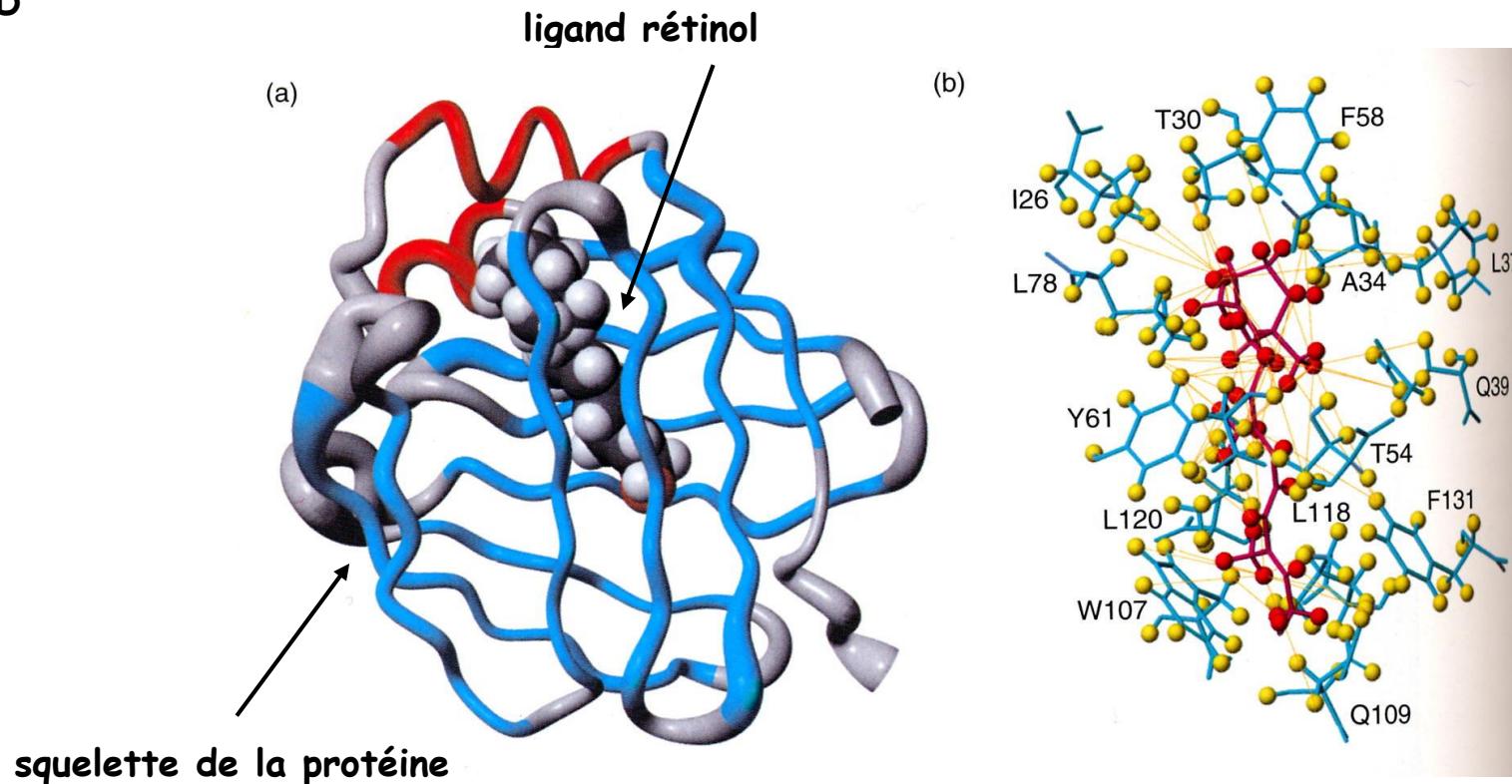
combinaison de :

DQ-COSY

TOCSY

NOESY → 3980 contraintes de distances

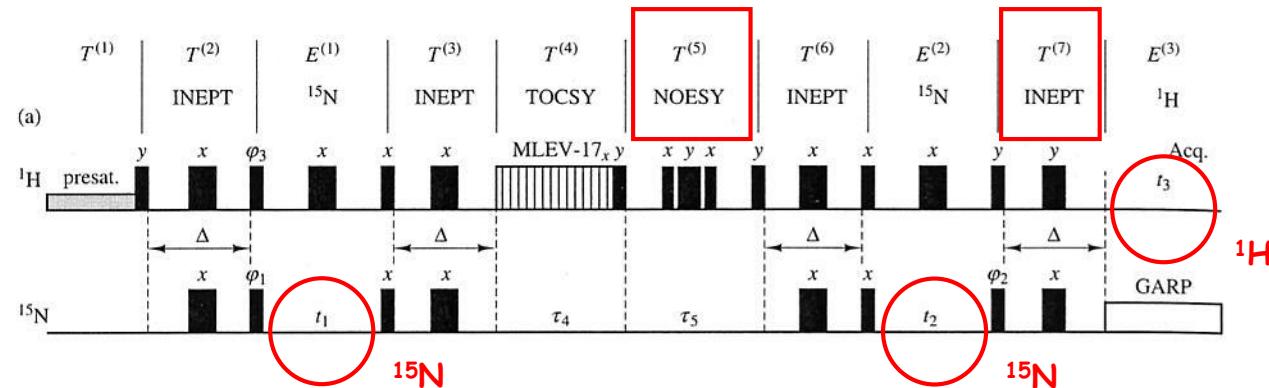
3D



cellular retinol binding protein II

RMN à 3 dimensions

rappel :



ex : COSY éditée par les corrélations $J_{\text{C}-\text{H}}$

